

Photoacoustics

Recent Developments of Quantum Cascade Laser based Trace Gas Sensor Technology: Opportunities and Challenges

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- Motivation: Wide Range of Chemical Sensing Applications
- Fundamentals of QE-Photoacoustic Spectroscopy
 - Comparison of QEPAS to L-PAS
- Selected Applications of QE-PAS
 - N₂O & CO Detection with a 4.6 μm LN₂ CW DFB Quantum Cascade Laser
 - H₂CO Detection with 3.5 μm LN₂ CW DFB Interband Cascade Laser
 - NH₃ Detection with 1.5 μm RT cw DFB Diode Laser
- Conclusions and Outlook

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OUTLINE

Photonics West
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Motivation: Wide Range of Gas Sensing Applications

- Urban and Industrial Emission Measurements**
 - Industrial Plants
 - Combustion Sources and Processes (eg. early fire sensing)
 - Automobile and Aircraft Emissions
- Rural Emission Measurements**
 - Agriculture and Animal Facilities
- Environmental Monitoring**
 - Atmospheric Chemistry (eg ecosystems and airborne)
 - Volcanic Emissions
- Chemical Analysis and Industrial Process Control**
 - Chemical, Pharmaceutical, Food & Semiconductor Industry
- Spacecraft and Planetary Surface Monitoring**
 - Crew Health Maintenance & Human Life Support Program
- Medical Diagnostics** (eg. breath analysis)
- Biohazard and Toxic Chemical Detection**
- Fundamental Science and Photochemistry**

Fundamentals of Laser Absorption Spectroscopy

Beer-Lambert's Law of Linear Absorption
 $I(\nu) = I_0 e^{-\alpha(\nu) P_0 L}$
 $\alpha(\nu)$ - absorption coefficient [$\text{cm}^{-1} \text{atm}^{-1}$]; L - path length [cm]
 ν - frequency [cm^{-1}]; P_0 - partial pressure [atm]

$\alpha(\nu) = C \cdot S(T) \cdot g(\nu - \nu_0)$
 C - total number of molecules of absorbing gas/atm cm^3 (molecule $\text{cm}^{-3} \text{atm}^{-1}$)
 S - molecular line intensity [$\text{cm}^2 \text{molecule}^{-1}$]
 $g(\nu - \nu_0)$ - normalized spectral lineshape function [cm], (Gaussian, Lorentzian, Voigt)

Optimum Molecular Absorbing Transition

- Overtones or Combination Bands (NIR)
- Fundamental Absorption Bands (MID-IR)

Long Optical Pathlengths

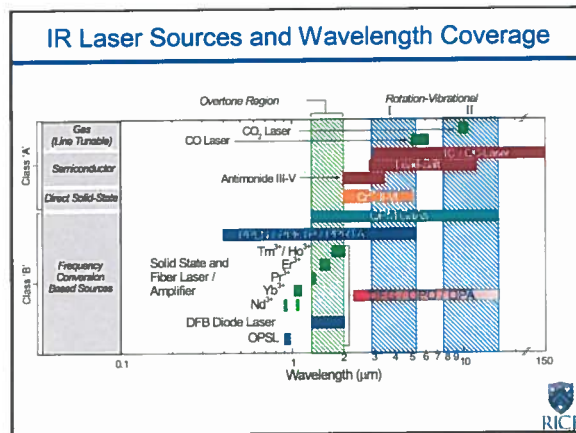
- Multipass Absorption Cell
- Cavity Enhanced and Cavity Ringdown Spectroscopy
- Open Path Monitoring (with retro-reflector)

Spectroscopic Detection Schemes

- Frequency or Wavelength Modulation
- Balanced Detection
- Zero-air Subtraction
- Photoacoustic Spectroscopy

CW IR Source Requirements for Laser Spectroscopy

REQUIREMENTS	IR SOURCE
Sensitivity (% to ppt)	Power
Selectivity	Narrow Linewidth
Multi-gas Components	Tunable Wavelengths
Directionality	Beam Quality
Rapid Data Acquisition	Fast Response
Room Temperature	No Consumables



Resonant Photoacoustic Spectroscopy

Laser beam, power P

Modulated (P or λ) at f or $f/2$

Absorption: $\alpha(\lambda)$

Cavity, resonant at f , volume V , quality factor Q

Microphone

$S_{PAS} \sim \frac{Q\alpha P}{fV}$

Sensitivity [k] = $\frac{\text{cm}^{-1} \times W}{\sqrt{\text{Hz}}}$

Quartz-Enhanced Photoacoustic Spectroscopy (QEPAS)

Laser beam, power P

Modulated (P or λ) at f or $f/2$

$$S_{QEPAS} \sim \frac{Q \alpha P}{f}$$

$$\text{Sensitivity [k]} = \frac{\text{cm}^{-1} \times \text{W}}{\sqrt{\text{Hz}}}$$

Absorption: $\alpha(\lambda)$

Piezoelectric quartz crystal (instead of microphone)

Resonant at f , quality factor Q is **>10,000** instead of 20-200 for PAS.

*Resonant Cavity in L-PAS
*Cell is OPTIONAL in QEPAS

A. Komarov et al, Optics Letters 27, 1912, 2002

Comparative Size of Absorbance Detection Modules (ADM)

Optical multipass cell (100 m):
 $l \sim 70 \text{ cm}, V \sim 3000 \text{ cm}^3$

Resonant photoacoustic cell (1000 Hz):
 $l \sim 60 \text{ cm}, V \sim 50 \text{ cm}^3$

QEPAS ADM:
 $l \sim 0.5 \text{ cm}, V \sim 0.05 \text{ cm}^3$

~5mm

Equivalent Electrical Circuit of a Quartz TF

Spring

Mass

Dashpot

C

L

R

$$\omega_0 = \sqrt{\frac{1}{LC}}$$

$$Q = \frac{1}{R} \sqrt{\frac{L}{C}}$$

$$\sqrt{\langle I_N^2 \rangle} = \sqrt{\frac{4k_B T}{R}}$$

"QUARTZ CRYSTAL RESONATORS AND OSCILLATORS For Frequency Control and Timing Applications", tutorial by John R. Vig, U.S. Army Communication-Electronics Command (July 2001)

TF & Trans-impedance Amplifier Noise Analysis

C

L

R

R_f

I_i

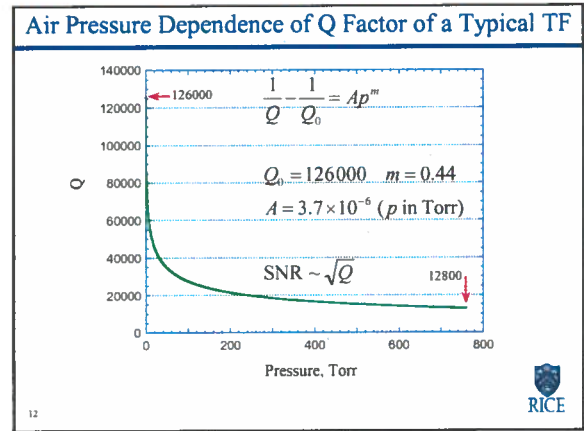
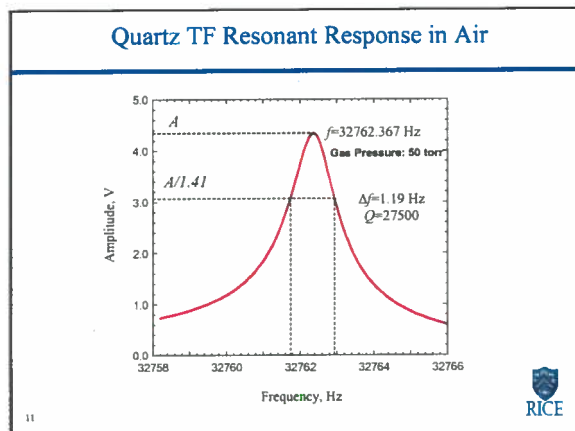
$U = I R_f$

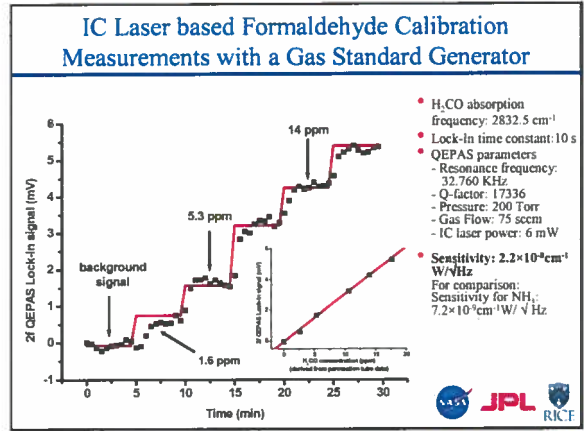
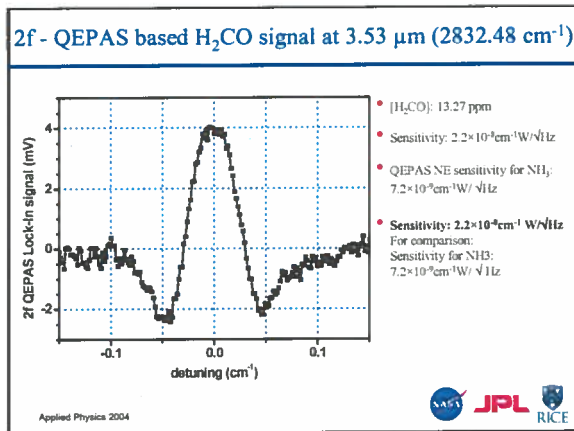
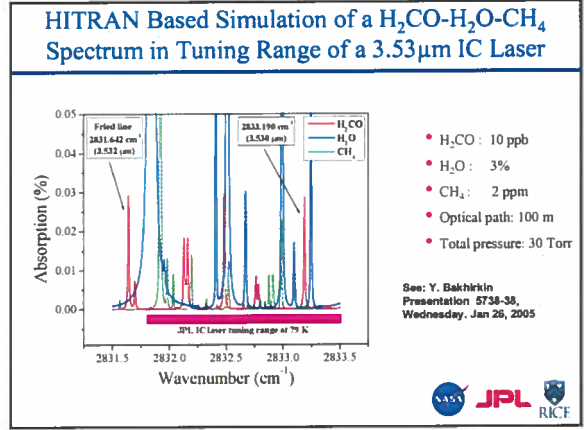
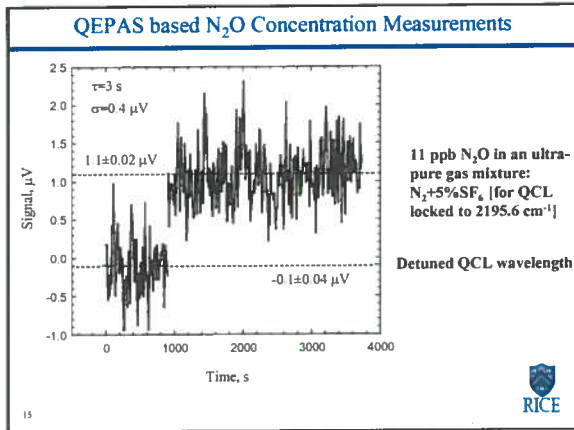
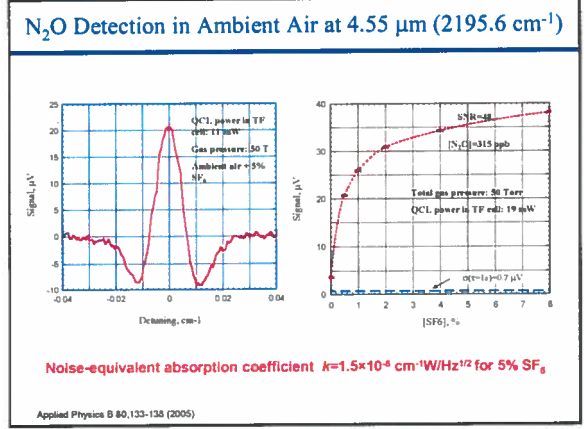
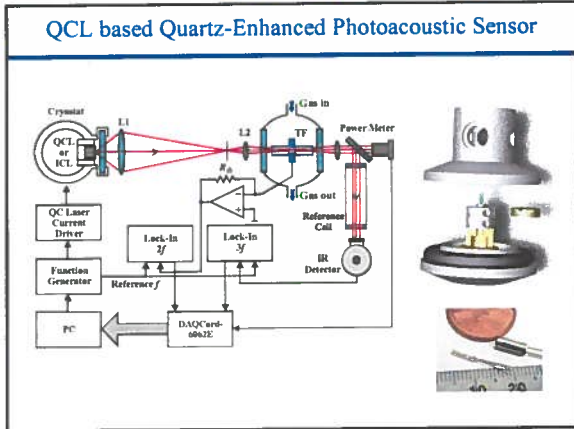
Pre-Amplifier

- Fast
- Low noise
- High impedance
- Low 1/f noise

$$S_1 = \sqrt{4k_B T R_f}; \quad R_f = 10 \text{ M}\Omega \Rightarrow S_1 = 4.1 \cdot 10^{-7} \frac{\text{V}}{\sqrt{\text{Hz}}}$$

$$S_2 = \sqrt{\frac{4k_B T}{R} R_f}; \quad R = 100 \text{ k}\Omega \Rightarrow S_2 = 4.1 \cdot 10^{-6} \frac{\text{V}}{\sqrt{\text{Hz}}} \text{ (at 760 Torr)}$$

$$S = \sqrt{S_1^2 + S_2^2} \approx S_2 \text{ (at resonance)} \quad \text{Noise goes up as } \sqrt{Q}.$$




Merits of QE Laser-PAS based Trace Gas Detection

- High sensitivity (ppm to ppb gas concentration levels) and excellent dynamic range
- Immune to ambient and flow acoustic noise, laser noise and etalon effects
- Significant reduction of sample volume (< 1 mm³)
- Applicable over a wide range of pressures
- Temperature, pressure and humidity insensitive
- Rugged and low cost compared to LAS that requires a multipass absorption cell and infrared detector(s)
- Potential for optically multiplexed concentration measurements



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QEPAS versus Traditional PAS

Parameter	Traditional PAS	QEPAS
f , Hz	100 to 4000	Presently ~32 760
Q	20 to 200	10 000 to 30 000
Q vs. pressure	INCREASES (high spectral resolution is problematic)	DECREASES (high spectral resolution is achievable)
Sample volume	>10 cm ³	<1 mm ³
Sensitivity to ambient acoustic and flow noise	Usually high	None observed
Pathlength involved	~10 cm	(a) 0.3mm, (b) 5mm

QEPAS Performance for 7 Trace Gas Species

Molecule (Host)	Frequency, cm ⁻¹	Pressure, Torr	NNEA, cm ⁻¹ W/Hz ^{1/2}	Power, mW	NEC ($\tau=1s$), ppmv
NH ₃ (N ₂)	6528.76	60	7.2×10^{-9}	38	0.65
H ₂ O (exhaled air)	6541.29	90	8×10^{-9}	5.2	580
CO ₂ (exhaled air)	6514.25	90	1.0×10^{-8}	5.2	890
N ₂ O (air+5%SF ₆)	2195.63	50	1.5×10^{-8}	19	0.007
CO (N ₂)	2196.66	50	5.3×10^{-7}	13	0.5
CO (propylene)	2196.66	50	7.4×10^{-8}	6.5	0.14
CH ₂ O (air)	2832.48	200	2.2×10^{-8}	3.4	0.55

NNEA – normalized noise equivalent absorption coefficient.

NEC – noise equivalent concentration for available laser power and $\tau=1s$ time constant.

Presently achieved QEPAS NH₃ sensitivity is 5.4×10^{-9} cm⁻¹W/√Hz (32,760 Hz)

For comparison: conventional PAS 2.2×10^{-9} cm⁻¹W/√Hz (1,800 Hz)*

* M. E. Webber, M. Poddarsky and C. K. N. Patel, Appl. Opt. 42, 2119-2126 (2003)

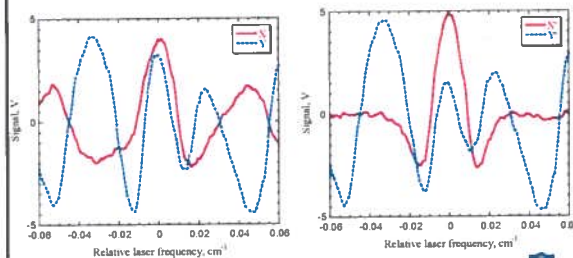


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Conclusions and Future Directions

- **Laser based Trace Gas Sensors**
 - Compact and robust sensors based on QE L-PAS and QC-LAS
 - QEL-PAS is immune to ambient noise.
 - TF sensitivity is limited by thermal excitation of symmetric mode.
 - Best demonstrated minimum detectable absorption coefficient is 5.4×10^{-9} cm⁻¹W/√Hz
 - Dramatic reduction of sample volume (~0.2 mm³) with QE L-PAS
 - Detected trace gases: NH₃, CH₄, N₂O, CO₂, CO, NO, H₂O, COS, C₂H₄, C₂H₂, C₂H₅OH, SO₂, H₂CO and several isotopic species of C, O, N & H
- **Applications in Trace Gas Detection**
 - Environmental & Spacecraft Monitoring (NH₃, CO, CH₄, C₂H₂, N₂O, CO₂ and H₂CO)
 - Biohazard and Toxic Chemical Detection
 - Medical Diagnostics (NO, CO, COS, CO₂, C₂H₄)
 - Industrial process control and chemical analysis (NO, NH₃)
- **Future Directions and Collaborations**
 - QE L-PAS and Cavity enhanced (ICOS) and spectroscopy based applications using novel thermoelectrically cooled cw and broadly wavelength tunable quantum cascade lasers
 - Applications using new near IR interband and far-IR intersub-band quantum cascade lasers

QEPAS based CO signal in C₃H₆ before and after Phase Rotation



Applied Physics B 76,673 (2004)