



Recent Developments of Chemical Sensors based on Quartz Enhanced Photoacoustic Spectroscopy

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OUTLINE

ES 2005

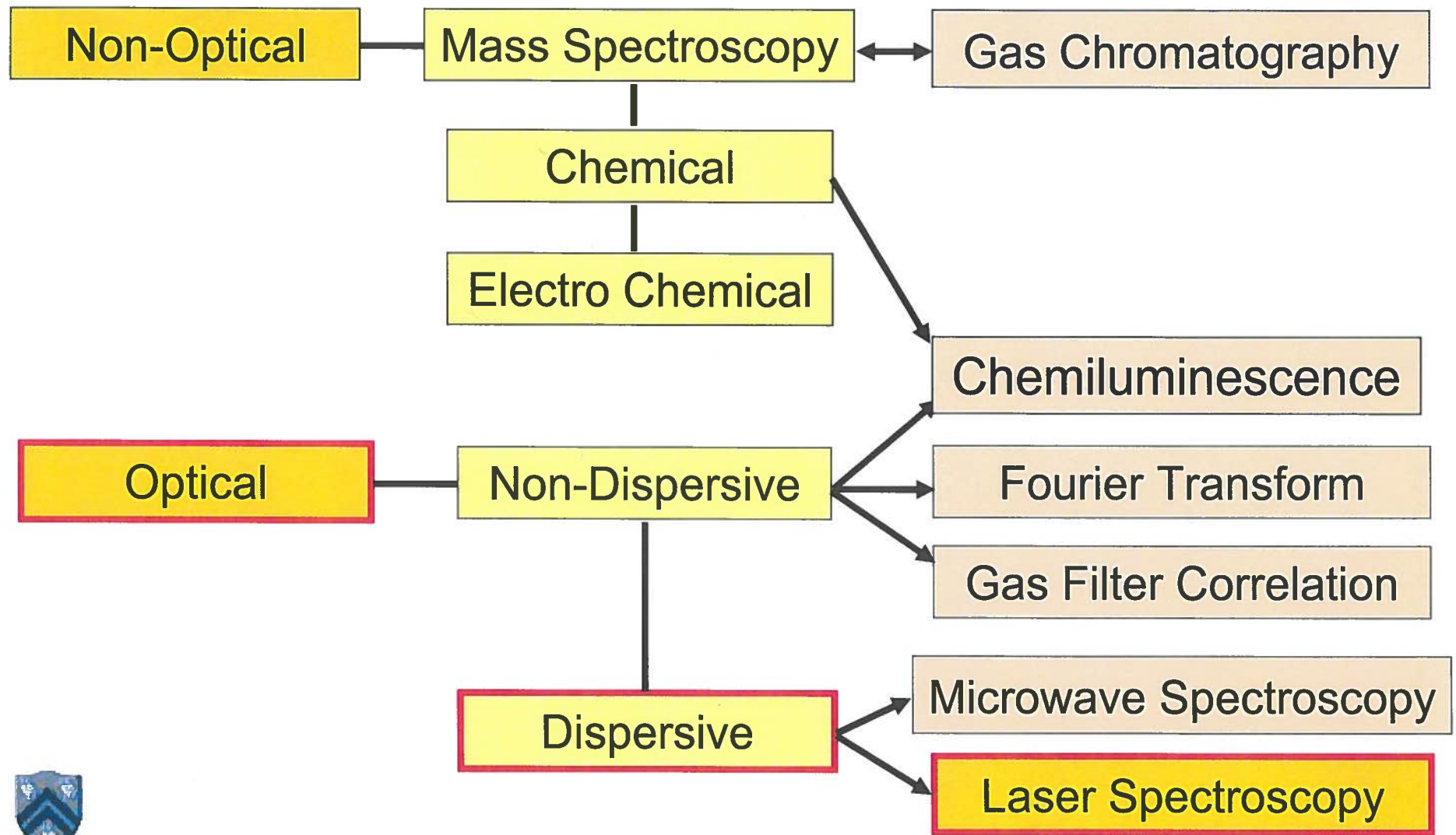
Houston, TX
June 1, 2005

- Motivation: Wide Range of Chemical Sensing
- Fundamentals of QE Photoacoustic Spectroscopy
- Selected Applications of QEPAS
 - NH₃ Detection with 1.5 μm RT cw DFB Diode Laser
 - H₂CO Detection with 3.5 μm LN₂ cw DFB interband cascade laser
 - N₂O Detection with 4.6 μm LN₂ cw DFB quantum cascade laser
- Merits, Summary and Outlook for QEPAS

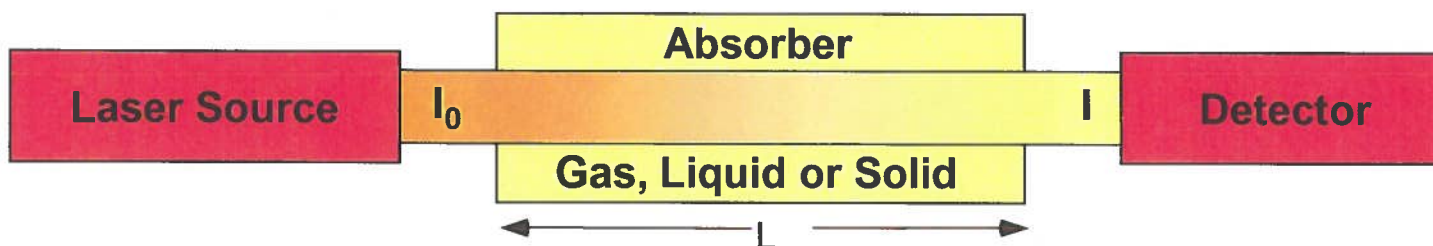
Gas sensing: wide range of applications

- **Urban and Industrial Emission Measurements**
 - Industrial Plants
 - Combustion Sources and Processes
 - Automobile
- **Rural Emission Measurements**
 - Agriculture
- **Environmental Monitoring**
 - Atmospheric Chemistry
 - Volcanic Emissions
- **Chemical Analysis and Industrial Process Control**
 - Chemical, Pharmaceutical & Semiconductor Industry
- **Spacecraft and Planetary Surface Monitoring**
 - Crew Health Maintenance & Life Support
- **Medical Applications**
- **Law enforcement**
- **Fundamental Science and Photochemistry**

Existing Methods for Trace Gas Detection



Direct Laser Absorption Spectroscopy

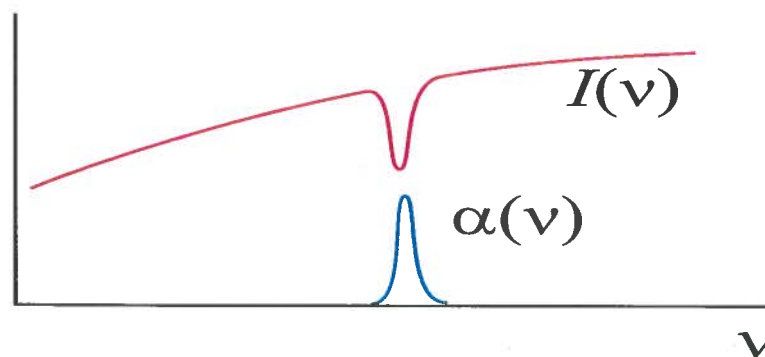


Beer-Lambert's Law of Linear Absorption

$$I(\nu) = I_0 \cdot e^{-\alpha(\nu) \cdot P_a \cdot L}$$

$\alpha(\nu)$ - absorption coefficient [$\text{cm}^{-1} \text{atm}^{-1}$]; L - path length [cm]

ν - frequency [cm^{-1}]; P_a - partial pressure [atm]



$$\alpha(\nu) = C \cdot S(T) \cdot g(\nu - \nu_0)$$

C - total number of molecules of absorbing gas/atm/ cm^3 [$\text{molecule} \cdot \text{cm}^{-3} \cdot \text{atm}^{-1}$]

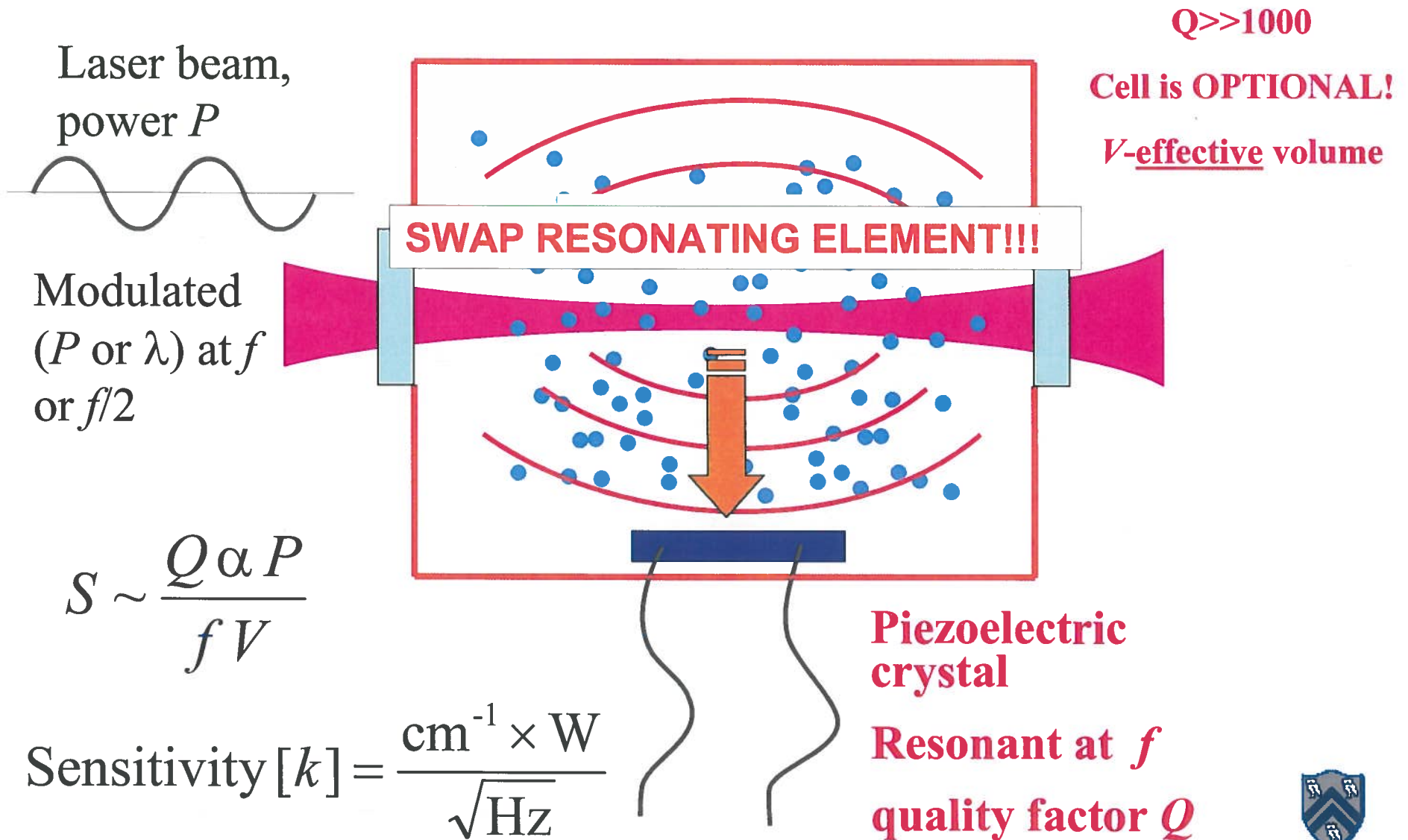
S - molecular line intensity [$\text{cm} \cdot \text{molecule}^{-1}$]

$g(\nu - \nu_0)$ - normalized lineshape function [cm], (Gaussian, Lorentzian, Voigt)

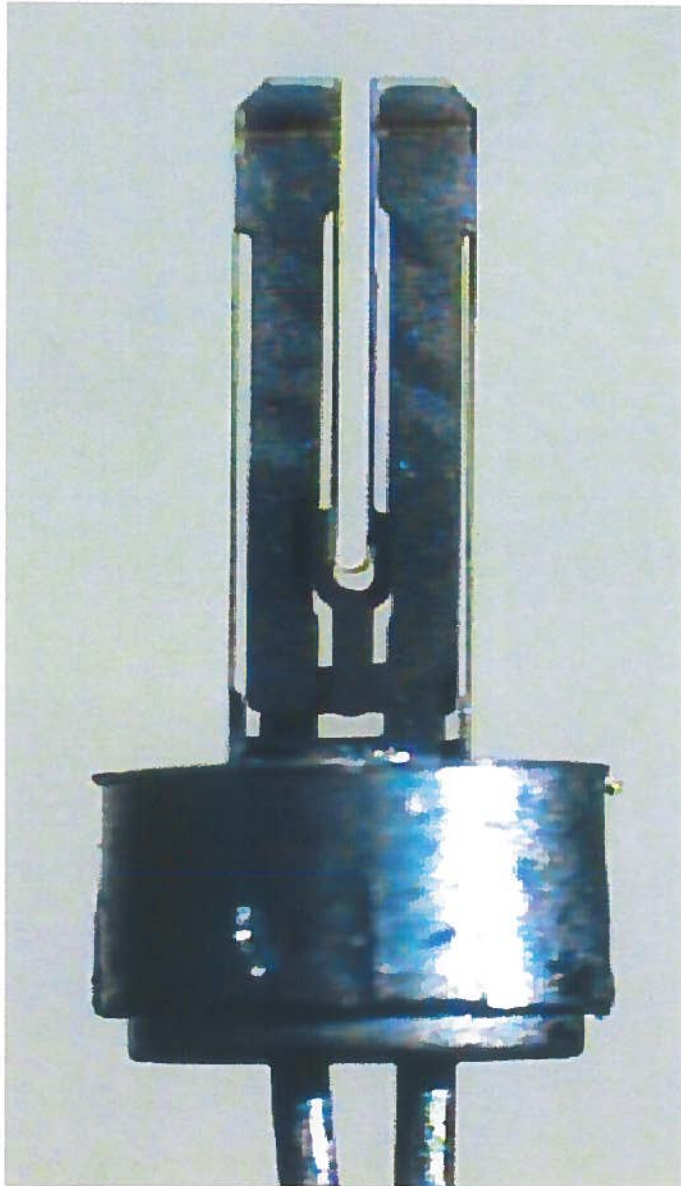
Sensitivity Enhancement Techniques

- **Optimum Molecular Absorbing Transition**
 - Overtone or Combination Bands (NIR)
 - Fundamental Absorption Bands (MID-IR)
- **Long Optical Pathlength**
 - Multipass Absorption Cell (White, Herriott)
 - Cavity Enhanced and Cavity Ringdown Spectroscopy
 - Open Path Monitoring (with retro-reflector)
 - Fiberoptic Evanescent Wave Spectroscopy
- **Spectroscopic Detection Schemes**
 - Frequency or Wavelength Modulation
 - Balanced Detection
 - Zero-air Subtraction
 - Photoacoustic Spectroscopy
 - Noise Immune Cavity Enhanced-Optical Heterodyne Molecular Spectroscopy (NICE-OHMS)

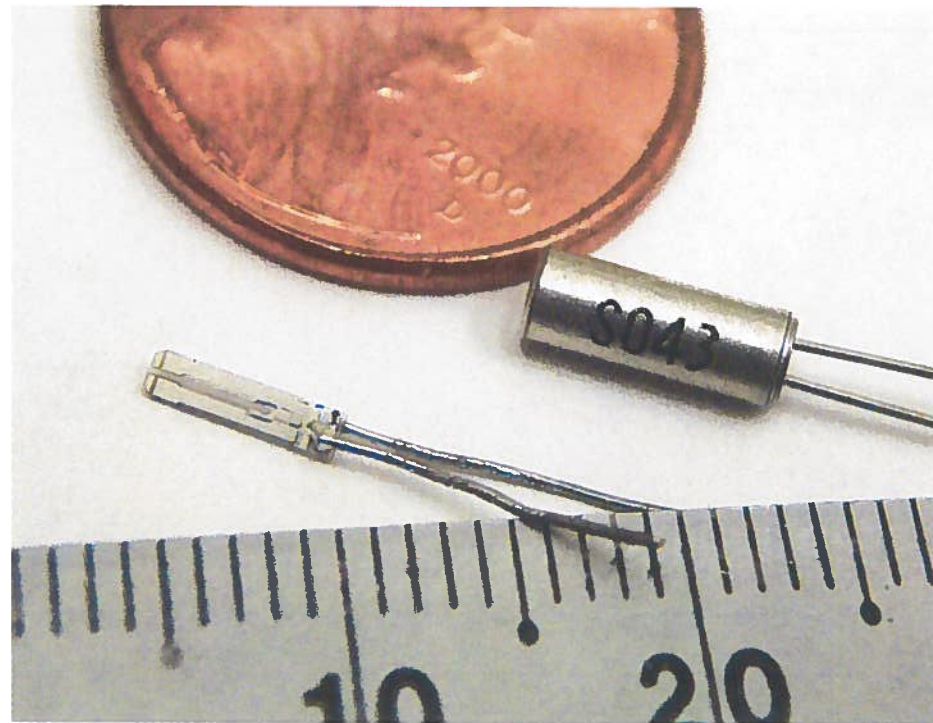
Resonant photoacoustic spectroscopy



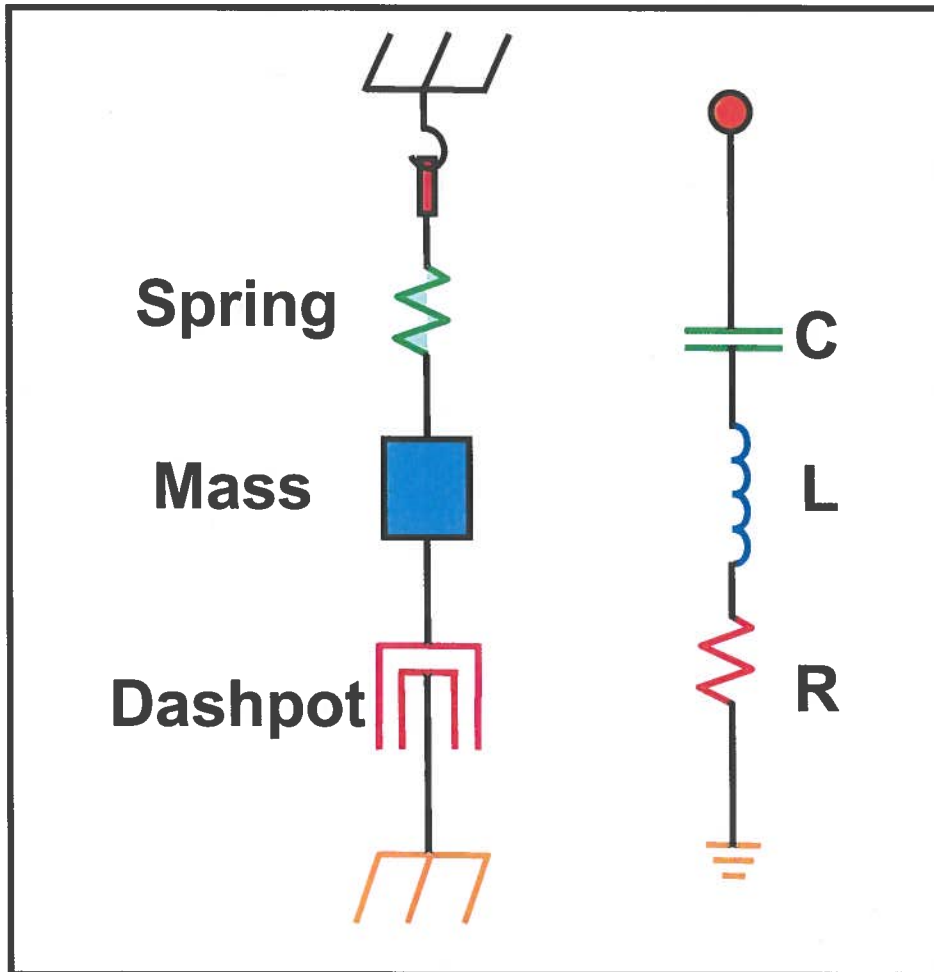
Quartz Tuning Fork as a Resonant Microphone



- Miniature size, 0.3 mm^3 detection volume
- Dimensions in mm: $l = 3.8$, $g = 0.3$, $t = 0.3$, $w = 0.58$
- Piezo-active material
- Signal currents $\approx \text{pA}$
- Intrinsically high Q factor, $\sim 10,000$ at ambient conditions; $Q_{\text{vacuum}} \sim 125,000$



Equivalent circuit of a quartz TF

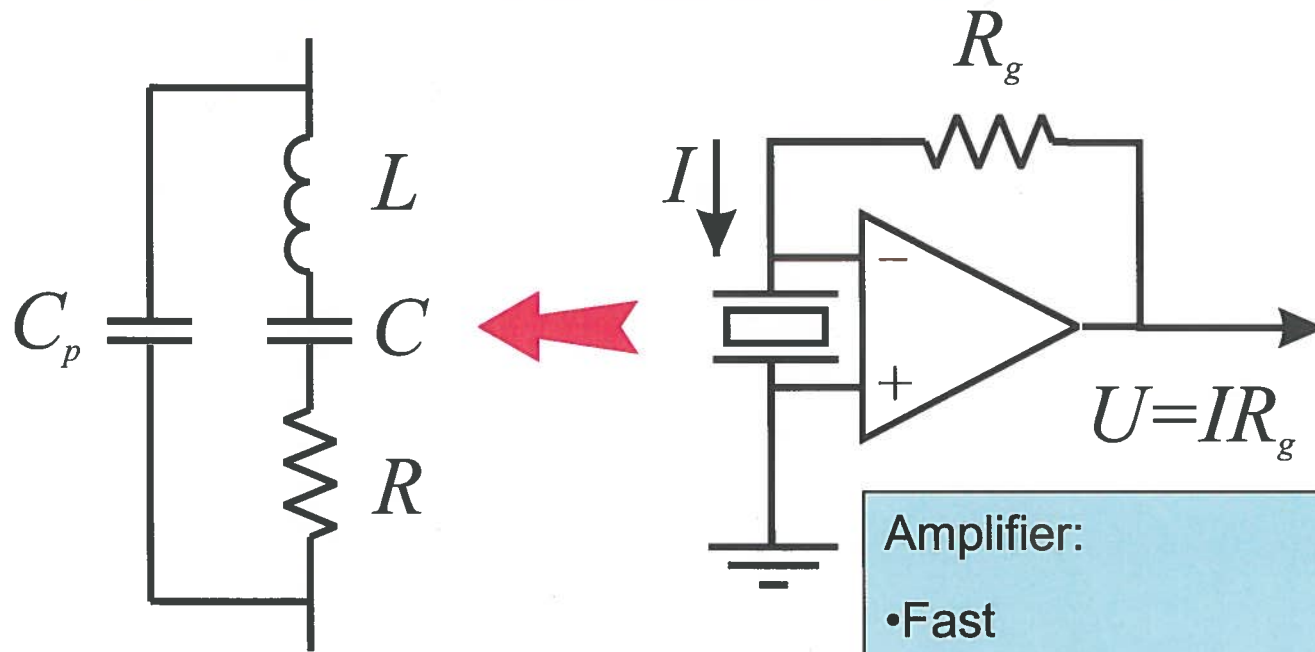


$$\omega_0 = \sqrt{\frac{1}{LC}}$$

$$Q = \frac{1}{R} \sqrt{\frac{L}{C}}$$

$$\sqrt{\langle I_N^2 \rangle} = \sqrt{\frac{4k_B T}{R}}$$

Noise analysis



Amplifier:

- Fast
- Low noise
- High impedance
- Low 1/f noise

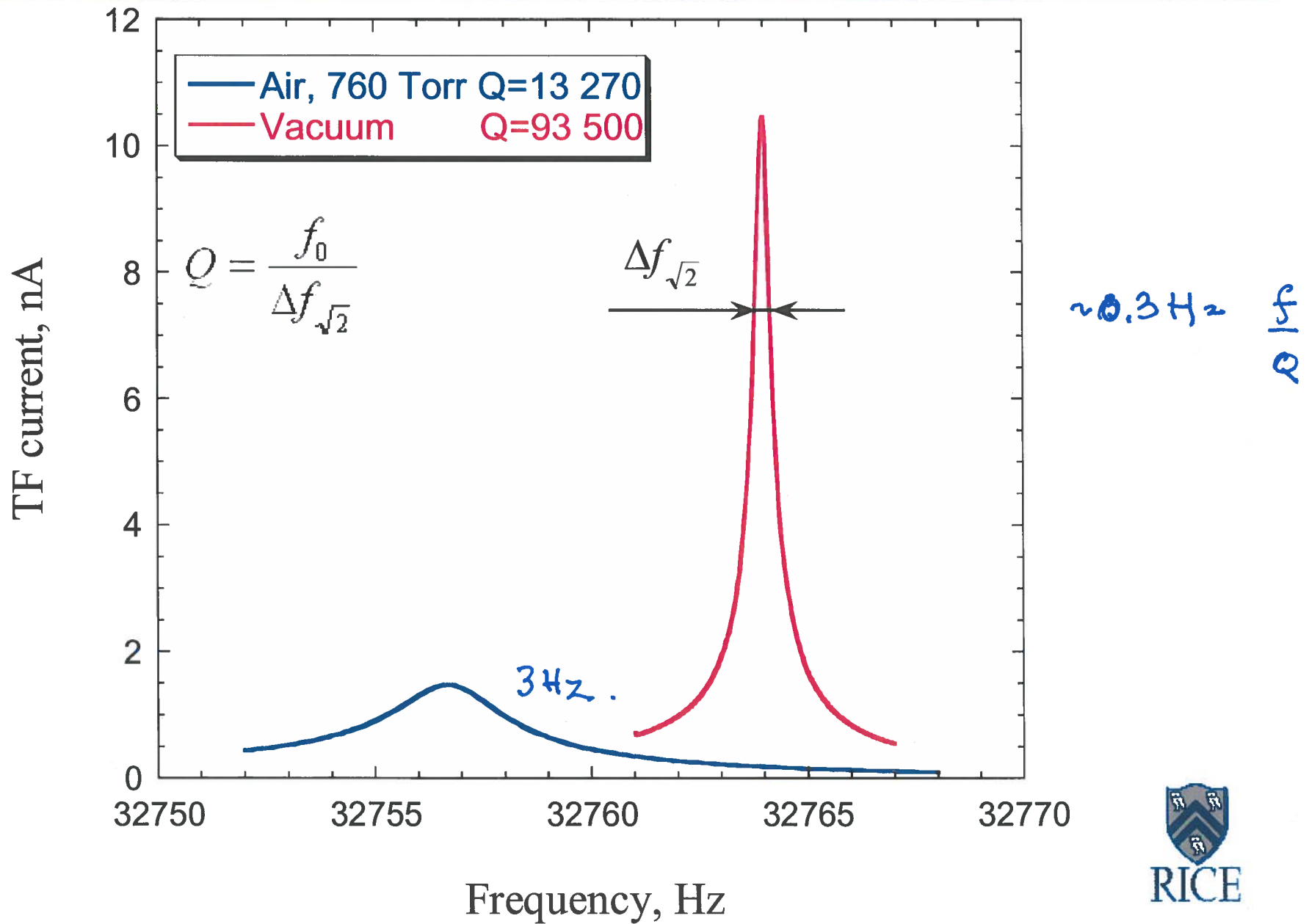
$$S_1 = \sqrt{4k_B T R_g}; \quad R_g = 10 \text{ M}\Omega \Rightarrow S_1 = 4.1 \cdot 10^{-14} \text{ V}/\sqrt{\text{Hz}}$$

$$S_2 = \sqrt{\frac{4k_B T}{R}} R_g; \quad R = 100 \text{ k}\Omega \Rightarrow S_2 = 4.1 \cdot 10^{-14} \frac{\text{V}}{\sqrt{\text{Hz}}} \text{ (at } 760 \text{ Torr)}$$

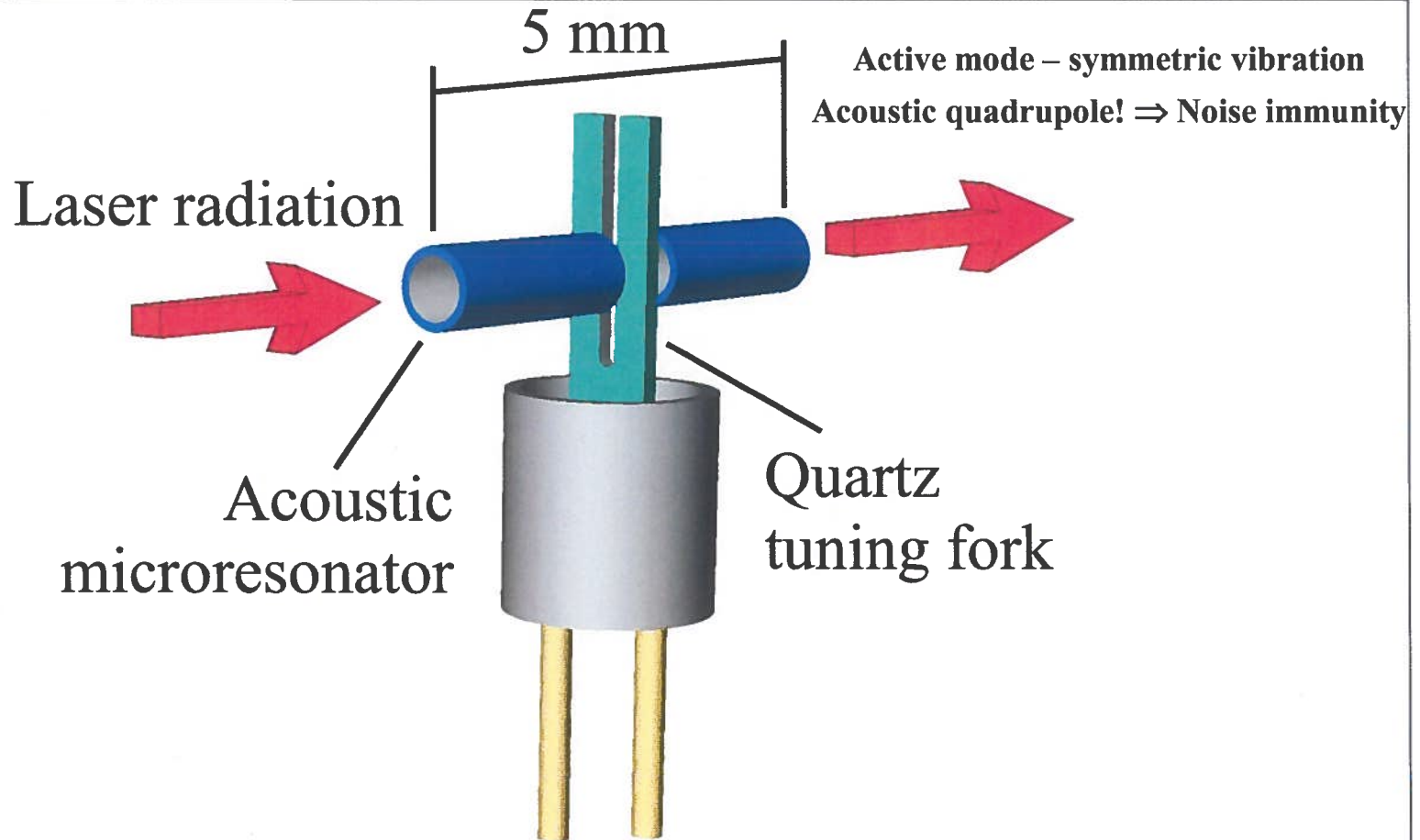
$$S = \sqrt{S_1^2 + S_2^2} \approx S_2 \quad \text{(at resonance)}$$

Noise goes up as \sqrt{Q} .

Typical QTF resonance curves



Quartz Enhanced PAS ADM



Acoustic micro-cavity is added to enhance sensitivity

NASA Target Gas Opportunity Matrix

Molecule	Detection Limit (ppb)	QEPAS detectable?	
		1.3-1.7 μm	2-5 μm
Formaldehyde	10	No	X
Acetaldehyde	20	Experiments required	
Ammonia	100	X	X
Carbon monoxide	1000	Probably not	X
Hydrogen cyanide	100	X	X
Carbon dioxide	<2%	X	X
Nitrogen dioxide	100	Probably not	X
HF	100	Experiments required	
Acrolein (2-Propenal)	5	Unlikely	
Water vapor	10-90%	X	X

X – Demonstrated

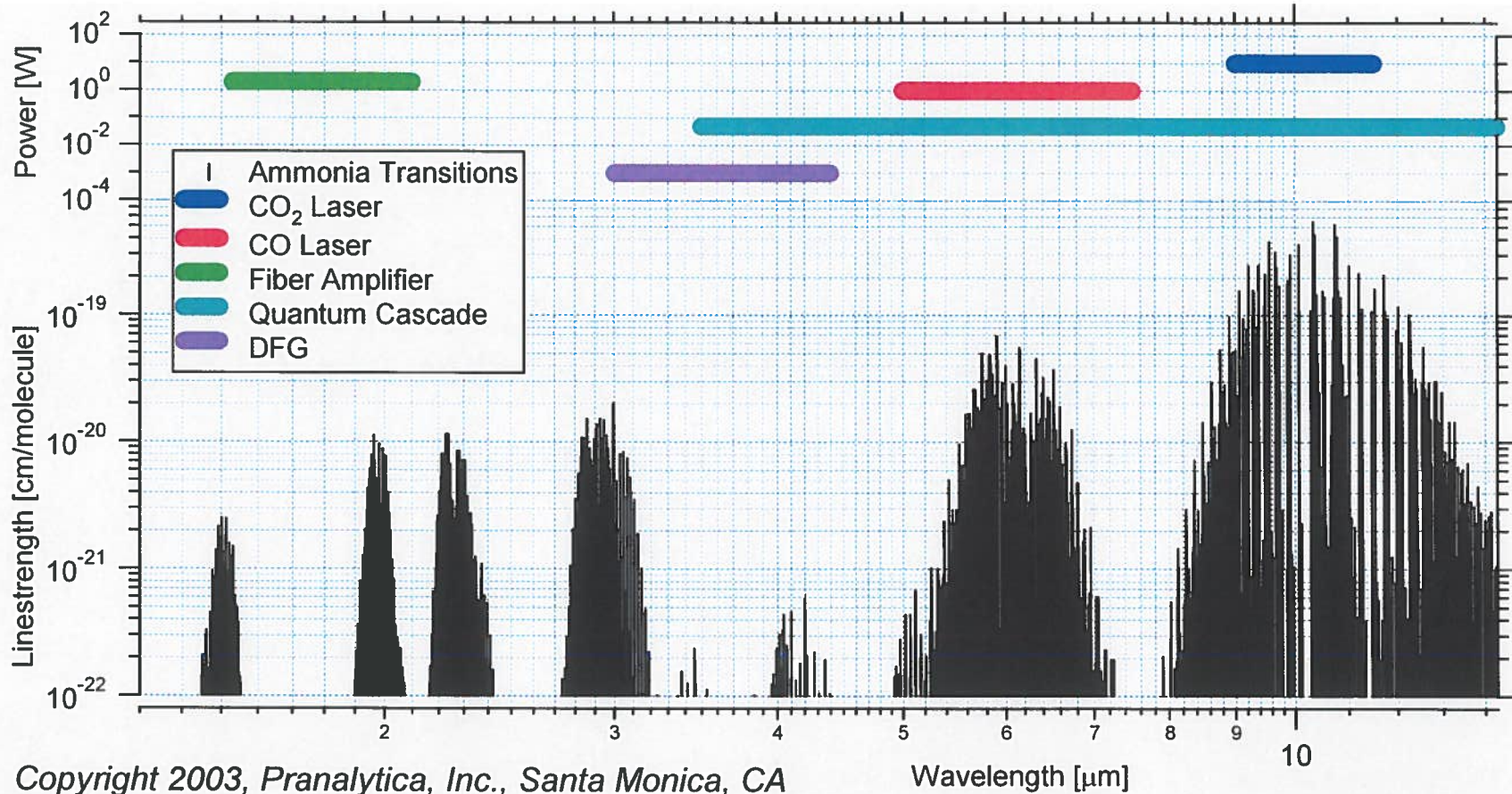
X – Highly expected based on the existing technology level

X – Expected with the technology advance

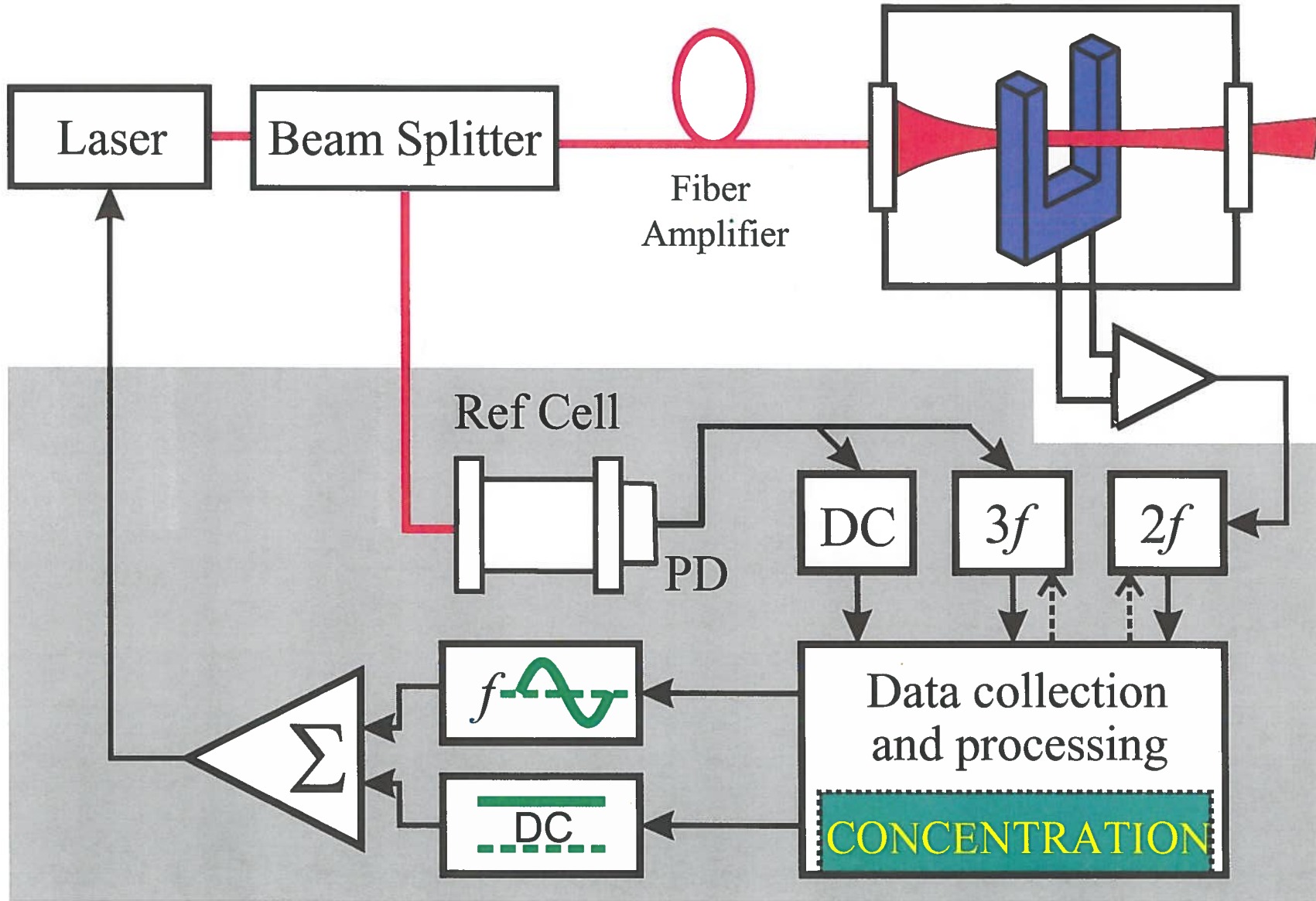
Motivation for NH₃ Detection

- Spacecraft related gas monitoring
- Monitoring NH₃ concentrations in the exhaust stream of NO_x removal systems based on selective catalytic reduction (SCR) techniques
- Semiconductor process monitoring & control
- Monitoring of industrial refrigeration facilities
- Pollutant gas monitoring
- Atmospheric chemistry
- Medical diagnostics (kidney & liver dysfunctions)

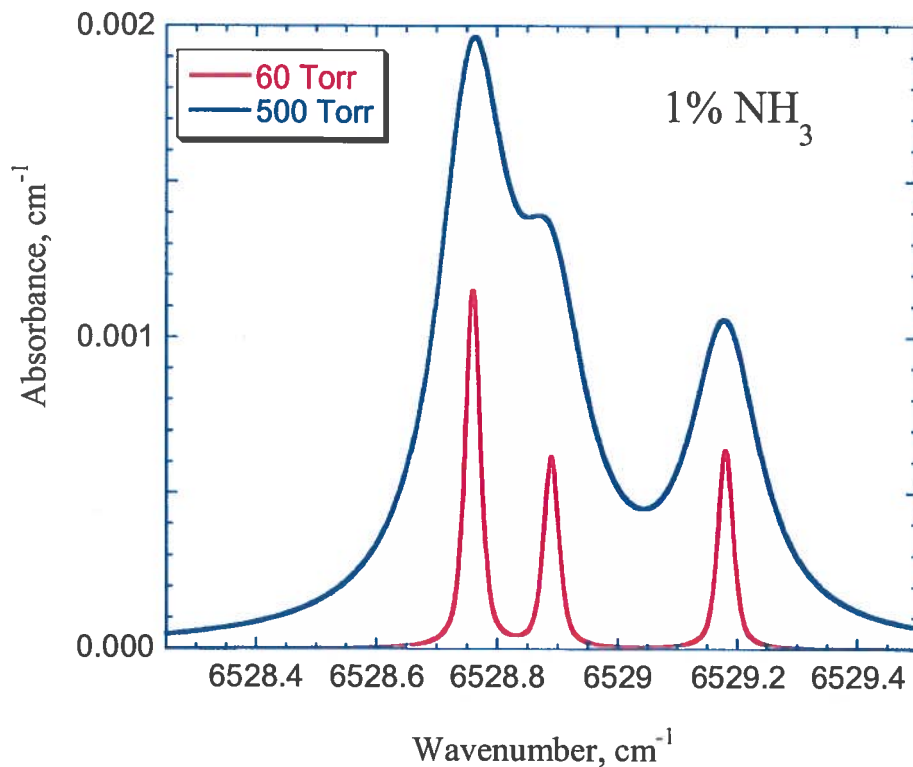
Molecular absorption spectra: NH₃



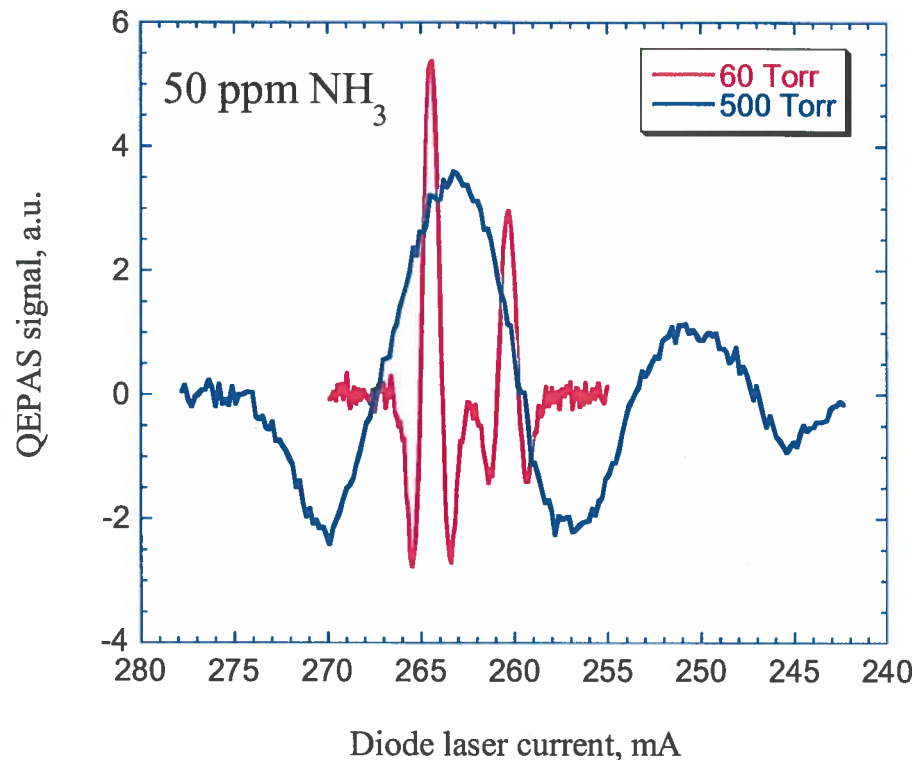
QEPAS fiber based gas sensor architecture



Ammonia Detection using a 1.53 μm Telecom Diode Laser

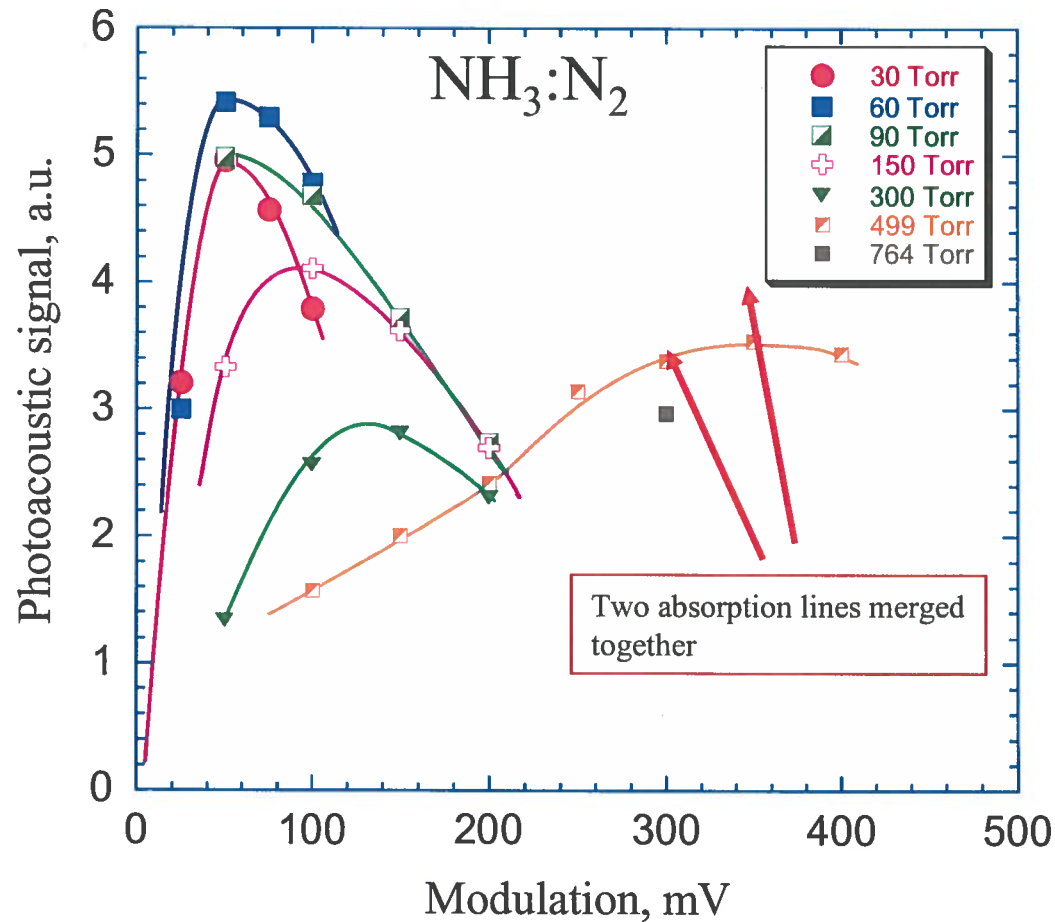


Spectral simulations based on data from Webber et al., APPLIED OPTICS 40, 2031-2042 (2001)]



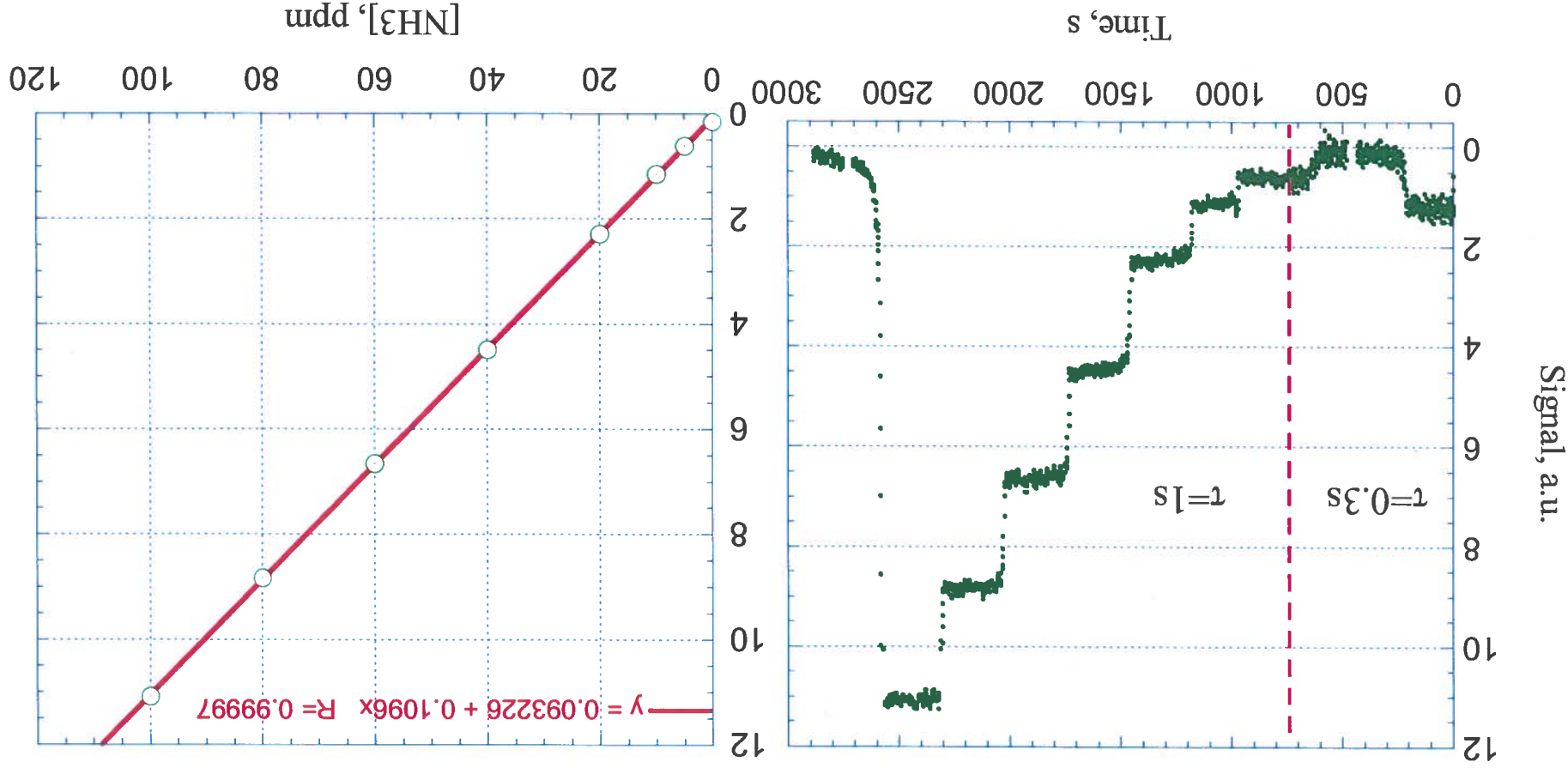
QEPAS spectra at different pressures of $\text{NH}_3:\text{N}_2$ gas mixture; $t=0.3\text{s}$, 38 mW diode laser excitation power at 6529 cm^{-1}

Pressure Dependence of QEPAS Sensitivity



- Peak optical absorption varies with pressure
- Q-factor decreases at higher pressure
- V-T relaxation is faster at higher pressure
- Acoustic resonator enhancement factor changes with pressure

Calibration and Linearity of QEPAS based NH₃ Sensor



Noise-equivalent (1s) concentration (NEC), for $\tau=1s$ time constant is 0.65 ppmv for 38 mW excitation power

90 last points of each step averaged

(Traditional PAS* - $1.5 \times 10^{-9} \text{ cm}^{-1} \text{ W} / \text{Hz}$)

*Webber et al., APPLIED OPTICS April 2003 Vol. 42, No. 12, p.2119

Noise-equivalent absorption (NEA) coefficient $k=7.2 \times 10^{-9} \text{ cm}^{-1} \text{ W} / \text{Hz}^{1/2}$

Case study: NH₃ (100 ppb target)

Presently demonstrated sensitivity – NIR: 490 ppb

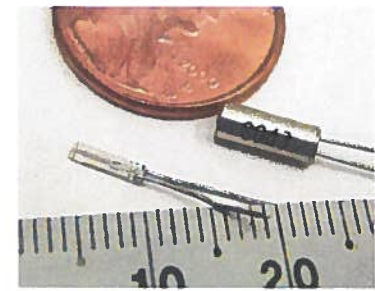
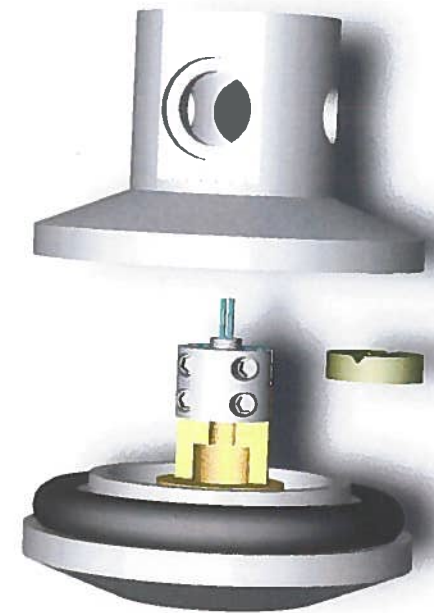
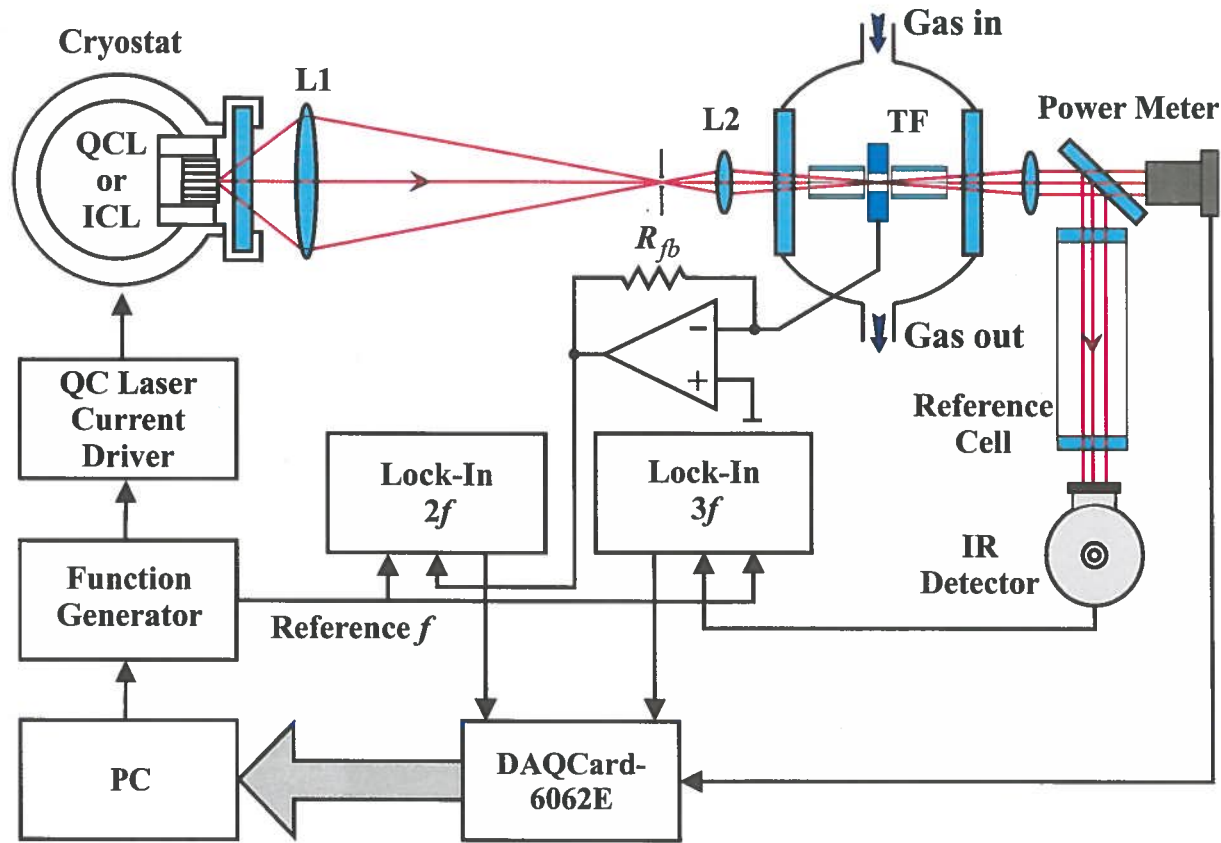
Parameter	Now	Modified	Scaling	Expected sensitivity, ppb
τ	1s	10s	3.2	155
W	38 mW	60 mW	1.6	100
f	32.7 kHz	10 kHz	3.2	35

Conclusion: QEPAS provides adequate sensitivity to NH₃ with NIR telecom lasers. Power consumption by the 63 mW JDS-Uniphase laser is (only) ~1 W.

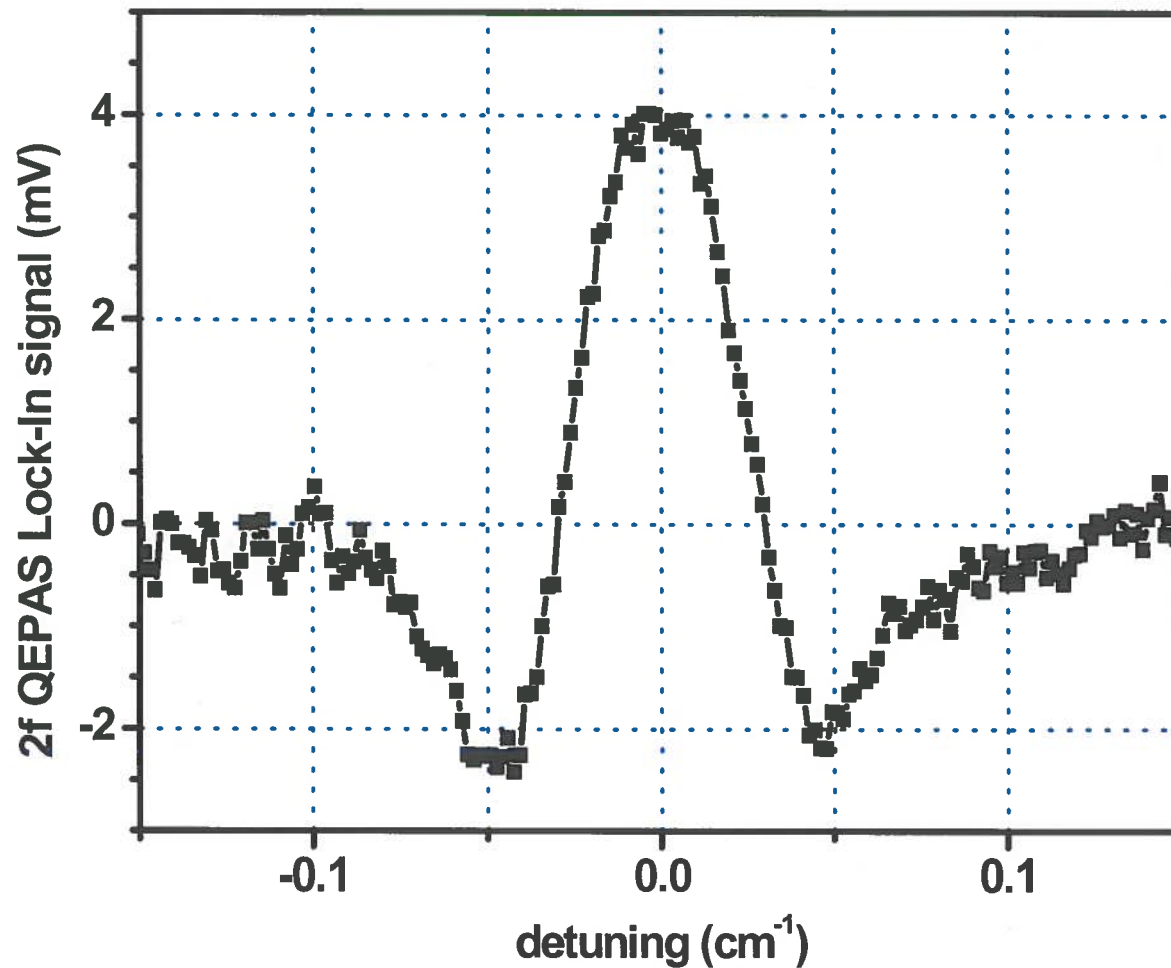
Motivation for Precision Monitoring of H₂CO

- Potential trace contaminant in industrial manufactured products
- Precursor to atmospheric O₃ production
- Pollutant due to incomplete fuel combustion processes
- Medically important gas

QCL based Quartz-Enhanced Photoacoustic Sensor

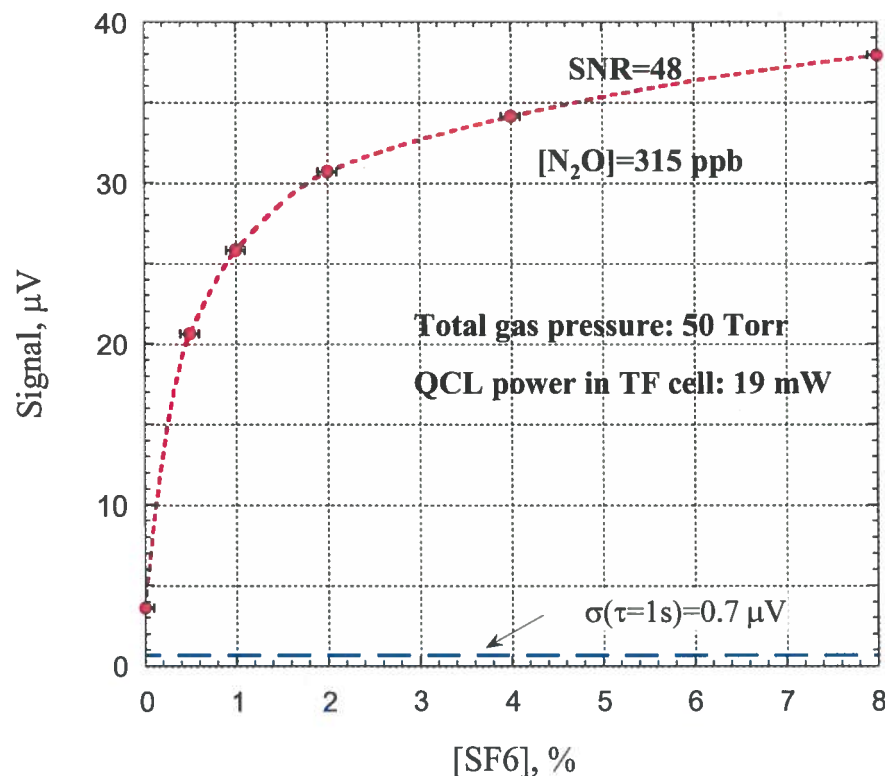
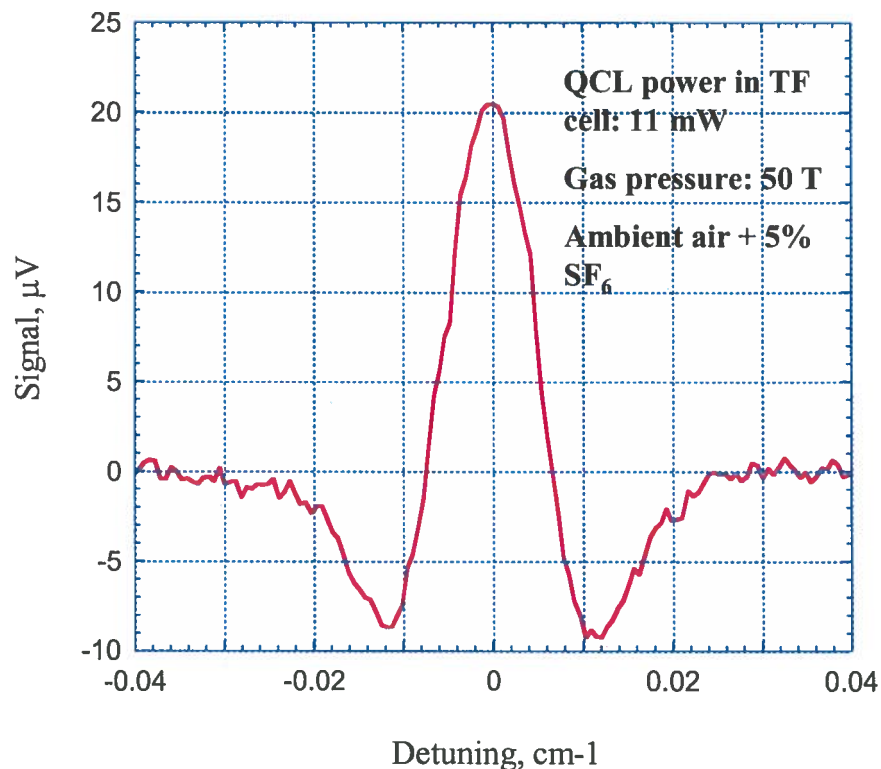


2f - QEPAS based H₂CO signal at 3.53 μm (2832.48 cm⁻¹)



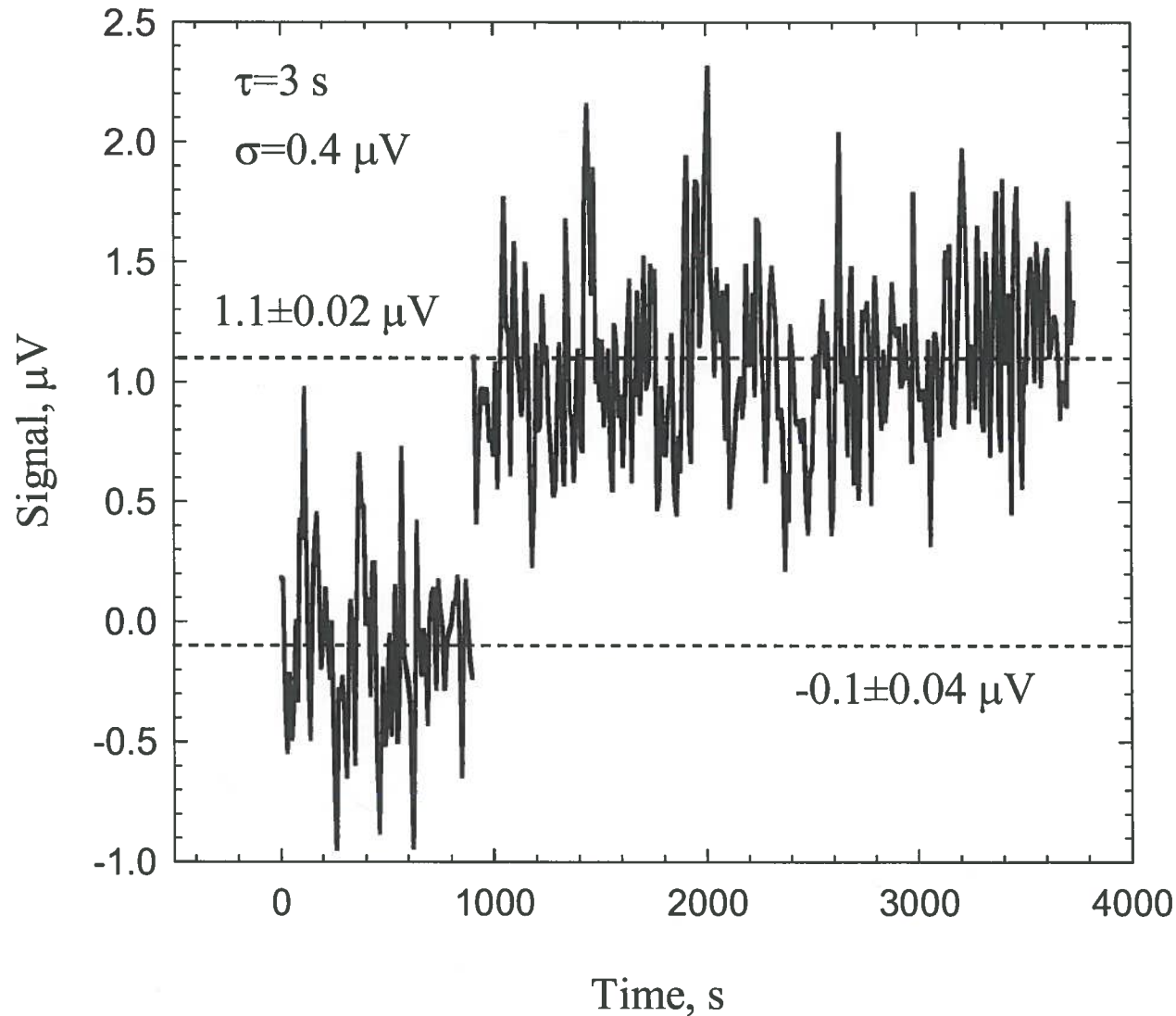
- [H₂CO]: 13.27 ppm
- Sensitivity: $2.2 \times 10^{-8} \text{ cm}^{-1} \text{ W} / \sqrt{\text{Hz}}$
- QEPAS NE sensitivity for NH₃:
 $7.2 \times 10^{-9} \text{ cm}^{-1} \text{ W} / \sqrt{\text{Hz}}$
- **Sensitivity: $2.2 \times 10^{-8} \text{ cm}^{-1} \text{ W} / \sqrt{\text{Hz}}$**
For comparison:
Sensitivity for NH₃:
 $7.2 \times 10^{-9} \text{ cm}^{-1} \text{ W} / \sqrt{\text{Hz}}$

N₂O Detection in Ambient Air at 4.55 μm (2195.6 cm⁻¹)



Noise-equivalent absorption coefficient is 1.5×10^{-8} cm⁻¹W/Hz^{1/2} for 5% SF₆ with a noise equivalent concentration of 4 ppbv for $\tau = 3$ sec

QEPAS based N₂O Concentration Measurements



**11 ppb N₂O in an ultra-pure gas mixture:
N₂+5%SF₆ [for QCL
locked to 2195.6 cm⁻¹]**

Detuned QCL wavelength

QEPAS performance for various species

Molecule (Host)	Frequency, cm^{-1}	Pressure, Torr	NNEA, $\text{cm}^{-1}\text{W}/\text{Hz}^{1/2}$	Power, mW	NEC ($\tau=1\text{s}$), ppmv
NH_3 (N_2)	6528.76	60	5.4×10^{-9}	38	0.65
H_2O (exhaled air)	6541.29	90	8×10^{-9}	5.2	580
CO_2 (exhaled air)	6514.25	90	1.0×10^{-8}	5.2	890
N_2O (air+5% SF_6)	2195.63	50	1.5×10^{-8}	19	0.007
CO (N_2)	2196.66	50	5.3×10^{-7}	13	0.5
CO (propylene)	2196.66	50	7.4×10^{-8}	6.5	0.14
CH_2O (air)	2832.48	200	2.2×10^{-8}	3.4	0.55

NNEA – normalized noise equivalent absorption coefficient.

NEC – noise equivalent concentration for available laser power and $\tau=1\text{s}$ time constant.

Presently achieved QEPAS NH_3 sensitivity is $5.4 \times 10^{-9} \text{ cm}^{-1}\text{W}/\sqrt{\text{Hz}}$ (32,760 Hz)

For comparison: conventional PAS $2.2 \times 10^{-9} \text{ cm}^{-1}\text{W}/\sqrt{\text{Hz}}$ (1,800 Hz)*

* M. E. Webber, M. Pushkarsky and C. K. N Patel, Appl. Opt. 42, 2119-2126 (2003)

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