



Mid-infrared semiconductor laser based trace gas sensor technologies: recent advances and applications

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OUTLINE

CUHK, 20th
Anniversary
Hong Kong,
China
August 29, 2014

- New Laser Based Trace Gas Sensor Technology
 - Novel Multipass Gas Absorption Cells & Electronics
 - Quartz Enhanced Photoacoustic Spectroscopy
- Examples of nine mid-infrared Trace Gas Species
 - C₂H₆, NH₃, NO, CO, SO₂, CH₄, N₂O, H₂O₂ & C₃H₆O
- Future Directions of Laser Based Gas Sensor Technology and Conclusions

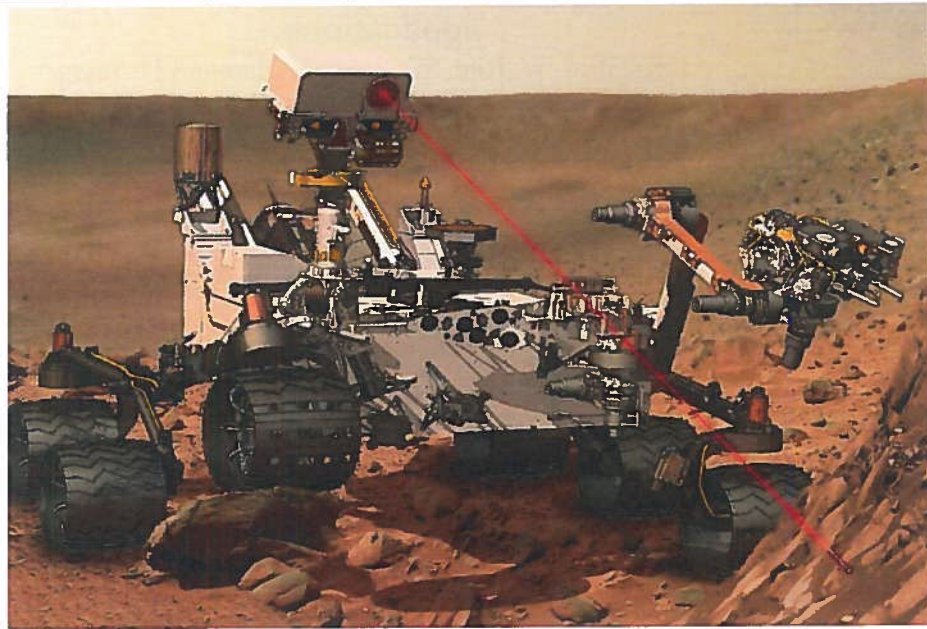
Research support by NSF ERC MIRTHE, NSF-ANR NexCILAS, the Robert Welch Foundation, and Sentinel Photonics Inc. via an EPA Phase 1 SBIR sub-award is acknowledged

Wide Range of Trace Gas Sensing Applications

- **Urban and Industrial Emission Measurements**
 - Industrial Plants
 - Combustion Sources and Processes (e.g. fire detection)
 - Automobile, Truck, Aircraft and Marine Emissions
- **Rural Emission Measurements**
 - Agriculture & Forestry, Livestock
- **Environmental Monitoring**
 - Atmospheric Chemistry (e.g isotopologues, climate modeling,...)
 - Volcanic Emissions
- **Chemical Analysis and Industrial Process Control**
 - Petrochemical, Semiconductor, Pharmaceutical, Metals Processing, Food & Beverage Industries, Nuclear Technology & Safeguards
- **Spacecraft and Planetary Surface Monitoring**
 - Crew Health Maintenance & Life Support
- **Applications in Medical Diagnostics and the Life Sciences**
- **Technologies for Law Enforcement, Defense and Security**
- **Fundamental Science and Photochemistry**



“Curiosity” Landed on Mars on August 6, 2012



Laser based Trace Gas Sensing Techniques

- **Optimum Molecular Absorbing Transition**
 - Overtone or Combination Bands (NIR)
 - Fundamental Absorption Bands (Mid-IR)
- **Long Optical Pathlength**
 - Multipass Absorption Cell (White, Herriot, Chernin, Sentinel Photonics)
 - Cavity Enhanced and Cavity Ringdown Spectroscopy
 - Open Path Monitoring (with retro-reflector): Standoff and Remote Detection
 - Fiberoptic Evanescent Wave Spectroscopy
- **Spectroscopic Detection Schemes**
 - Frequency or Wavelength Modulation
 - Balanced Detection
 - Zero-air Subtraction
 - Photoacoustic & Quartz Enhanced Photoacoustic Spectroscopy (QEPAS)

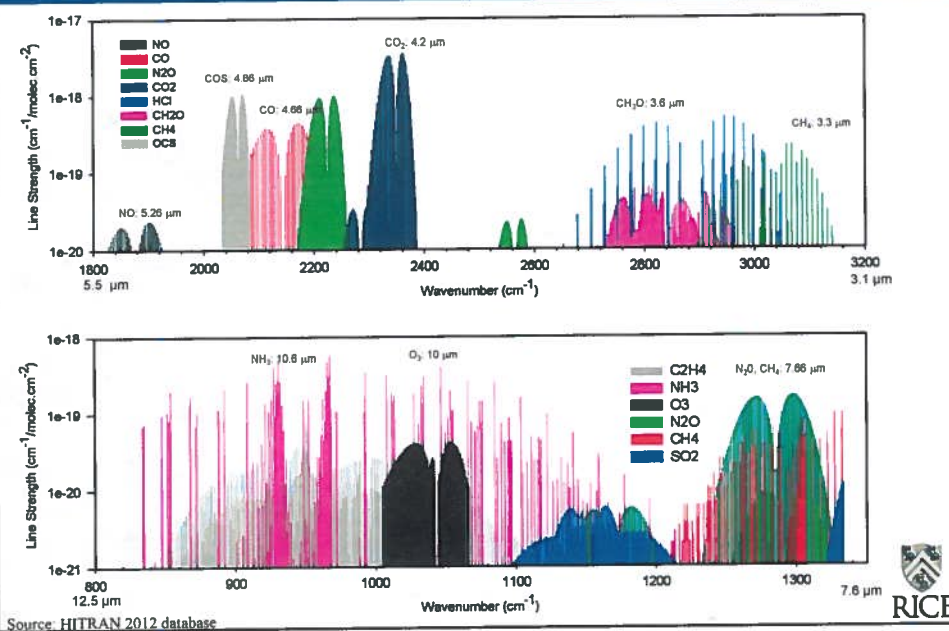


Other Spectroscopic Methods

- Faraday Rotation Spectroscopy (limited to paramagnetic chemical species)
- Differential Optical Dispersion Spectroscopy (DODiS)
- Noise Immune Cavity Enhanced-Optical Heterodyne Molecular Spectroscopy (NICE-OHMS)
- Frequency Comb Spectroscopy
- Laser Induced Breakdown Spectroscopy (LIBS)



HITRAN Simulated Mid-Infrared Molecular Absorption Spectra



Mid-IR Source Requirements for Laser Spectroscopy

<u>REQUIREMENTS</u>	<u>IR LASER SOURCE</u>
Sensitivity (% to pptv)	Optimum Wavelength, Power
Selectivity (Spectral Resolution)	Stable Single Mode Operation and Narrow Linewidth
Multi-gas Components, Multiple Absorption Lines and Broadband Absorbers	Mode Hop-free Wavelength Tunability
Directionality or Cavity Mode Matching	Beam Quality
Rapid Data Acquisition	Fast Time Response
Room Temperature Operation	High wall plug efficiency, no cryogenics or cooling water
Field deployable in harsh environments	Compact & Robust

Key Characteristics of Mid-IR QCL & ICL Sources – Aug 2014

- **Band – structure engineered devices**
Emission wavelength is determined by layer thickness – MBE or MOCVD; Type I QCLs operate in the 3 to 24 μm spectral region; Type II and GaSb based ICLs can cover the 3 to 6 μm spectral range.

- Compact, reliable, stable, long lifetime, and commercial availability
- Fabry-Perot (FP), single mode (DFB) and multi-wavelength devices

- **Wide spectral tuning ranges in the mid-IR**

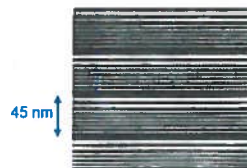
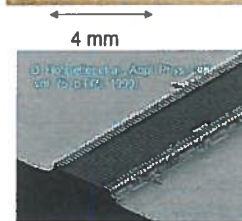
- 1.5 cm^{-1} using injection current control for DFB devices
- 10-20 cm^{-1} using temperature control for DFB devices
- ~100 cm^{-1} using current and temperature control for QCL DFB Array
- ~525 cm^{-1} (22% of c.w.) using an external grating element and FP chips with heterogeneous cascade active region design; also QCL DFB Array

- **Narrow spectral linewidths**

- CW: 0.1 - 3 MHz & <10kHz with frequency stabilization (0.0004 cm^{-1})
- Pulsed: ~ 300 MHz

- **High pulsed and CW powers of QCLs at TEC/RT temperatures**

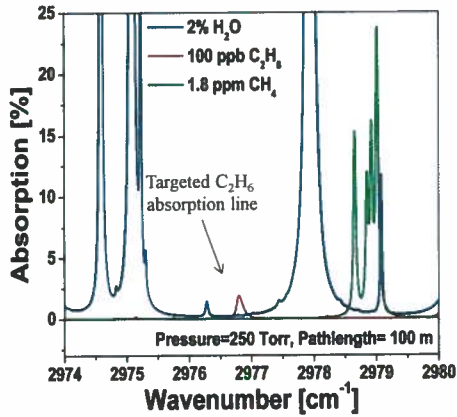
- Room temperature pulsed power of > 30 W with 44% wall plug efficiency and CW powers of ~ 5 W with 23% wall plug efficiency at 293 °K
- > 1W, TEC CW DFB @ 4.6 μm ~~EC QCLs & ICLs~~
- > 600 mW CW ~~DFB~~ @ RT; wall plug efficiency of ~ ~~8%~~ % at 4.6 μm ;



DFB @ 2859 2.5%

Motivation for Mid-infrared C₂H₆ Detection

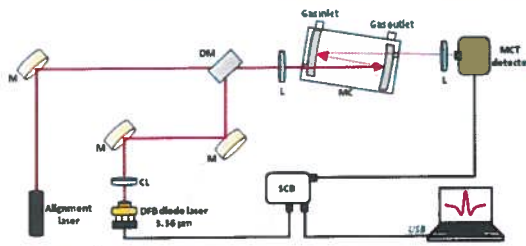
- **Application in medical breath analysis**
 - Asthma
 - Schizophrenia
 - Lung cancer
 - Vitamin E deficiency
- **Atmospheric chemistry and climate**
 - Fossil fuel and biofuel consumption
 - Biomass burning
 - Vegetation/soil
 - Natural gas loss



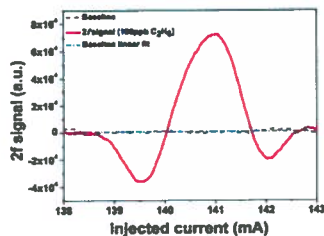
HITRAN absorption spectra of C₂H₆, CH₄, and H₂O



C₂H₆ Detection with a 3.36 μm CW DFB LD using a Novel Compact Multipass Absorption Cell and Control Electronics



Schematic of a C₂H₆ gas sensor using a Nanoplus 3.36 μm DFB laser diode M – mirror, CL – collimating lens, DM – dichroic mirror, MC – multipass cell, L – lens, SCB – sensor control board.



2f WMS signal for a C₂H₆ line at 2976.8 cm⁻¹ at 200 Torr

Minimum detectable C₂H₆ concentration:
~ 740 pptv (1σ; 1 s time resolution)



Innovative long path, small volume multipass gas cell: **57.6 m with 459 passes**



MGC dimensions: **17 x 6.5 x 5.5 (cm)**
Distance between the MGC mirrors: 12.5 cm

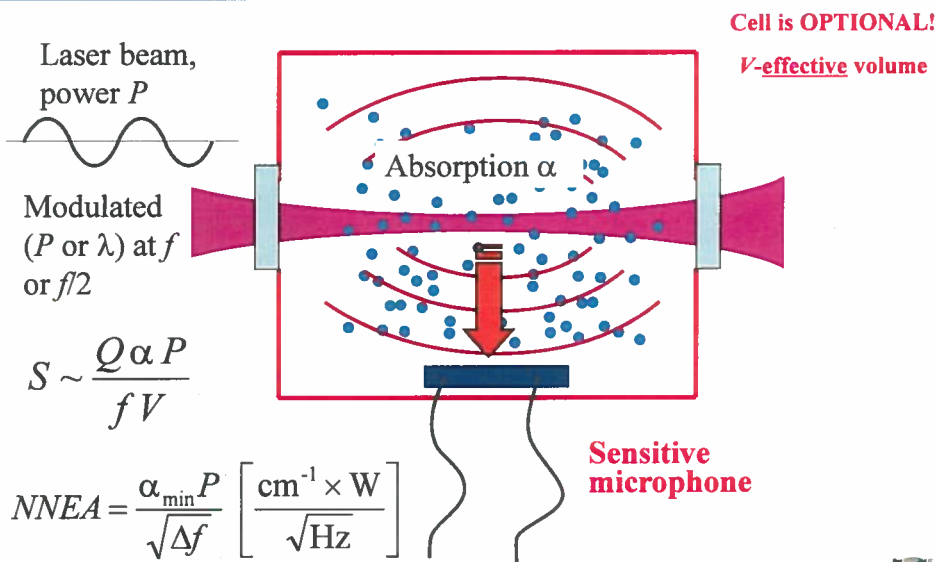
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Motivation for NH₃ Detection

- **Medical diagnostics**
 - Kidney disease
 - Liver failure and Cirrhosis
 - Brain Cells dysfunction
 - Drowsiness and Coma
- **Atmospheric chemistry**
- **Pollutant gases monitoring**
- **Monitoring NH₃ concentrations in the exhaust stream of NO_x removal systems based on selective catalytic reduction (SCR) techniques associated with electric power plants**
- **Spacecraft related trace gas monitoring**

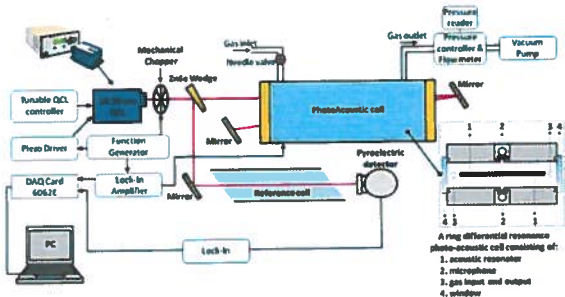


Conventional Photoacoustic Spectroscopy (PAS)



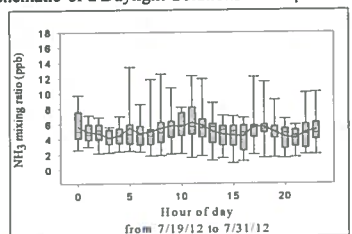
Unrestricted Remote detection of

Atmospheric NH₃ Measurements using an EC-QCL PAS Sensor

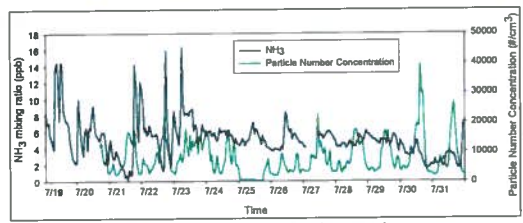


NH₃ sensor deployed at the UH Moody Tower rooftop monitoring site.

Schematic of a Daylight Solutions 10.36 μm CW TEC EC-QCL based PAS NH₃ Sensor.

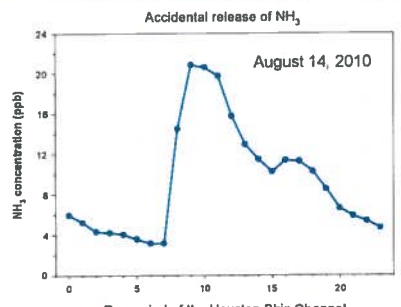


Diurnal profile of atmospheric NH₃ levels in Houston, TX.

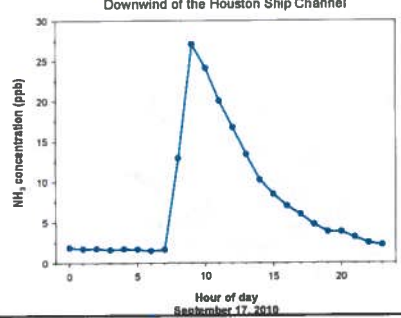


Comparison between NH₃ and particle number concentration time series from July 19 to July 31 2012.

NH₃ Detection due to a Fire resulting from a Truck Collision



A chemical incident occurred at ~ 6 a.m. after two trucks collided on I-59. Both trucks caught fire. [www.chron.com]

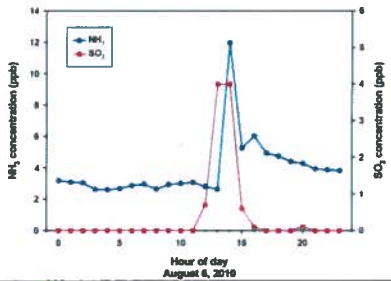
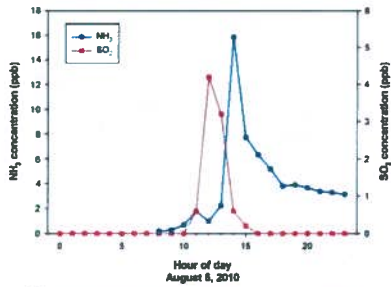


Estimated hourly NH₃ emission from the Houston Ship Channel area is about 0.25 ton. Mellqvist et al., (2007) Final Report, HARC Project H-53.



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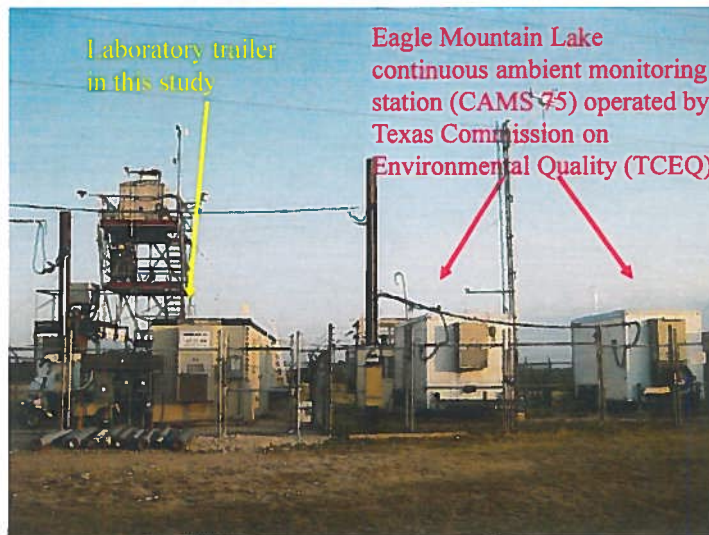
Sporadic increase in NH_3 concentration levels related to emissions by the Parish Electric Power Plant, TX



The Parish electric power plant is located near the Brazos River in Fort Bend County, Texas (~27 miles SW from downtown Houston)



Fort-Worth, Dallas(TX) CAMS 75 & TCEQ monitoring site



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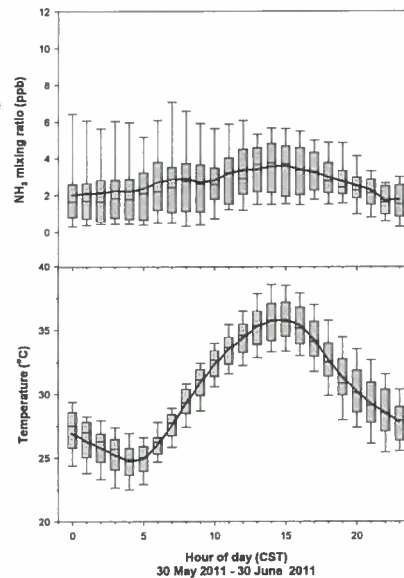
Instrumentation available at CAMS 75 & TCEQ monitoring site

Species/parameter	Measurement technique
NH ₃	Daylight Solutions External Cavity Quantum Cascade Laser (Photo-acoustic Spectroscopy)
CO	Thermo Electron Corp 48C Trace Level CO Analyzer (Gas Filter Correlation)
SO ₂	Thermo Electron Corp 43C Trace Level SO ₂ Analyzer (Pulsed Fluorescence)
NO _x	Thermo Electron Corp 42C Trace Level NO-NO ₂ -NO _x Analyzer (Chemiluminescence)
NO _y	Thermo Electron Corp 42C-Y NO _y Analyzer (Molybdenum Converter)
HNO ₃	Mist Chamber coupled to Ion Chromatography (Dionex, Model CD20-1)
HCl	Mist Chamber coupled to Ion Chromatography (Dionex, Model CD20-1)
VOC _s	IONICON Analytik Proton Transfer Reaction Mass Spectrometer and TCEQ Automated Gas Chromatograph
PBL height	Vaisala Ceilometer CL31 with updated firmware to work with Vaisala Boundary Layer View software
Temperature	Campbell Scientific HMP45C Platinum Resistance Thermometer
Wind speed	Campbell Scientific 05103 R. M. Young Wind Monitor
Wind direction	Campbell Scientific 05103 R. M. Young Wind Monitor

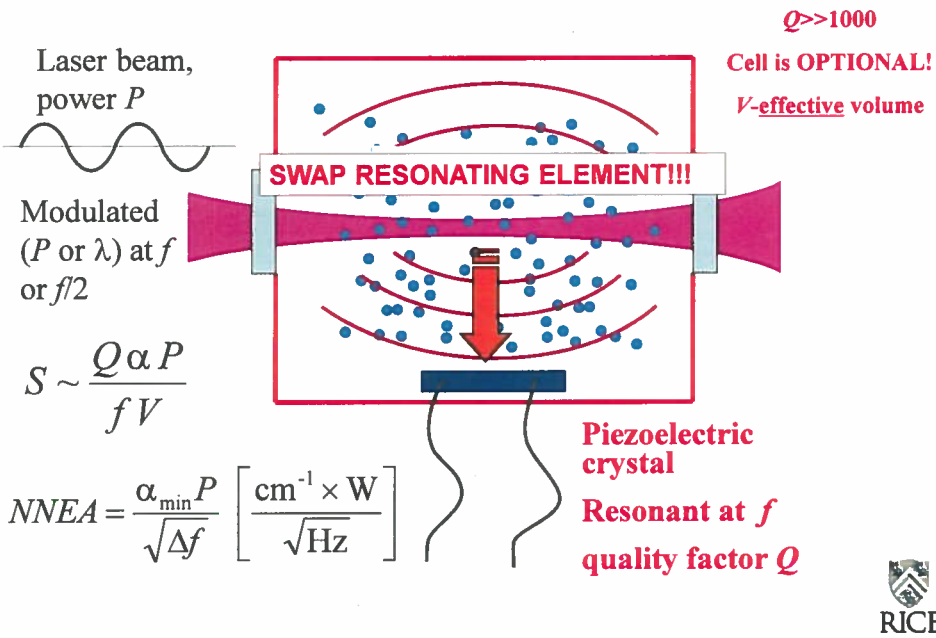


NH₃ Source Attribution & Temperature Variations

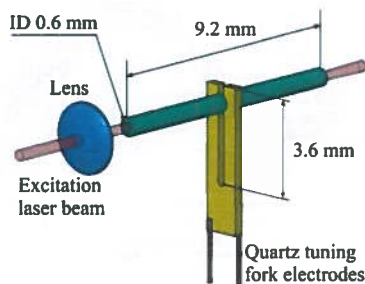
- Emission events from specified point sources (i.e., industrial facilities)
- Estimated NH₃ emissions from cows (1.3 tons/day)
- Estimated NH₃ emissions from soil and vegetation (0.15 tons/day)
- EPA PMF (biogenic:74.1%; light duty vehicles:12.1%; natural gas/industry: 9.4%; heavy duty vehicles:4.4%)
- Livestock might account for approximately 66.4% of total NH₃ emissions
- Increased contribution from industry (→18.9%)



From Conventional PAS to Quartz Enhanced PAS (QEPAS)



Quartz Tuning Fork as a Resonant Microphone for QEPAS



Unique Properties

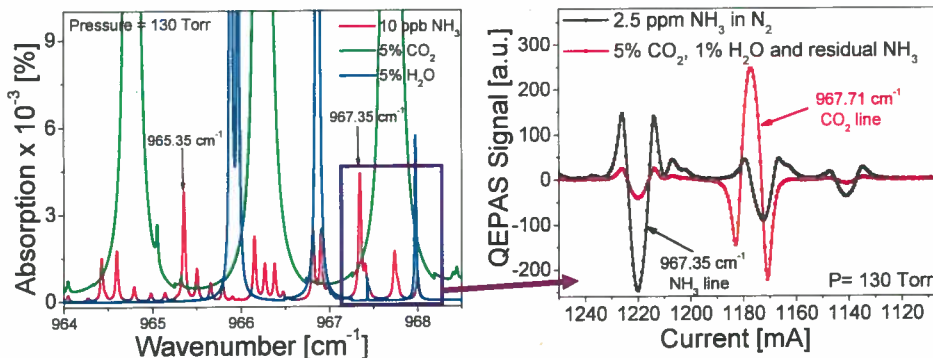
- Extremely low internal losses:
 - $Q \sim 10\,000$ at 1 atm
 - $Q \sim 100\,000$ in vacuum
- Acoustic quadrupole geometry
 - Low sensitivity to external sound
- Large dynamic range ($\sim 10^6$) – linear from thermal noise to breakdown deformation
 - 300K noise: $x \sim 10^{-11}$ cm
 - Breakdown: $x \sim 10^{-2}$ cm
- Wide temperature range: from 1.6K to ~ 700 K

Acoustic Micro-resonator (μ R) Tubes

- Optimum inner diameter: 0.6 mm; μ R-QTF gap is 25-50 μ m
- Optimum mR tubes must be ~ 4.4 mm long ($\sim \lambda/4 < l < \lambda/2$ for sound at 32.8 kHz)
- SNR of QTF with μ R tubes: $\times 30$ (depending on gas composition and pressure)

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Optimum NH₃ Line Selection for a 10.34 μm CW TEC DFB QCL

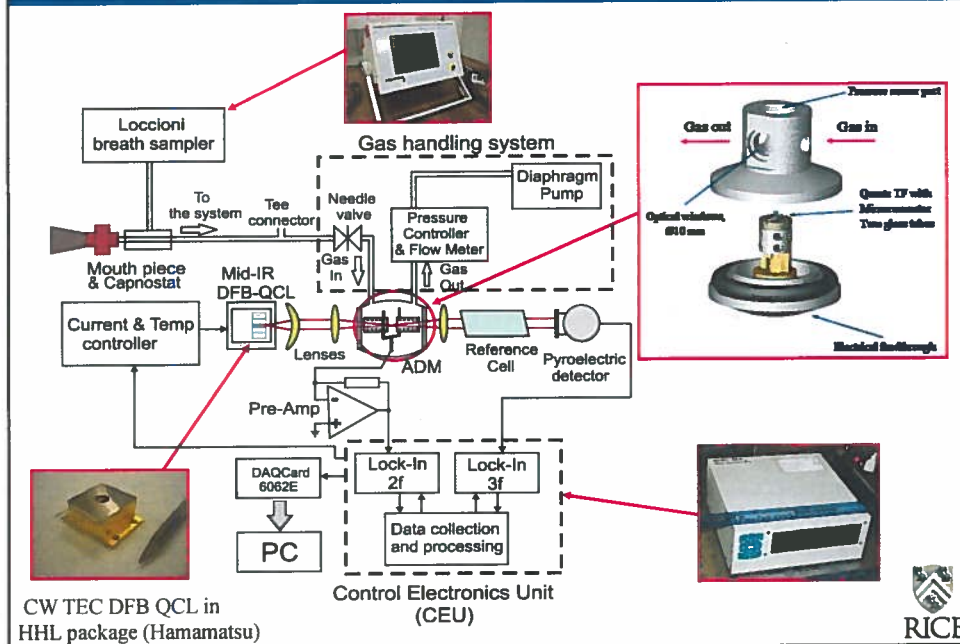


Simulated HITRAN high resolution spectra @ 130 Torr indicating two NH₃ absorption lines of interest

No overlap between NH₃ and CO₂ absorption lines was observed for the selected **967.35 cm⁻¹** NH₃ absorption line in the ν₂ R-band.

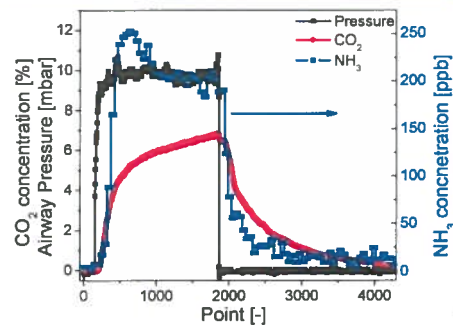
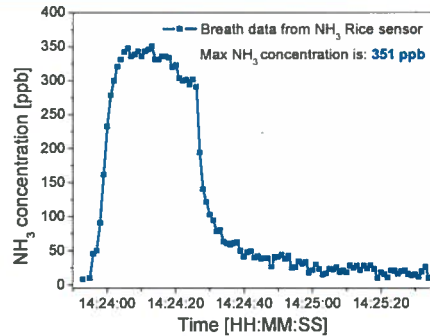


QEPAS based NH₃ Gas Sensor Architecture



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Real-time exhaled human NH₃ Breath Measurements



Airway pressure (black), CO₂ (red), and NH₃ (blue) profiles of a single breath exhalation lasting 40sec.



Successful testing of a 2nd generation breath ammonia monitor installed in a clinical environment. (Johns Hopkins, Baltimore, MD and St. Luke's Hospital, Bethlehem, PA)

Minimum detectable concentration of NH₃ is:
~ 6 ppbv at 967.35 cm⁻¹ (1σ; 1 s time resolution)

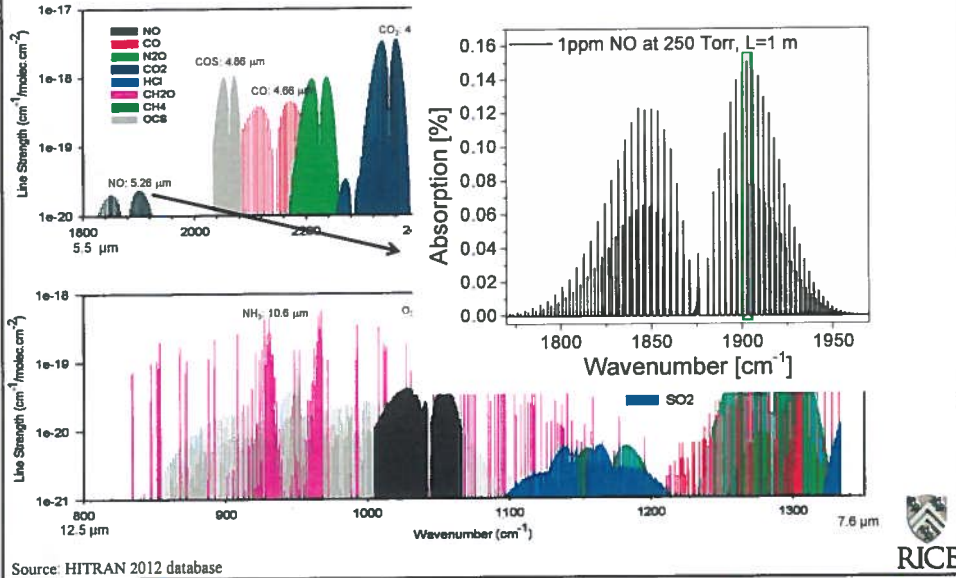
Motivation for Nitric Oxide Detection

- **NO in medicine and biology**
 - Important signaling molecule in physiological processes in humans and mammals (1998 Nobel Prize in Physiology/Medicine)
 - Treatment of asthma, chronic obstructive pulmonary disease (COPD) & lung rejection
- **Environmental pollutant gas monitoring**
 - Ozone depletion
 - Precursor of smog and acid rain
 - NO_x monitoring from automobile exhaust and power plant emissions
- **Atmospheric Chemistry**

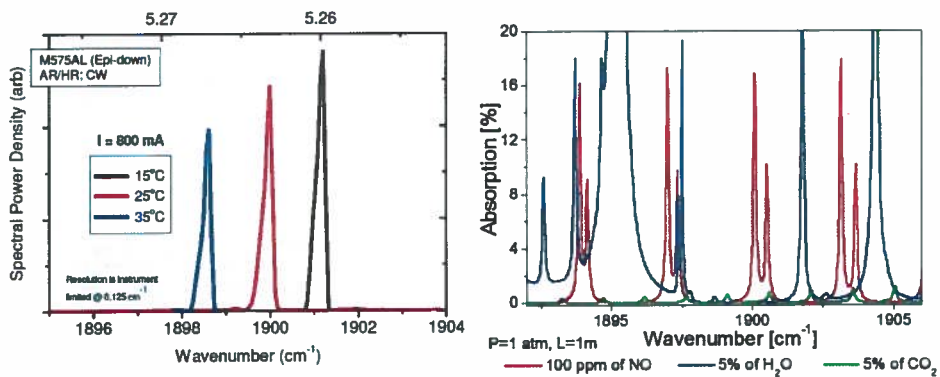


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Molecular Absorption Spectra within two Mid-IR Atmospheric Windows and NO absorption @ 5.26 μm



Emission spectra of a 1900 cm^{-1} TEC DFB QCL and HITRAN simulated spectra of NO, H₂O & CO₂

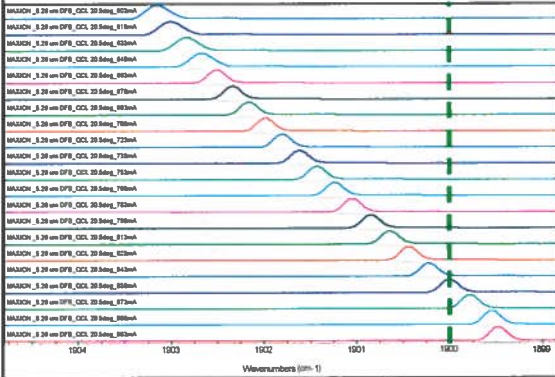


Output power: 117 mW @ 25 C
Thorlabs/Maxion

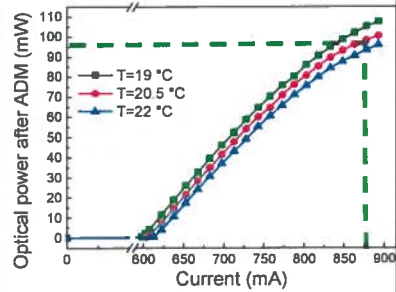


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Performance of a 5.26 μm CW HHL TEC DFB-QCL



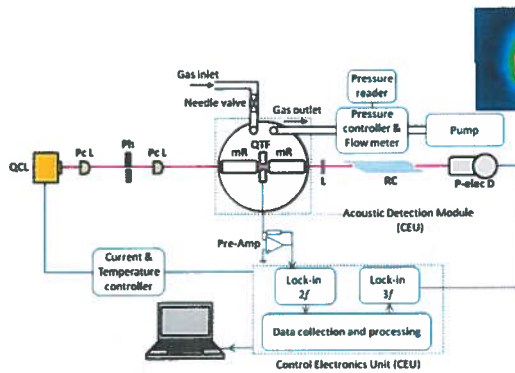
Single frequency QCL radiation recorded with FTIR for different laser current values at a QCL temperature of 20.5°C.



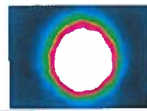
CW DFB-QCL optical power and current tuning at three different temperatures.



CW TEC DFB QCL based QEPAS NO Gas Sensor



Schematic of a DFB-QCL based Gas Sensor.
 PcL – plano-convex lens, Ph – pinhole,
 QTF – quartz tuning fork, mR – microresonator,
 RC- reference cell, P-elec D – pyro electric detector



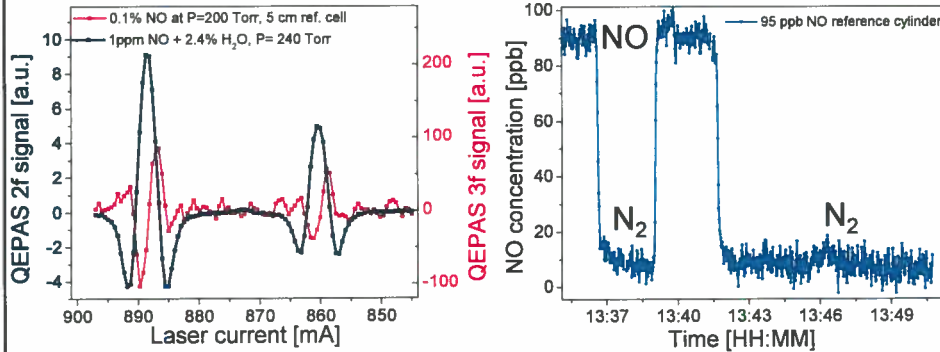
CW HHL TEC DFB-QCL package and IR camera image of the laser beam at 630 mA and 20.5 deg C through tubes after ADM



Compact Prototype NO Sensor (September 2012)



Performance of CW DFB-QCL based WMS QEPAS NO Sensor Platform



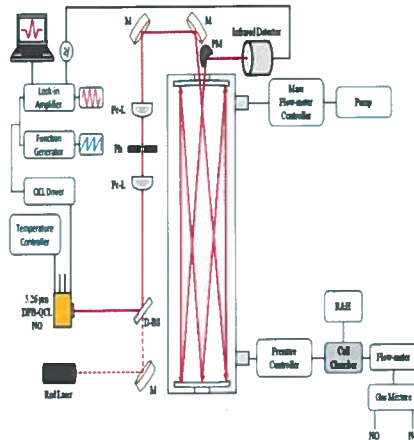
2f QEPAS signal (navy) and reference 3f signal (red) when DFB-QCL was tuned across 1900.08 cm^{-1} NO line.

2f QEPAS signal amplitude for 95 ppb NO when DFB-QCL was locked to the 1900.08 cm^{-1} line.

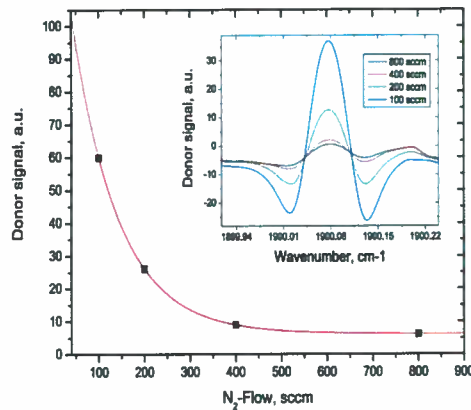
Minimum detectable NO concentration is:
 $\sim 3 \text{ ppbv}$ (1σ ; 1 s time resolution)



QCL based TDLAS Sensor for Detection of NO Emission from Cancer Cells



Schematic drawing of the sensor setup



Dependence of the sensor signal from biological samples on the overall gas flow of the sensor system (black squares). The inset shows spectra corresponding to the data points.

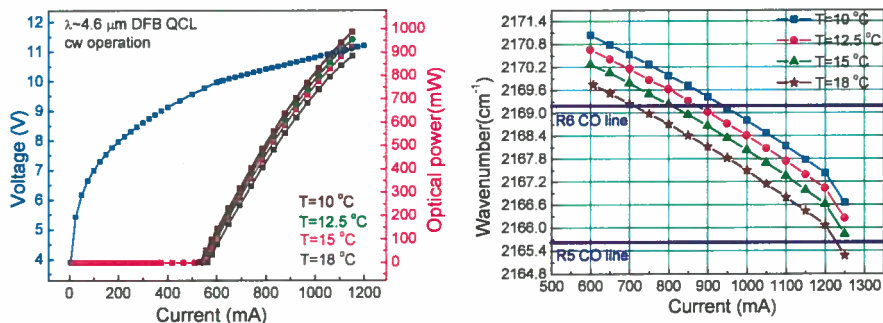
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Motivation for Carbon Monoxide Detection

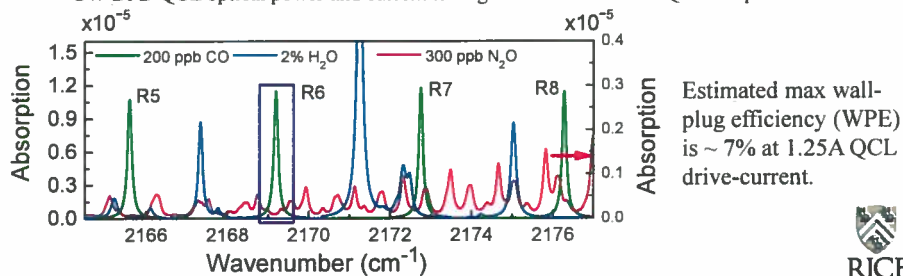
- **CO in Medicine and Biology**
 - Hypertension and abnormality in heme metabolism
- **Public Health**
 - Extremely dangerous to human life even at a low concentrations. Therefore CO must be carefully monitored at low concentration levels (<35 ppm).
- **Atmospheric Chemistry**
 - Incomplete combustion of natural gas, fossil fuel and other carbon containing fuels.
 - Impact on atmospheric chemistry through its reaction with hydroxyl (OH) for troposphere ozone formation and changing the level of greenhouse gases (e.g. CH₄).



Performance of a 4.61 μm high power CW TEC DFB QCL

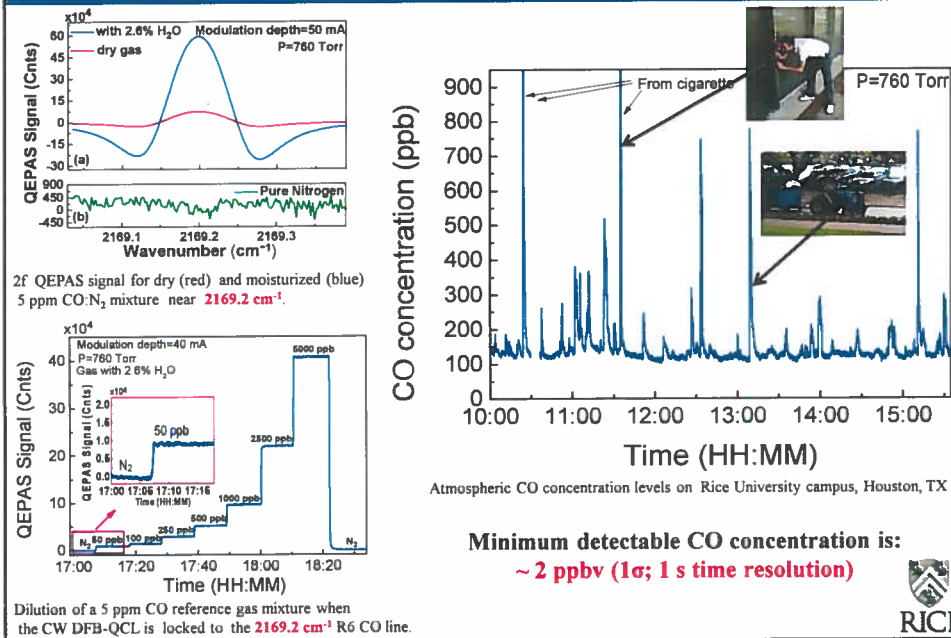


CW DFB-QCL optical power and current tuning at a four different QCL temperatures.



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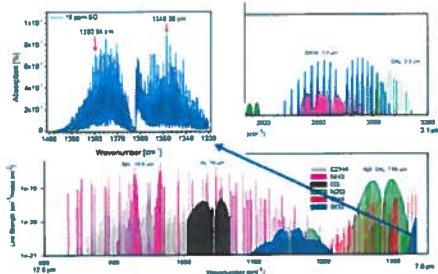
CW DFB-QCL based CO QEPAS Sensor Results



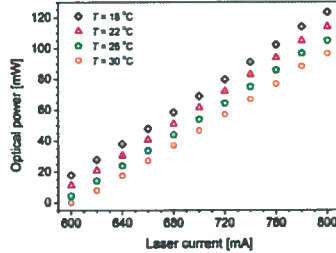
CW DFB-QCL based SO₂ QEPAS Results

Motivation for Sulfur Dioxide Detection

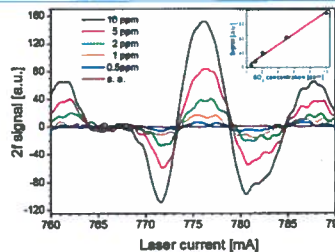
- [SO₂ exposure affects lungs and causes breathing difficulties, bronchitis, cardiovascular disease](#)
- Currently, reported annual average atmospheric SO₂ concentrations range from ~ 1 - 6 ppb
- Prominent air pollutant
- Emitted from coal fired power plants (~73%) and other industrial facilities (~20%)
- In the atmosphere SO₂ converts to sulfuric acid → primary contributors to acid rain
- SO₂ reacts to form sulfate aerosols
- Primary SO₂ exposure for 1 hour is 75 ppb



Molecular Absorption Spectra within two Mid-IR Atmospheric Windows



7.24 μm CW DFB-QCL optical power and current tuning at three different operating temperatures.



2f WMS QEPAS signals for different SO₂ concentrations when laser was tuned across 1380.9 cm⁻¹ line.

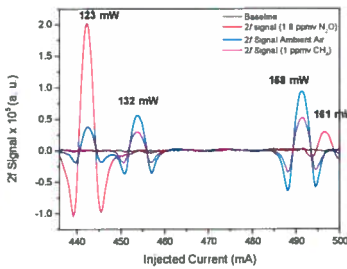
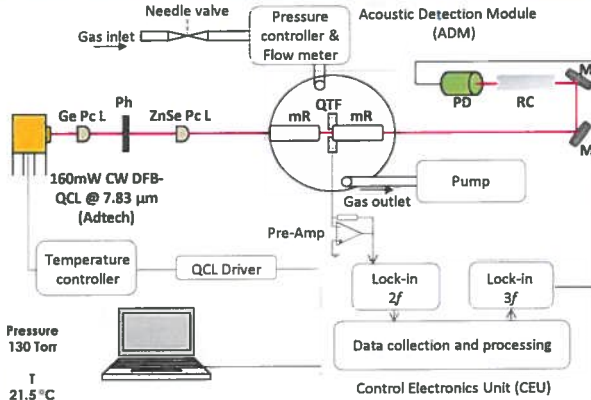
Minimum detectable SO₂ concentration is: **~ 100 ppbv (1σ; 1 s time resolution)**

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QEPAS based CH₄ and N₂O Gas Sensor

Motivation for CH₄ and N₂O Detection

- **Medical Diagnostics**
 - Nausea, blurred vision, vomiting
- Prominent greenhouse gases
- Sources: wetlands, leakage from natural gas systems, fossil fuel production and agriculture



Pressure
130 Torr
T
21.5 °C
AM
4 mA
f
32760 Hz
f_{mod}
16380 Hz

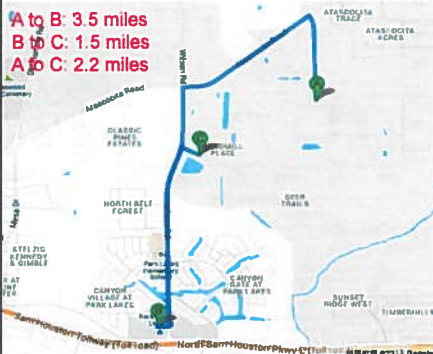
Detection Limit (1σ) with a 1-sec averaging time
Methane (CH₄) (1275.04 cm⁻¹) **13 ppbv**
Nitrous Oxide (N₂O) (1275.5 cm⁻¹) **6 ppbv**

Deduced N₂O concentration in the ambient laboratory air: **331 ppbv**

The Analyst Aug. 2013



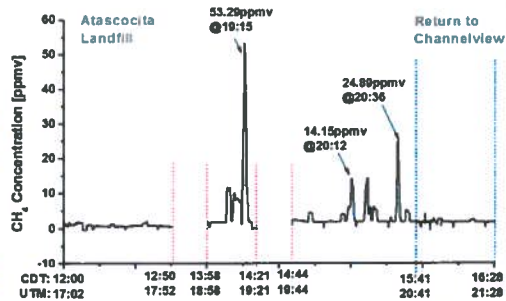
CH₄ Measurements performed with a DFB-QCL based QEPAS Sensor installed in Aerodyne Mobile Laboratory (Sept 7, 2013)



A: 29.9599° North, 95.2334° West
B: 29.9364° North, 95.2508° West
C: 29.9547° North, 95.2462° West (Landfill)

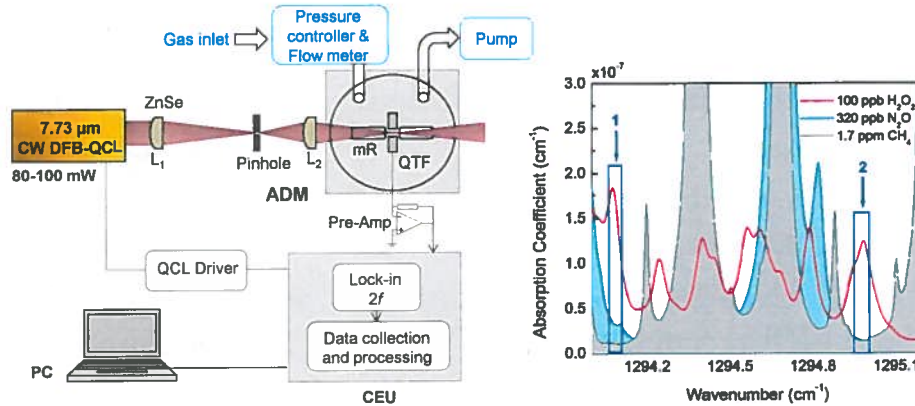
Jahjah et. al., Opt. Lett., 39, 957-960, 2014

Atascocita Landfill, Humble, TX 77396 CH₄ Perimeter Measurements



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QCL based QEPAS Sensor for Hydrogen Peroxide (H₂O₂)



Schematic of QCL based QEPAS sensor:
ADM – acoustic detection module; CEU – control electronics unit; PC – personal computer.

Simulated spectra (HITRAN) of H₂O₂ at 296 K and 130 Torr, along with atmospheric interfering molecules of CH₄ and N₂O; two target wavelengths at 1294.1 and 1294.9 cm⁻¹ are shown.



H₂O₂ Exposure limit is set at 1 ppmv by OSHA

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QEPAS Performance for Trace Gas Species (August 2014)

	Molecule (Host)	Frequency, cm ⁻¹	Pressure, Torr	NNEA, cm ¹ W/Hz ^{1/2}	Power, mW	NEC (τ=1s), ppmv
VIS	O ₂ (air)	35087.70	700	3.0 × 10 ⁻⁸	0.8	1.27
	O ₂ (N ₂)	13099.30	158	4.74 × 10 ⁻⁷	1228	13
	C ₂ H ₂ (N ₂)*	6523.88	720	4.1 × 10 ⁻⁹	57	0.03
NIR	NH ₃ (N ₂)*	6528.76	575	3.1 × 10 ⁻⁹	60	0.06
	C ₂ H ₄ (N ₂)*	6177.07	715	5.4 × 10 ⁻⁹	15	1.7
	CH ₄ (N ₂ +1.2% H ₂ O)*	6057.09	760	3.7 × 10 ⁻⁹	16	0.24
	N ₂ H ₄	6470.00	700	4.1 × 10 ⁻⁸	16	1
	H ₂ S (N ₂)*	6357.63	780	5.6 × 10 ⁻⁸	45	5
	HCl (N ₂ dry)	5739.26	760	5.2 × 10 ⁻⁸	15	0.7
	CO ₂ (N ₂ +1.5% H ₂ O)*	4991.26	50	1.4 × 10 ⁻⁴	4.4	18
	CH ₂ O (N ₂ :75% RH)*	2804.90	75	8.7 × 10 ⁻⁶	7.2	0.12
	CO (N ₂ +2.2% H ₂ O)	2176.28	100	1.4 × 10 ⁻⁷	71	0.002
	CO (propylene)	2196.66	50	7.4 × 10 ⁻⁴	6.5	0.14
Mid-IR	N ₂ O (air+5%SF ₆)	2195.63	50	1.5 × 10 ⁻⁸	19	0.007
	C ₂ H ₅ OH (N ₂)**	1934.2	770	2.2 × 10 ⁻⁷	10	90
	NO (N ₂ +H ₂ O)	1900.07	250	7.5 × 10 ⁻⁸	100	0.003
	C ₂ HF ₅ (N ₂)***	1208.62	770	7.8 × 10 ⁻⁹	6.6	0.009
	NH ₃ (N ₂)*	1046.39	110	1.6 × 10 ⁻⁸	20	0.006
	SF ₆	948.62	75	2.7 × 10 ⁻¹⁰	18	5 × 10 ⁻³ (50 ppt)

* - Improved microresonator
** - Improved microresonator and double optical pass through ADM
*** - With amplitude modulation and metal microresonator
NNEA – normalized noise equivalent absorption coefficient.
NEC – noise equivalent concentration for available laser power and τ=1s time constant, 18 dB/oct filter slope.

For comparison: conventional PAS 2.2 × 10⁻⁹ cm⁻¹W/√Hz for NH₃

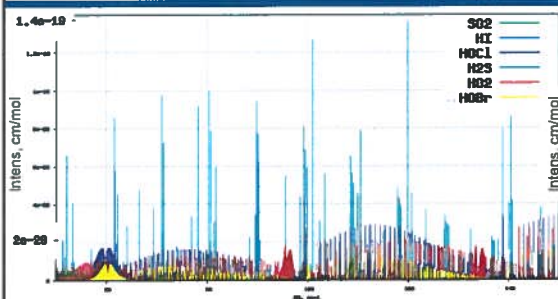
Future Directions and Outlook

- New target analytes: formaldehyde (CH₂O), ethylene (C₂H₄), ozone (O₃), nitrate (NO₃),
- Ultra-compact, low cost, robust sensors (e.g. C₂H₆, NO, CO...) CH₄
- QCL based ultra-portable atmospheric carbon isotope monitor for ¹²CH₄ & ¹³CH₄
- **Monitoring of broadband absorbers: acetone (C₃H₆O)-**, ~ MDL:1.5 ppm with a 7mW ICL & AM, or 20ppb with a 100mW QCL @ 8.23μm; benzene (C₆H₆)...
- **Optical power build-up cavity designs (I-QEPAS)**
- **THz QEPAS based sensors**
- Development of trace gas sensor networks

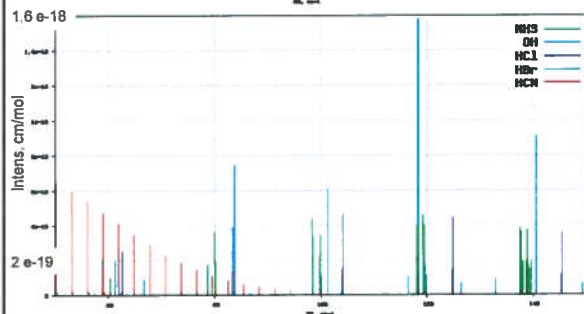


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Why is THz important for gas sensing ?



Several gas species such as HF, OH, HCN, HCL, HBr, NH₃, H₂O₂, H₂S, H₂O, explosive (vapor phase) show strong absorption bands in the THz spectral region

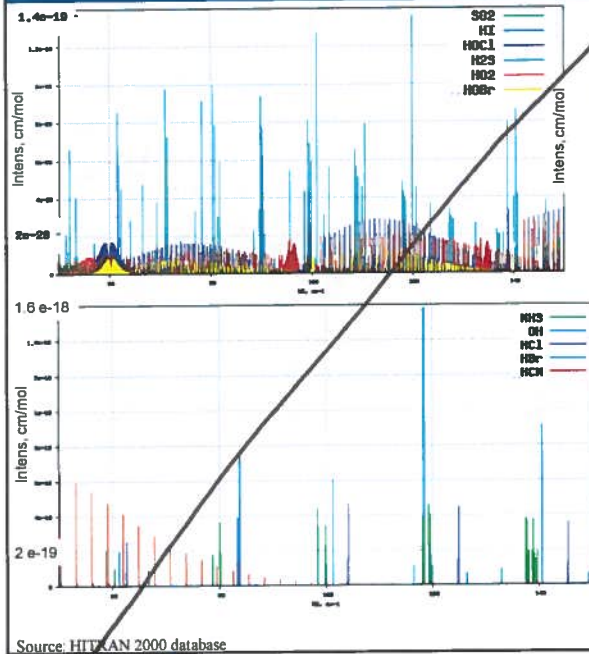


Mainly rotational levels are involved in THz absorption processes and rotational-translational (R-T) relaxation rates are up to **three order of magnitude faster** with respect to vibrational-translational (V-T) relaxation rates.

Source: HITRAN 2000 database

Handwritten notes:
 NH₃
 OH

Why is THz important for gas sensing ?



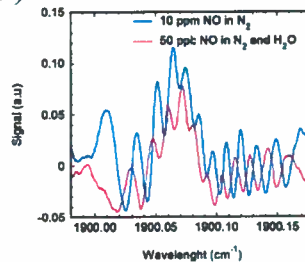
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Mainly rotational levels are involved in THz absorption processes and rotational-translational (R-T) relaxation rates are up to *three order of magnitude faster* with respect to vibrational-translational (V-T) relaxation rates.

Why QEPAS sensors have not been developed in the THz range to-date?

Standard QTFs are characterized by a very small sensitive volume ($\sim 0.3 \times 0.3 \times 3 \text{ mm}^3$)

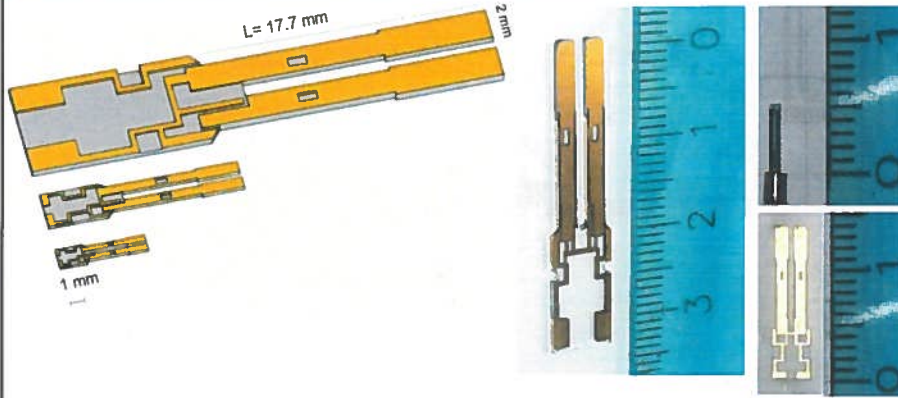
In QEPAS experiments, it is critical to avoid laser illumination of the QTF, since the radiation blocked by the QTF prongs results in an undesirable non-zero background which is associated with a shifting fringe-like interference pattern.



The narrow space ($300 \mu\text{m}$) between the QTF prongs is comparable with the wavelength of THz sources, which has so far represented the main limitation preventing the use of QEPAS system in THz range.

Hence, larger sized QTFs are required for operating in the THz regime.

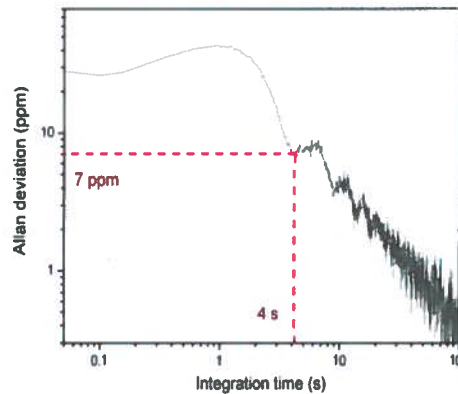
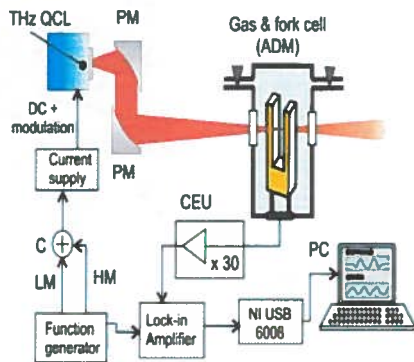
Custom QTFs can be scaled in dimension with respect to a standard 32.76 KHz QTF (7 and 3 times larger) for THz trace gas sensing



- Standard photolithographic techniques can be used to etch custom QTFs, starting from a z-cut quartz wafer.
- Chromium/gold contacts can be deposited on both sides of a QTF.



THz QEPAS based Sensor for Methanol (CH_3OH) Detection



Schematic of THz QCL-based QEPAS sensor. PM – paraboloidal mirror; C – power combiner; LM, HM – low-frequency modulation (triangular ramp), high-frequency modulation; CEU – control electronics unit; PC – personal computer.

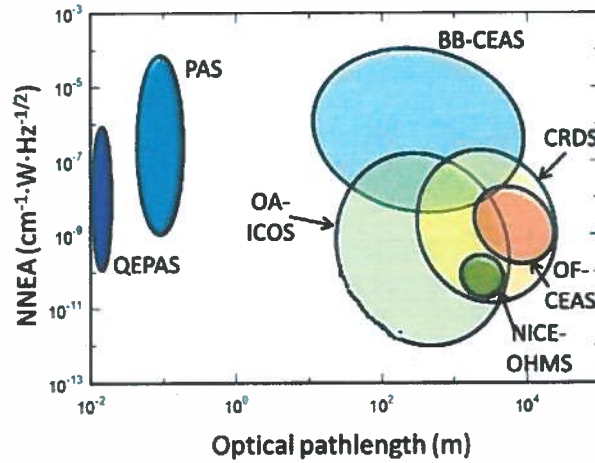
Allan deviation in ppm of the QEPAS signal as a function of the integration time. The initial growth from 0.1 to 1s reflects delays due to the signal sampling time (200 ms).

Spangolo et. al. Appl. Phys. Lett. 103, 021105 (2013)



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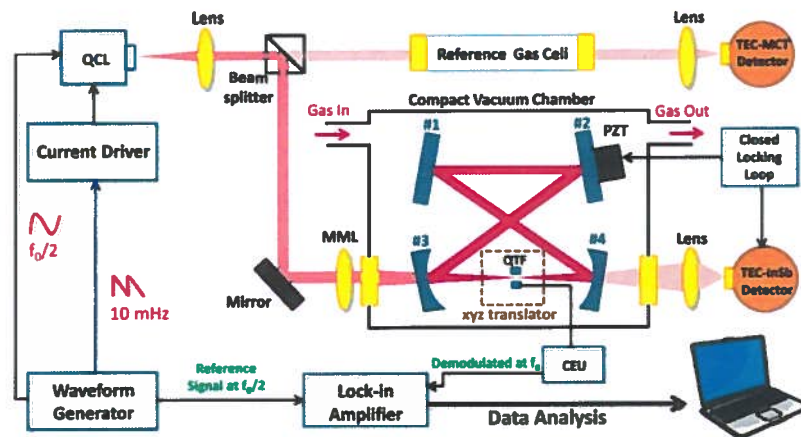
Performance of mid-infrared gas sensing techniques



NEEA for categories of gas detection techniques as a function of optical path-length. Key: BB-CEAS— broadband cavity-enhanced spectroscopy, CRDS—cavity ring-down spectroscopy, OA-ICOS—off-axis integrated cavity output spectroscopy, OF-CEAS—optical feedback cavity-enhanced absorption spectroscopy, NICE-OHMS—noise-immune cavity-enhanced optical heterodyne spectroscopy, PAS photoacoustic spectroscopy, QEPAS—Quartz-enhanced photoacoustic spectroscopy.

P. Fatimisco, G. Scamarcio, F.K. Tittel & V. Spagnolo, "Quartz-enhanced photoacoustic spectroscopy: a review", *Sensors*, 14, 6165-6206, (2014)

Proposed Intracavity-QEPAS (I-QEPAS) Sensor System



Optical power build up cavity can provide:

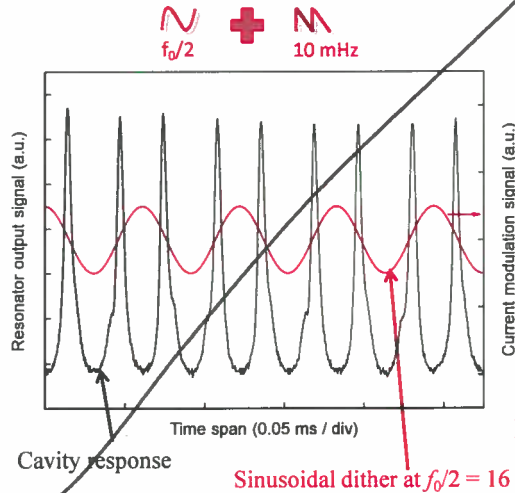
- RT CW DFB QCL, $\lambda=4.33$ microns
- Low noise current driver \rightarrow narrow QC laser linewidth ~ 1 MHz
- Bow-tie cavity \rightarrow 4 high reflectivity mirrors, $R=99.9\%$
- Electronic Control Loop + PZT motor lock to cavity resonant frequency to QCL frequency

P. Fatimisco, G. Scamarcio, F.K. Tittel & V. Spagnolo, "Quartz-enhanced photoacoustic spectroscopy: a review", *Sensors*, 14, 6165-6206 (2014)

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Optical Properties of I-QEPAS Bow-Tie Cavity

- Cavity response:
- Voltage ramp + modulation dither applied to OCL
 - under vacuum
 - locking loop ON



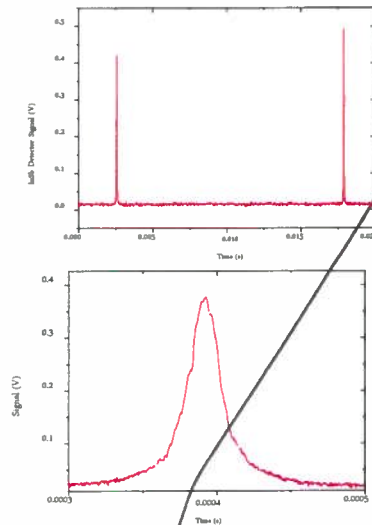
- The close-locked loop acts on the PZT tuning the cavity length
- It was not fast enough to follow the fast dither at $f_0/2 = 16$ kHz
- It maintains the optical cavity resonant with the laser frequency at the center of the fast dither



Mechanical chopper at f_0

Optical Properties of I-QEPAS Bow-Tie Cavity (cont.)

- Cavity response:
- Voltage ramp + modulation dither applied to OCL
 - under vacuum
 - locking loop ON



Cavity length $L = 174$ mm

$$\text{FSR} = c/L = 1.725 \text{ GHz}$$

$$\Delta\nu(\text{FWHM}) = 1.15 \text{ MHz}$$

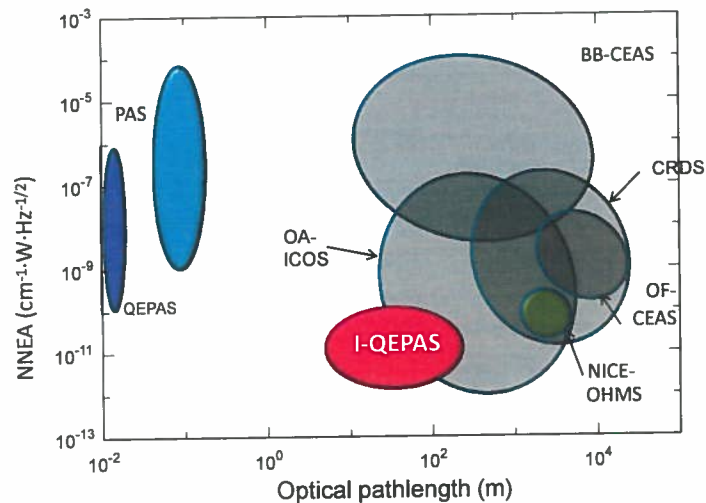
$$F = \text{FSR} / \Delta\nu = 1505$$

$$G = F/\pi = 480$$

Mode matching 50%

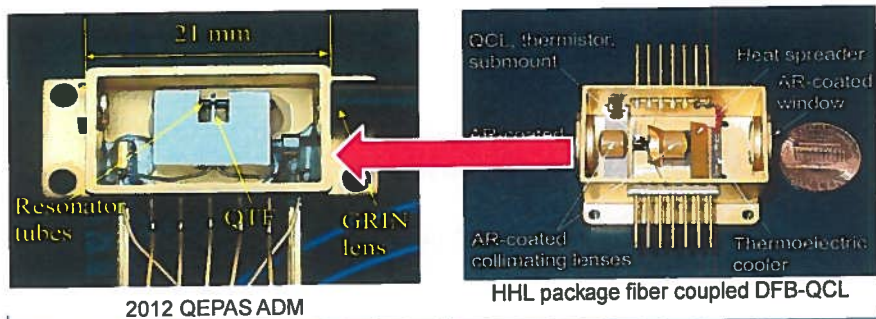
Intracavity optical power enhancement factor = 240

Comparison of I-QEPAS with Other Trace Gas Sensing Techniques



P. Patimisco, G. Scamarcio, F.K. Tittel & V. Spagnolo, "Quartz-enhanced photoacoustic spectroscopy: a review", *Sensors*, 14, 6165-6206 (2014)

Potential Integration of a CW DFB-QCL and QEPAS Absorption Detection Module



A. Lyakh, et al "1.6 W high wall plug efficiency, continuous-wave room temperature quantum cascade laser emitting at 4.6 μm ", *Appl. Phys. Lett.* 92, 111110 (2008)

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Summary and Conclusions

- Development of robust, compact, sensitive, selective mid-infrared trace gas sensor technology based on room temperature, continuous wave DFB laser diodes and high performance QCLs for [medical diagnostics](#) and environmental monitoring.
- Interband cascade and quantum cascade lasers were used in [TDLAS, PAS and QEPAS based sensor platforms](#)
- Six target trace gas species were detected with a 1 sec sampling time:
 - C₂H₆: ~3.36 μm , detection sensitivity of 740 pptv using TDLAS
 - NH₃: ~10.4 μm , detection sensitivity of ~1 ppbv (200 sec averaging time)
 - NO: ~5.26 μm , detection limit of 3 ppbv
 - CO: ~4.61 μm , minimum detection limit of 2 ppbv
 - SO₂: ~7.24 μm , detection limit of 100 ppbv
 - CH₄ and N₂O: ~7.28 μm , detection limits of 13 and 6 ppbv, respectively
 - H₂O₂: ~7.73 μm , detection limit of 75 ppb
 - C₂H₆O:
- New target analytes: CH₂O, C₆H₆, and C₃H₈



Merits of QEPAS based Trace Gas Detection

- Very small sensing module and sample volume (a few mm³ to ~2cm²)
- Extremely low dissipative losses
- Optical detector is not required
- Wide dynamic range
- Frequency and spatial selectivity of acoustic signals
- Rugged transducer – quartz monocrystal; can operate in a wide range of pressures and temperatures
- Immune to environmental acoustic noise, sensitivity is limited by the fundamental thermal TF noise: $k_B T$ energy in the TF symmetric mode
- Absence of low-frequency noise: SNR scales as \sqrt{t} , up to $t=3$ hours as experimentally verified

QEPAS: some challenges

- Cost of Spectrophone assembly
- Sensitivity scales with laser power
- Effect of H₂O
- Responsivity depends on the speed of sound and molecular energy transfer processes
- Cross sensitivity issues

