



# Development of QCL Based Sensor Technology for Trace Gas Monitoring Applications

F.K. Tittel, Yu. Bakhirkin, R.F.Curl, A.A.

Kosterev, S. So, D. Weidmann & G. Wysocki

Rice Quantum Institute, Rice University, Houston, TX

<http://ece.rice.edu/lasersci/>

## OUTLINE

5<sup>th</sup> QCL  
Workshop

Freiburg  
Sept. 23-24,  
2004

- Motivation and Technology Issues
- Fundamentals of Laser Absorption Spectroscopy
- Selected Applications of Trace Gas Detection
  - Off Axis-ICOS based detection of NO
  - Direct Absorption Spectroscopy of OCS
  - Quartz Enhanced Laser-PAS of N<sub>2</sub>O and H<sub>2</sub>CO
  - CO<sub>2</sub> Flux and Isotopic Ratio Measurements
- Conclusions and Outlook

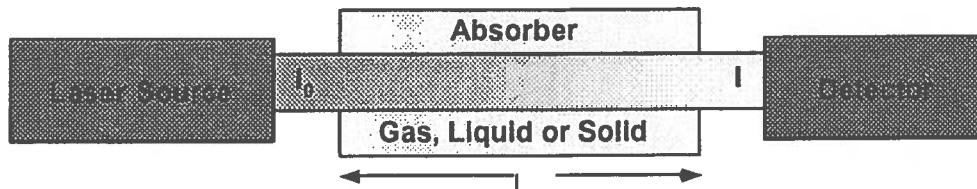
# Motivation: Wide Range of Gas Sensing Applications

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- **Urban and Industrial Emission Measurements**
  - Industrial Plants
  - Combustion Sources and Processes (eg. early fire sensing)
  - Automobile and Aircraft Emissions
- **Rural Emission Measurements**
  - Agriculture and Animal Facilities
- **Environmental Monitoring**
  - Atmospheric Chemistry (eg ecosystems and airborne)
  - Volcanic Emissions
- **Chemical Analysis and Industrial Process Control**
  - Chemical, Pharmaceutical, Food & Semiconductor Industry
- **Spacecraft and Planetary Surface Monitoring**
  - Crew Health Maintenance & Human Life Support Program
- **Medical Diagnostics (eg. breath analysis)**
- **Biohazard and Toxic Chemical Detection**
- **Fundamental Science and Photochemistry**



# Fundamentals of Laser Absorption Spectroscopy

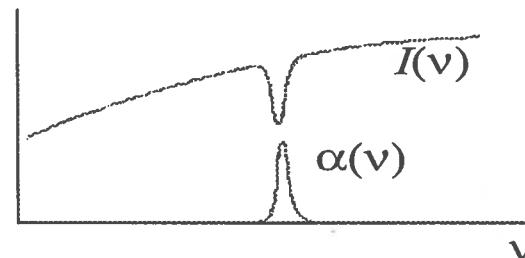


## Beer-Lambert's Law of Linear Absorption

$$I(v) = I_0 e^{-\alpha(v) P_a L}$$

$\alpha(v)$  - absorption coefficient [ $\text{cm}^{-1} \text{ atm}^{-1}$ ]; L – path length [cm]

v - frequency [ $\text{cm}^{-1}$ ];  $P_a$  - partial pressure [atm]



$$\alpha(v) = C \cdot S(T) \cdot g(v - v_0)$$

C - total number of molecules of absorbing gas/atm/cm<sup>3</sup> [molecule · cm<sup>-3</sup> · atm<sup>-1</sup>]

S – molecular line intensity [ $\text{cm} \cdot \text{molecule}^{-1}$ ]

$g(v - v_0)$  – normalized spectral lineshape function [cm], (Gaussian, Lorentzian, Voigt)

## Optimum Molecular Absorbing Transition

- Overtone or Combination Bands (NIR)
- Fundamental Absorption Bands (MID-IR)

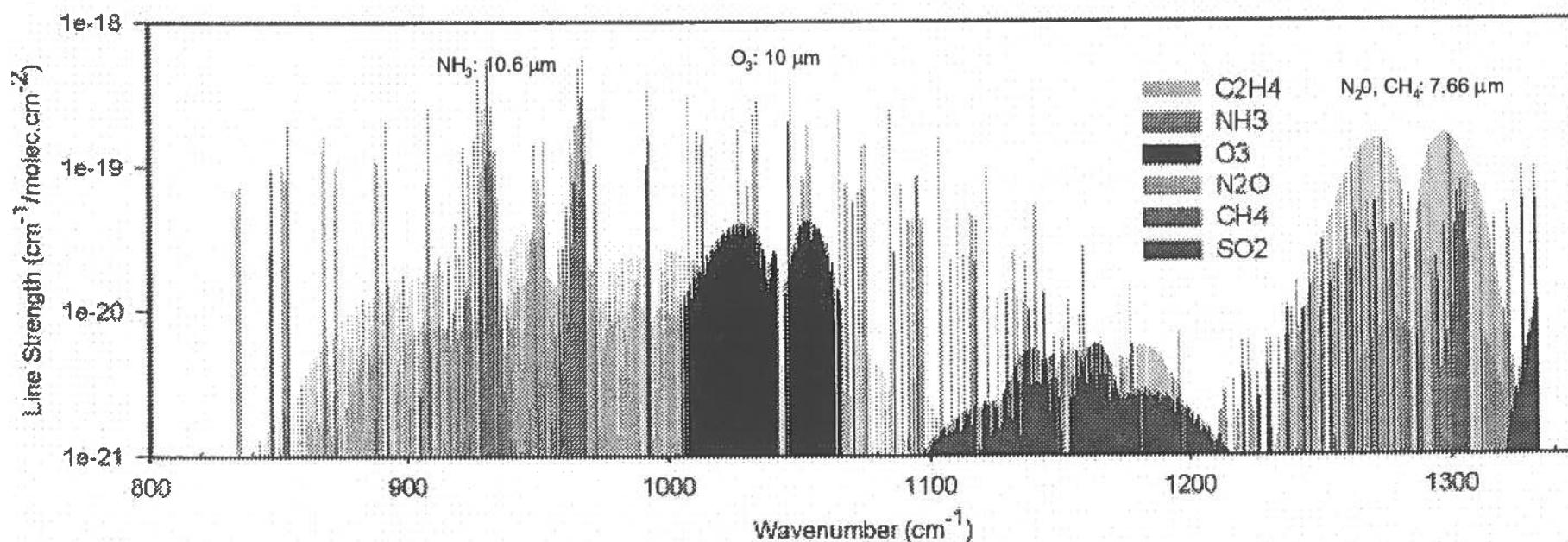
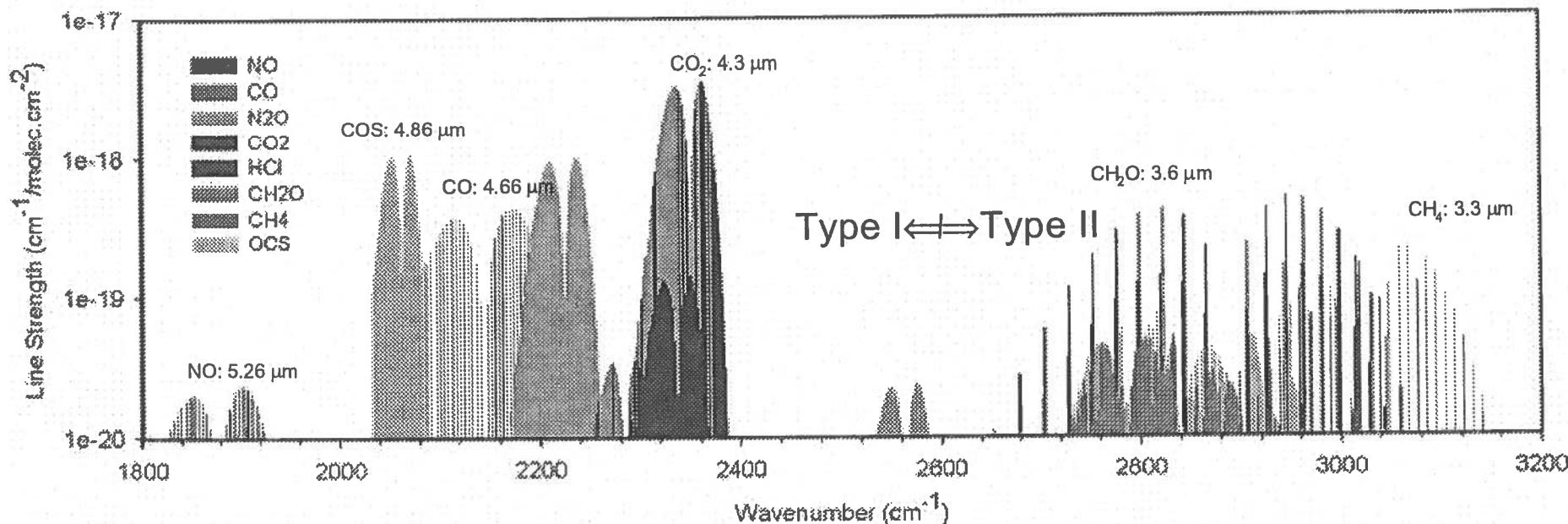
## Long Optical Pathlengths

- Multipass Absorption Cell
- Cavity Enhanced and Cavity Ringdown Spectroscopy
- Open Path Monitoring (with retro-reflector)

## Spectroscopic Detection Schemes

- Frequency or Wavelength Modulation
- Balanced Detection
- Zero-air Subtraction
- Photoacoustic Spectroscopy

# HITRAN Simulation of Absorption Spectra (3.1-5.5 & 7.6-12.5 $\mu\text{m}$ )



# CW IR Source Requirements for Laser Spectroscopy

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## REQUIREMENTS

- Sensitivity (% to ppt)
- Selectivity
- Multi-gas Components
- Directionality
- Rapid Data Acquisition
- Room Temperature

## IR SOURCE

- Power
- Narrow Linewidth
- Tunable Wavelength
- Beam Quality
- Fast Response
- No Consumables



# Key Characteristics of InGa-In Quantum Cascade Lasers for Spectroscopy

QC laser wavelengths cover entire mid-IR range from 3.3 to 24  $\mu\text{m}$  determined by thickness of the quantum well and barrier layers of the active region

Intrinsically high power lasers (determined by number of stages of injector-active quantum well gain regions)

- CW:~100 mW @ 80°K and 1- 640 mWs @ 295 °K
- Pulsed: >1 W peak at room temperature, ~50 mW avg. @ 0 °C (up to 80 % duty cycle) to 100 mWs (56% d.c)

High spectral purity (single frequency:<kHz - 330MHz)

Wavelength tunable by current ( $\sim 1 \text{ cm}^{-1}$ ) or temperature scanning ( $\sim 10 \text{ cm}^{-1}$ );  $\sim 150 \text{ cm}^{-1}$  with external cavity grating

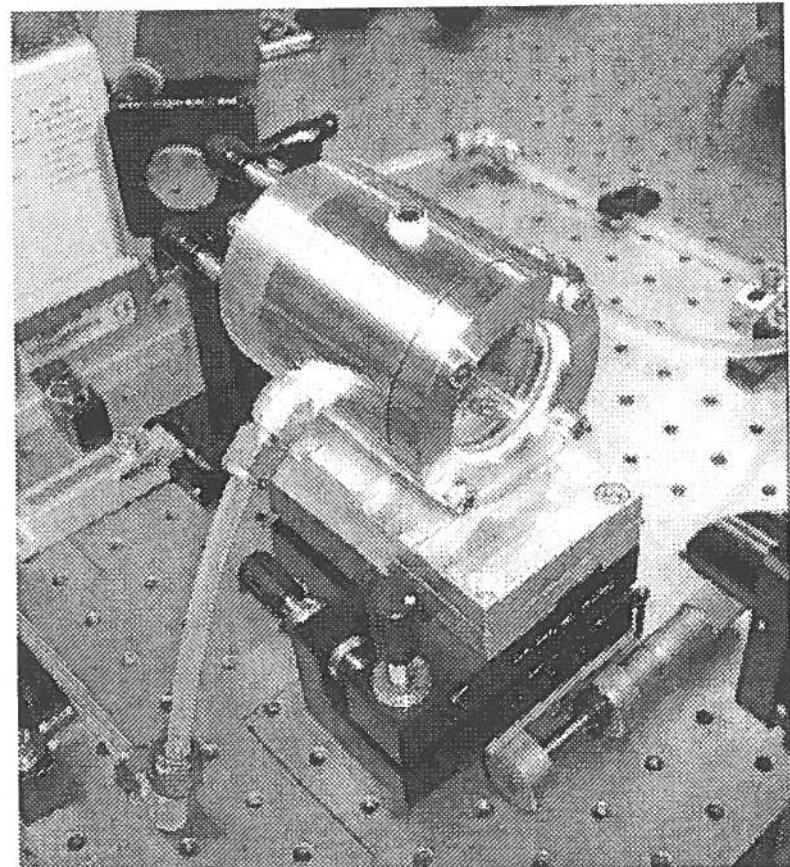
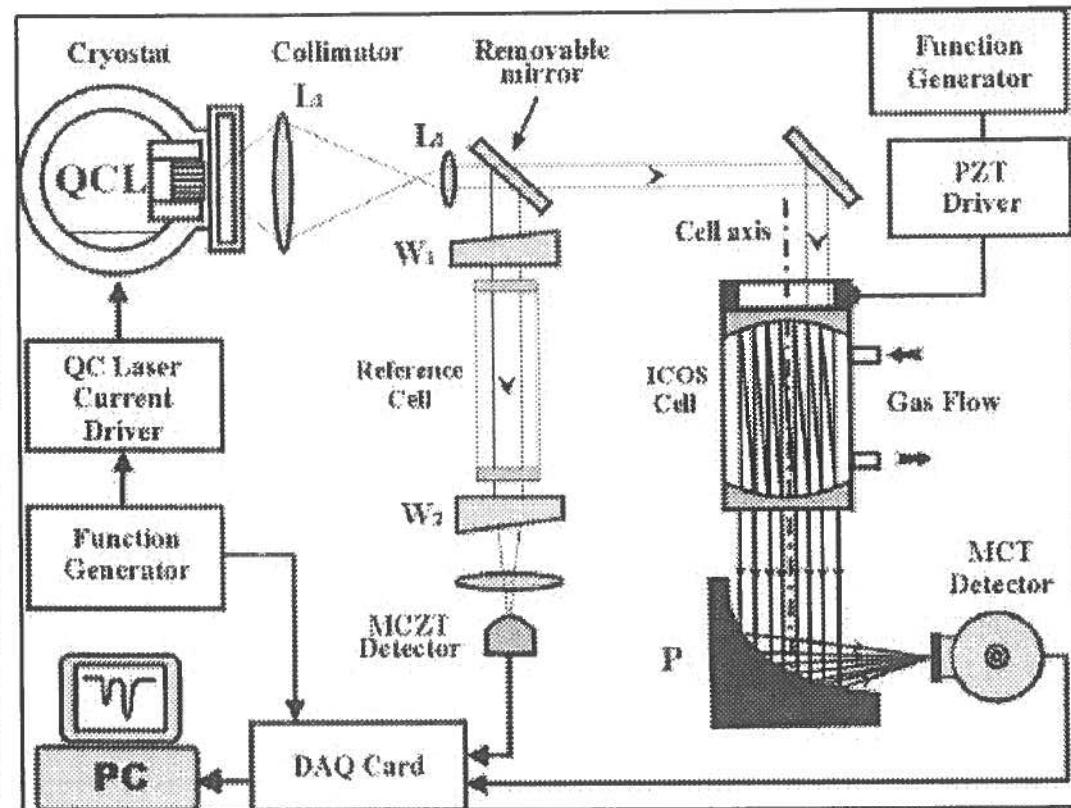
High reliability: long lifetime, robust operation and reproducible emission wavelengths



# Important Biomedical Target Gases

Molecule	Formula	Biological/Pathology Indication
Pentane	$\text{CH}_3(\text{CH}_2)_3\text{CH}_3$	Lipid peroxidation, oxidative stress associated with inflammatory diseases, transplant rejection, breast and lung cancer
Ethane	$\text{C}_2\text{H}_6$	Lipid peroxidation and oxidative stress
$\text{CO}_2$ isotope ratio	$^{13}\text{CO}_2 / ^{12}\text{CO}_2$	Marker for Helicobacter pylori infection, Gastrointestinal and hepatic function
Carbonyl Sulfide	COS	Liver disease and acute rejection in lung transplant recipients (10-500 ppb?)
Carbon disulfide	$\text{CS}_2$	Schizophrenia
Ammonia	$\text{NH}_3$	Hepatic encephalopathy, liver and renal diseases, fasting response
Formaldehyde	HCHO	Cancerous tumors, breast cancer (400-1500 ppb)
Nitric Oxide	NO	Inflammatory and immune responses (e.g., asthma) and vascular smooth muscle response (6-100 ppb)
Hydrogen Peroxide	$\text{H}_2\text{O}_2$	Airway Inflammation, Oxidative stress (1-5 ppb)
Carbon Monoxide	CO	Smoking response, CO poisoning, vascular smooth muscle response, platelet aggregation (400-3000 ppb)
Ethylene	$\text{H}_2\text{C}=\text{CH}_2$	Oxidative stress, cancer

# Off-Axis Integrated Cavity Output Spectroscopy Based Gas Sensor



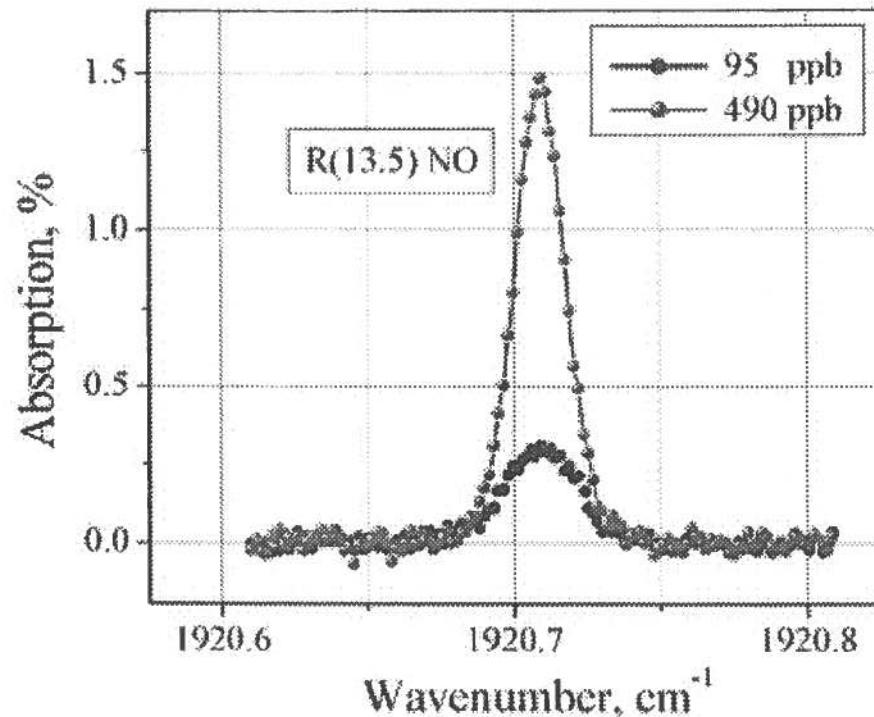
- Novel compact gas cell design of length: 3.8 – 5.3 cm and cell volumes < 80 cm<sup>3</sup>;
- Low loss mirrors (ROC 1m):~60-250 ppm, R~99.975,  $L_{\text{eff}}=170-800$  m
- Rapid eNO concentration measurements during a single breath cycle feasible

ISI

LGR

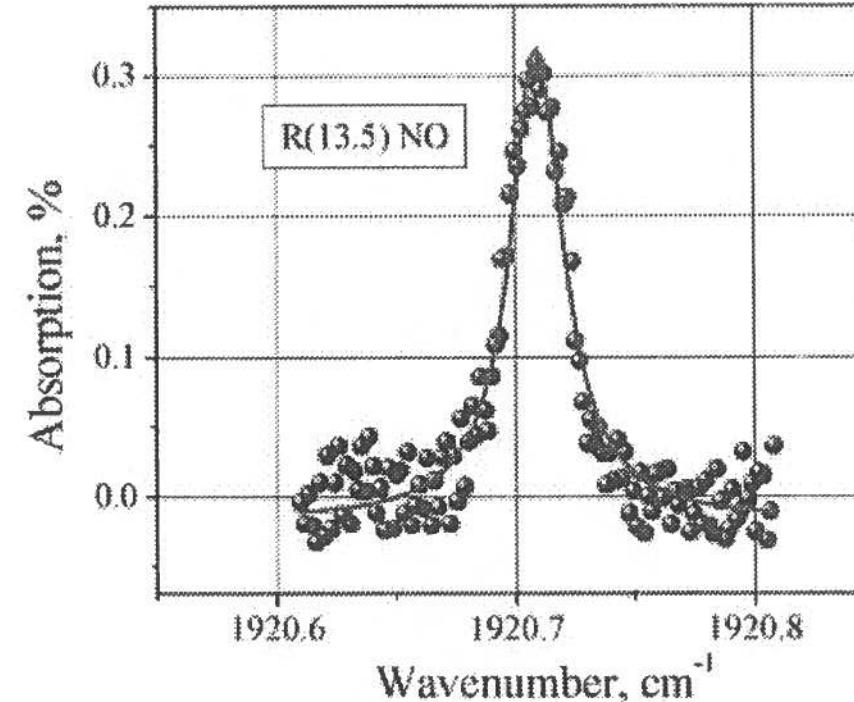


# Off-axis ICOS Detection of NO



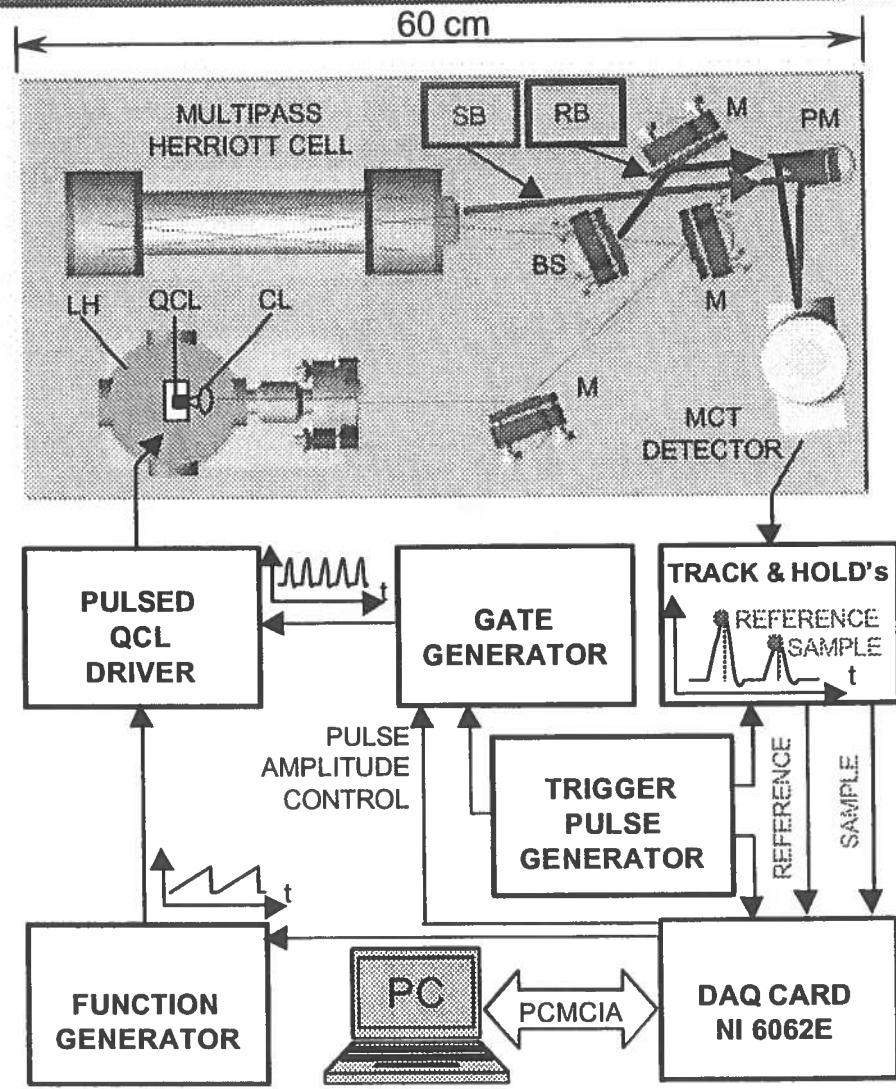
- 95 and 490 ppb NO/N<sub>2</sub> calibration mixture at 100 Torr total pressure
- Effective optical path ~ 70 m (1,350 passes)

Noise-equivalent sensitivity is 10 ppb for  $1\sigma$  deviation of the best fit coefficient.  
Detection sensitivity:  $1.0 \times 10^{-7} \text{ cm}^{-1} \text{ Hz}^{-1/2}$



Voigt fit of measured NO absorption line at  $1920.7 \text{ cm}^{-1}$  for a concentration of 95 pp

# OCS Sensor Architecture



QCL – quantum cascade laser chip

LH - laser housing

CL - collimating lens

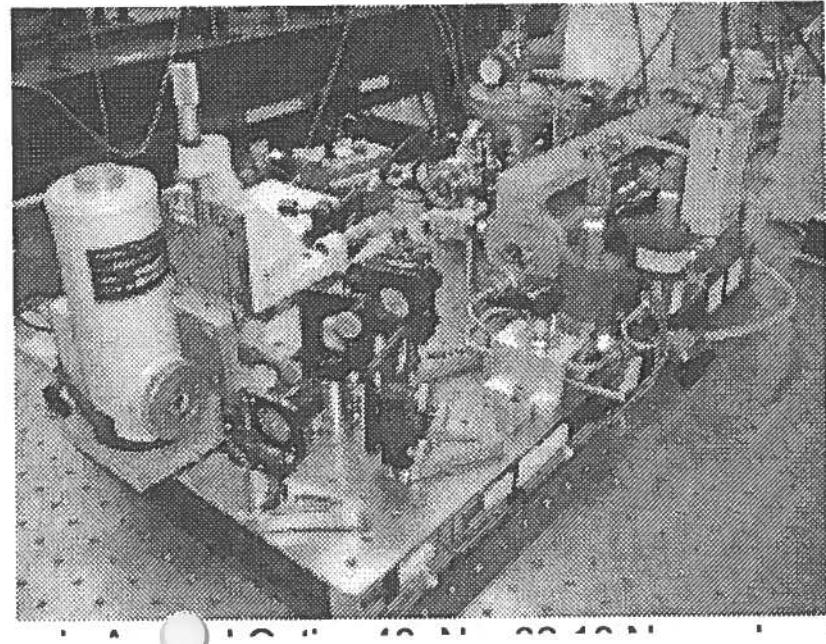
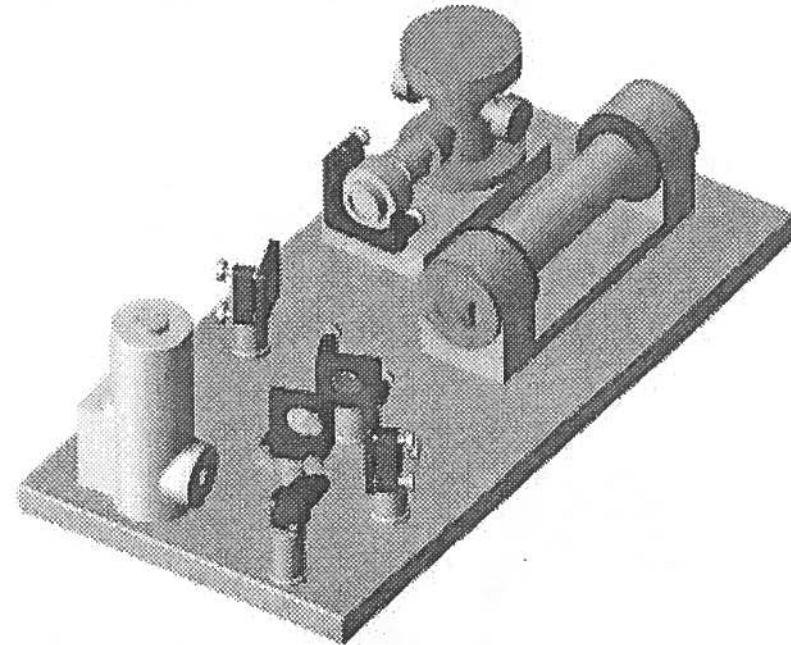
SB - sample beam

RB – reference beam

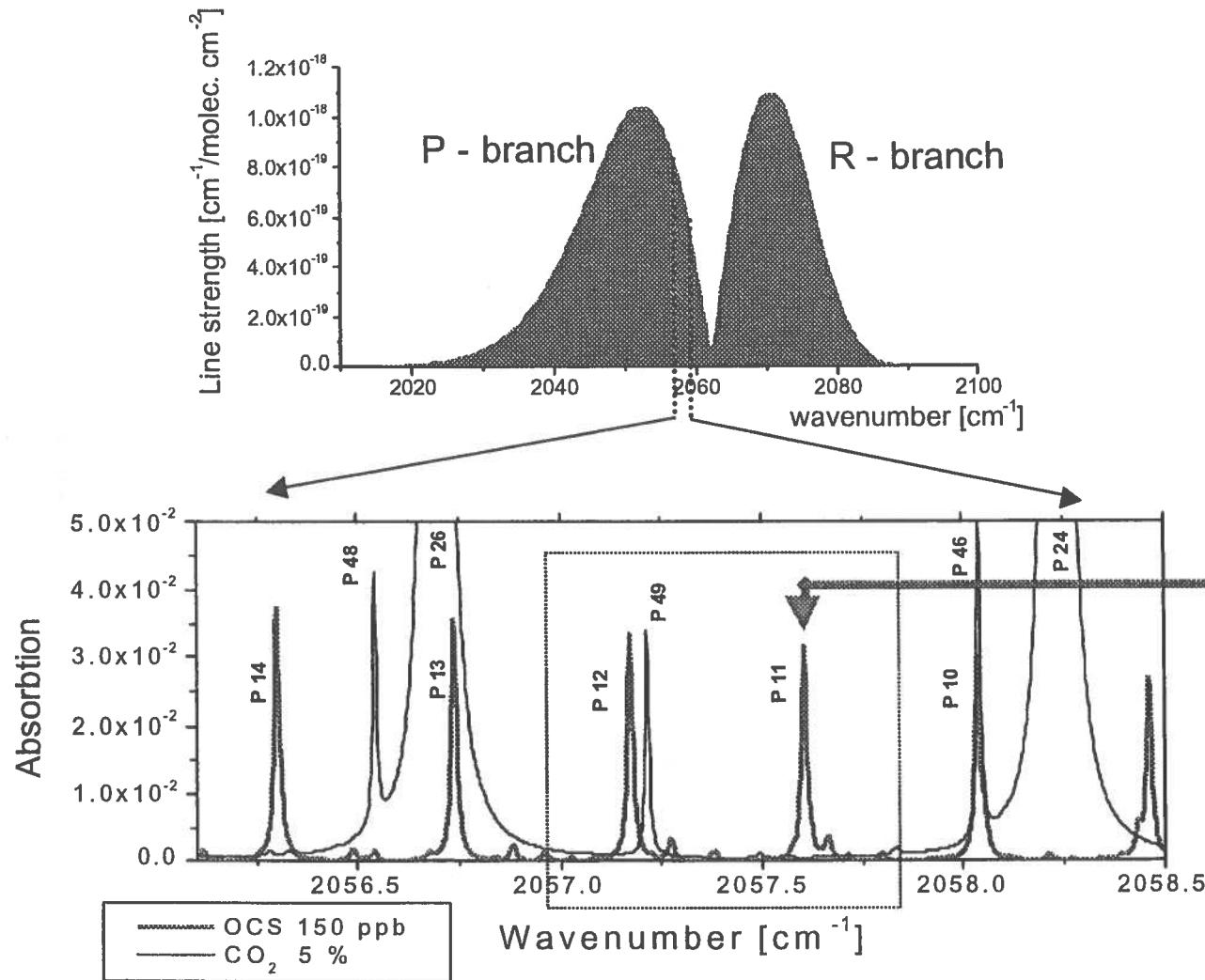
M – mirror

BS – beam splitter

PM – off-axis parabolic mirror

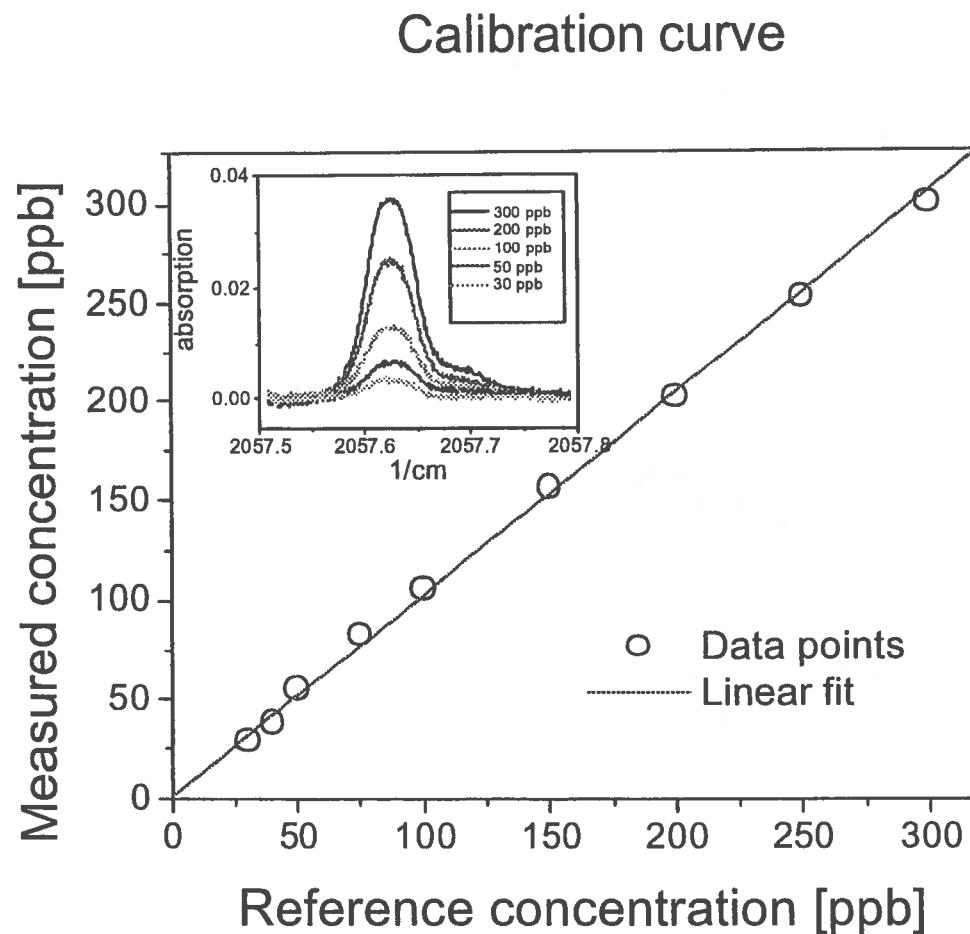


# OCS ro-vibrational Spectrum



- Line intensity:  $7.49 \cdot 10^{-19} \text{ cm}^{-1}/\text{molecule}\cdot\text{cn}$
- Minimal spectral interference by nearby CO<sub>2</sub> and H<sub>2</sub>O absorption lines
- Availability of a CO<sub>2</sub> line with the fast tuning range of the QCL for ventilation monitoring simultaneously with an OCS measurement

# OCS Concentration Calibration of QCL Sensor

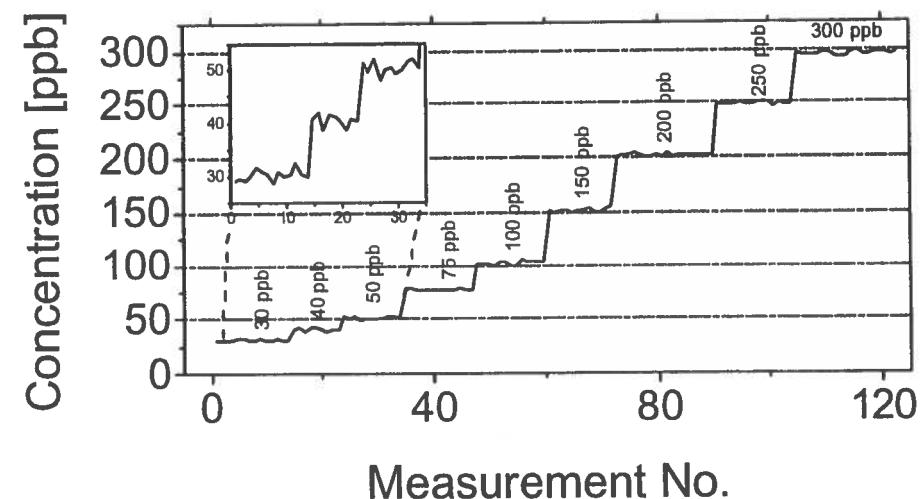


1000 spectra averaged acquired within  $t = 4$  s  
and fitted to 300 ppb OCS reference spectrum

Theoretical sensitivity:

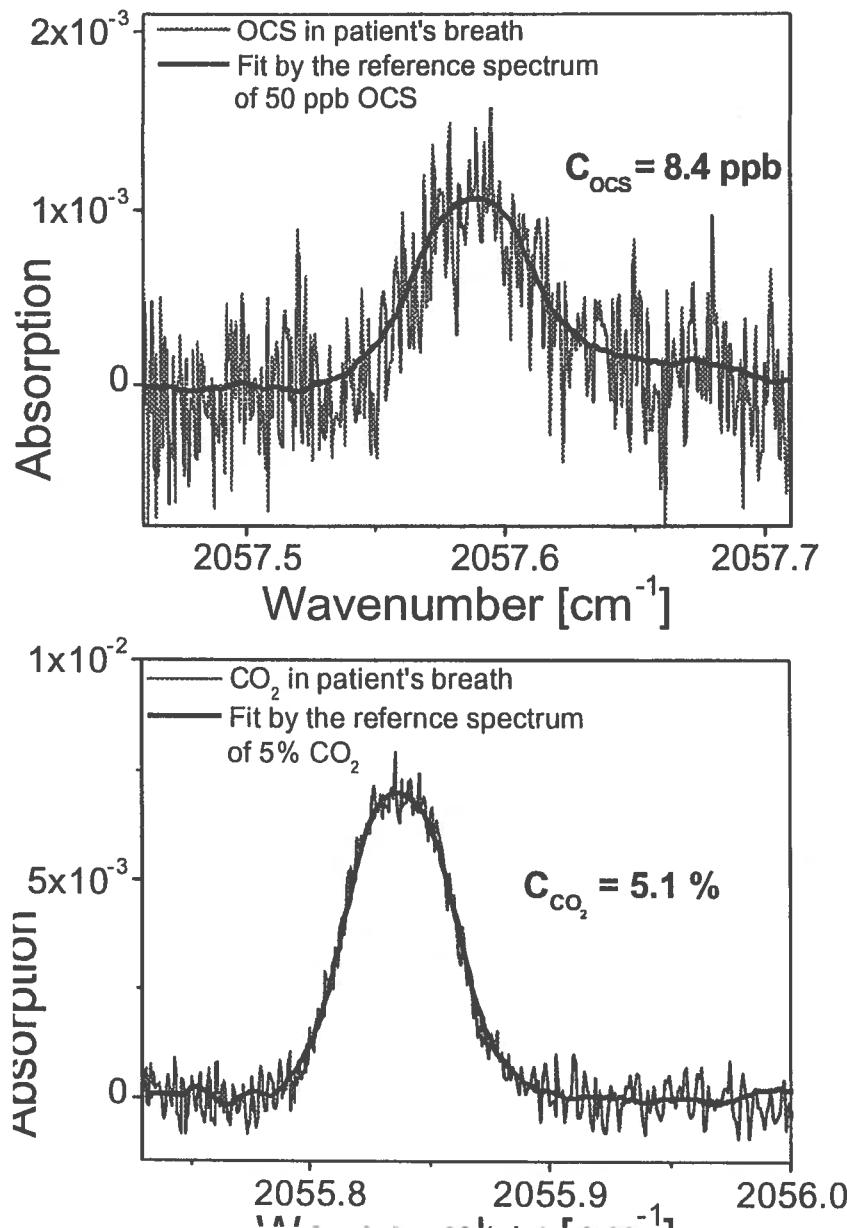
$$0.27 \text{ ppb} \cdot \sqrt{1000/100} = 0.85 \text{ ppb}$$

Scattering of the concentration measurement:  $\sigma = 1.2 \text{ ppb}$



100 spectra averaged acquired within  $t = 0.4$  s  
and fitted to 300 ppb OCS reference spectrum

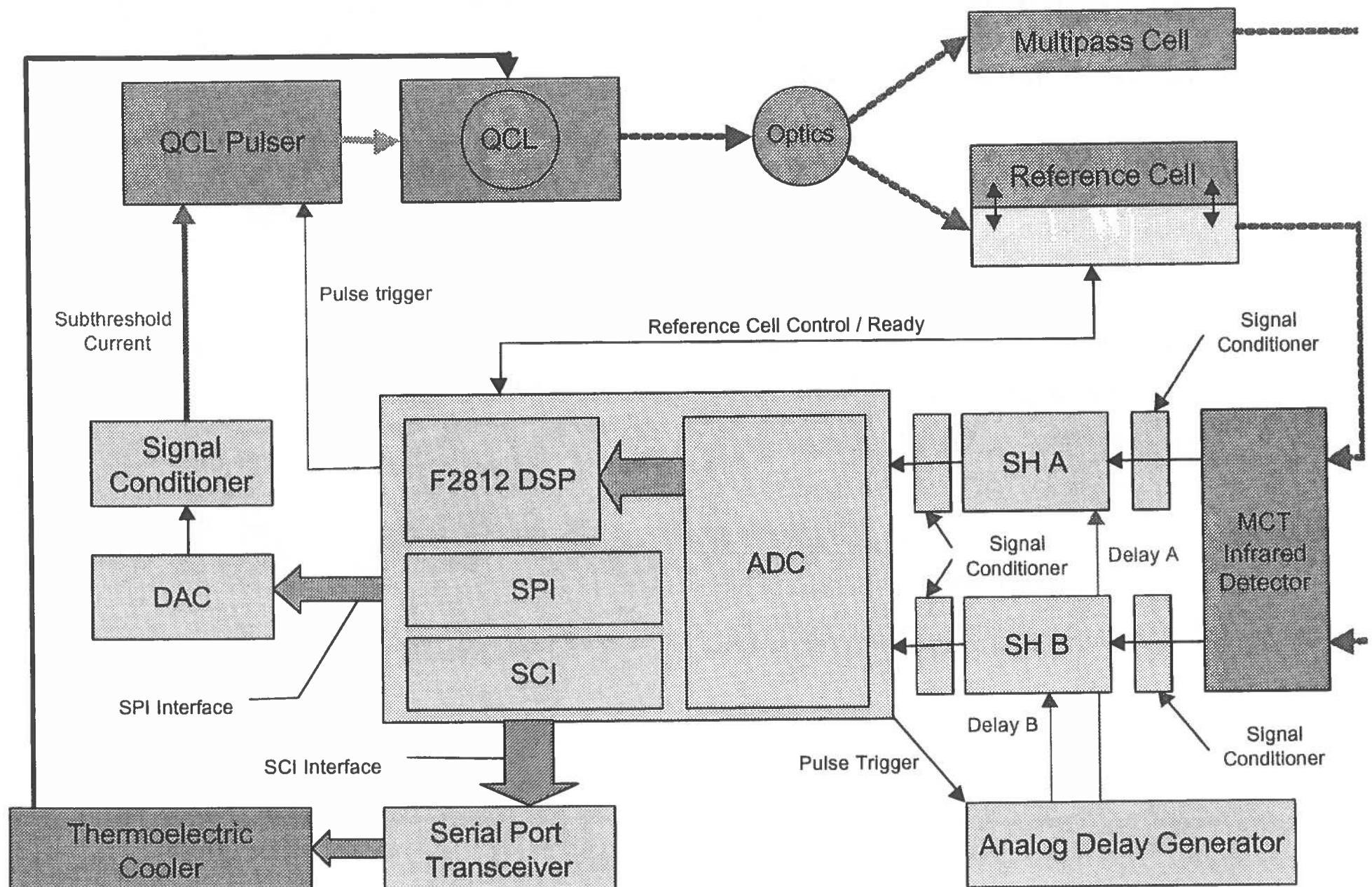
# OCS and CO<sub>2</sub> Concentration Measurements in Exhaled Breath



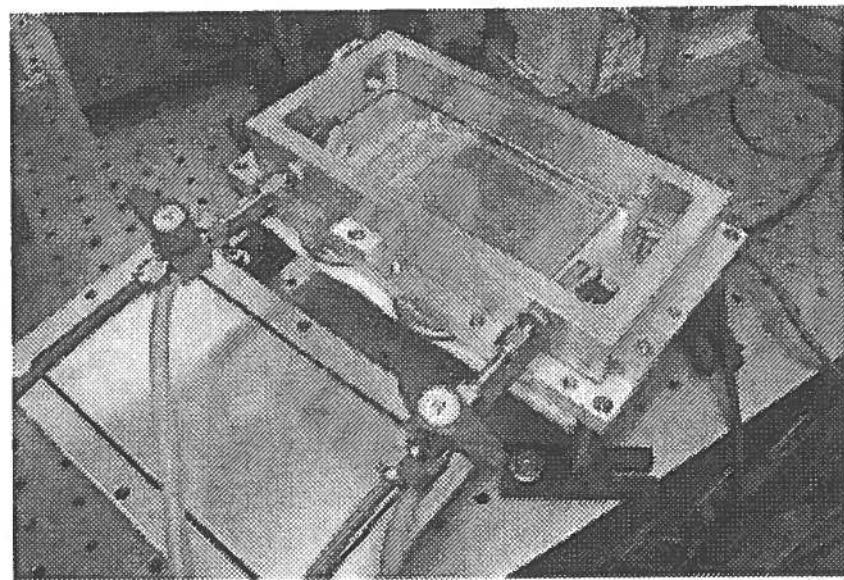
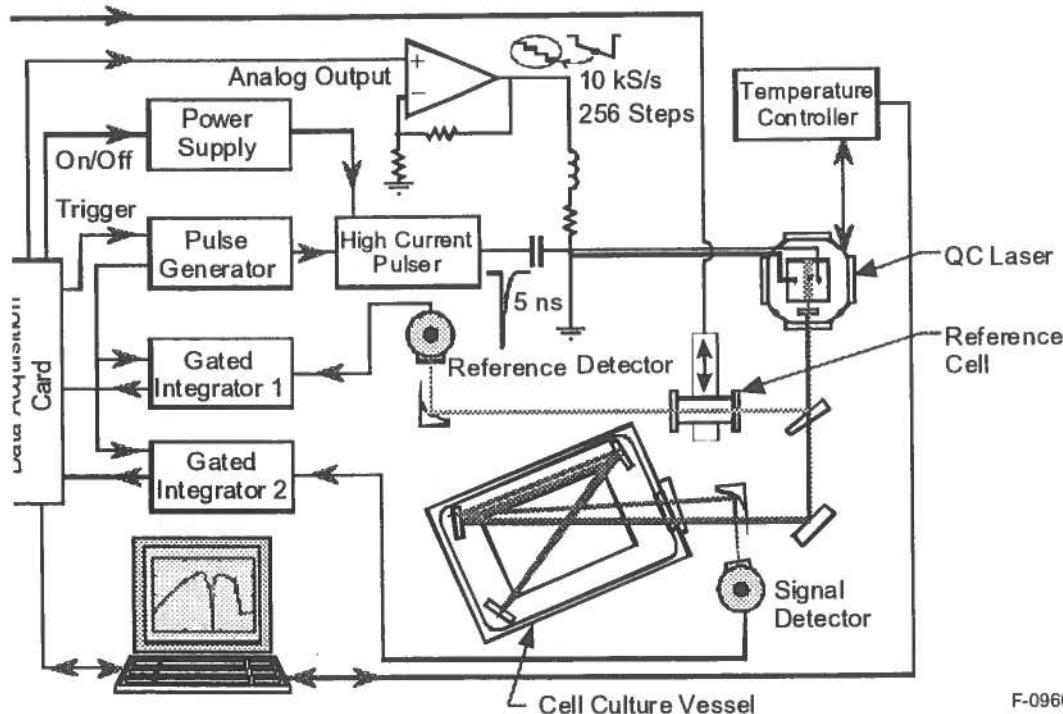
- Sample was taken from lung transplant patient with suspected bronchiolitis\*
- Sampling was performed using chemically inert 1 liter tedlar sampling bags and analyzed within 2 hours after collection
- Spectrum was measured at a total pressure of 60 torr

\*The authors wish to thank Dr. Remzi Bag and Carolyn M. Paraguaya from Baylor College of Medicine, Houston, TX for supplying breath samples

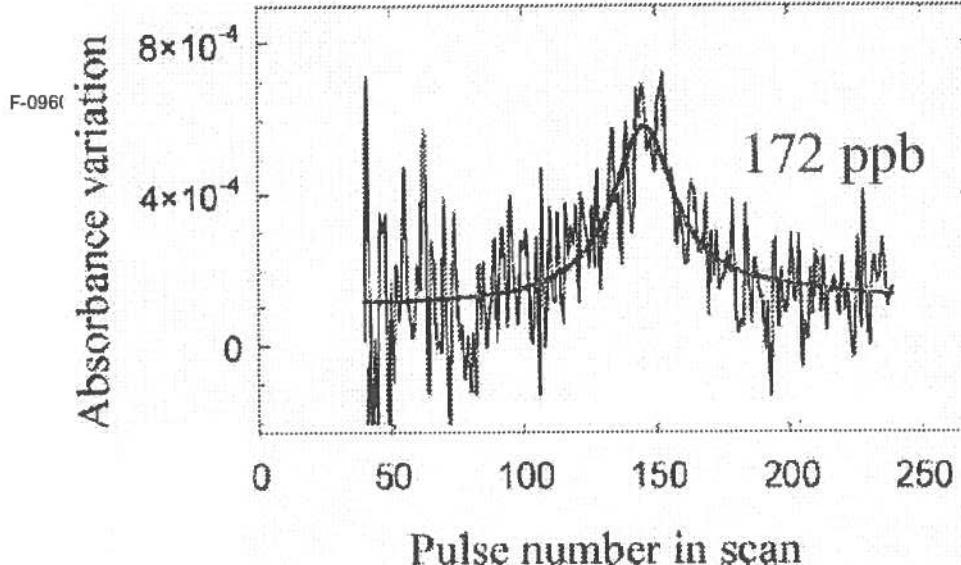
# DSP Fast Data Acquisition Architecture\*



# QC laser based measurements of CO trace gas above cell cultures



- Measured CO production rates of viable cultures of vascular smooth muscle cells
- Achieved a detection limit of for CO of ~20 ppb

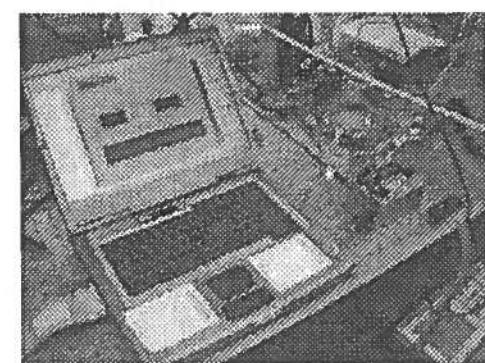
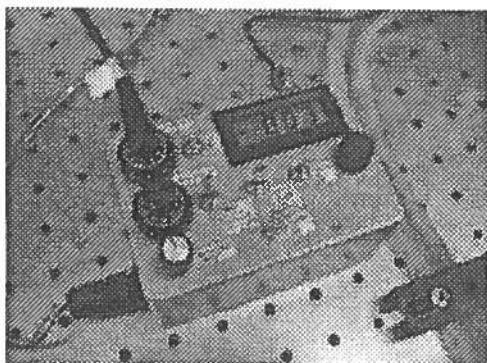
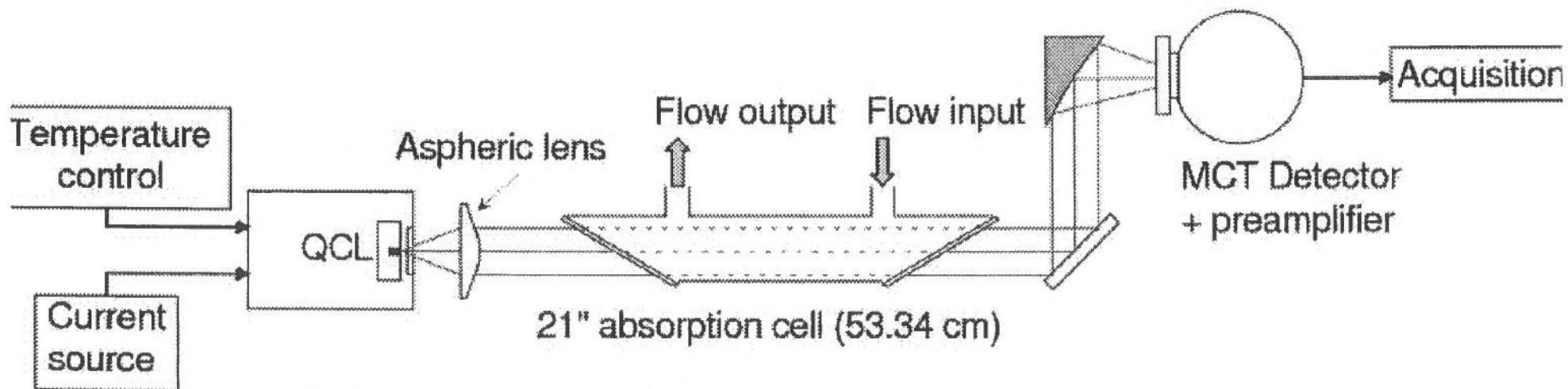
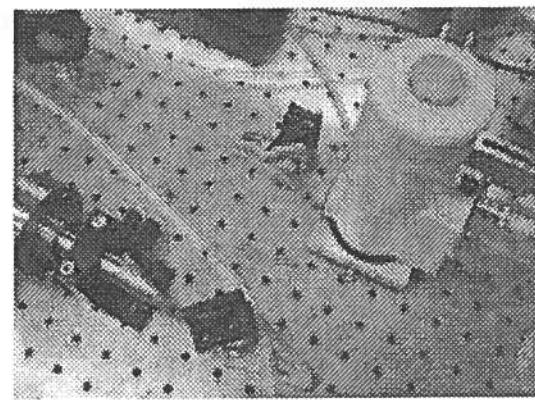
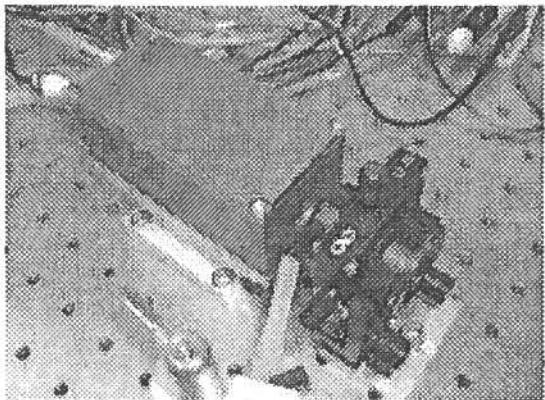


# Towards CW Mode Operation of DFB QCLs

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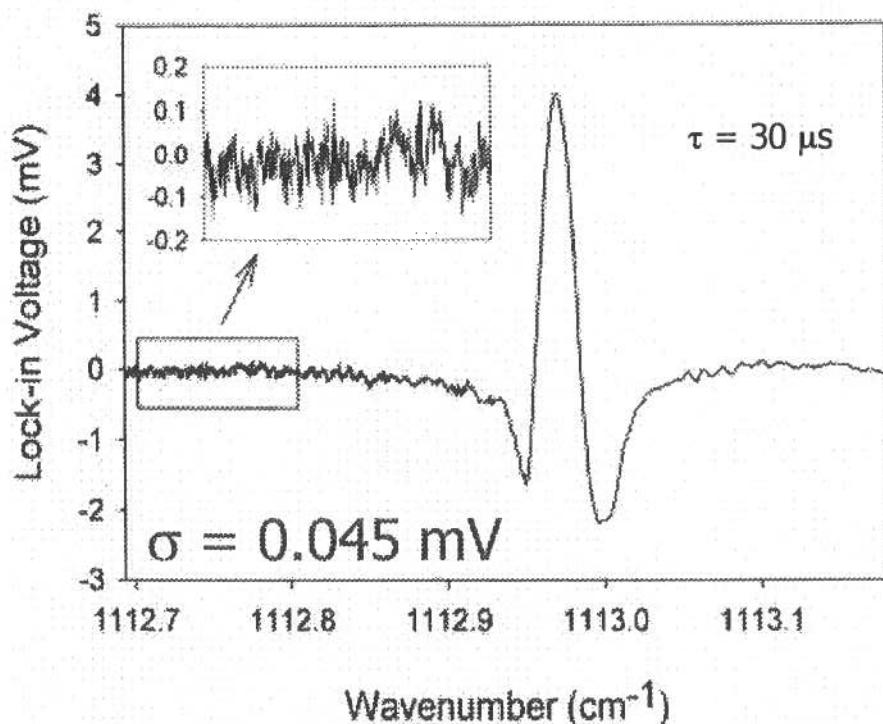
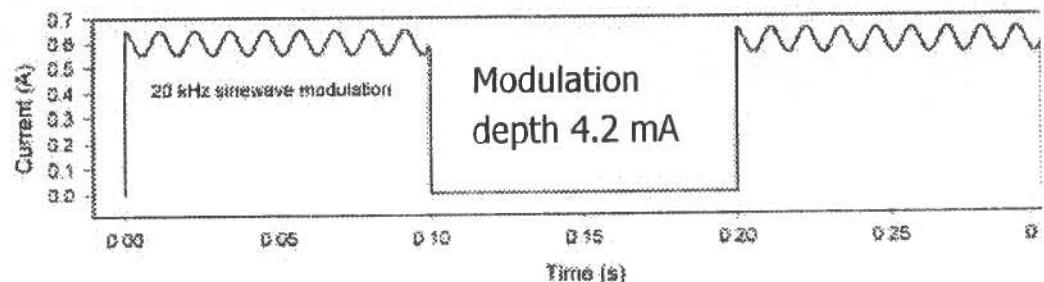
- Main drawbacks of QCL pulsed operation
  - Pulse to pulse intensity variation
  - Linewidth broadening by thermal chirp
  - Requirement of nanosecond electronics
- Efforts towards achieving quasi-RT CW DFB QCLs
  - M. Beck *et al.*, Science, **295**, 301-305, 2002
  - T. Aellen *et al.*, Applied Physics Letters, **83**, 1929, 2003
  - A. Evans, *et al.*, Applied Physics Letters, **84**, 314, 2004 [NON DFB QCL]

# Direct Absorption Based Gas Sensor Architecture



# Wavelength Modulation Spectroscopy

- QCL Drive Current :  
Quasi CW +  
Wavelength modulation



- Calibration with a  
1038 ppm  $\text{NH}_3:\text{N}_2$  mixture  
1 $\sigma$  extrapolated sensitivity  
82 ppb.m/ $\sqrt{\text{Hz}}$   
⇒ Improvement by a factor of  
3 compared to direct  
absorption spectroscopy

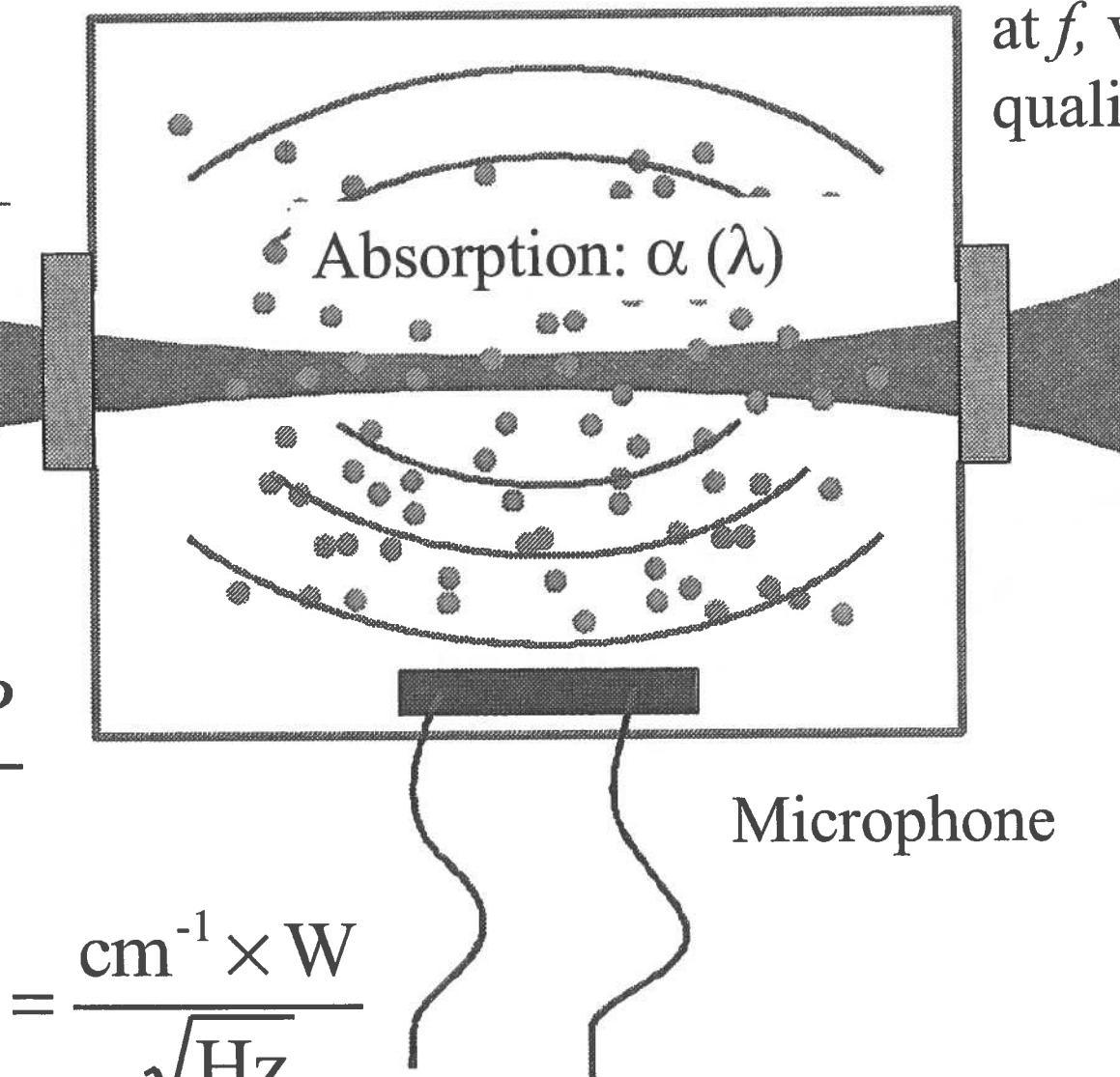
# Resonant photoacoustic spectroscopy

Laser beam,  
power  $P$

Modulated  
( $P$  or  $\lambda$ ) at  $f$   
or  $f/2$

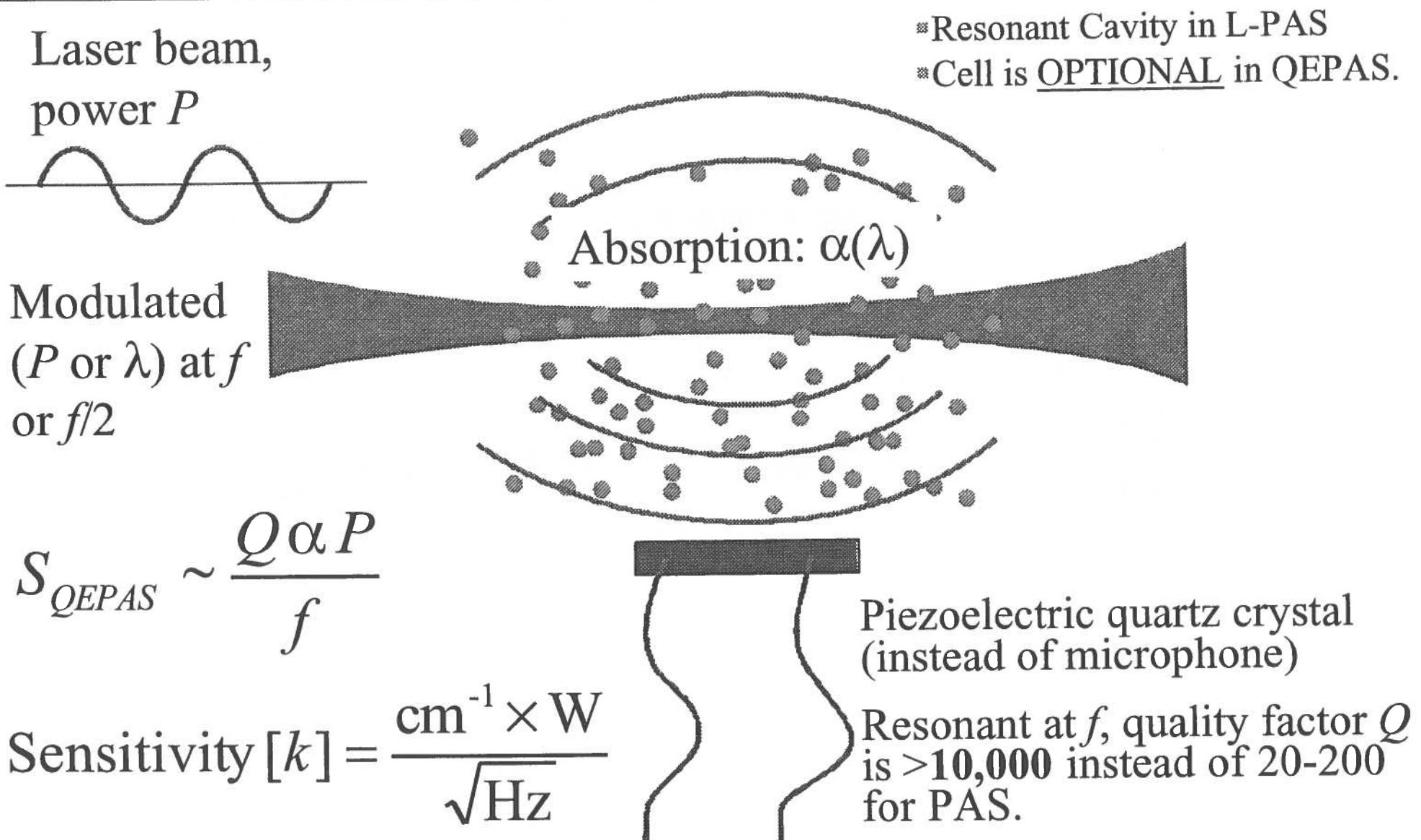
$$S_{PAS} \sim \frac{Q\alpha P}{fV}$$

$$\text{Sensitivity } [k] = \frac{\text{cm}^{-1} \times \text{W}}{\sqrt{\text{Hz}}}$$

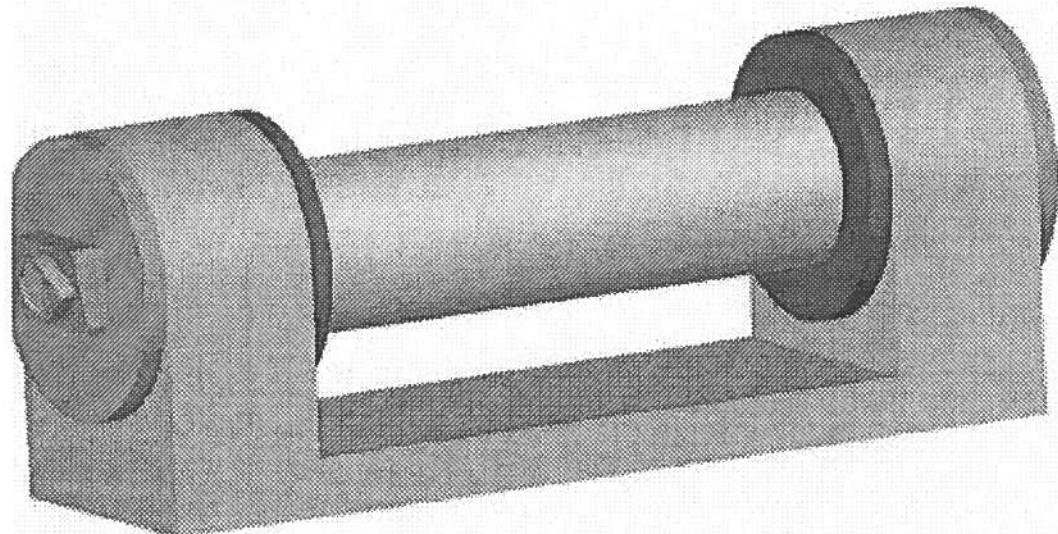


Cavity, resonant  
at  $f$ , volume  $V$ ,  
quality factor  $Q$

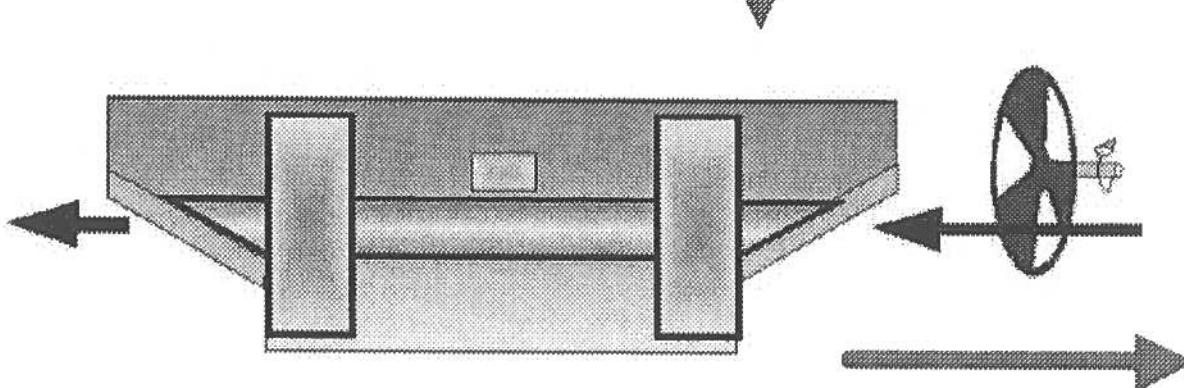
# Quartz-Enhanced Photoacoustic Spectroscopy (QEPAS)



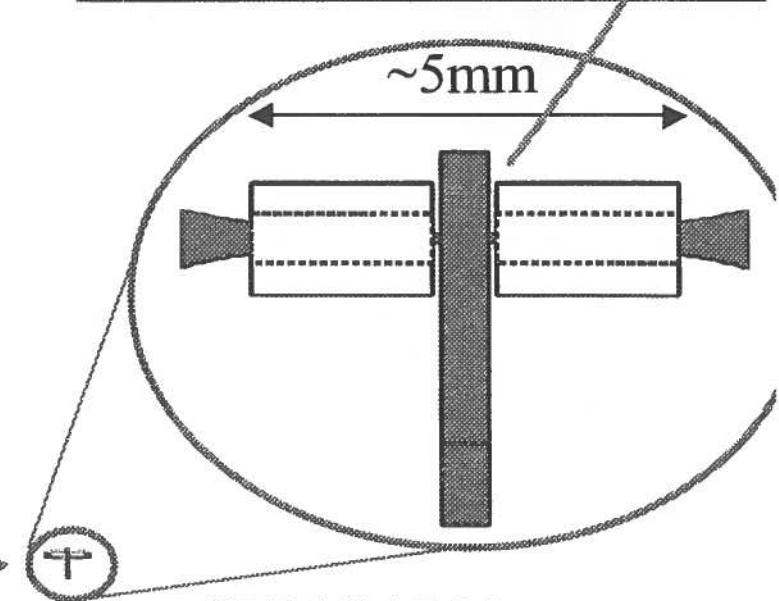
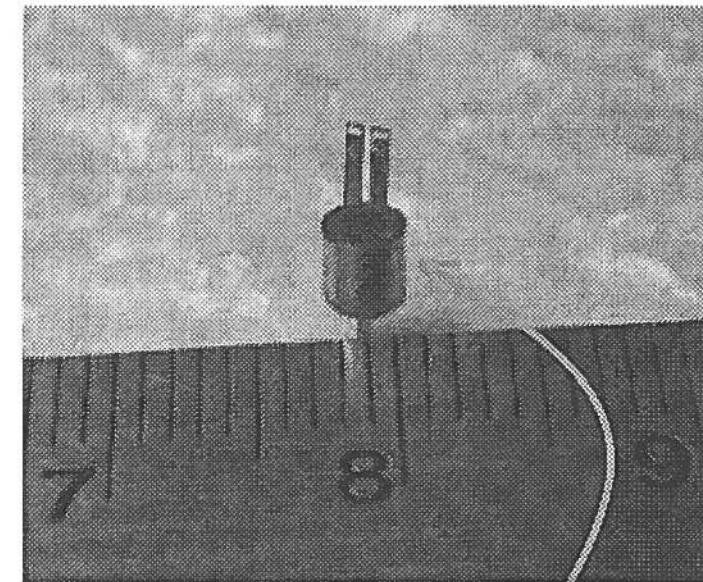
# Comparative Size of Absorbance Detection Modules (ADM)



Optical multipass cell (100 m):  
 $l \sim 70 \text{ cm}$ ,  $V \sim 3000 \text{ cm}^3$

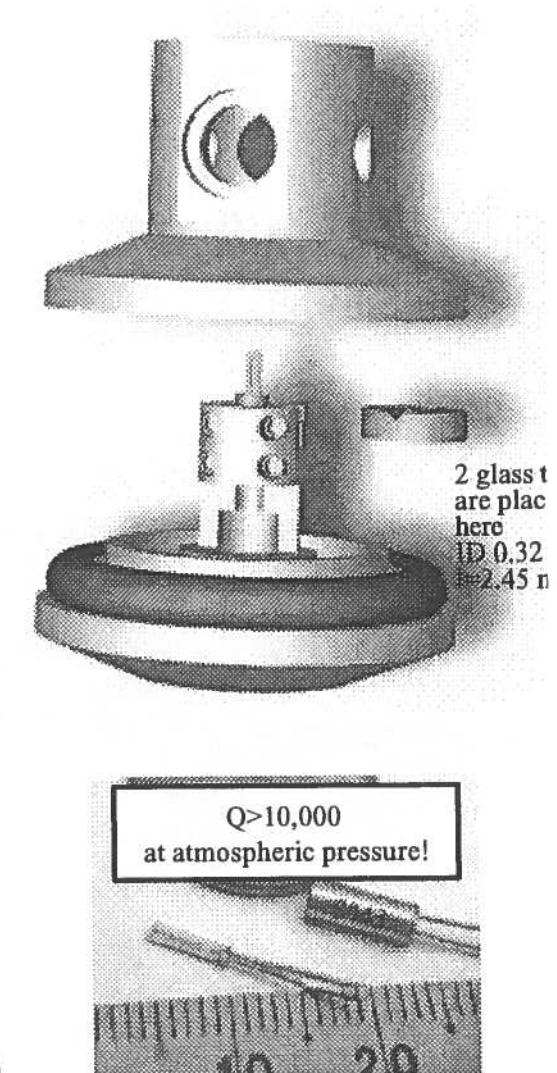
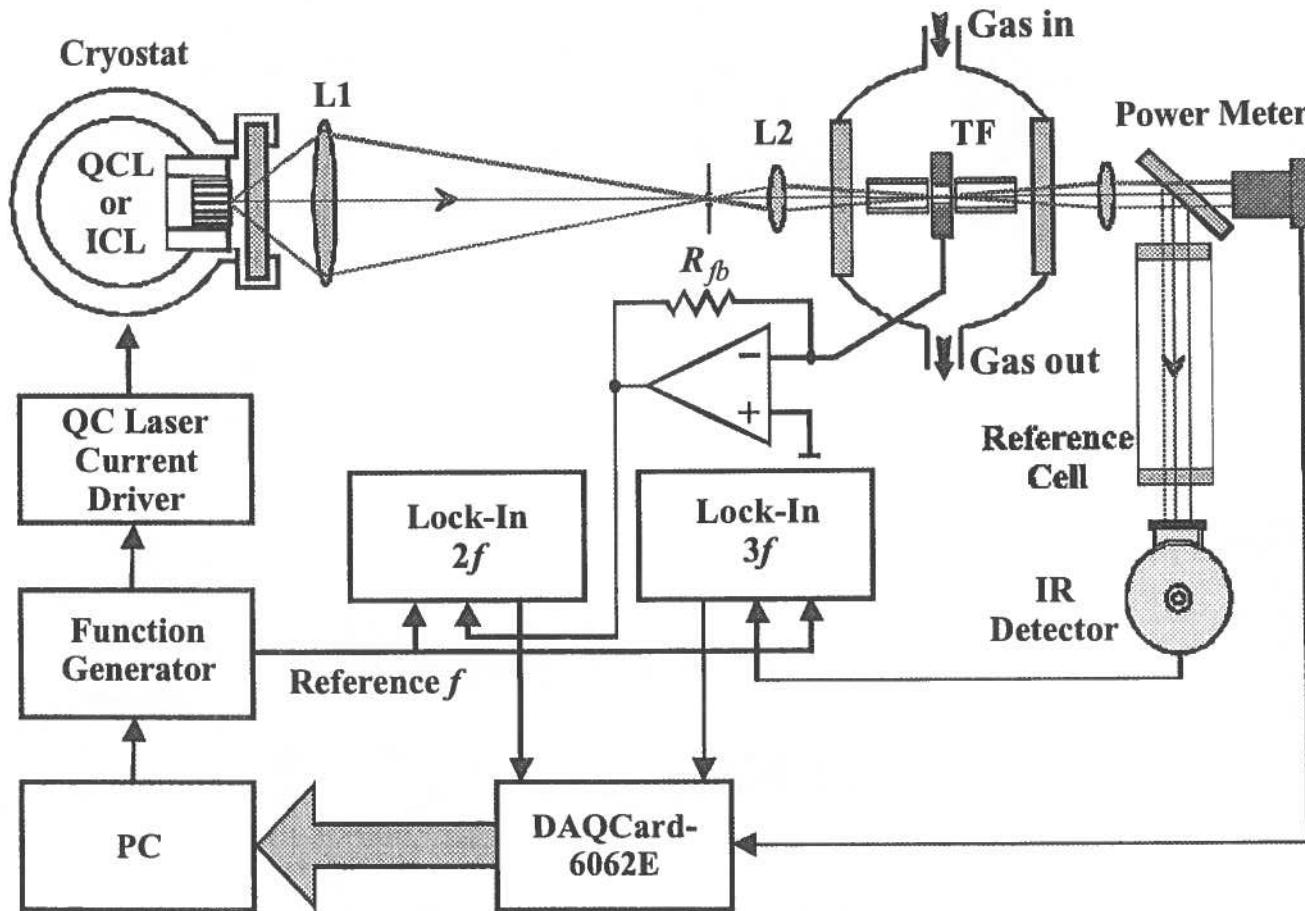


Resonant photoacoustic cell (1000 Hz):  
 $l \sim 60 \text{ cm}$ ,  $V \sim 50 \text{ cm}^3$



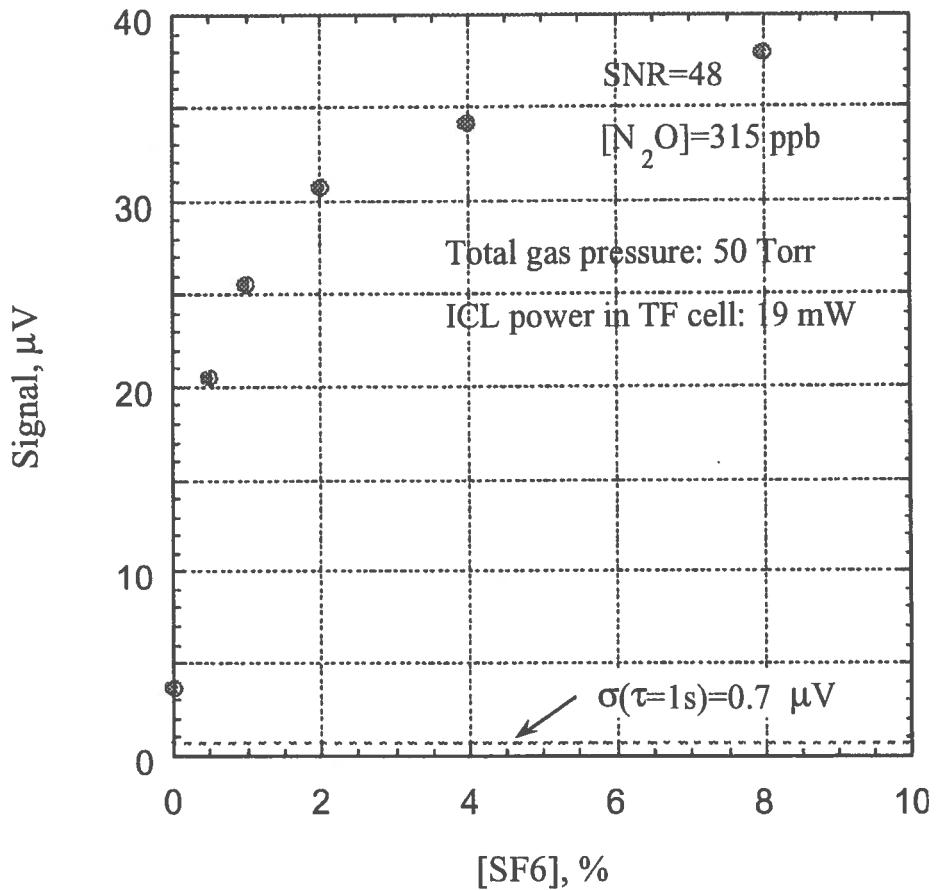
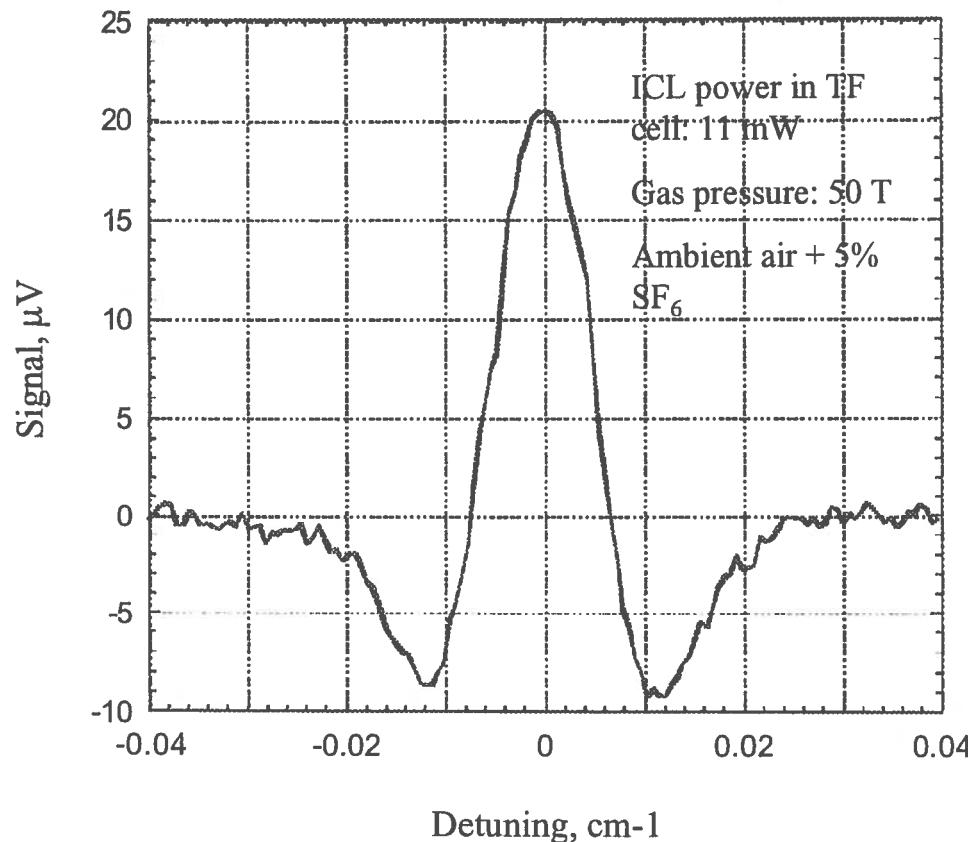
QEPAS ADM:  
 $l \sim 0.5 \text{ cm}$ ,  $V \sim 0.05 \text{ cm}^3$

# QCL based Quartz-Enhanced Photoacoustic Spectrometer



Noise-equivalent sensitivity ( $\alpha_{\min}$ ):  $8.1 \times 10^{-9} \text{ cm}^{-1} \text{W}/\sqrt{\text{Hz}}$  (Rice Dec. 2003)  
f. traditional PAS  $1.1 \times 10^{-8} \text{ cm}^{-1} \text{W}/\sqrt{\text{Hz}}$  (Webber et al. Appl. Phys.B 77, 381, 2003)

# $\text{N}_2\text{O}$ Detection in Ambient Air at 4.6 $\mu\text{m}$ (2195.6 $\text{cm}^{-1}$ )



# Merits of QE Laser-PAS based Trace Gas Detection

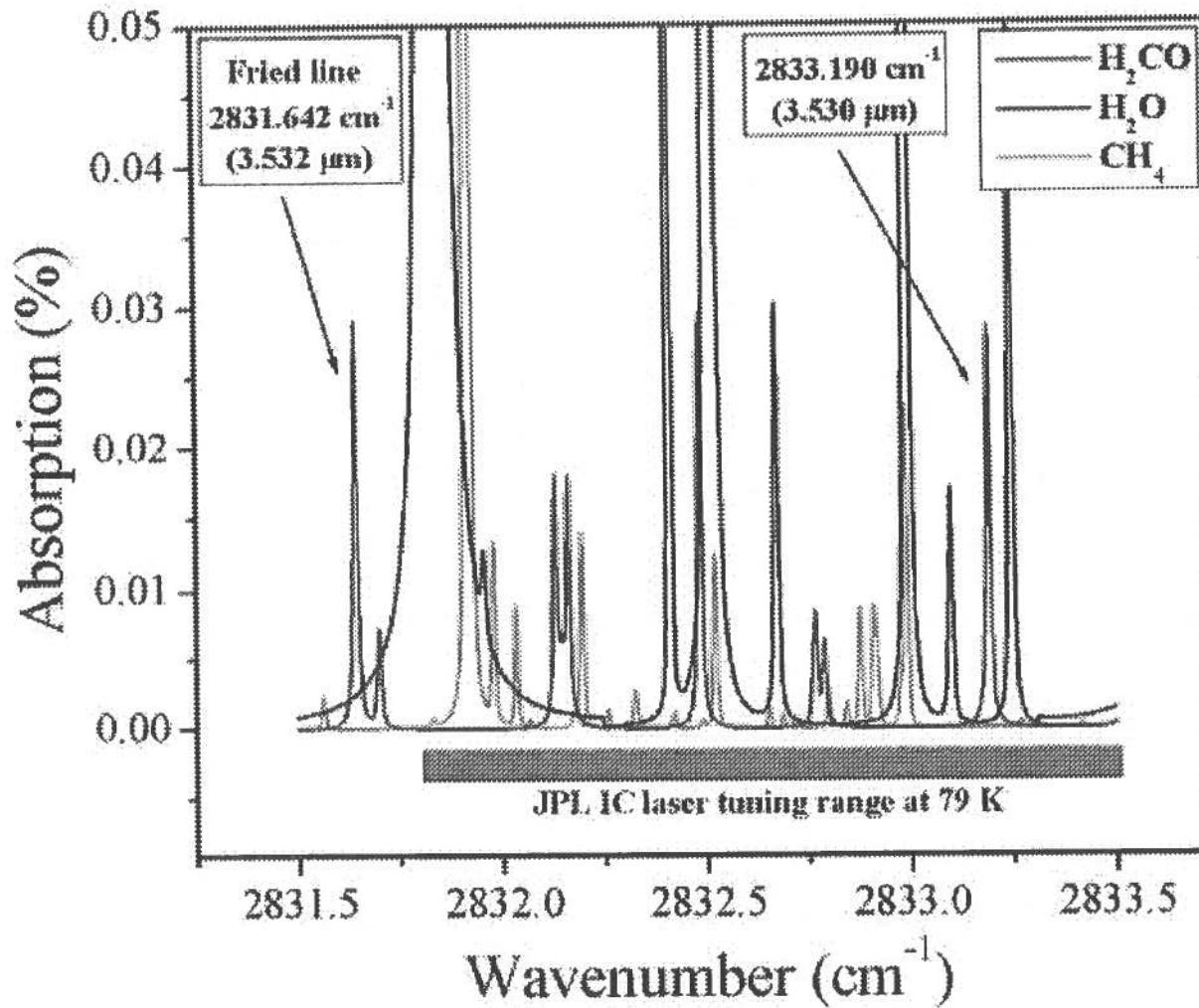
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- High sensitivity (ppm to ppb gas concentration levels) and excellent dynamic range
- **Immune to ambient and flow acoustic noise, laser noise and etalon effects**
- **Significant reduction of sample volume (< 1 mm<sup>3</sup>)**
- Applicable over a wide range of pressures
- Temperature, pressure and humidity insensitive
- Rugged and low cost compared to LAS that requires a multipass absorption cell and infrared detector(s)
- Potential for optically multiplexed concentration measurements

# QEPAS vs. traditional PAS

Parameter	Traditional PAS	QEPAS
$f$ , Hz	100 to 4000	Presently ~32 760
Q	20 to 200	10 000 to 30 000
Q vs. pressure	INCREASES (high spectral resolution is problematic)	DECREASES (high spectral resolution is achievable)
Sample volume	$>10\text{ cm}^3$	$<1\text{ mm}^3$
Sensitivity to ambient acoustic and flow noise	Usually high	None observed
Pathlength involved	$\sim 10\text{ cm}$	(a) 0.3mm, (b) 5mm

# HITRAN Based Simulation of a H<sub>2</sub>CO-H<sub>2</sub>O-CH<sub>4</sub> Spectrum in Tuning Range of a 3.53μm IC Laser



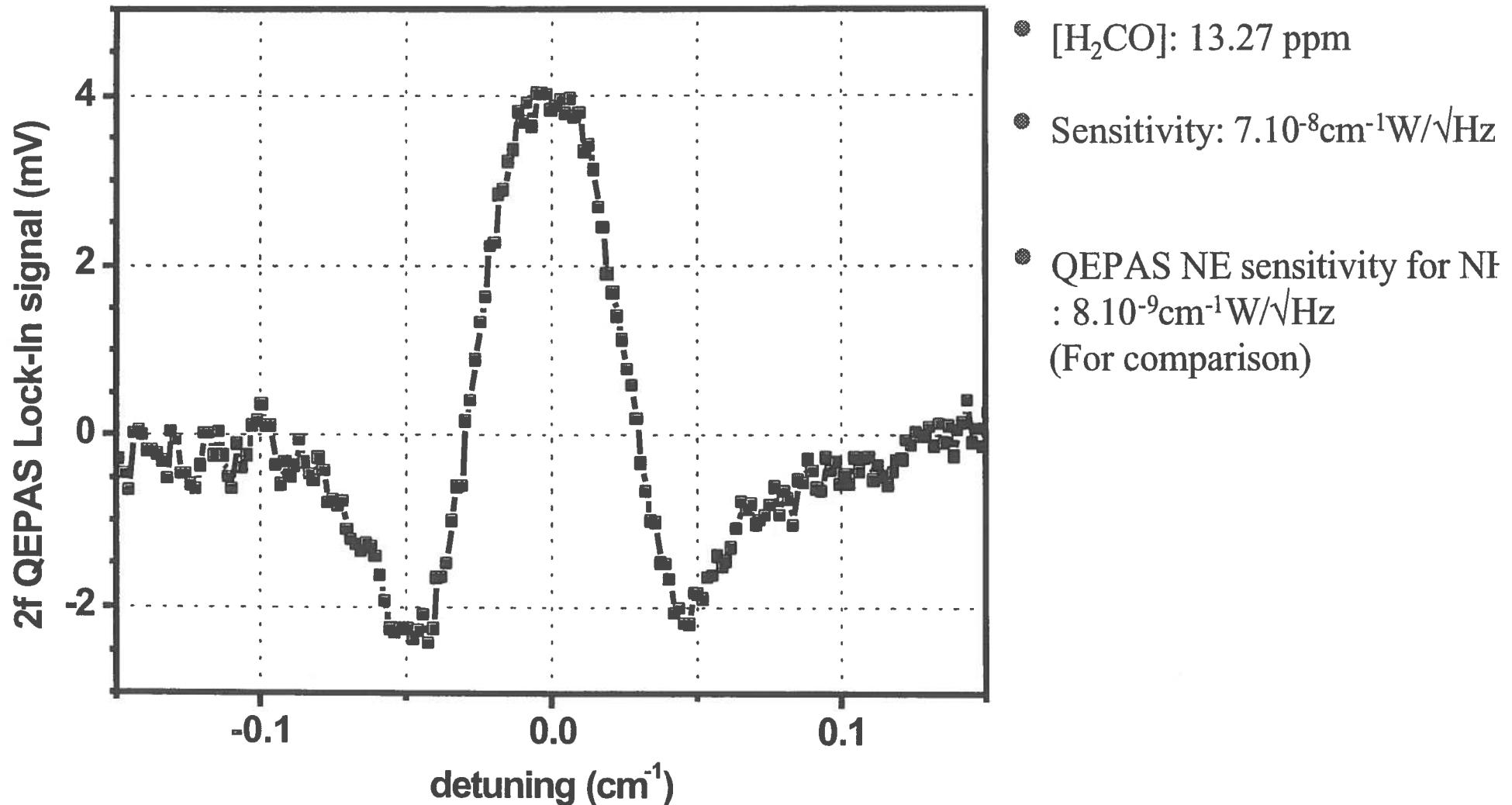
- H<sub>2</sub>CO : 10 ppb
- H<sub>2</sub>O : 3%
- CH<sub>4</sub> : 2 ppm
- Optical path: 100 m
- Total pressure: 30 Torr



JPL



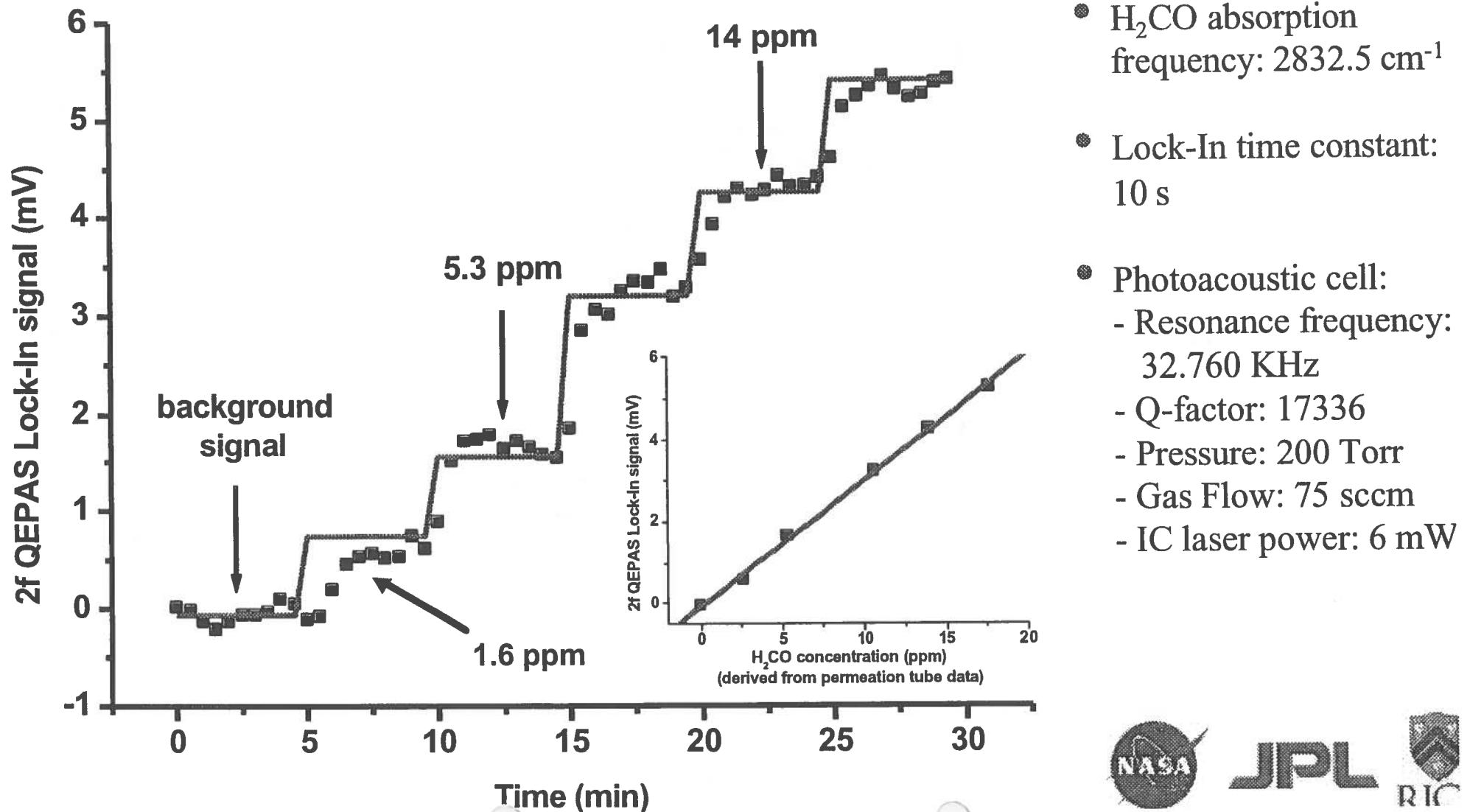
# 2f QEPAS based H<sub>2</sub>CO signal at 3.53 μm (2832.48 cm<sup>-1</sup>)



JPL



# C laser based formaldehyde calibration measurement with a gas standard generator



JPL



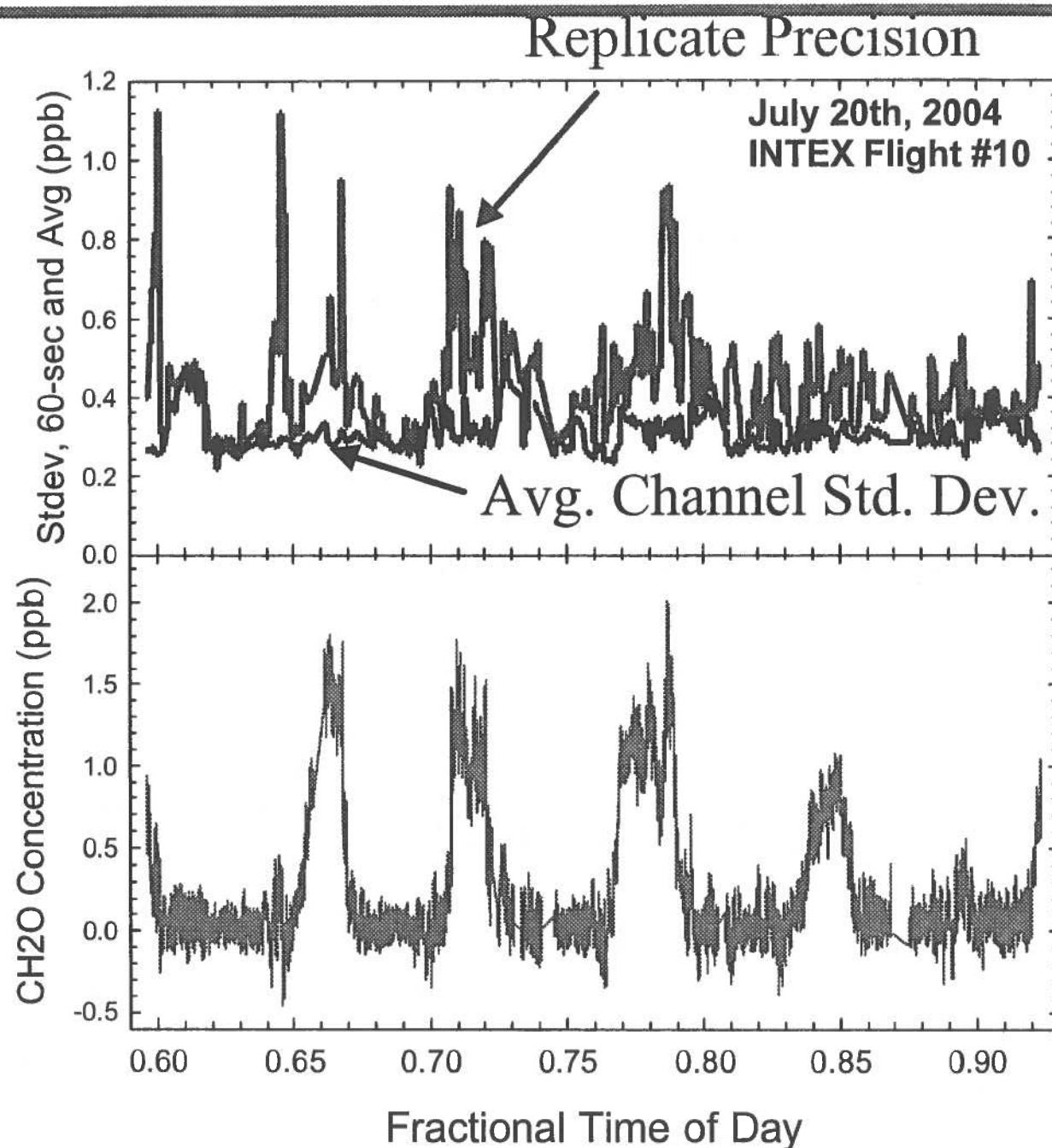
# Intercontinental Chemical Transport Experiment 2004

Each Point  
**Replicate Precision**  
Over 1 min, N = 60  
And  
**Avg. Channel Std. Dev.**

$$\sigma_{1\text{min}} = \frac{\sigma_{1\text{sec}}}{\sqrt{n}} \approx \frac{200 \text{ ppt}}{\sqrt{60}} \approx 26 \text{ ppt}$$

**Each Point**  
**1-second data points**

Concentration determined  
by least squares fit of  
Calibration spectrum  
to Sample spectra



# Motivation for Measuring $^{13}\text{CO}_2/^{12}\text{CO}_2$ Isotopic Ratios

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- Volcano eruption forecasting and gas emission studies ( $\text{CO}_2$ ,  $\text{HCl}$ ,  $\text{SO}_2$ ,  $\text{HF}$ ,  $\text{H}_2\text{S}$ ,  $\text{CO}$ ,  $\text{H}_2\text{O}$ )
- Atmospheric Chemistry: Environmental monitoring of  $\text{C}_y$  gases ( $\text{CO}_2, \text{H}_2\text{O}, \text{CO}, \text{N}_2\text{O}, \text{CH}_4$ )
  - Global warming studies
    - Temporal and spatial variations of the isotopic ratios
    - Identification of carbon sources and sinks
  - Global carbon budget studies
- Study of planetary gases (e.g. for Mars:  $\text{CO}$ ,  $\text{CO}_2$ ,  $\text{H}_2\text{O}$ ,  $\text{CH}_4$ ,  $\text{O}_3$ ,  $\text{OCS}$ )
- Medical applications (non-invasive human health monitoring)



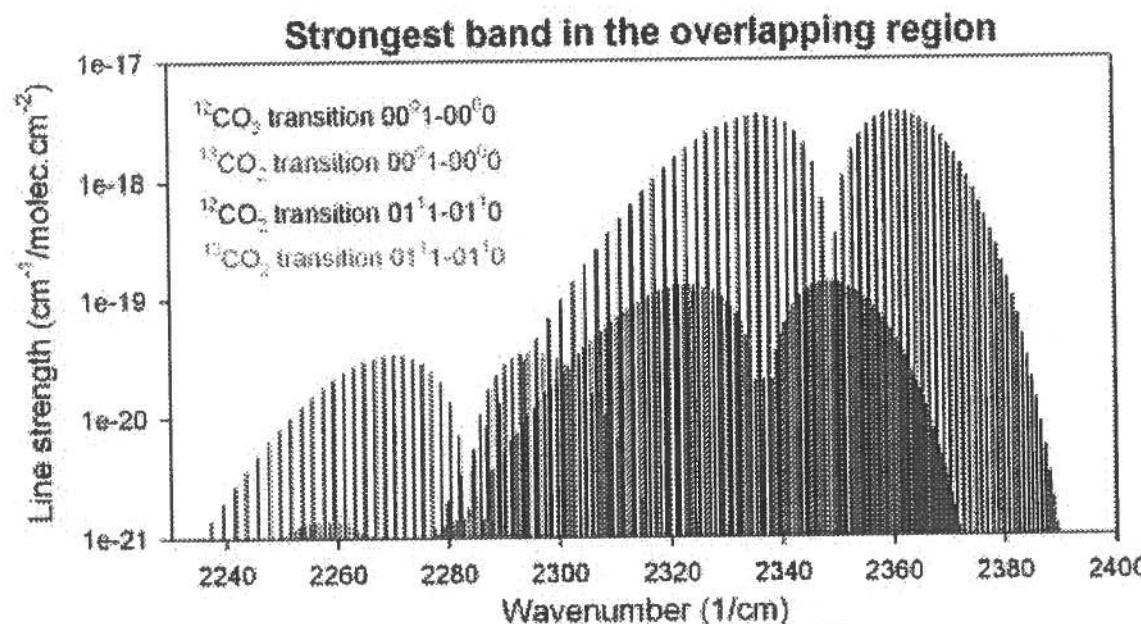
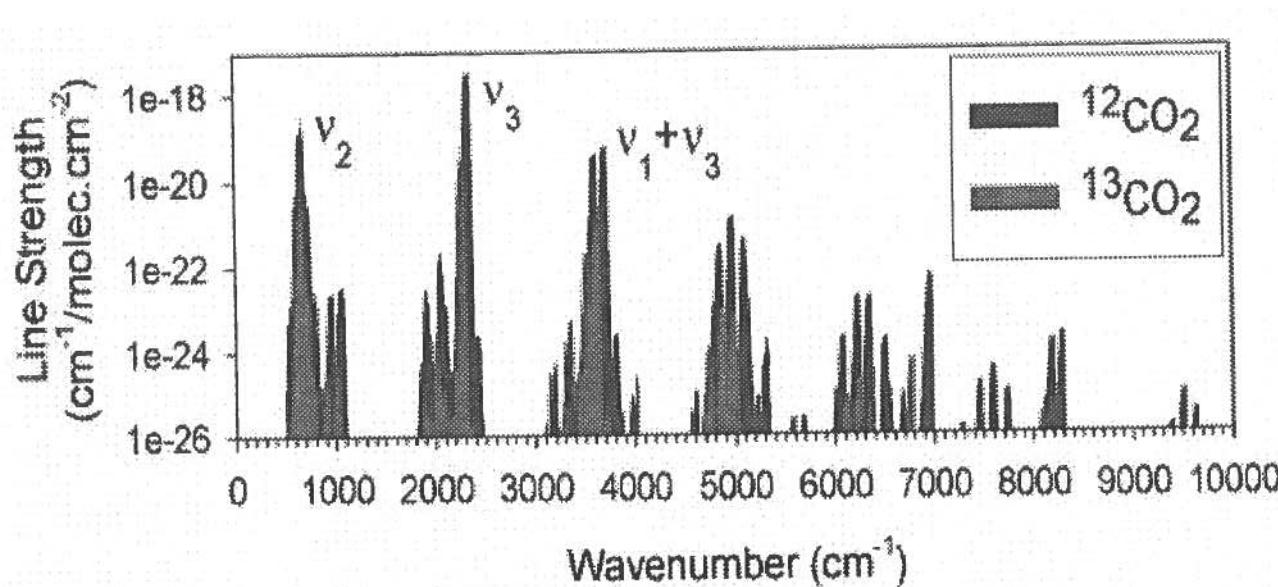
# CO<sub>2</sub> Absorption Line Selection Criteria

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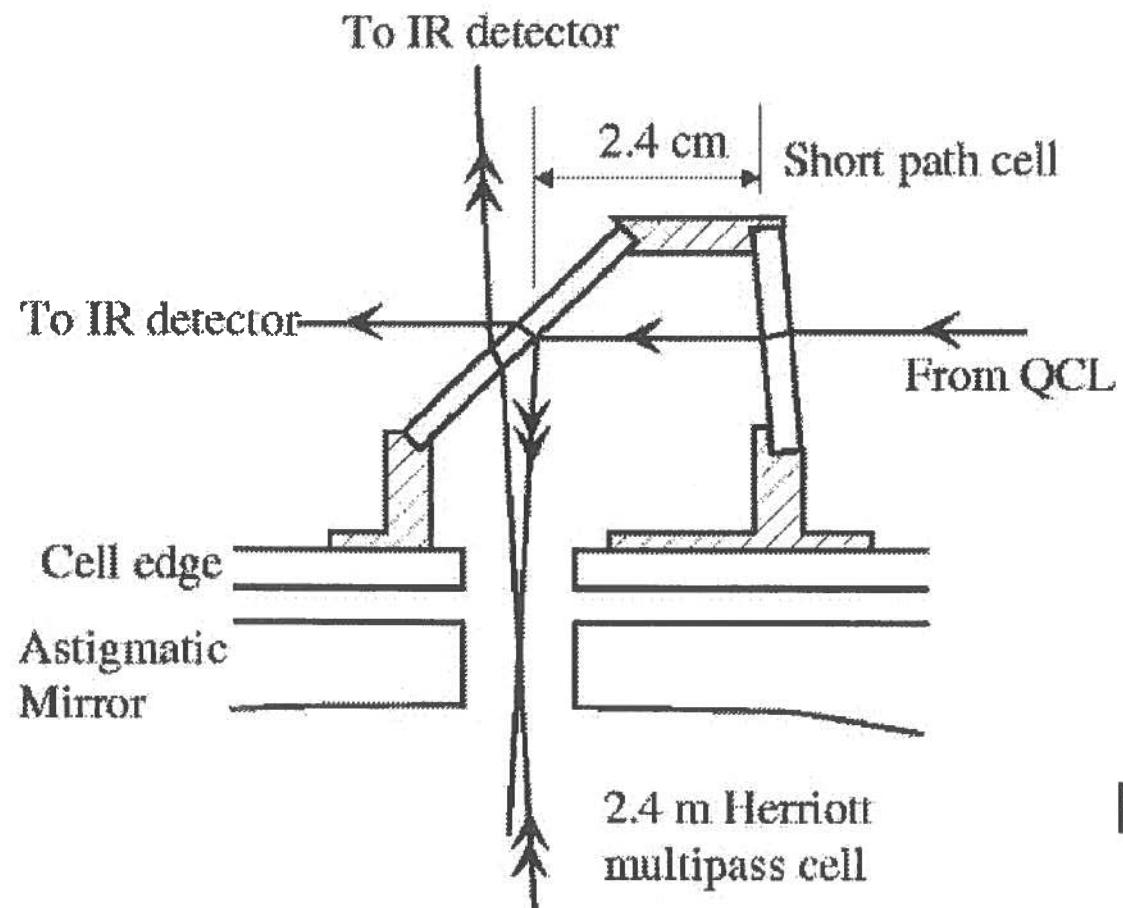
- Three strategies:
  - Similar strong absorption of <sup>12</sup>CO<sub>2</sub> and <sup>13</sup>CO<sub>2</sub> lines
    - Very sensitive to temperature variations
  - Similar transition lower energies
    - Requires a dual path length approach to compensate for the large difference in concentration between major and minor isotopic species-or-
    - Can be realized if different vibrational transitions are selected for the two isotopes ( 4.35 μm for <sup>13</sup>CO<sub>2</sub> and 2.76 μm for <sup>12</sup>CO<sub>2</sub>)\*
- For the first 2 strategies both absorption lines must lie in a laser frequency scan window
- Avoid presence of other interfering atmospheric trace gas species

\* Proposed scheme by Curl, Uehara, Kosterev, et al, Oct. 2002

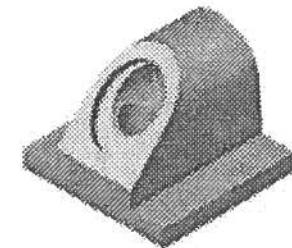
# Ro-vibrational bands suitable for $^{12}\text{CO}_2/^{13}\text{CO}_2$ ratio measurements



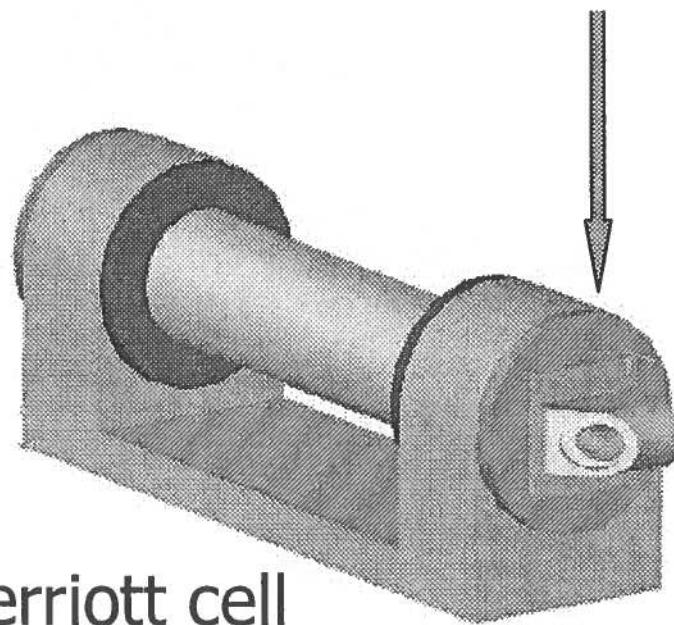
# Dual path length gas cell design



Short cell

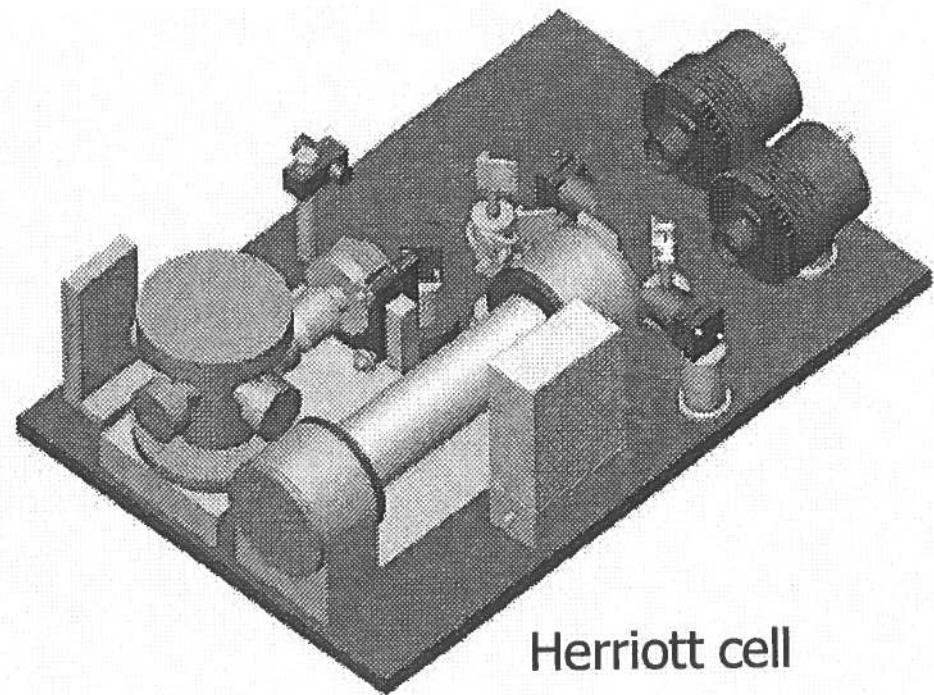
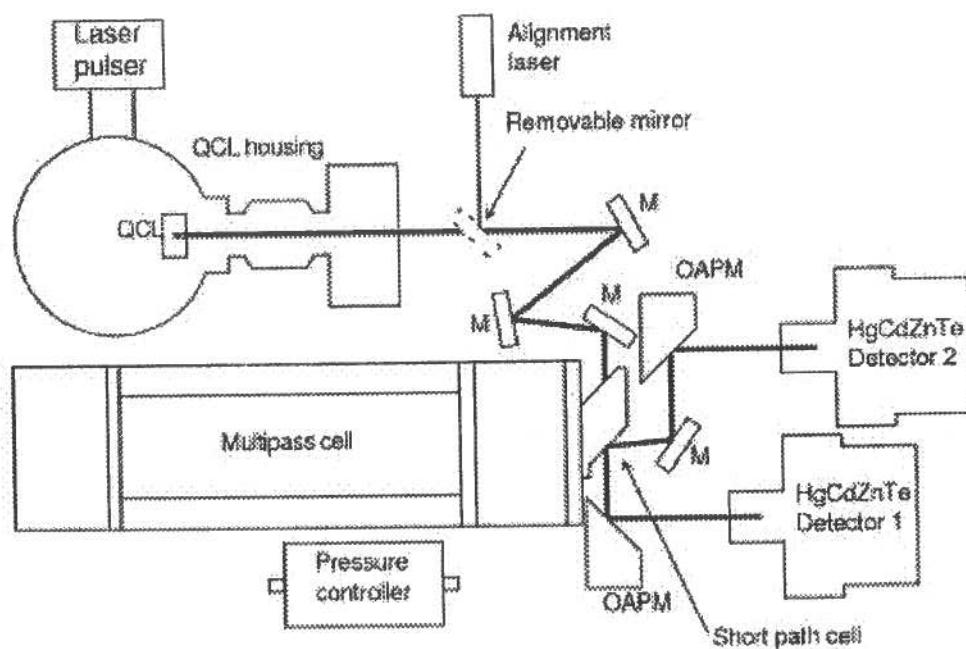


Herriott cell



# QC laser based Isotopic Ratio Sensor Layout

TEC IR detectors

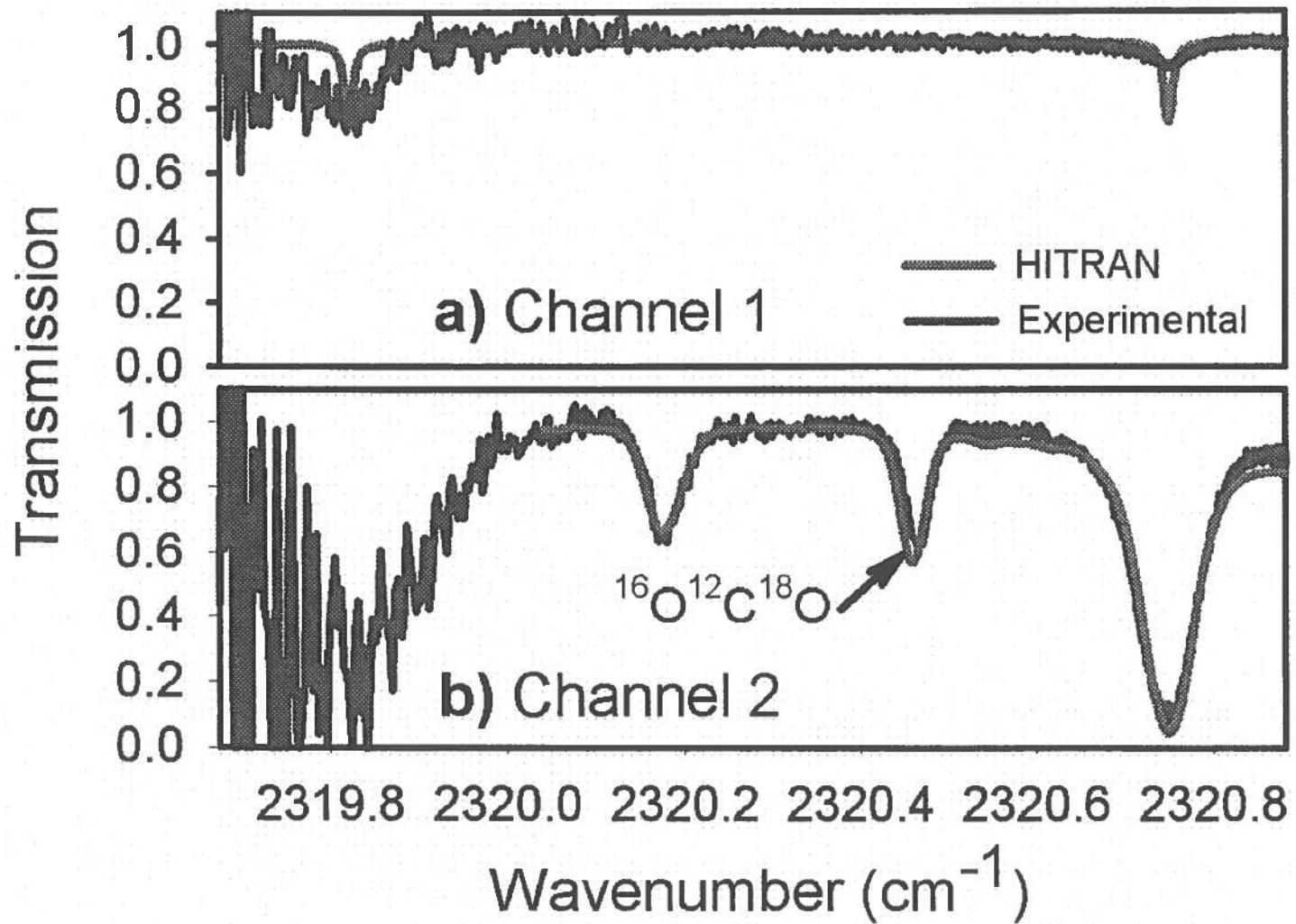


Herriott cell

Bread board: 12x18" (30x45 cm)

The sensor must be operated in a dry nitrogen atmosphere  
to eliminate atmospheric CO<sub>2</sub> background

# $^{16}\text{O}^{12}\text{C}^{18}\text{O}$ Spectra at 2320.2 $\text{cm}^{-1}$



# Conclusions and Future Directions

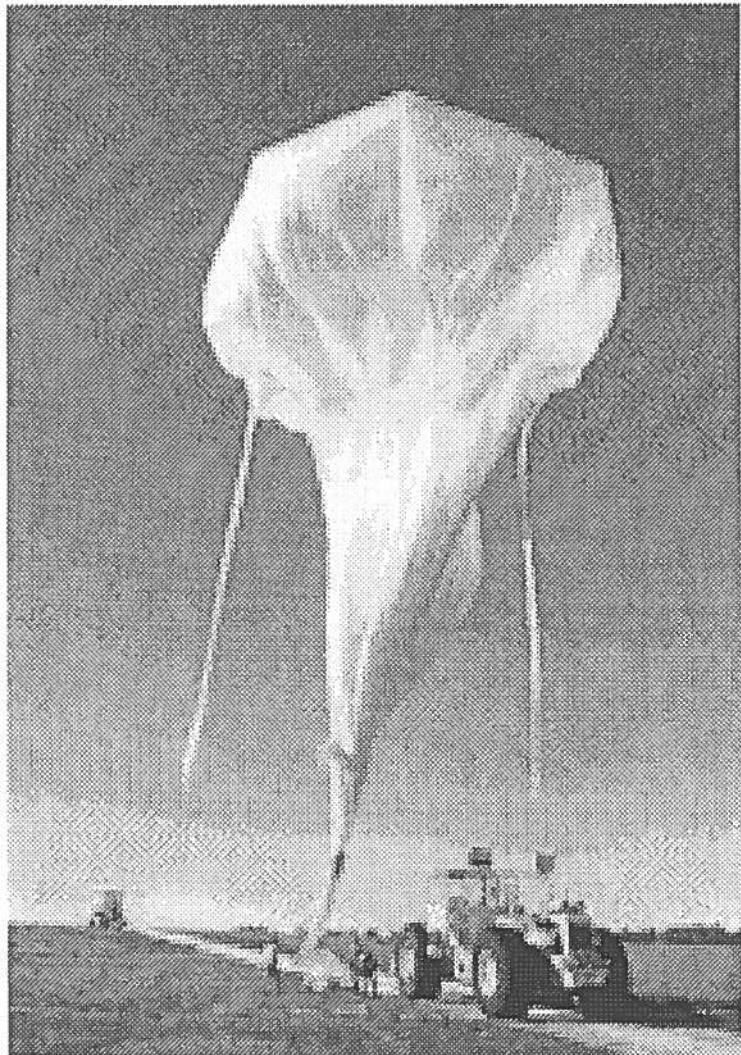
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- **Quantum Cascade Laser based Trace Gas Sensors**
  - Compact and robust sensors based on QC-LAS and QE L-PAS
  - High sensitivity ( $10^{-4}$ - $10^{-5}$ ) and selectivity (3 to 500 MHz)
  - Dramatic reduction of sample volume ( $\sim 0.2 \text{ mm}^3$ )
  - Detected trace gases:  $\text{NH}_3$ ,  $\text{CH}_4$ ,  $\text{N}_2\text{O}$ ,  $\text{CO}_2$ , CO, NO,  $\text{H}_2\text{O}$ , COS,  $\text{C}_2\text{H}_4$ ,  $\text{C}_2\text{H}_5\text{OH}$ ,  $\text{SO}_2$ ,  $\text{H}_2\text{CO}$  and several isotopic species of C, O, N and H.
- **Applications in Trace Gas Detection**
  - Environmental monitoring ( $\text{NH}_3$ , CO,  $\text{CH}_4$ ,  $\text{C}_2\text{H}_4$ ,  $\text{N}_2\text{O}$ ,  $\text{CO}_2$ )
  - Industrial process control and chemical analysis (NO,  $\text{NH}_3$ )
  - Medical Diagnostics (NO, CO, COS,  $\text{CO}_2$ ,  $\text{C}_2\text{H}_4$ )
- **Future Directions and Collaborations**
  - Cavity enhanced (ICOS) and QE L-PAS spectroscopy based applications using novel thermoelectrically cooled cw and broadly wavelength tunable quantum cascade lasers
  - Applications using new near IR interband and far-IR intersub-band quantum cascade lasers



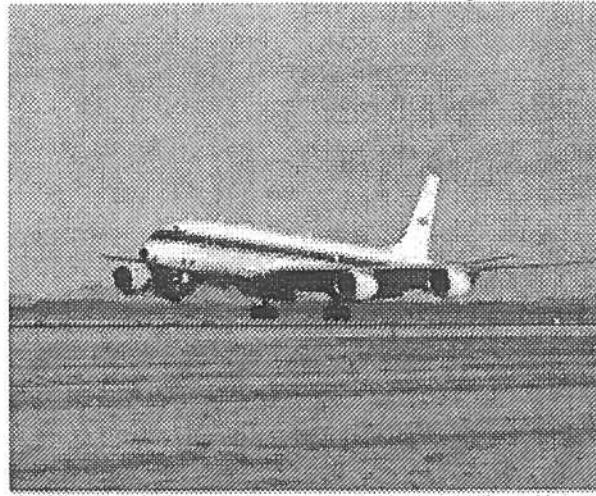
# NASA Atmospheric & Mars Gas Sensor Platforms

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Tunable laser based sensor  
for stratospheric measurements

Aircraft based laser absorption spectrometer



Dryden Flight Research Center 8123-9430-2. Photographed 29/01/1987  
Boeing 727-8 Airborne Laboratory, owned by Dryden (NASA/Tony Landis)

Tunable laser planetary spectrometer

