



# Real Time Ammonia Gas Sensor for Exhaled Human Breath Diagnostic Based on Quantum Cascade Laser

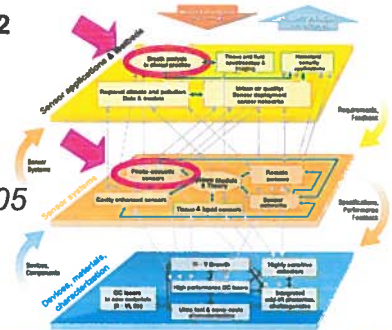


R. Lewicki,<sup>1</sup> A.A. Kosterev,<sup>1</sup> D.M. Thomazy,<sup>1</sup> L. Dong,<sup>1</sup> T.H. Risby,<sup>2</sup> S. Solga,<sup>3</sup> and F.K. Tittel<sup>1</sup>

1 – Electrical and Computer Engineering Department, Rice University  
 website: <http://www.ece.rice.edu/lasersci>

2 – Bloomberg School of Public Health, The Johns Hopkins University, Baltimore, MD 21205

3 – St. Luke's Hospital, 701 Ostrum St, Suite 604, Bethlehem, PA 18015



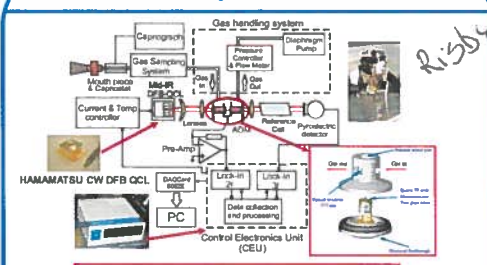
## Outline

- Motivation: Mid-IR QCL sensor for trace gas detection in exhaled breath
  - Laser sources: External Cavity (EC) and CW Distributed Feedback (DFB) QCLs
  - Detection technique: Quartz Enhanced Photoacoustic Spectroscopy (QEPAS)
- Exhaled breath** is a mixture of molecules, some of which are present in our body at very low concentrations, that have both:
- endogenous origin (normal and abnormal physiological processes)
  - exogenous origin (e.g. inspiratory air, ingested food and beverages.)
- Exhaled human breath contains ~ 400 different molecules, which can serve as biomarkers for the identification and monitoring of various types of human diseases or wellness states.

## Important Biomedical Target Gases

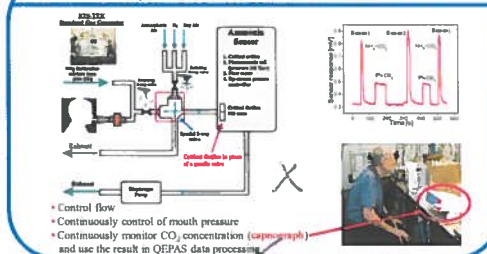
Molecule	Formula	Biological/Pathology Indication	Center wavelength [μm]
Pentane	C <sub>5</sub> H <sub>12</sub>	Inflammatory diseases, transplant rejection	6.8
Ethane	C <sub>2</sub> H <sub>6</sub>	Lipid peroxidation and oxidative stress, lung cancer (low ppbv range)	6.8
Carbon Dioxide isotopic ratio	<sup>13</sup> CO <sub>2</sub> / <sup>12</sup> CO <sub>2</sub>	Helicobacter pylori Infection (peptic ulcers, gastric cancer)	4.4
Carbonyl Sulfide	COS	Liver disease, acute rejection in lung transplant recipients (10-500 ppbv)	4.8
Carbon Disulfide	CS <sub>2</sub>	Disulfiram treatment for alcoholism	6.6
Ammonia	NH <sub>3</sub>	Liver and renal disease, ascariasis physiology	10.3
Formaldehyde	CH <sub>2</sub> O	Cancerous tumors (400-1500 ppbv)	5.7
Nitric Oxide	NO	Nitric oxide synthase activity, inflammatory and immune responses (e.g. asthma) and vascular smooth muscle response (8-100 ppb)	5.3
Hydrogen Peroxide	H <sub>2</sub> O <sub>2</sub>	Airway inflammation, oxidative stress (1-5 ppbv)	7.9
Carbon Monoxide	CO	Smoking response, lipid peroxidation, CO poisoning, vascular smooth muscle response	4.7
Ethylene	C <sub>2</sub> H <sub>4</sub>	Oxidative stress, cancer	10.6
Acetone	C <sub>3</sub> H <sub>6</sub> O	Ketoacid, diabetes mellitus	7.3

## QEPAS based NH<sub>3</sub> Gas Sensor Architecture

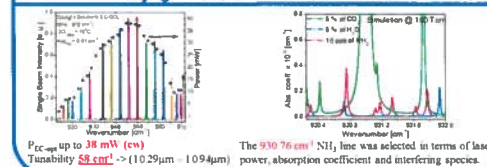


- Advantages of using CW DFB-QCL in the sensor architecture
- Small laser package → system compactness.
  - DFB-QCL RT operation & free-of using water-cooling system.
  - Performing WM spectroscopy of optimum modulation depth.
  - Baseline reduction with 2f WM

## Components of a real-time breath monitor



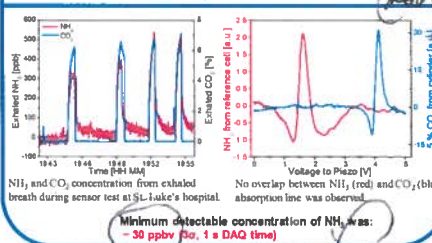
## NH<sub>3</sub> line selection with previously used 10.6 μm CW Daylight Solutions EC-QCL source



P<sub>EC</sub> = up to 38 mW (cm<sup>-1</sup>) Tunability 59 cm<sup>-1</sup> → (1029 μm - 1094 μm)

The 930.76 cm<sup>-1</sup> NH<sub>3</sub> line was selected in terms of laser power, absorption coefficient and interfering species.

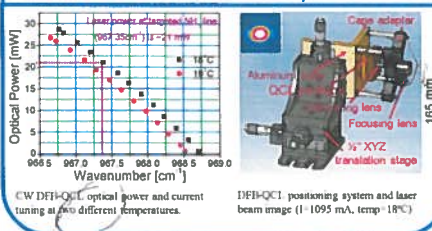
## Real-time breath sampling data based on EC-QCL - St. Luke's Hospital, Bethlehem, PA



NH<sub>3</sub> and CO<sub>2</sub> concentration from exhaled breath during sensor test at St. Luke's hospital. No overlap between NH<sub>3</sub> (red) and CO<sub>2</sub> (blue) absorption line was observed.

Minimum detectable concentration of NH<sub>3</sub> is: ~ 30 ppbv (3σ, 1 s DAQ time)

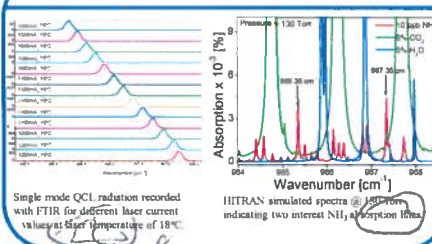
## Performance of HAMAMATSU 10.34 μm CW DFB-QCL



CW DFB-QCL optical power and current tuning at two different temperatures.

DFB-QCL positioning system and laser beam image (I = 1095 mA, temp = 18°C)

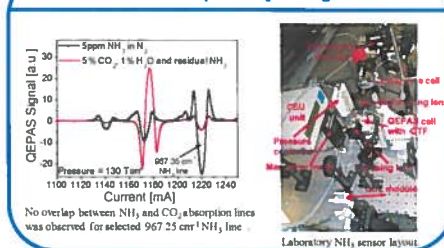
## Line selection for HAMAMATSU CW DFB QCL



Single mode QCL radiation recorded with FTIR for different laser current values at laser temperature of 18°C

HITRAN simulated spectra @ 150 Torr, indicating two interest NH<sub>3</sub> absorption lines

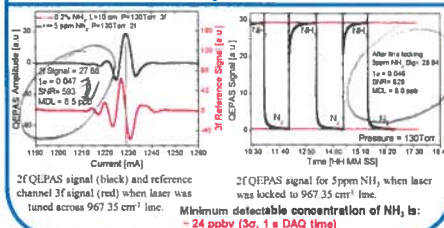
## Lab tests for developed NH<sub>3</sub> trace gas sensor



5 ppm NH<sub>3</sub> in N<sub>2</sub>, 5% CO<sub>2</sub>, 1% H<sub>2</sub>O and residual NH<sub>3</sub>. No overlap between NH<sub>3</sub> and CO<sub>2</sub> absorption lines was observed for selected 967.25 cm<sup>-1</sup> NH<sub>3</sub> line.

Laboratory NH<sub>3</sub> sensor layout

## Results for a 10.34 μm CW DFB-QCL based NH<sub>3</sub> Gas Sensor

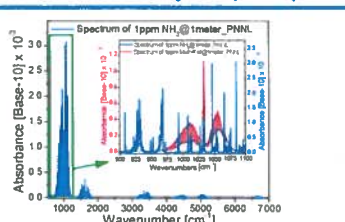


2f QEPAS signal (black) and reference channel 3f signal (red) when laser was tuned across 967.35 cm<sup>-1</sup> line. Minimum detectable concentration of NH<sub>3</sub> is: ~ 24 ppbv (3σ, 1 s DAQ time)

## Summary

- Monitoring of ammonia concentration in exhaled breath using laser spectroscopy techniques provides a fast, non-invasive diagnostic method for patients with liver and kidney disorders, and helicobacter pylori infections
- With EC-QCL based sensor platform the minimum detectable concentration of NH<sub>3</sub> in human breath was demonstrated at ~ 30 ppbv (3σ, 1 s DAQ time)
- Fast time response of ≤ 1 s was obtained by keeping sensor enclosure at 45°C (to minimize ammonia adsorption effects)
- By using a commercial capnograph the CO<sub>2</sub> concentration measurements are performed independently
- With DFB-QCL based sensor platform the minimum detectable concentration of NH<sub>3</sub> was observed at ~ 24 ppbv (3σ, 1 s DAQ time)
- Current tasks include enclosing existing NH<sub>3</sub> sensor platform into a compact and robust system and improving sensor response time by heating up sensing chamber
- Laser spectroscopy in combination with a mid-infrared, continuous wave, high performance QCL is a promising analytical approach for real time breath analysis and the quantification of breath metabolites

## Simulated Mid-IR NH<sub>3</sub> Absorption Spectra



Line strength in the NH<sub>3</sub> region (~1000 cm<sup>-1</sup>) is 200 times higher than in the near IR region (~4000 cm<sup>-1</sup>).



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*Handwritten signatures and notes:*