



Recent Advances and Applications of Semiconductor Laser based Gas Sensor Technology

F.K. Tittel, Yu. Bakhirkin, A.A. Kosterev
M. McCurdy, S.G. So, G. Wysocki & R.F. Curl
Rice Quantum Institute, Rice University, Houston, TX
<http://ece.rice.edu/lasersci/>

OUTLINE

SRNL
Athens, GA
Dec. 3, 2005

- Motivation: Wide Range of Chemical Sensing
- Fundamentals of Laser Absorption Spectroscopy
- Selected Applications of Trace Gas Detection
 - LAS with a Multipass absorption Cell (NH_3 , H_2CO)
 - Quartz Enhanced Laser-PAS (H_2CO)
 - OA-ICOS NO based Sensor Technology
- Outlook and Conclusions

Work supported by NASA, PNNL, NSF, NIH and Welch Foundation

Motivation: Wide Range of Gas Sensing Applications

- **Urban and Industrial Emission Measurements**
 - Industrial Plants
 - Combustion Sources and Processes (eg. early fire detection)
 - Automobile and Aircraft Emissions
- **Rural Emission Measurements**
 - Agriculture and Animal Facilities
- **Environmental Gas Monitoring**
 - Atmospheric Chemistry (eg. ecosystems and airborne)
 - Volcanic Emissions
- **Chemical Analysis and Industrial Process Control**
 - Chemical, Pharmaceutical, Food & Semiconductor Industry
 - Toxic Industrial Chemical Detection
- **Spacecraft and Planetary Surface Monitoring**
 - Crew Health Maintenance & Advanced Human Life Support Technology
- **Biomedical and Clinical Diagnostics** (eg. breath analysis)
- **Forensic Science and Security**
- **Fundamental Science and Photochemistry**



Trace Gas Monitoring in a Petrochemical Plant



University of Szeged, Hungary

Worldwide Megadirty Mega Cities

	Population, m		Sulphur dioxide	Particulate matter	Lead	Carbon monoxide	Nitrogen dioxide	Ozone
	1990, est.	2000, proj.						
Bangkok	7.16	10.26	0	0	0	0	0	0
Beijing	9.74	11.47	0	0	0	0	0	0
Bombay	11.13	15.43	0	0	0	0	0	0
Buenos Aires	11.58	13.05	-	0	0	-	-	-
Cairo	9.08	11.77	-	0	0	-	-	-
Calcutta	11.83	15.94	0	0	0	0	0	0
Doha	8.62	12.77	0	0	0	0	0	0
Jakarta	9.42	13.23	0	0	0	0	0	0
Karachi	7.67	11.57	0	0	0	-	-	-
London	10.57	10.79	0	0	0	0	0	0
Los Angeles	10.47	10.91	0	0	0	0	0	0
Manila	8.40	11.48	0	0	0	-	-	-
Mexico City	19.37	24.44	0	0	0	0	0	0
Moscow	9.39	10.11	-	0	0	0	0	-
New York	15.65	16.10	0	0	0	0	0	0
Rio de Janeiro	11.12	13.00	0	0	0	0	-	-
Sao Paulo	16.42	23.90	0	0	0	0	0	0
Seoul	11.33	12.97	0	0	0	0	0	0
Shanghai	13.30	14.99	0	0	0	-	-	-
Tokyo	20.52	21.32	0	0	0	-	-	-

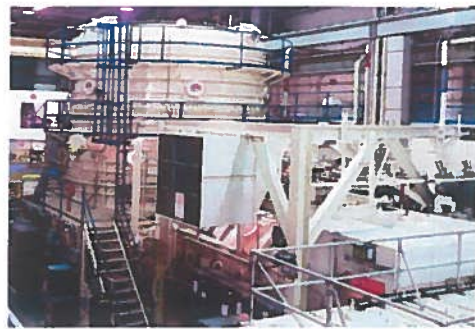
Source: United Nations. 0 = high pollution, 1/2 = Moderate to heavy pollution, 1/4 = Low pollution, - = No data available

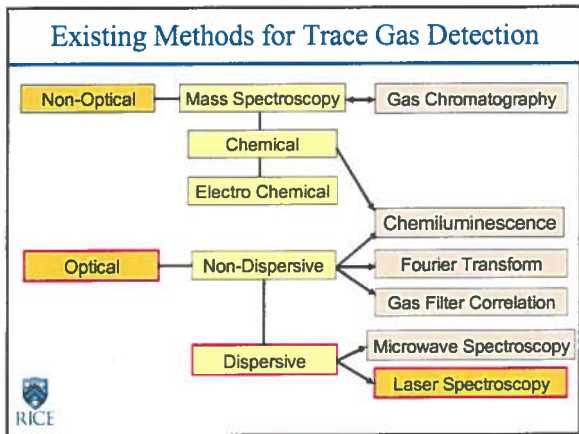


Megacity Air Pollution: Houston, TX



NASA-JSC Human-Rated Simulation Chamber





Fundamentals of Laser Absorption Spectroscopy

Beer-Lambert's Law of Linear Absorption
 $I(\nu) = I_0 e^{-\alpha(\nu) P_s L}$
 $\alpha(\nu) = \text{absorption coefficient [cm}^{-1} \text{ atm}^{-1}]$; $L = \text{path length [cm]}$
 $\nu = \text{frequency [cm}^{-1}]$; $P_s = \text{partial pressure [atm]}$

Optimum Molecular Absorbing Transition

- Overtone or Combination Bands (NIR)
- Fundamental Absorption Bands (MID-IR)

Long Optical Pathlengths

- Multipass Absorption Cell
- Cavity Enhanced, Cavity Ringdown & Intracavity Spectroscopy
- Open Path Monitoring (with retro-reflector)
- Evanescent Field Monitoring (fibers & waveguides)

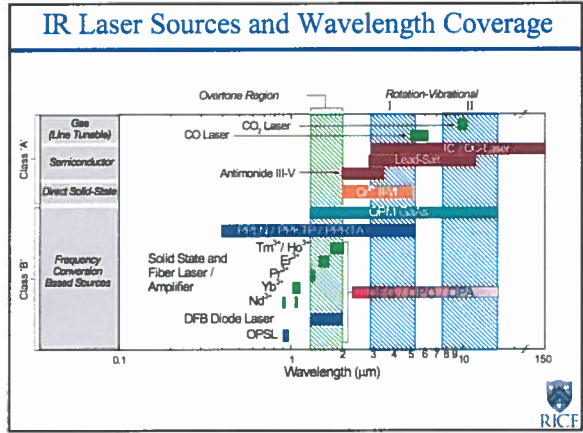
Spectroscopic Detection Schemes

- Frequency or Wavelength Modulation
- Balanced Detection
- Zero-air Subtraction
- Photoacoustic Spectroscopy

$C = \text{total number of molecules of absorbing gas/atm cm}^{-3}$ [molecule $\text{cm}^{-3} \text{ atm}^{-1}$]
 $S = \text{molecular line intensity [cm molecule}^{-1}]$
 $g(\nu - \nu_0) = \text{normalized spectral lineshape function [cm]}$
 (Gaussian, Lorentzian, Voigt)

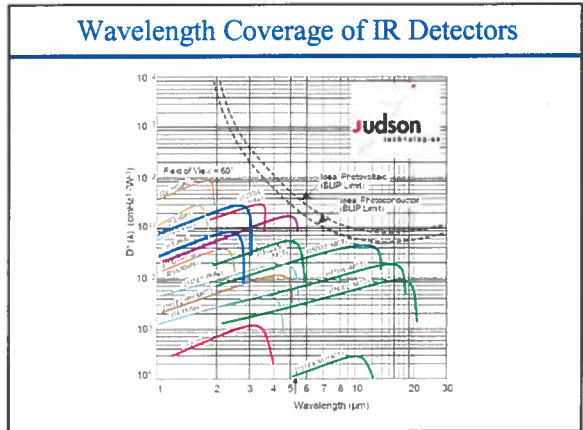
Mid-IR Source Requirements for Laser Spectroscopy

REQUIREMENTS	IR LASER SOURCE
Sensitivity (% to ppt)	Power
Selectivity	Single Mode Operation and Narrow Linewidth
Multi-gas Components, Multiple Absorption Lines and Broadband Absorbers	Tunable Wavelengths
Directionality or Cavity Mode Matching	Beam Quality
Rapid Data Acquisition	Fast Time Response
Room Temperature Operation	No Consumables
Field deployable	Compact & Robust

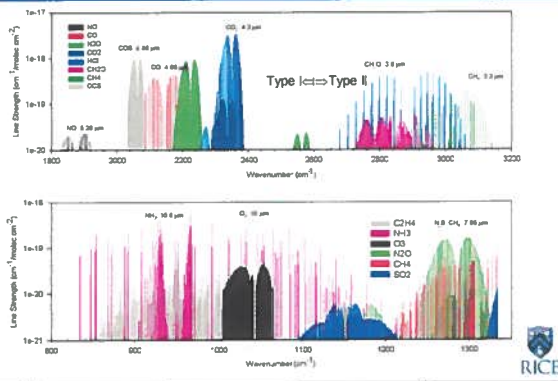


Quantum and Interband Cascade Laser: Basic Facts

- Band-structure engineered devices (emission wavelength is determined by layer thickness - MBE or MOCVD) QCLs operate from 4 to 160 μm (limited by the CB offset on the short wavelength side)
 - Unipolar devices
 - Cascading (each electron creates N laser photons and the number of periods N determines laser power)
- Compact, reliable, stable, long lifetime, commercial availability Fabry-Perot (FP) or single mode (DFB)
- Broad spectral tuning range in the mid-IR (4-24 μm for QCLs and 3-5 μm for ICLs)
 - 1.5 cm^{-1} using current
 - 10-20 cm^{-1} using temperature
 - > 150 cm^{-1} using an external grating element
- Narrow spectral linewidth cw, 0.1 - 3 MHz & <10KHz with frequency stabilization
 Linewidth is ~ 300 MHz of pulsed QCLs (chirp from heating)
- High output powers at TEC/RT temperatures
 - Pulsed peak powers of 1.6 W, high temperature operation ~ 425 K
 - Average power levels: 1-600 mW
 - ~50 mW, TEC CW DFB @ 5 and 10 microns (Alpes & L'nine)
 - >600 mW (CW FP) and >150 mW (CW DFB) at 298 K (Northwestern)



HITRAN Simulation of Absorption Spectra (3.1-5.5 & 7.6-12.5 μm)



Representative Trace Gas Detection Limits

Species	cm ⁻¹	Precision 1 s RMS (ppt)	LOD 100 s (ppt)
NH ₃	967	50	20
NO ₂	1600	80	40
HONO	1700	200	80
CO	2180	120	50
N ₂ O	2240	100	50
HNO ₃	1720	200	80
O ₃	1050	500	200
NO	1905	200	100
CH ₄	1270	400	200
SO ₂	1370	310	120
C ₂ H ₄	960	360	140
HCHO	1765	350	100
H ₂ O ₂	1267	1000	400

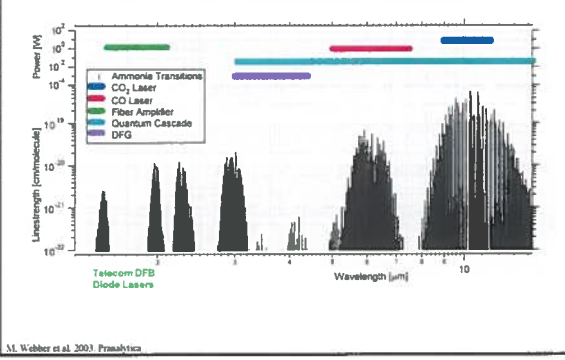
Limit of Detection (LOD) for S/N = 2
Pathlength: 210 m
Typical data acquisition time: 1-100 s

Mark S. Zahniser, SFRIG 2004, September 2004

Motivation for NH₃ Detection

- Monitoring of gas separation processes
- Spacecraft related gas monitoring
- Monitoring NH₃ concentrations in the exhaust stream of NO_x removal systems based on selective catalytic reduction (SCR) techniques
- Semiconductor process monitoring & control
- Monitoring of industrial refrigeration facilities
- Pollutant gas monitoring
- Atmospheric chemistry
- Medical diagnostics (kidney & liver dysfunctions)

Infrared NH₃ Absorption Spectra



Fundamentals of Laser Absorption Spectroscopy

B Beer-Lambert's Law of Linear Absorption
 $I(v) = I_0 e^{-\alpha(v) P_a L}$
 $\alpha(v)$ - absorption coefficient [cm⁻¹ atm⁻¹]; L - path length [cm]
 v - frequency [cm⁻¹]; P_a - partial pressure [atm]

$\alpha(v) = C \cdot S(T) \cdot g(v - \nu_0)$

C - total number of molecules of absorbing gas/atm cm² [molecule cm⁻² atm⁻¹]
 S - molecular line intensity [cm² molecule⁻¹]
 g(v - ν₀) - normalized spectral lineshape function [cm⁻¹]
 (Gaussian, Lorentzian, Voigt)

Optimum Molecular Absorbing Transition

- Overtone or Combination Bands (NIR)
- Fundamental Absorption Bands (MID-IR)

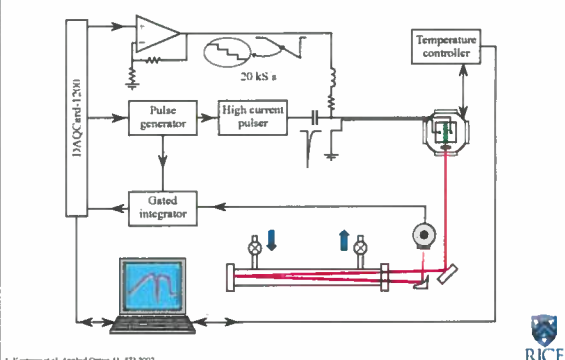
Long Optical Pathlengths

- Multipass Absorption Cell
- Cavity Enhanced, Cavity Ringdown & Intracavity Spectroscopy
- Open Path Monitoring (with retro-reflector)
- Evanescent Field Monitoring (fibers & waveguides)

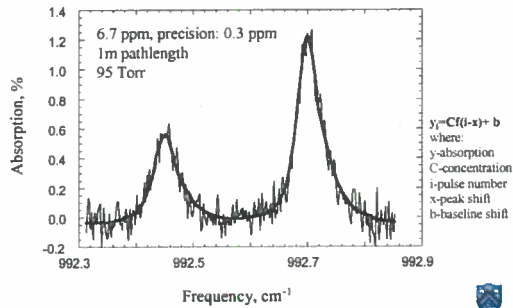
Spectroscopic Detection Schemes

- Frequency or Wavelength Modulation
- Balanced Detection
- Zero-air Subtraction
- Photoacoustic Spectroscopy

Pulsed QC Laser Based Gas Sensor



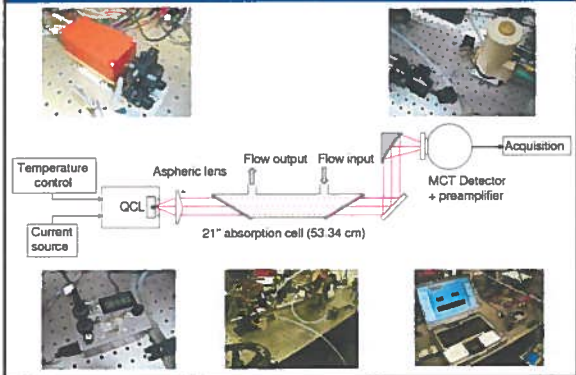
Ammonia Absorption Spectrum @ 993 cm⁻¹



A. Kosterev et al. Applied Optics 41, 573 (2002)

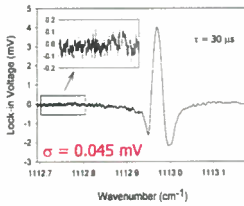


CW RT DFB QC laser based NH₃ Sensor @ 9 μm (1113 cm⁻¹)



Wavelength Modulation Spectroscopy of NH₃

- QCL Drive Current : Quasi CW + Wavelength modulation



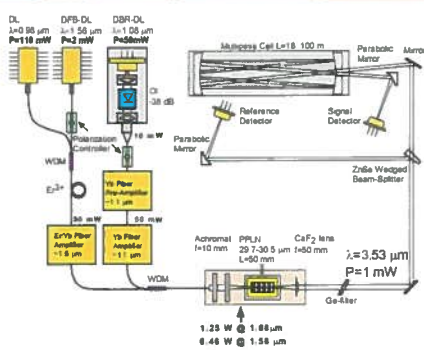
- Calibration with a 1038 ppm NH₃:N₂ mixture
- 1σ extrapolated sensitivity 82 ppb.m/√Hz
- ⇒ Improvement by a factor of 3 compared to direct absorption spectroscopy

87

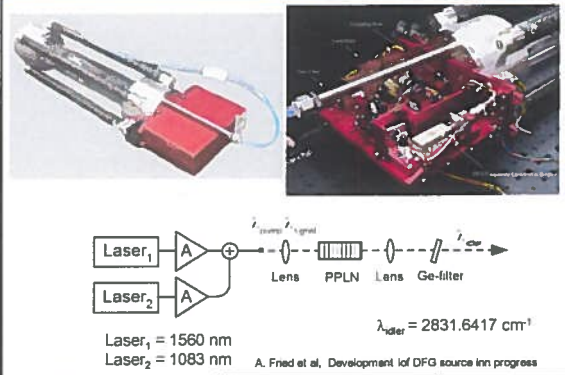
Motivation for Precision Monitoring of H₂CO

- Pollutant due to incomplete fuel combustion processes
- Potential trace contaminant in industrial manufactured products
- Precursor to atmospheric O₃ production
- Medically important gas

Mid-IR DFG Based H₂CO Sensor

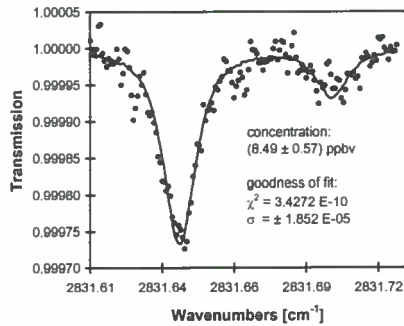


Advanced DFG System for H₂CO Detection



A. Fried et al. Development of DFG source in progress

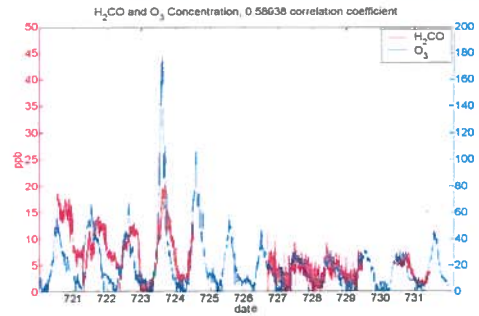
H₂CO Detection in Ambient Air at 3.53 μm



D. Rehk et al. Applied Physics B71, 947-952 (2001)



H₂CO and O₃ Concentrations at Deer Park, TX for July 20-31, 2003



43



Fundamentals of Laser Absorption Spectroscopy

Bear-Lambert's Law of Linear Absorption
 $I(\nu) = I_0 e^{-\alpha(\nu) P_s L}$
 $\alpha(\nu)$ - absorption coefficient ($\text{cm}^{-1} \text{ atm}^{-1}$), L - path length (cm)
 ν - frequency (cm^{-1}), P_s - partial pressure (atm)

Optimum Molecular Absorbing Transition

- Overtone or Combination Bands (NIR)
- Fundamental Absorption Bands (MID-IR)

Long Optical Pathlengths

- Multipass Absorption Cell
- Cavity Enhanced, Cavity Ringdown & Intracavity Spectroscopy
- Open Path Monitoring (with retro-reflector)
- Evanescent Field Monitoring (fibers & waveguides)

Spectroscopic Detection Schemes

- Frequency or Wavelength Modulation
- Balanced Detection
- Zero-air Subtraction
- Photoacoustic Spectroscopy

$\alpha(\nu) = C \cdot S(T) \cdot g(\nu - \nu_0)$
 C - total number of molecules of absorbing gas/atoms cm^{-3} (molecule $\text{cm}^{-3} \text{ atm}^{-1}$)
 S - molecular line intensity ($\text{cm}^2 \text{ molecule}^{-1}$)
 $g(\nu - \nu_0)$ - normalized spectral lineshape function (cm), (Gaussian, Lorentzian, Voigt)

From conventional PAS to QEPAS

Laser beam, power P

Modulated (P or λ) at f or $f/2$

$S \sim \frac{Q \alpha P}{f V}$

$NNEA = \frac{\alpha_{\text{min}} P}{\sqrt{\Delta f}} \left[\frac{\text{cm}^{-1} \times \text{W}}{\sqrt{\text{Hz}}} \right]$

Cell is OPTIONAL!
 $Q \gg 1000$
 effective volume

SWAP RESONATING ELEMENT!!!

Resonant at f
 quality factor Q

Bioelectric microphone crystal

28



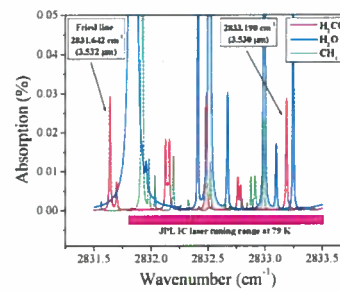
Comparative Size of Absorbance Detection Modules (ADM)

Optical multipass cell (100 m):
 $l=70 \text{ cm}$, $V=3000 \text{ cm}^3$

Resonant photoacoustic cell (1000 Hz):
 $l=60 \text{ cm}$, $V=50 \text{ cm}^3$

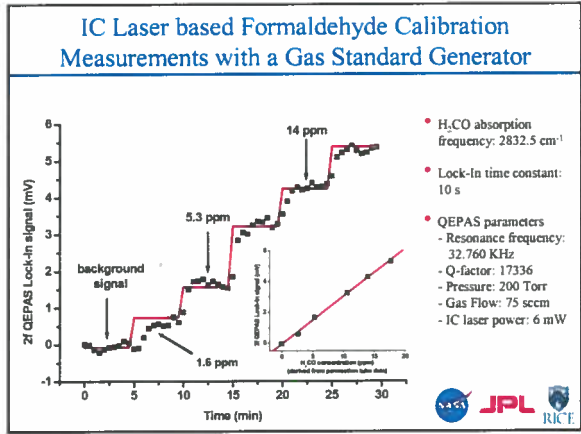
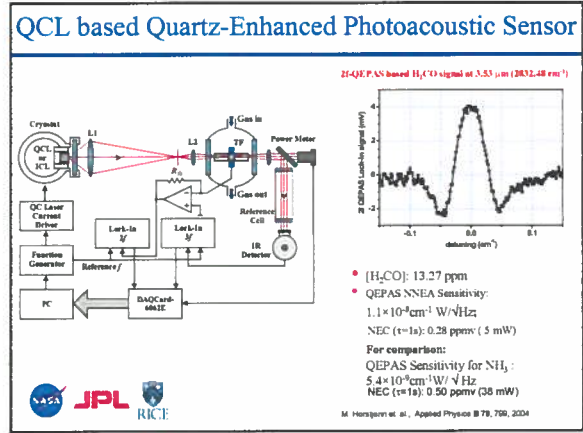
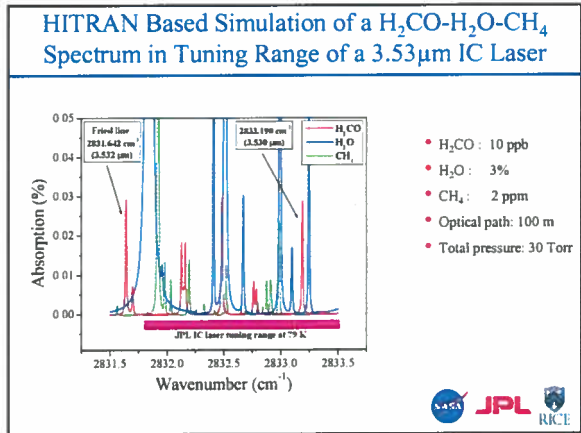
QEPAS ADM:
 $l=0.5 \text{ cm}$, $V=0.05 \text{ cm}^3$

HITRAN Based Simulation of a H₂CO-H₂O-CH₄ Spectrum in Tuning Range of a 3.53 μm IC Laser



- H₂CO : 10 ppb
- H₂O : 3%
- CH₄ : 2 ppm
- Optical path: 100 m
- Total pressure: 30 Torr





Merits of QE Laser-PAS based Trace Gas Detection

- High sensitivity (ppm to ppb gas concentration levels) and excellent dynamic range
- Immunity to ambient and flow acoustic noise, laser noise and etalon effects, which allows applications that involve harsh operating environments
- Required sample volume is very small. The volume is ultimately limited by the gap size between the TF prongs, which is < 1 mm³ for the presently used QTF.
- No spectrally selective elements are required
- Applicable over a wide range of pressures, including atmospheric pressure
- Sensitive to phase shift introduced by vibrational to translational (V-T) relaxation processes and hence the potential of concentration measurements of spectrally interfering species
- Ultra-compact, rugged and low cost compared to LAS that requires a multipass absorption cell and infrared detector(s)
- Potential for optically multiplexed concentration measurements

NASA JPL RICE

QEPAS Performance for 10 Trace Gas Species (Dec'05)

Molecule (Host)	Frequency, cm ⁻¹	Pressure, Torr	NNEA, cm ⁻¹ W/Hz ^{1/2}	Power, mW	NEC (τ=1s), ppmv
H ₂ O (N ₂)**	7181.17	60	2.1 × 10 ⁻⁷	5.8	0.18
HCN (air: 50% hum)**	6539.11	60	< 2.6 × 10 ⁻⁷	50	0.1
C ₂ H ₂ (N ₂)**	6529.17	75	-2.5 × 10 ⁻⁷	-40	0.06
NH ₃ (N ₂)*	6528.76	60	5.4 × 10 ⁻⁷	38	0.50
CO ₂ (exhaled air)	6514.25	90	1.0 × 10 ⁻⁷	5.2	890
CO ₂ (N ₂)***	4990.00	300	1.5 × 10 ⁻⁷	4.6	130
CH ₂ O (N ₂)*	2832.48	100	1.1 × 10 ⁻⁷	4.6	0.28
CO (N ₂)	2196.66	50	5.3 × 10 ⁻⁷	13	0.5
CO (propylene)	2196.66	50	7.4 × 10 ⁻⁷	6.5	0.14
N ₂ O (air+5%SF ₆)	2195.63	50	1.5 × 10 ⁻⁷	19	0.007

* - Improved microresonator
 ** - Improved microresonator and double optical pass through QTF
 *** - Without microresonator

NNEA - normalized noise equivalent absorption coefficient.
 NEC - noise equivalent concentration for available laser power and τ=1s time constant.

For comparison: conventional PAS 2.2 × 10⁻⁶ cm⁻¹W/√Hz (1,800 Hz) for NH₃*
 * M. E. Webber, N. Probstsky and C. K. N. Patel, Appl. Opt. 42, 2119-2128 (2003)

NASA JPL RICE

Fundamentals of Laser Absorption Spectroscopy

Beer-Lambert's Law of Linear Absorption
 $I(\nu) = I_0 e^{-\alpha(\nu) P_s L}$

$\alpha(\nu)$ - absorption coefficient [cm⁻¹ atm⁻¹]; L - path length (cm)
 ν - frequency [cm⁻¹]; P_s - partial pressure (atm)

$\alpha(\nu) = C \cdot S(\nu) \cdot g(\nu - \nu_0)$

C - total number of molecules of absorbing gas/atm/cm³ [molecule cm⁻³ atm⁻¹]
 S - molecular line intensity [cm molecule⁻¹]
 g(ν - ν₀) - normalized spectral lineshape function (cm), (Gaussian, Lorentzian, Voigt)

Optimum Molecular Absorbing Transition

- Overtone or Combination Bands (NIR)
- Fundamental Absorption Bands (MID-IR)

Long Optical Pathlengths

- Multipass Absorption Cell
- Cavity Enhanced, Cavity Ringdown & Intracavity Spectroscopy
- Open Path Monitoring (with retro-reflector)
- Evanescent Field Monitoring (fibers & waveguides)

Spectroscopic Detection Schemes

- Frequency or Wavelength Modulation
- Balanced Detection
- Zero-air Subtraction
- Photoacoustic Spectroscopy

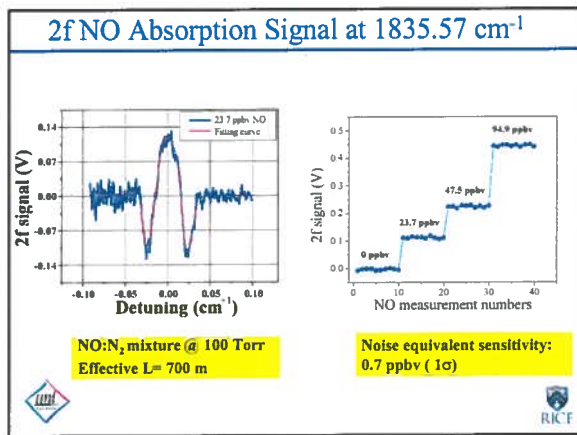
Off-Axis Integrated Cavity Output Spectroscopy (ICOS) Based Gas Sensor

* Novel compact gas cell design of length: 3.8 – 5.3 cm and cell volume $\approx 80 \text{ cm}^3$
 * Low loss mirrors (ROC 1m) ~ 60 -250 ppm, R ~ 99.975 , $L_{\text{eff}} = 170$ -800 m
 * Rapid eNO concentration measurements during a single breath cycle are feasible

ISI LGR RICE

TEC – CW-DFB QCL based Nitric Oxide OA-ICOS Sensor

ISI RICE



ICOS vs. CRDS

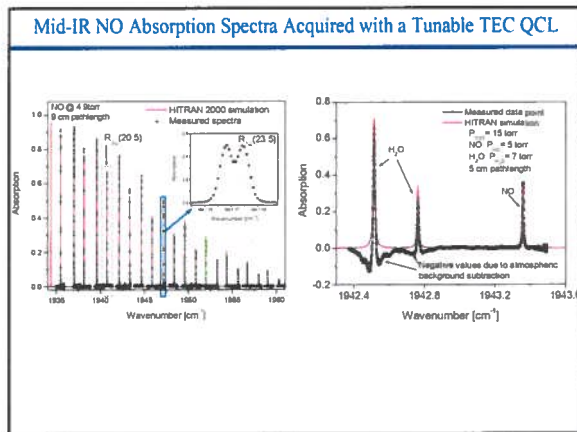
ICOS	CRDS
<ul style="list-style-type: none"> High sensitivity High time resolution not required, slow detector is sufficient Multiple high-order transverse modes, off-axis propagation Relies on quasi-random mode structure, non-critical alignment Low throughput [(1-R)/2 max] No need for narrow line laser Sensitive to the source power fluctuations 	<ul style="list-style-type: none"> Extremely high sensitivity possible – 10^{-11} cm^{-1} demonstrated in NIR Time resolved measurements, fast detector needed Single transverse mode, on-axis propagation – critical alignment Laser must be locked to the cavity mode High throughput in resonance for a narrow line (~kHz) laser Insensitive to the source power fluctuations

ISI RICE

External Cavity QCL Based Spectrometer

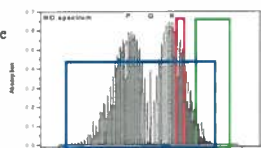
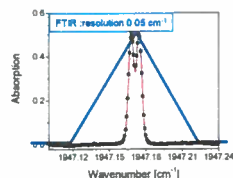
• PZT controlled EC-length
 • PZT controlled grating angle
 • Optimization of cavity alignment performed by means of lens positioning using electrically controlled 3D translation stage

G. Wysocki et al. Applied Physics B, 81, p.769 (2005)



Important facts of novel EC-QCL Technology

- Laser spectroscopy provides superior resolution compared to other techniques e.g. FTIR
- Single mode operation of the laser is required
- Wavelength tunability of single mode (DFB) mid-IR semiconductor lasers is $\sim 10\text{cm}^{-1}$
- Demonstrated wavelength tunability of the Rice EC QCL is $\sim 35\text{cm}^{-1}$ (limited by the gain chip properties and not by the designed EC configuration)
- Gain chips, which can provide tunability of $>200\text{cm}^{-1}$ are already reported in the literature

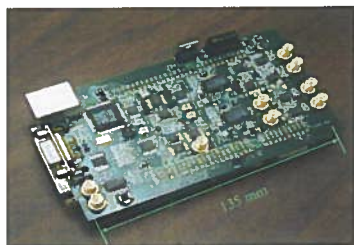


Sensor control and data processing

- Computer control of a laser-based spectroscopic sensor using PC (Windows, LabView) is convenient but not reliable and often does not allow to achieve the optimum sensor performance
- Reliable systems such as NI Real-Time devices are expensive, in part because of their multifunction abilities
- Dedicated electronic modules for autonomous sensor control and data processing are reliable, small, and consist of inexpensive part
- Today's technology such as DSP and FPGA offers convenience and flexibility of design



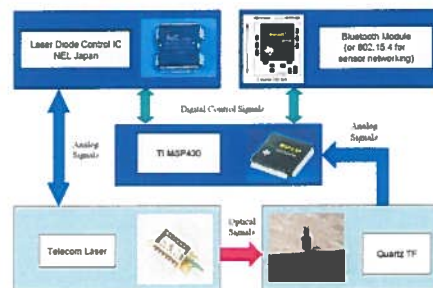
Dedicated DSP-based electronics for trace gas sensing using a pulsed QC laser



Pulsed laser requires high speed pulsed processing system for minimum detection limits



Concept of a ultra-miniature QEPAS gas sensor



Conclusions and Future Directions

- **Laser based Trace Gas Sensors**
 - Ultra compact ($\sim 0.2\text{mm}^3$), robust & low cost sensors based on QE L-PAS
 - QEL-PAS is immune to ambient noise. The measured noise level coincides with the thermal noise of the QTF
 - Best to date demonstrated QEPAS sensitivity is $2.1 \times 10^{-9}\text{cm}^{-1}\text{W}/\sqrt{\text{Hz}}$ for $\text{H}_2\text{O}/\text{N}_2$
 - QEPAS exhibits a low $1/f$ noise level, allowing data averaging for more than 3 hours
 - Detected 14 trace gases to date: NH_3 , CH_4 , N_2O , CO_2 , CO , NO , H_2O , COS , HCN , C_2H_2 , C_2H_4 , $\text{C}_2\text{H}_5\text{OH}$, SO_2 , H_2CO and several isotopic species of C, O, N & H
- **Applications in Trace Gas Detection**
 - Environmental & Spacecraft Monitoring (NH_3 , CO , CH_4 , C_2H_2 , N_2O , CO_2 and H_2CO)
 - Medical Diagnostics (NO , CO , COS , CO_2 , NH_3 , C_2H_2)
 - Industrial process control and chemical analysis (NO , NH_3 , H_2O)
- **Future Directions and Collaborations**
 - QE L-PAS based applications using novel thermoelectrically cooled cw and broadly wavelength tunable quantum and interband cascade lasers
 - Investigate QTFs with lower resonant frequencies
 - Investigate amplitude modulation QEPAS potential and limitations
 - New target gases, in particular VOCs and HCs
 - Development of optically multiplexed gas sensor networks based on QE L-PAS

NASA Atmospheric & Mars Gas Sensor Platforms



Tunable laser sensors for earth's stratosphere

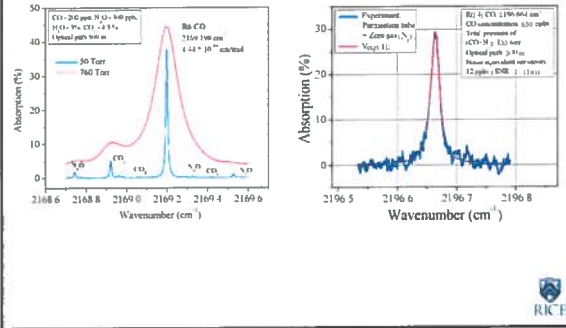
Aircraft laser absorption spectrometers



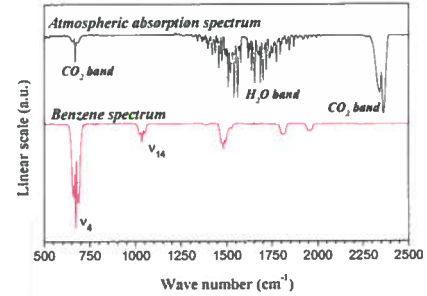
Tunable laser planetary spectrometer



OA-ICOS based CO Concentration Measurements at 2196.66 cm⁻¹



FT-IR survey absorption spectrum of benzene vapor (C₆H₆)



W. Chen, F. Cazzer, F. K. Tittel and D. Boucher, Appl. Optics 39, 6238, 2000