## Quantum Cascade Laser Atmospheric Monitoring

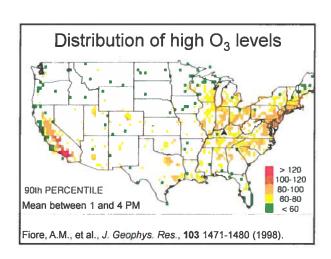
Robert Curl, Frank Tittel, Rafal Lewicky, Gerard Wysocki, University of Louisville May 6, 2013

#### Talk outline

- · A primer on urban smog
- · Quantum cascade lasers
- · Magnetic rotation spectroscopy
- Monitoring NO and NO<sub>2</sub>

# The components of smog

- H<sub>2</sub>SO<sub>3</sub> and H<sub>2</sub>SO<sub>4</sub> in killer London smogs. Different from U.S. urban smog.
- NO<sub>x</sub>, organics, and sunlight produce classic summer urban smogs of Los Angeles and Houston
- Particulates



 $\begin{array}{c} \text{The NO}_{x} \text{ cycle} \\ \text{Compared with other relevant chemistry, the} \end{array}$ reactions below are fast.

(1) 
$$NO_2 + hv \rightarrow NO + O$$
  
(2)  $O + O_2 \rightarrow O_3$   
(3)  $O_3 + NO \rightarrow NO_2 + O_2$ 

(3) 
$$O_3 + NO \rightarrow NO_2 + O_2$$

Setting up a steady state in O and O<sub>3</sub> gives.

$$[O_3] = \left(\frac{k_1[h\nu]}{k_3}\right) \frac{[NO_2]}{[NO]}$$

On a summer day, the factor in () is about 1/100 in ppm units. [NO<sub>2</sub>]/[NO] almost 10.

# Sources of NO and NO2

Although reduced by catalytic converters, NO comes primarily from vehicle exhaust.

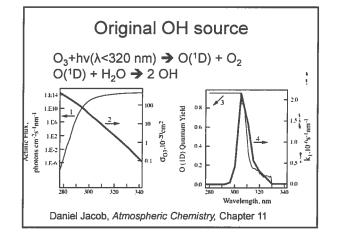
NO<sub>2</sub> is formed from NO by a series of reactions involving radicals.

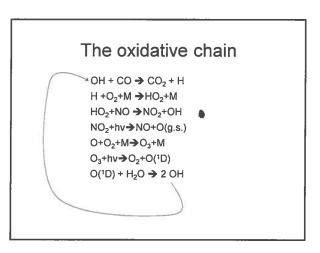
$$OH + RH \rightarrow RG + H_2O$$

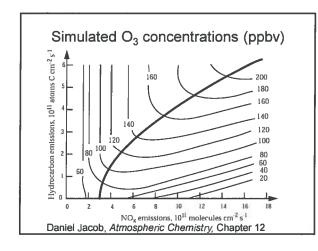
$$RG + O_2 \rightarrow RO_2$$

$$RO_2 + NO \rightarrow ROG + NO_2$$

When R=H, OH is regenerated among otherprocesses making OH. HO2 arises from the oxidation of CO.







#### Spectroscopic monitoring techniques

- ➤OH (UV laser fluorescence)
- ➤ Radicals (NO,NO<sub>2</sub>,HO<sub>2</sub>?) IR Faraday Magnetic Rotation
- ➤ CO, CO<sub>2</sub>, O<sub>3</sub> IR Photoacoustic Spectroscopy





# **OH** monitoring

OH is extremely reactive. This almost requires an in situ monitoring scheme.

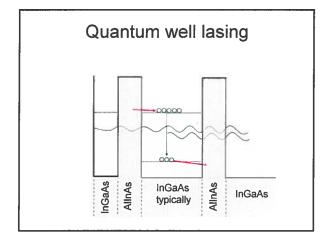
In situ detection by UV laser induced fluorescence can be done even though the fluorescence quantum yield at atmospheric pressure is tiny.

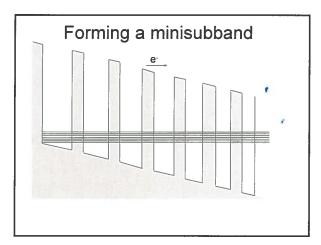
Converting its fluorescence signal into an OH concentration is difficult.

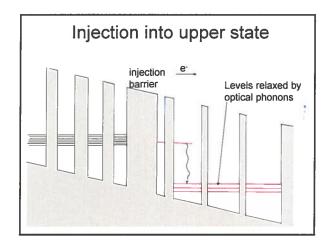
I believe no one is currently monitoring OH concentrations.

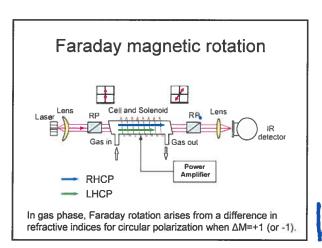
### Quantum cascade laser (QCL)

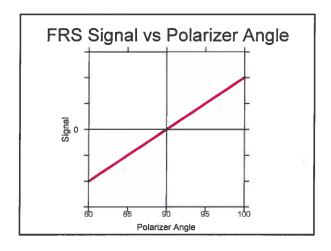
"Quantum Cascade Laser", Faist, J; Capasso, F; Sivco, DI; Sirtori, C; Hutchinson, AI; Cho, Ay. Science **264**, 553 (1994).

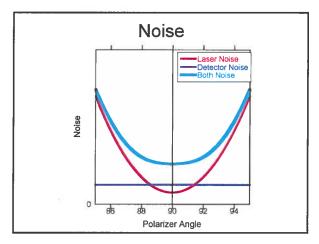


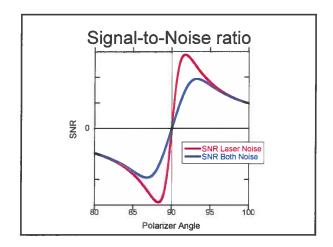










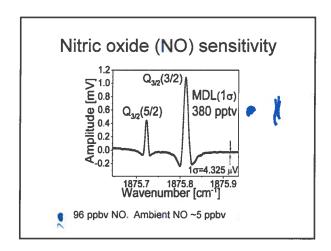


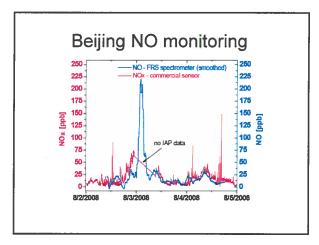
#### FRS and QCLs

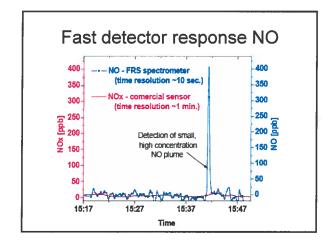
Quantum cascade lasers are very quiet. Modulating at 1 kHz, for a QCL producing a few mW, sensitivity is determined by detector noise.

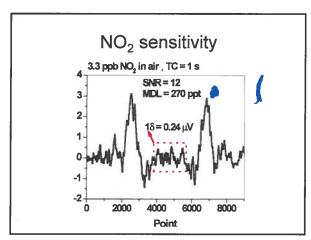
Recently QCL's producing 0.5 to 1 W cw are becoming available. For these sensitivity should be determined either by polarizer quality or power saturation.

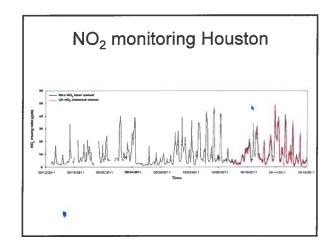
FRS observes refractive indices. These can power saturate, but far less easily than absorptions do.

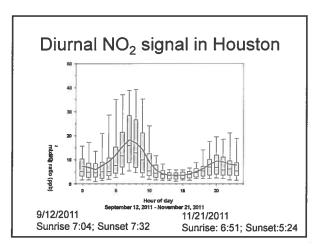


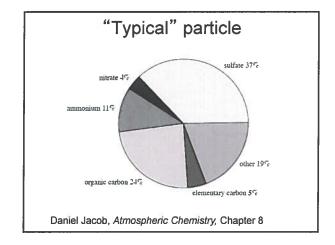










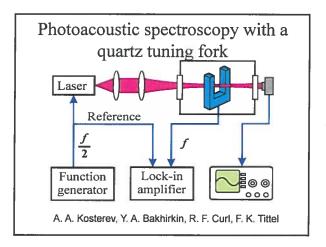


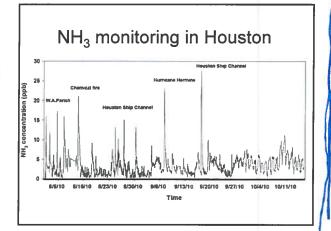
## Houston area particle sources

- ≻ From elsewhere
  - >Pollution from other urban areas
  - ➤ Remote fires: U.S. wildfires and Yucatan agricultural
  - >Sea salt
- >Very fine sand from as distant as Sahara
- > Our own particles
  - ➤ Vehicle exhaust (diesel especially)
  - ➤Industrial emissions (SO<sub>2</sub>, NH<sub>3</sub>, organics)

# Photoacoustic spectroscopy of nonmagnetic small molecules

- ➤ Salt particle formation
  - ≻NH<sub>3</sub> (done)
  - >SO₂ (not yet done)
- >Small molecule components of smog
  - >CO (done)
  - >O₃ (not yet done)





#### **QCL IR Monitoring**

- > Is sufficiently sensitive
- > Reliably selective for species
- ➤ Can be applied to most small molecules
- > Responds rapidly
- ➤ Can detect artifacts caused by pulses
- >Can be matched with wind direction
- > Is suitable for remote operations
- ➤ Needs little downtime