

Mid-infrared interband and quantum cascade laser based trace gas sensor technologies: recent advances and applications



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http://www.ece.rice.edu/~lasersci/

- Novel Laser-Based Trace Gas Sensor Technologies
 - Quartz Enhanced Photoacoustic Spectroscopy (QEPAS)
 - Mid-IR TDLAS based on a Novel Multipass Gas Cell Design
- Motivation for CH₄ and C₂H₆ Detection
- HITRAN Simulation for CH₄ and C₂H₆
- Performance Evaluation of CH₄ & C₂H₆ Sensor Platform
- Future Directions of Mid-IR Laser based Trace Gas Sensor Technologies and Conclusions

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Laser-Based Trace Gas Sensing Techniques

- Optimum Molecular Absorbing Transition
 - Overtone or Combination Bands (NIR)
 - Fundamental Absorption Bands (Mid-IR)
- Long Optical Pathlength
 - Multipass Absorption Gas Cell (e.g., White, Herriot, Chernin, Aeris Technologies, and Circular Cylindrical Multipass Cell
 - Cavity Enhanced and Cavity Ringdown Spectroscopy
 - Open Path Monitoring (with retro-reflector or back scattering from topographic target): Standoff and Remote Detection
 - Fiberoptic Evanescent Wave Spectroscopy
- Spectroscopic Detection Schemes
 - Frequency or Wavelength Modulation
 - Balanced Detection
 - Zero-air Subtraction



 Photoacoustic & Quartz Enhanced Photoacoustic Spectroscopy (QEPAS)

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Mid-IR Source Requirements for Laser Spectroscopy

REQUIREMENTS	IR LASER SOURCE
Sensitivity (% to pptv)	Optimum Wavelength, Power
Selectivity (Spectral Resolution)	Stable Single Mode Operation and Narrow Linewidth
Multi-gas Components, Multiple Absorption Lines, and Broadband Absorbers	Mode Hop-Free Wavelength Tunability
Directionality or Cavity Mode Matching	Beam Quality
Rapid Data Acquisition	Fast Time Response
Room Temperature Operation	High Wall Plug Efficiency, No Cryogenics or Cooling Water
Field Deployable in Harsh Environments	Compact and Robust

Key Characteristics of Mid-IR QCL & ICL Sources - March 2015

Band - structure engineered devices

Emission wavelength is determined by layer thickness – MBE or MOCVD; QCLs operate in the 3 to 24 µm spectral region and ICLs can cover the 3 to 6 μm spectral range.
 Compact, reliable, stable, long lived, and commercially available
 Fabry-Perot (FP), single mode (DFB) and multi-wavelength devices

Wide spectral tuning ranges in the mid-IR

- 1.5 cm⁻¹ using injection current control for DFB devices
- 10-20 cm⁻¹ using temperature control for DFB devices
- ~100 cm⁻¹ using current and temperature control for QCL DFB Array ~ 525 cm⁻¹ (22% of c.w.) using an external grating element and FP chips with heterogeneous cascade active region design; also QCL DFB Array

Narrow spectral linewidths

- CW: 0.1 3 MHz & <10kHz with frequency stabilization Pulsed: $\sim\!300$ MHz

High pulsed and CW powers of QCLs at TEC/RT **temperatures**

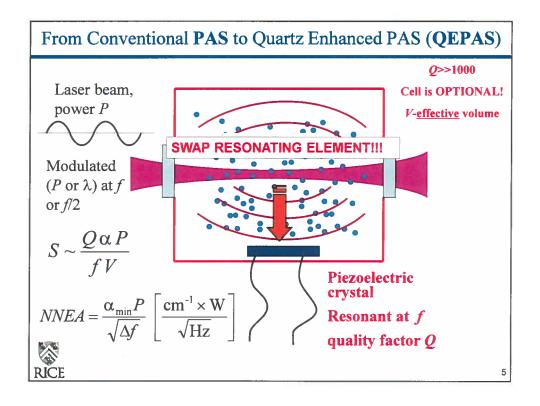
- Room temperature pulsed peak power of > 30 W with 44% wall plug
- CW powers of ~ 5 W with 23% wall plug efficiency at 293 K
- > 600 mW CW DFB @ 285 K; wall plug efficiency 23% at 4.6 μm

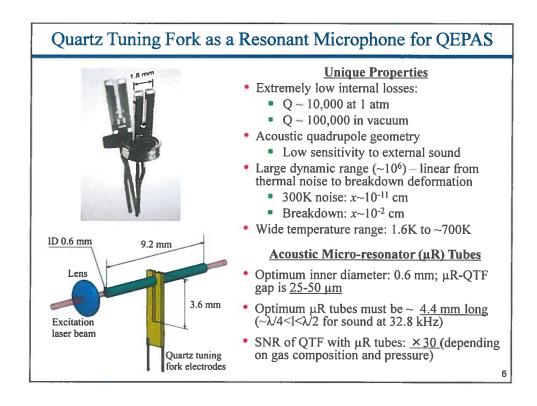












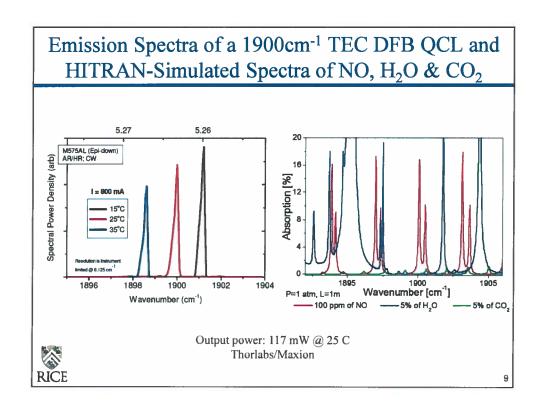
Chemicals of interest in defense & security applications

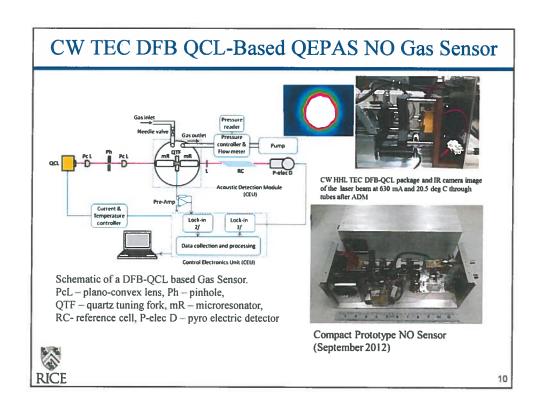
- Chemical warfare agents
 - Nerve agents
 - Blister agents
 - Pulmonary agents
 - Blood agents
- Explosives
- TNT, RDX, PETN, RTN
- High Threat Toxic Industrial Chemicals (TICs)
 - Ammonia, Arsine, Boron trichloride, Carbon disulphide, Chlorine, Ethylene oxide, Fluorine, Formaldehyde, Hydrogen bromide, Hydrogen chloride, Hydrogen cyanide, Hydrogen fluoride, Hydrogen sulphide, Nitric acid, Phosgene, Sulphur dioxide, Sulphuric acid, Tungsten hexafluoride
- Environmental pollutant gas monitoring
 - Naval vessels in particular Submarines
 - · Aircraft and drones
 - Exposure to combat personnel in battle field.

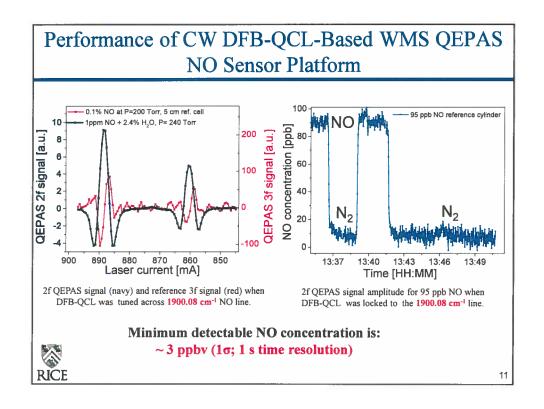


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HITRAN Simulated Mid-Infrared Molecular Absorption Spectra 1e-17 1e-18 1e-18 1e-18 1e-18 1e-18 1e-20 1e-20







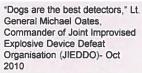
Torr cm \text{ tw} Hz\tilde{k} mW ppbv
C ₂ H ₂ (N ₂)* 6523.88 70 4.1×10* 57 30 NH ₃ (N ₃)* 6528.76 575 3.1×10* 60 60 60 C ₂ H ₂ (N ₃)* 6177.07 715 5.4×10* 15 1,700 CH ₄ (N ₃ +1.2% H ₂ O)* 6057.09 760 3.7×10* 16 240 N ₂ H ₄ 6470.00 700 4.1×10* 16 1,000 H ₂ S (N ₂)* 6357.63 780 5.6×10* 45 5,000 HCI (N ₂ dry) 5739.26 760 5.2×10* 15 700 CO ₂ (N ₃ +1.5% H ₂ O)* 4991.26 50 1.4×10* 4.4 18,000 CH ₂ O (N ₃ :75% RH)* 2804.90 75 8.7×10* 7.2 120 CO (N ₃ +2.2% H ₂ O) 2176.28 100 1.4×10* 71 2
NH ₂ (N ₂)* 6528.76 575 3.1×10* 60 60 ChH ₄ (N ₂)* 6177.07 715 5.4×10* 15 1,700 CH ₄ (N ₂ +1.2% H ₂ O)* 6057.09 760 3.7×10* 16 240 N ₂ H ₄ 6470.00 700 4.1×10* 16 1,000 H ₂ S (N ₁)* 6357.63 780 5.6×10* 45 5,000 HC1 (N ₂ dry) 5739.26 760 5.2×10* 15 700 CO ₂ (N ₂ +1.5% H ₂ O)* 4991.26 50 1.4×10* 4.4 18,000 CH ₂ O (N ₁ :75% RH)* 2804.90 75 8.7×10* 7.2 120 CO (N ₂ +2.2% H ₂ O) 2176.28 100 1.4×10* 71 2
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CH ₂ (N ₂ +1.2% H ₂ O)* 6057.09 760 3.7×10-3 16 240 N ₂ H ₄ 6470.00 700 4.1×10-3 16 1,000 H ₂ S (N ₂)* 6357.63 780 5.6×10-3 45 5,000 HCI (N ₂ dry) 5739.26 760 5.2×10-4 15 700 CO ₂ (N ₂ +1.5% H ₂ O)* 4991.26 50 1.4×10-4 4.4 18,000 CH ₂ O (N ₂ :75% RH)* 2804.90 75 8.7×10-3 7.2 120 CO (N ₂ +2.2% H ₂ O) 2176.28 100 1.4×10-7 71 2
N3H4 6470.00 700 4.1×10 ⁻³ 16 1,000 H±5 (N1) ² 6357.63 780 5.6×10 ⁻³ 45 5,000 HCl (N2 dry) 5739.26 760 5.2×10 ⁻³ 15 700 CO ₂ (N1+1.5% H2O) ² 4991.26 50 1.4×10 ⁻³ 4.4 18,000 CH ₂ O (N2:75% RH) ² 2804.90 75 8.7×10 ⁻³ 7.2 120 CO (N2 +2.2% H ₂ O) 2176.28 100 1.4×10 ⁻³ 71 2
H ₂ S (N ₂)* 6357 63 780 5.6×10* 45 5,000 HCl (N ₂ dry) 5739 26 760 5.2×10-3 15 700 CO ₂ (N ₂ +1.5% H2O)* 4991 26 50 1.4×10-4 4.4 18,000 CH ₂ O (N ₂ :75% RH)* 2804 90 75 8.7×10-8 7.2 120 CO (N ₂ +2.2% H ₂ O) 2176 28 100 1.4×10-7 71 2
HCl (N ₂ dry) 5739 26 760 5.2 × 10 ⁻¹ 15 700 CO ₂ (N ₁ +1.5% H2O) * 4991 26 50 1.4 × 10 ⁻¹ 4.4 18,000 CH ₂ O (N ₁ :75% RH) * 2804 90 75 8.7 × 10 ⁻³ 7.2 120 CO (N ₁ +2.2% H ₂ O) 2176.28 100 1.4 × 10 ⁻³ 71 2
CO ₂ (N ₂ +1.5% H ₂ O) * 4991.26 50 1.4×10 ⁴ 4.4 18,000 CH ₂ O (N ₂ :75% RH) * 2804.90 75 8.7×10 ³ 7.2 120 CO (N ₁ +2.2% H ₂ O) 2176.28 100 1.4×10 ⁻⁷ 71 2
CH ₂ O (N ₂ :75% RH)* 2804 90 75 8.7×10° 7.2 120 CO (N ₁ +2.2% H ₂ O) 2176.28 100 1.4×10 ⁻⁷ 71 2
CO (N ₂ +2.2% H ₂ O) 2176.28 100 1.4×10 ⁻⁷ 71 2
CO (
CO (propylene) 2196.66 50 7.4×10-8 6.5 140
N ₂ O (air+5%SF ₆) 2195.63 50 1.5×10 ⁻⁸ 19 7
C ₂ H ₅ OH (N ₂)** 1934.2 770 2.2×10-7 10 90,000
d-IR NO (N ₂ +H ₂ O) 1900.07 250 7,5×10 ⁻⁹ 100 3
H ₂ O ₂ 1295.6 150 4.6×10 ⁹ 100 12
C ₁ HF ₅ (N ₂)*** 1208.62 770 7.8×10 ⁻⁹ 6.6 9
NH ₃ (N ₃)* 1046,39 110 1.6×10 ^{-‡} 20 6
SF4 948.62 75 2.7x10-10 18 0.05 (50 ppt)

Use of Canines for Sensitive Detection of Explosives and Toxic Industrial Chemicals











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Advantages & Disadvantages of Canines in Explosive and TICs Detection

Advantages

- Non-invasive, safe and easy sample collecting
- Relatively easy training and interpretation of dogs' indications
- Odor samples can be tested several times
- Extremely high detection sensitivity and specificity
- Potentially useful in search, rescue and emergency applications

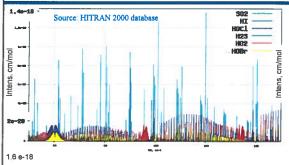
Disadvantages

- To-date a "black-box technology"
- It is a method based on earning a reward, which becomes unreliable after ~ 4 years
- Variation of sensitivity and specificity
- · Re-training of dogs is not effective



Breath 2014, Torum, Prof. T. Jezierski et al., Institute of Genetics and Animal Breeding PAS, Polar 14





Several gas species such as HF, OH, HCN, HCl, HBr, NH₃, H₂O₂, H₂S, H₂O & explosives (in the vapor phase) show strong absorption bands in the THz spectral range.

Mainly rotational levels are involved in THz absorption processes and rotational-translational (R-T) relaxation rates are up to three order of magnitude faster with respect to vibrational-translational (V-T) in the mid-infrared

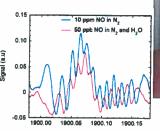
QEPAS signal strongly depends on the energy relaxation rates due to the possibility to operate at low pressure, & thereby taking advantages of the typically very high QTF Q-factors.

Why have QEPAS sensors not been developed in the THz spectral range so far?

Standard QTFs have a very small volume

 $(\sim 0.3 \times 0.3 \times 3 \text{ mm}^3)$

In QEPAS sensor systems, it is critical to avoid laser illumination of the QTF, since the radiation blocked by the QTF prongs results in an undesirable non-zero background as well as a shifting fringe-like interference pattern.

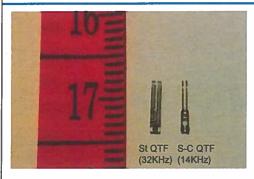




Standard QTF

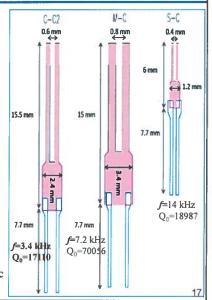
The standard QTF prong separation of 330 μm is comparable with the THz wavelength which prevents the use of a QEPAS sensor architecture in the THz range unless we use large sized QTFs.

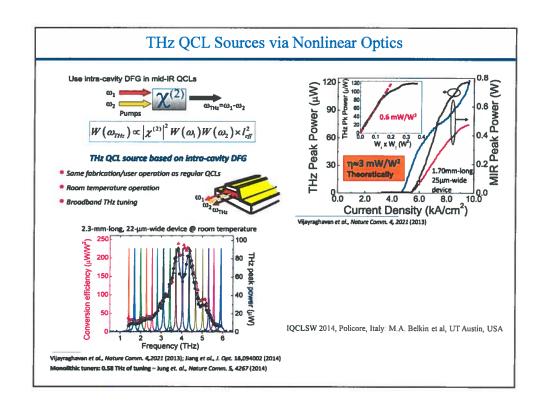
Custom fabricated QTFs with new shapes and dimensions optimized for QEPAS

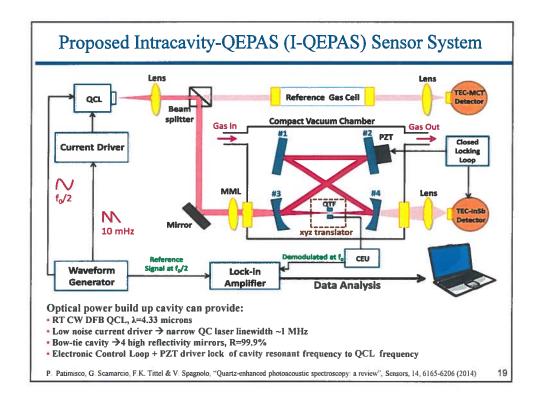


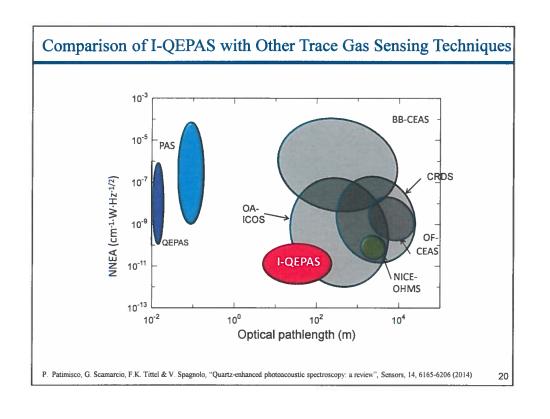
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- Standard photolithographic techniques were used to etch custom QTFs. Chromium/gold layer was deposited on both sides of the custom QTFs for electrical contacts.
- Preliminary results show that this new generation of custom QTFs behave similar to "standard" QTFs in terms of vibrational mode. Response to photoacoustic excitation is in progress.









Summary and Conclusions

- Development of robust, compact, sensitive, selective mid-infrared trace gas sensor technology based on room temperature, continuous wave high performance DFB ICLs & QCLs for detection of explosives and TICs as well as environmental monitoring and medical diagnostics.
- Interband cascade and quantum cascade lasers were used in <u>QEPAS and TDLAS based</u> sensor platforms
- To date six target trace gas species were detected with a 1 sec sampling time:
 - NO: ~5.26 μm, detection limit of 3 ppbv
 - CO: ~4.61 μm, minimum detection limit of 2 ppbv
 - SO₂: ~7.24 μm, detection limit of 100 ppbv
 - CH₄ and N₂O: \sim 7.28 µm, detection limits of 13 and 6 ppbv, respectively
 - H₂O₂: ~7.73 μm, detection limit of 75 ppb
- New target analytes: CH₂O and C₃H₆O
- · Mid-Infrared Optical Frequency Comb

