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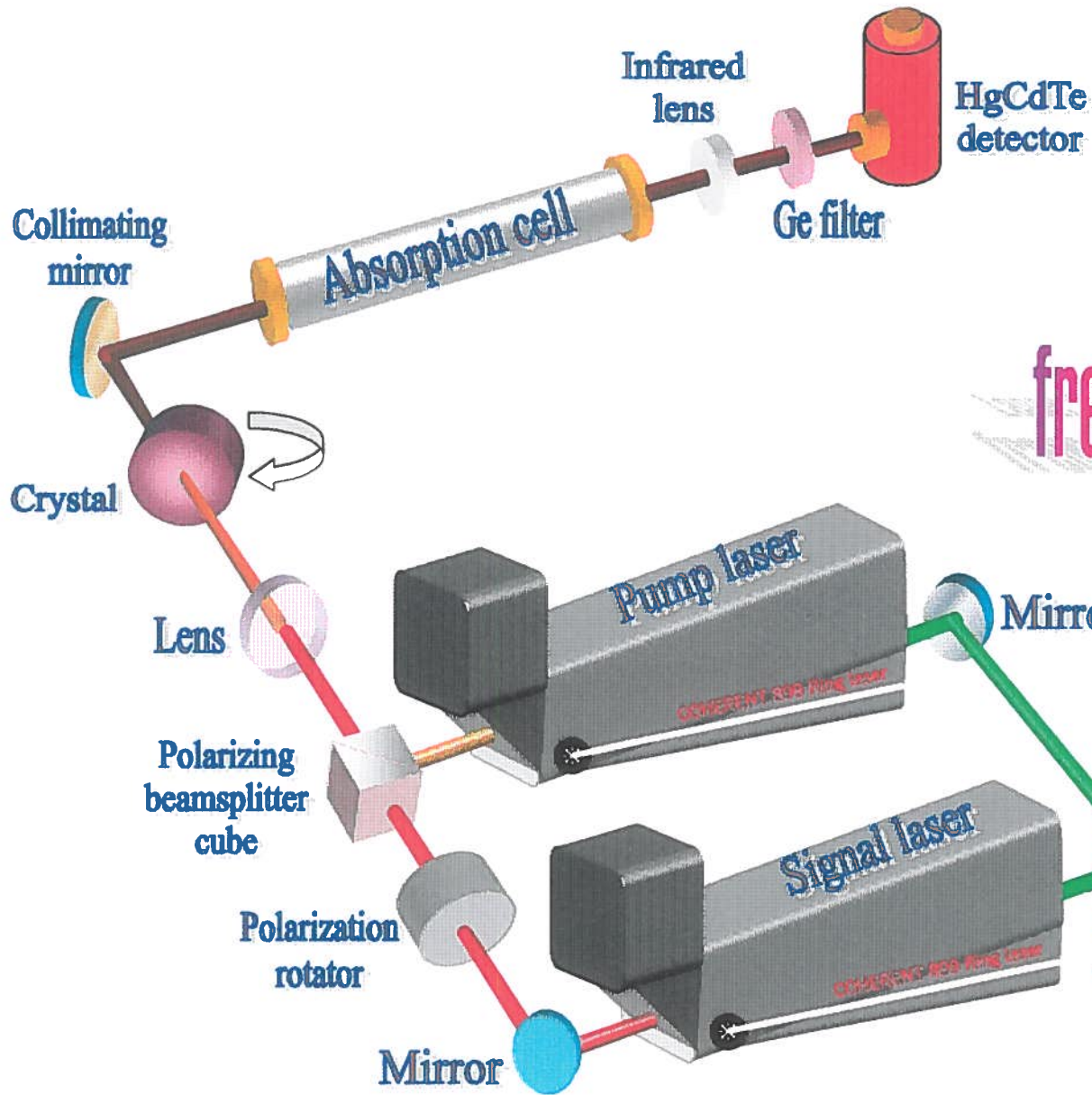
OUTLINE

- ⇒ *Motivation*
- ⇒ *Infrared spectrometer based on laser difference-frequency generation*
- ⇒ *Spectroscopic performances*
- ⇒ *Trace gas measurements by laser absorption spectroscopy*
- ⇒ *Conclusion & outlook*

Motivation

⇒ *Spectroscopic investigation of molecular line parameters ;*

⇒ *Environmental monitoring of trace gas, in particular volatile organic compounds, by laser absorption spectroscopy*



Laser difference

frequency spectrometer

Laser difference-frequency spectrometer

Spectral coverage

3.5-6.5 μm (DFG in QPM-PPRTA and AgGaS₂)

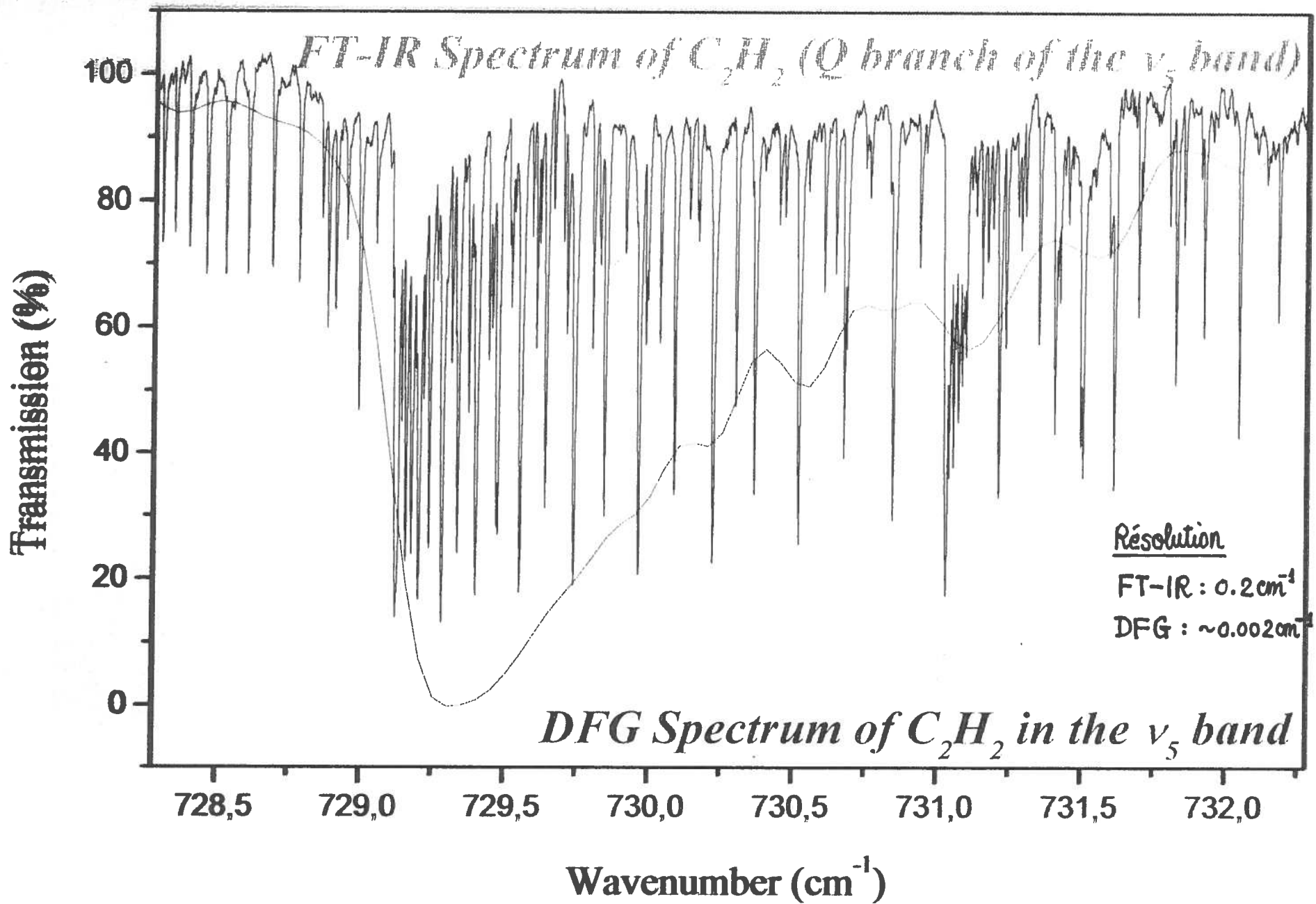
8-19 μm (DFG in GaSe)

cw power conversion

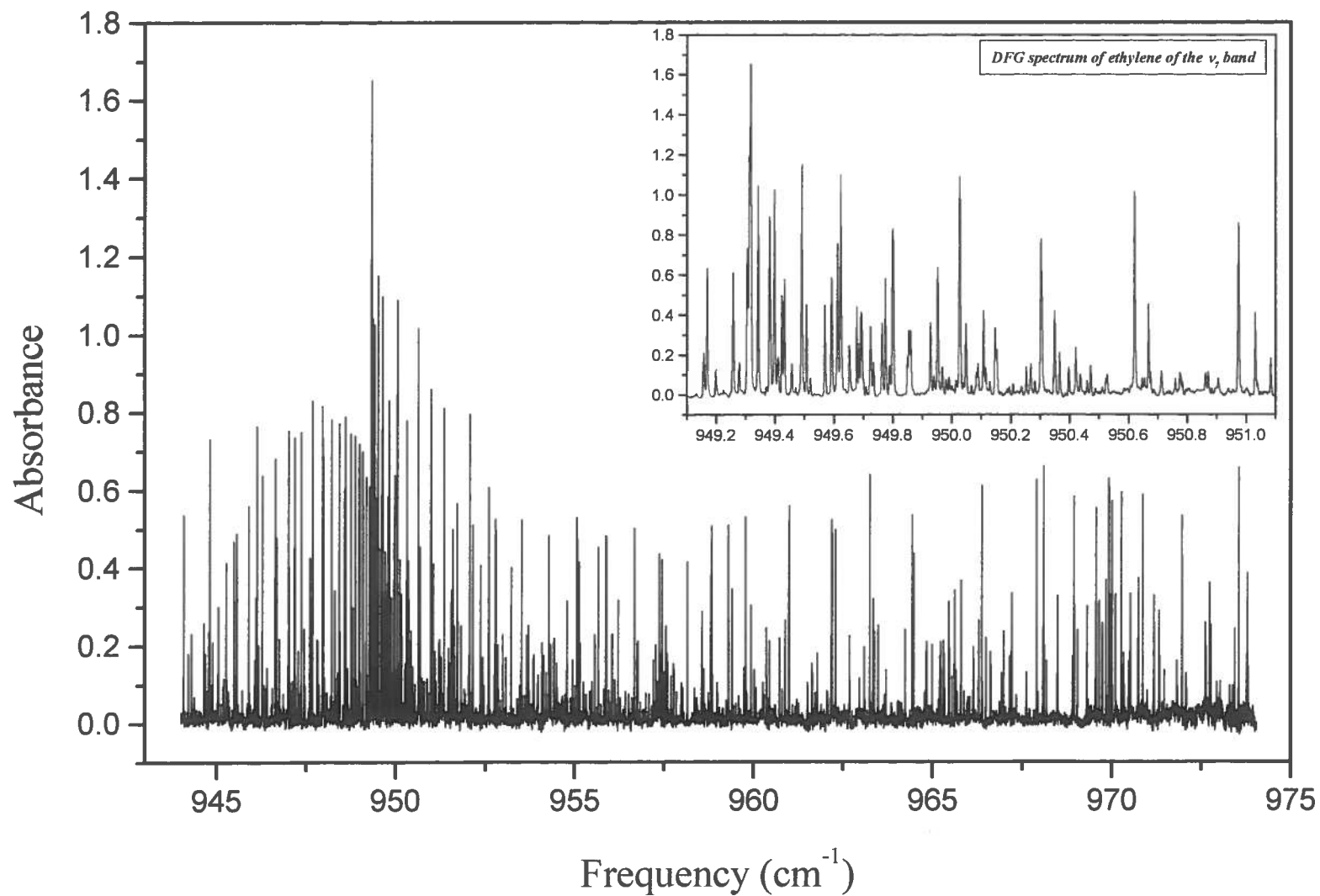
10-500 $\mu\text{W}/(\text{W}^2 \cdot \text{cm})$

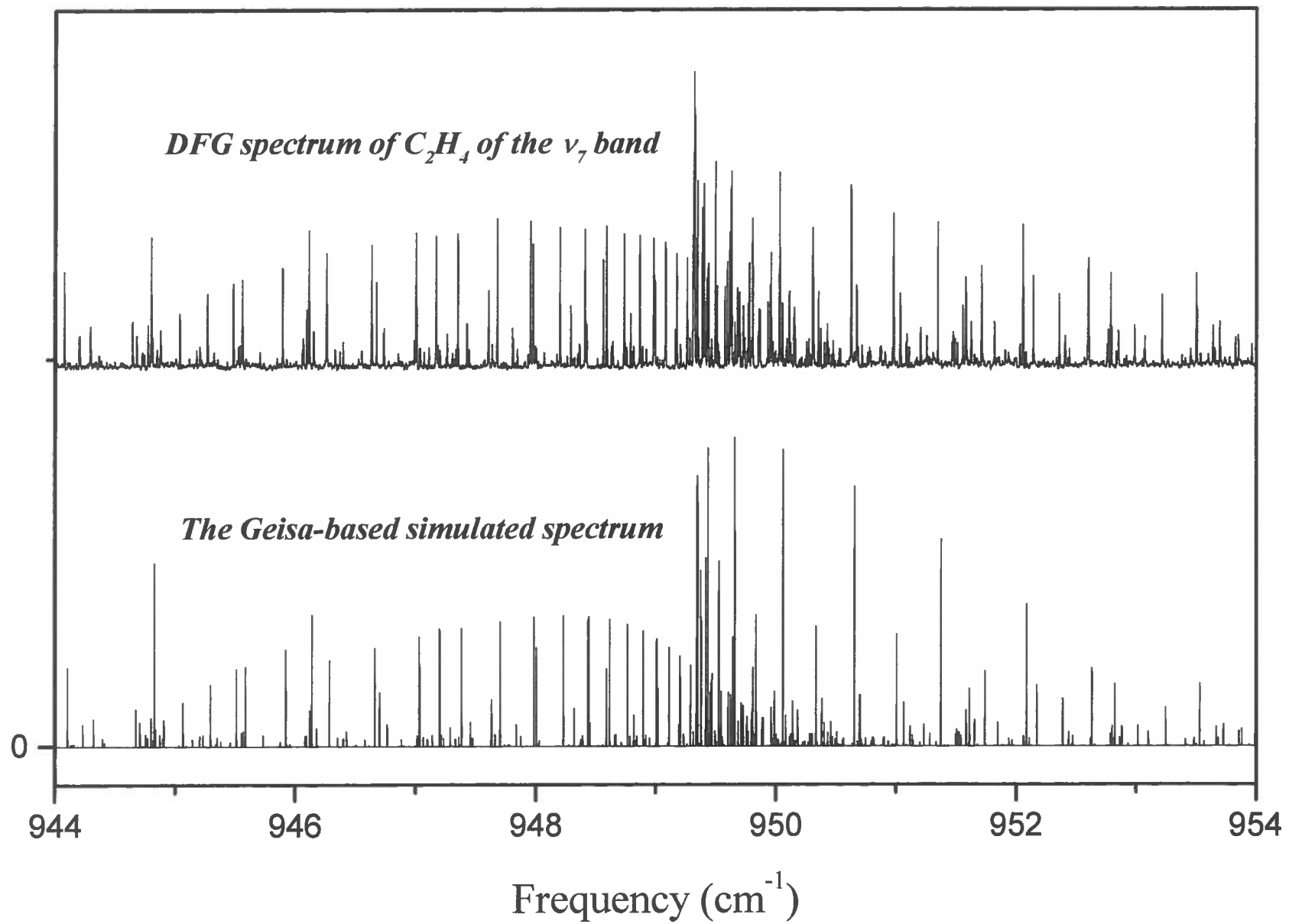
Spectral purity of the DFG source

< 1 MHz ($\sim 10^{-4} \text{ cm}^{-1}$)

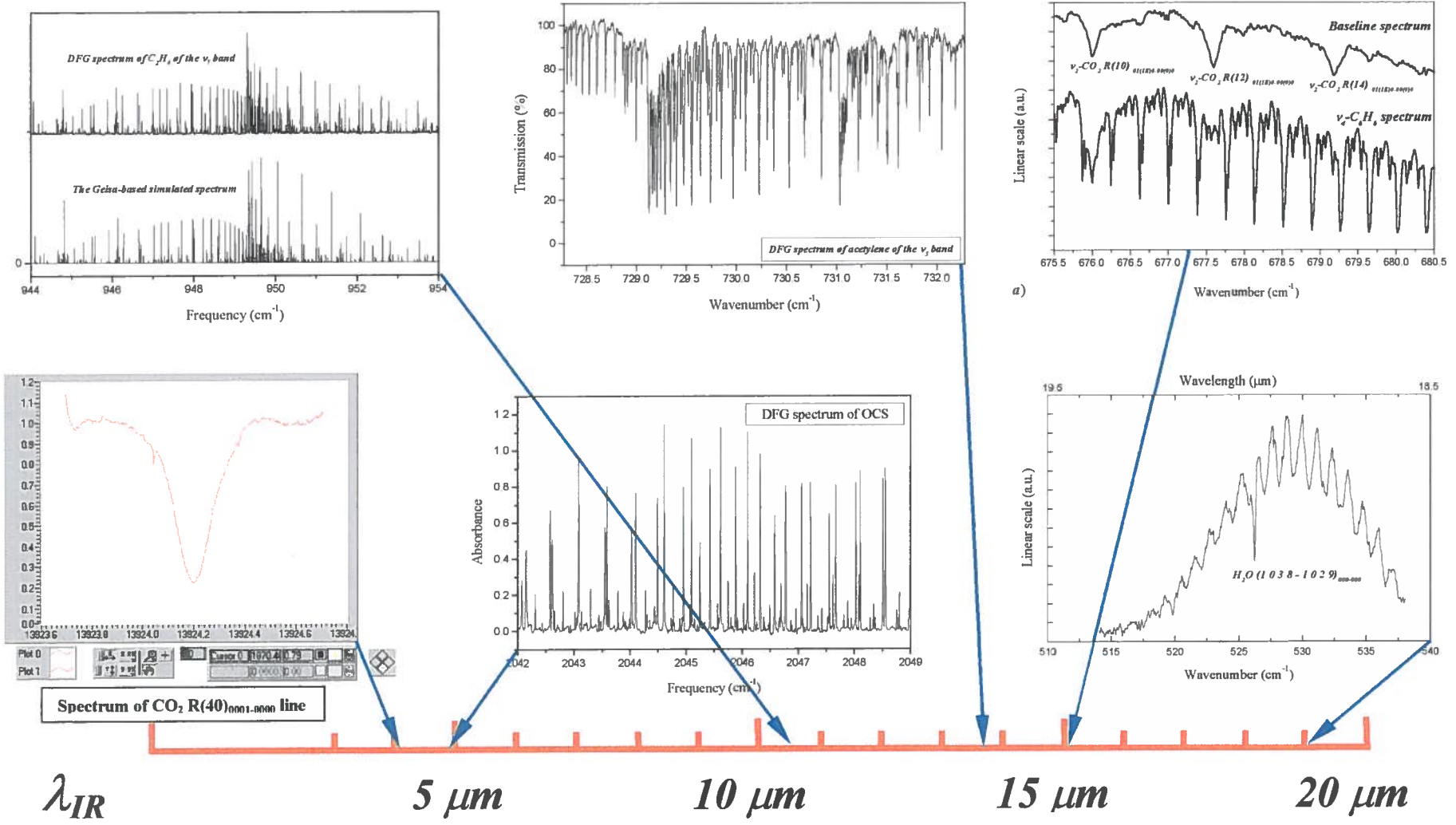


High-resolution DFG spectrum of ν_7 C_2H_4 (@ 3.1 mbar & $L=10$ cm)

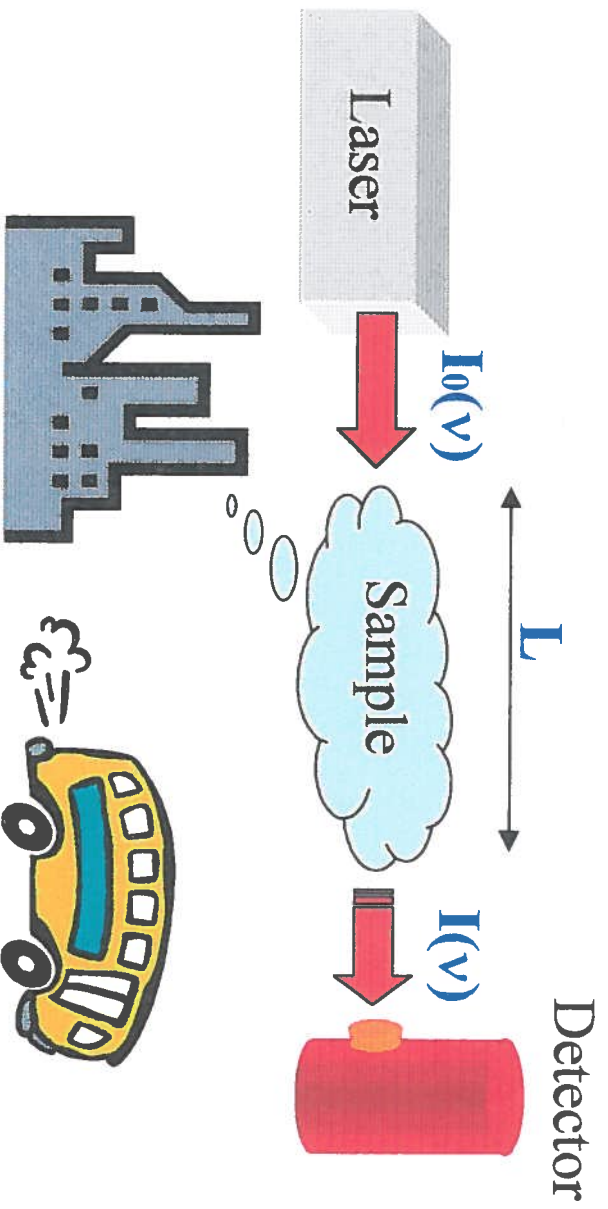




High-resolution Laser DFG Absorption Spectra



Trace Gas Detection Using Laser Absorption Spectroscopy



The Beer's law:

$$I(\nu) = I_0(\nu) \exp(-\sigma(S, \gamma, \nu) C L)$$

Then the concentration:

$$C = \ln(I_0/I)(\sigma(S, \gamma, \nu) L)$$

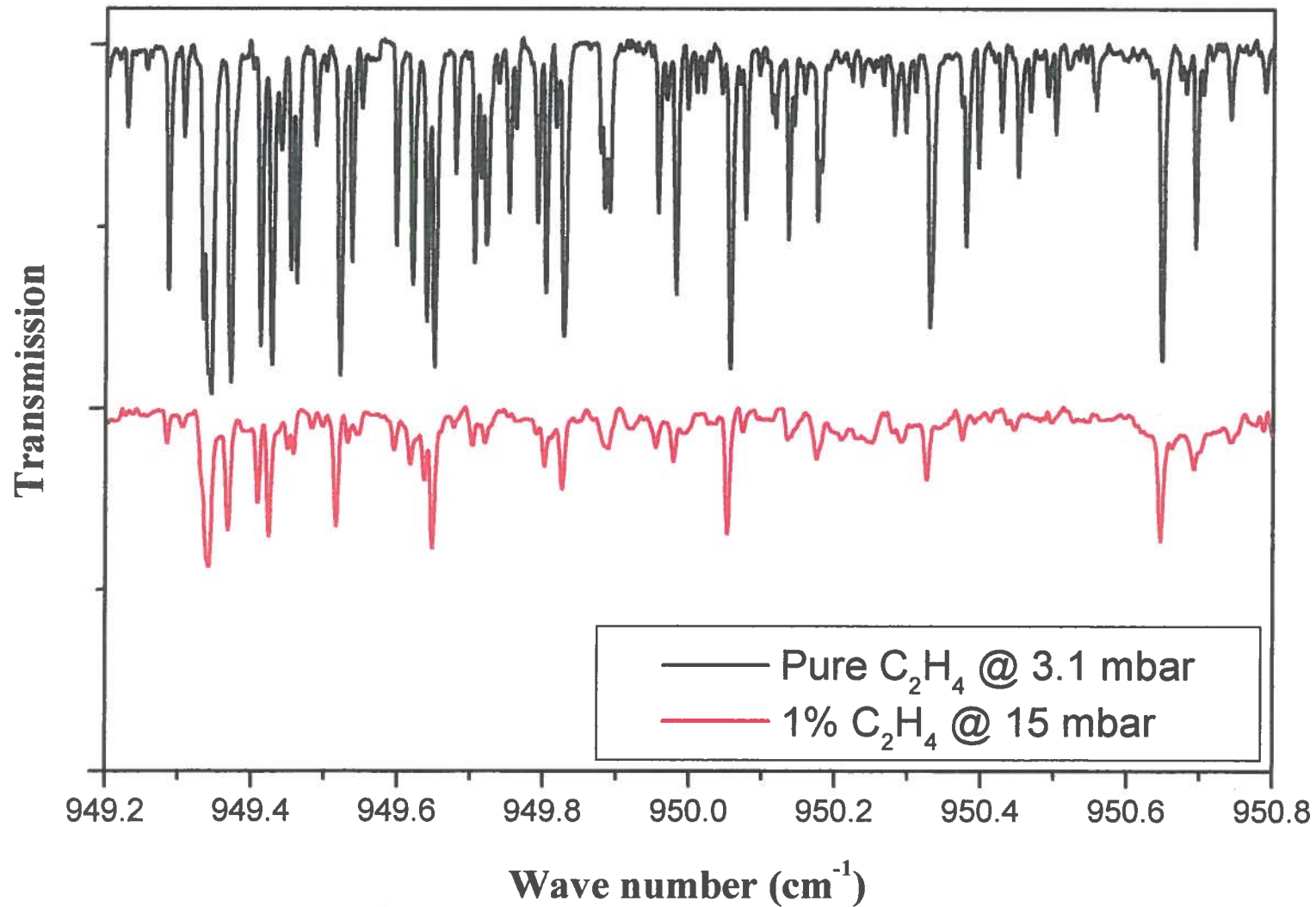
$I(\nu)$: the intensity after traversing a sample thickness L

$I_0(\nu)$: the incident intensity

$\sigma(S, \gamma, \nu)$: the absorption cross-section

C : the concentration of the absorbing species

High-resolution DFG spectra of ethylene of the ν_7 band



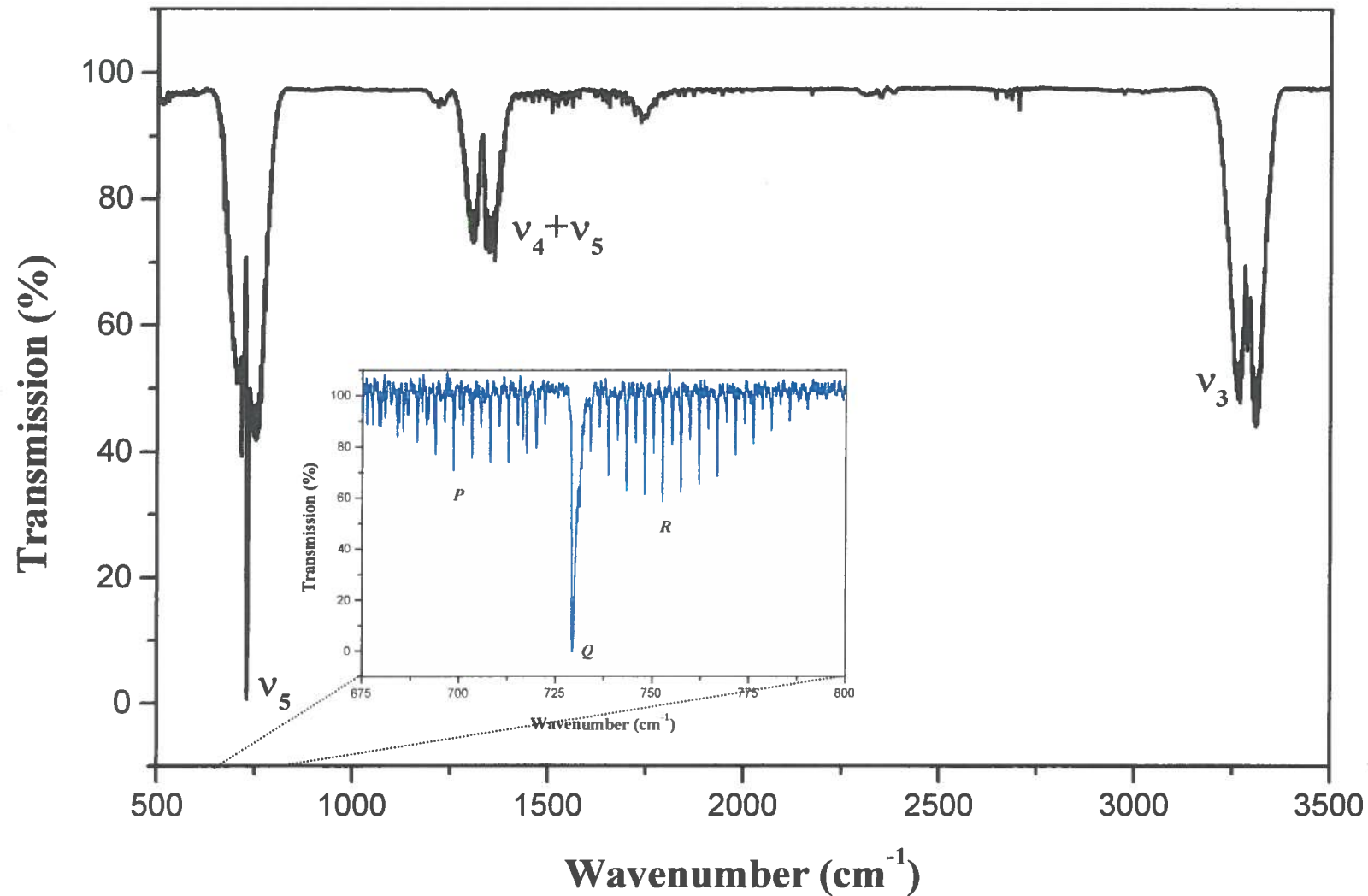
Absorption line selection

- ⇒ **Strong** for highly sensitive detection
- ⇒ **Isolated** for highly selective measurement



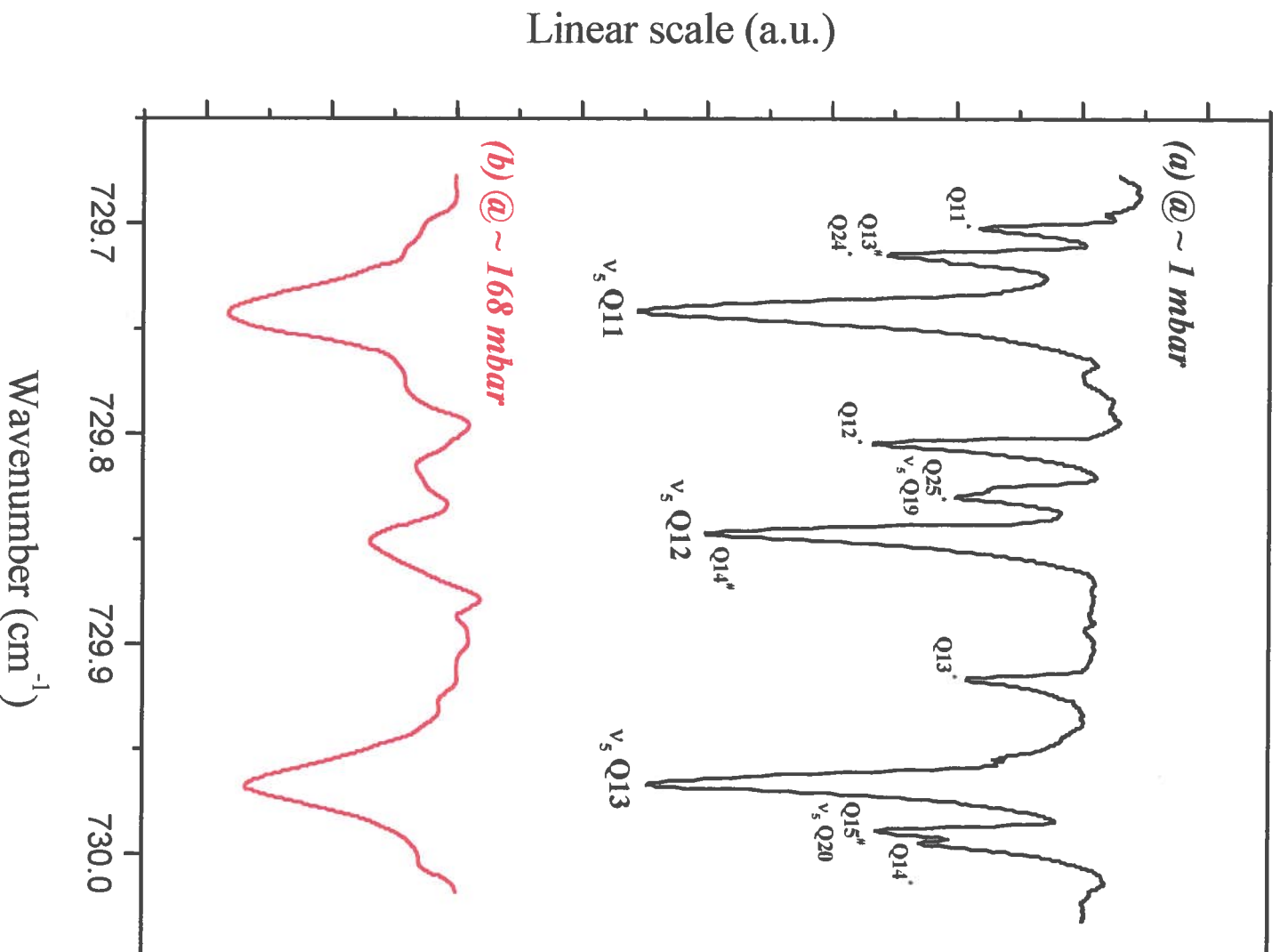
Selecting a strong absorption line for high sensitivity which at the same time should be isolated from interfering lines due to other gas species or from the same species.

FT-IR absorption spectrum of C_2H_2 @ ~ 40 mbar ($R=2\text{ cm}^{-1}$ with 64 scans)
The inset spectrum was obtained @ ~ 3 mbar ($R=0.2\text{ cm}^{-1}$ with 16 scans)

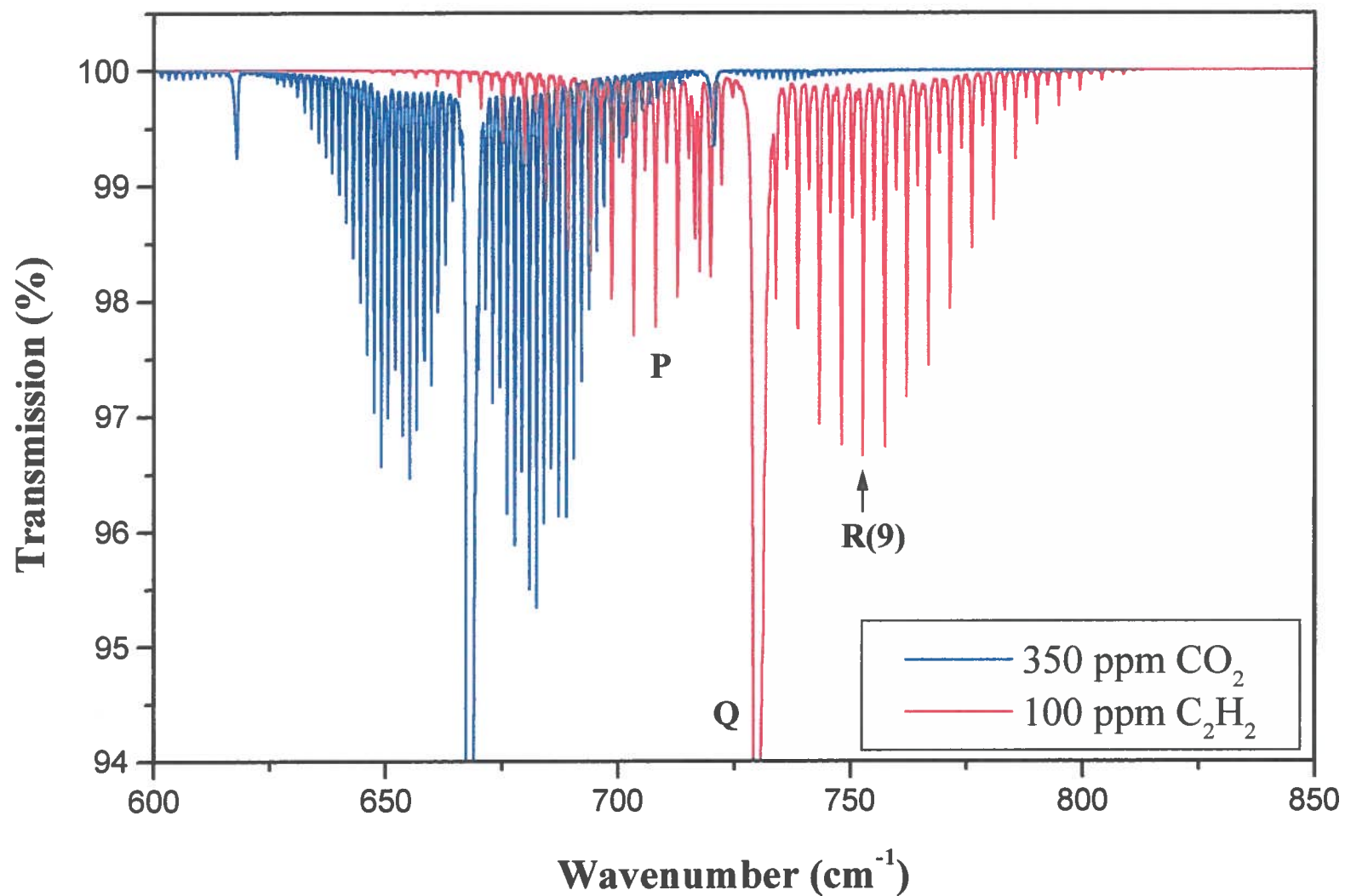


77 ppm C₂H₂ measurement

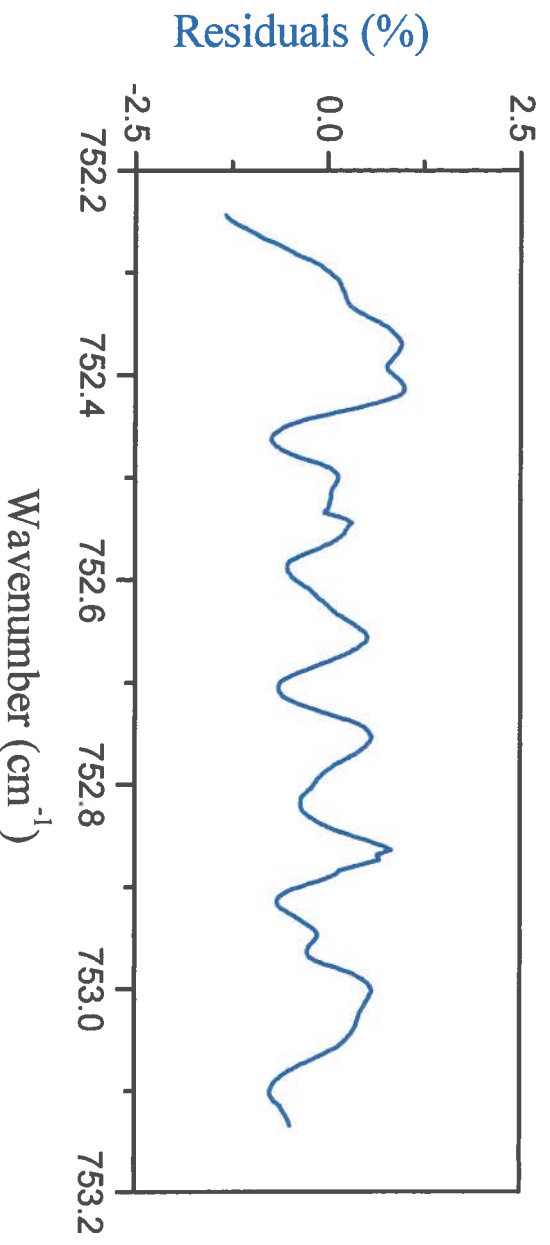
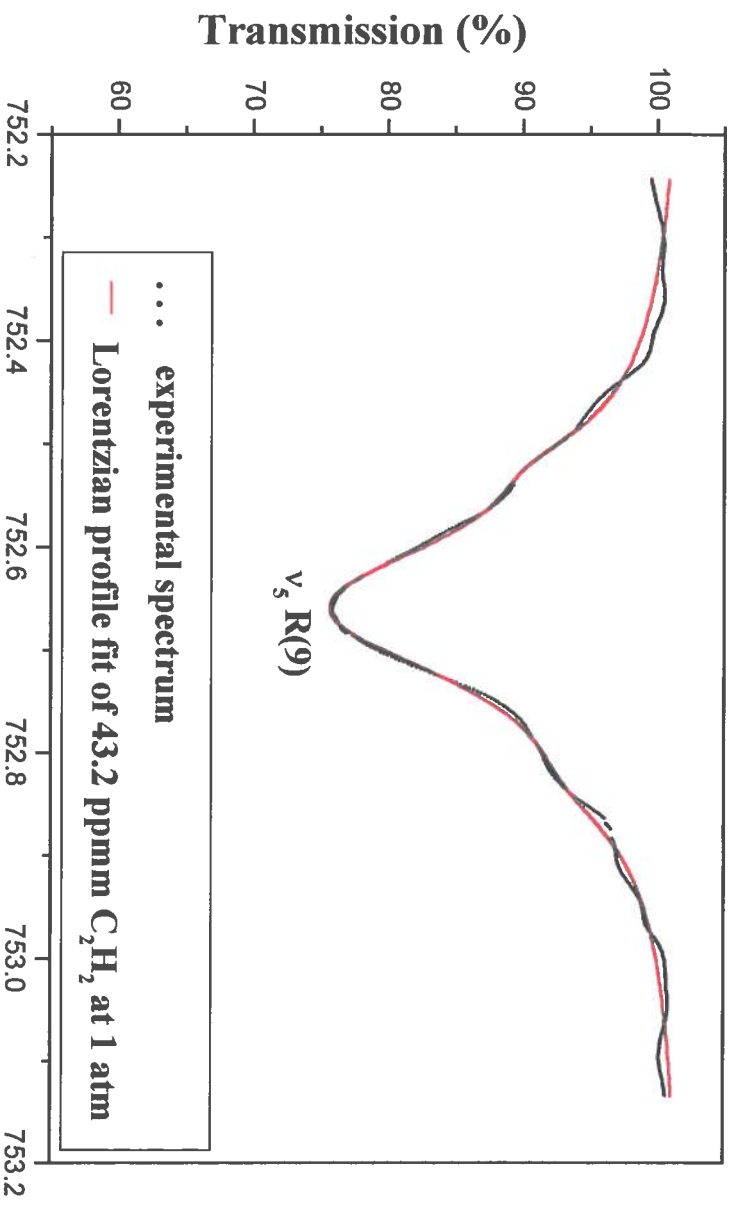
using ν_5 -Q11 and ν_5 -Q13 lines **(b)**



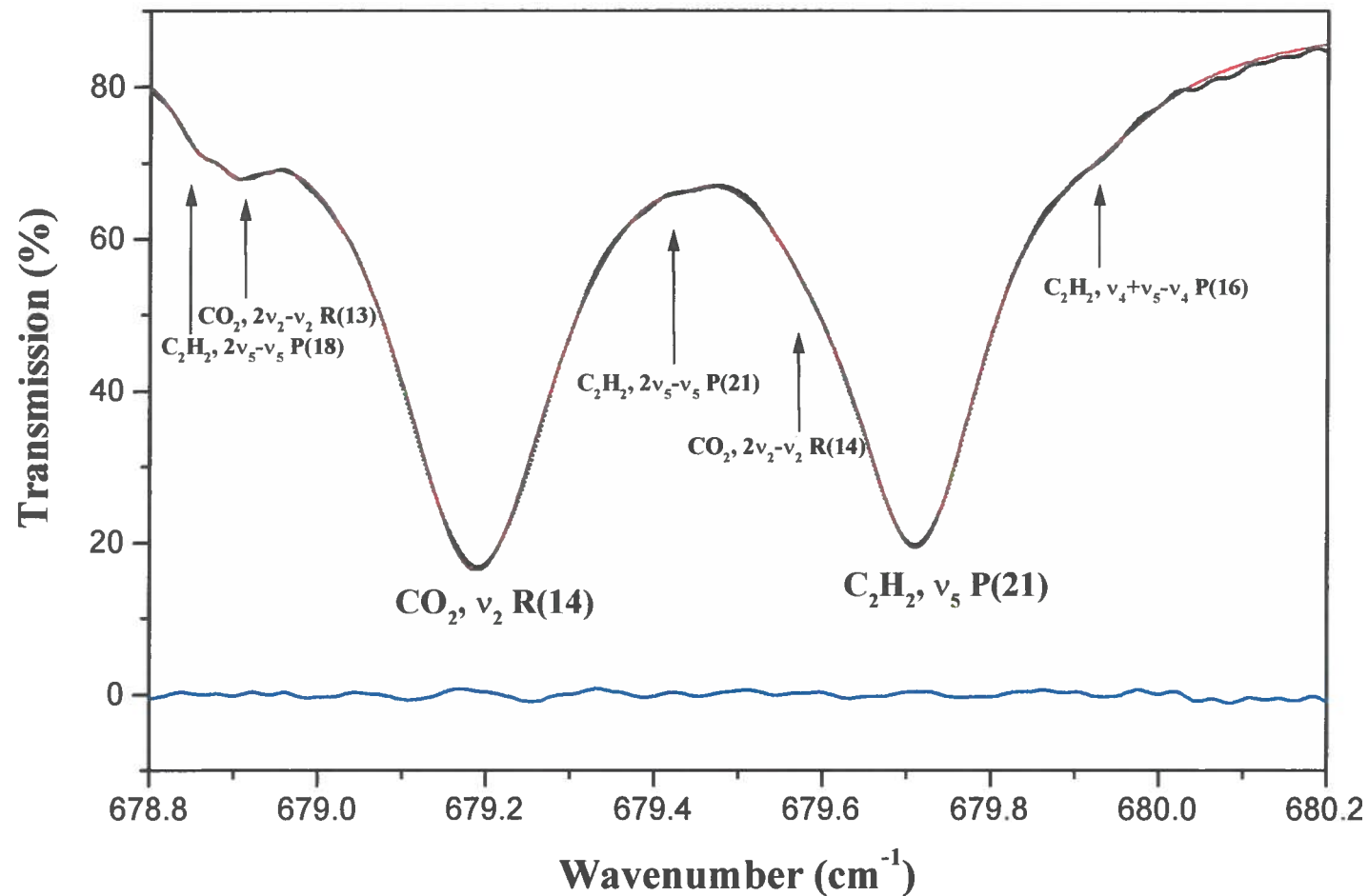
Simulated spectra of CO₂ and C₂H₂ @ atmospheric pressure



Spectrum of the ν_5 R(9) transition of C_2H_2 trace at atmospheric pressure



Spectrum of calibrated 1% CO₂ and C₂H₂ mixture around 679.5 cm⁻¹ @ atmospheric pressure



Comparison of C₂H₂ trace detection using the P(21), Q(11) and R(9) lines of the ν_5 band

Rotational assignment (ν_5 band)	Transition frequency (cm ⁻¹)	Line strength (cm/mol.)	Absorption coefficient (ppm ⁻¹ m ⁻¹)	Measured concentration-path (ppm m)	Pressure (Torr)	MDC* (ppb)
P (21)	679.7095	1.32^E-19	6.65^E-4	96.1	760	84.7
Q (11)	729.7380	1.08^E-18	2.68^E-2	7.7	126	2.4
R (9)	752.6589	6.82^E-19	2.88^E-3	43.2	760	20.7

* MDC is the 3 σ -detection-limited minimum detectable concentration in ppb (part per billion, 10⁻⁹ atm.) using an optical path of 100-m.

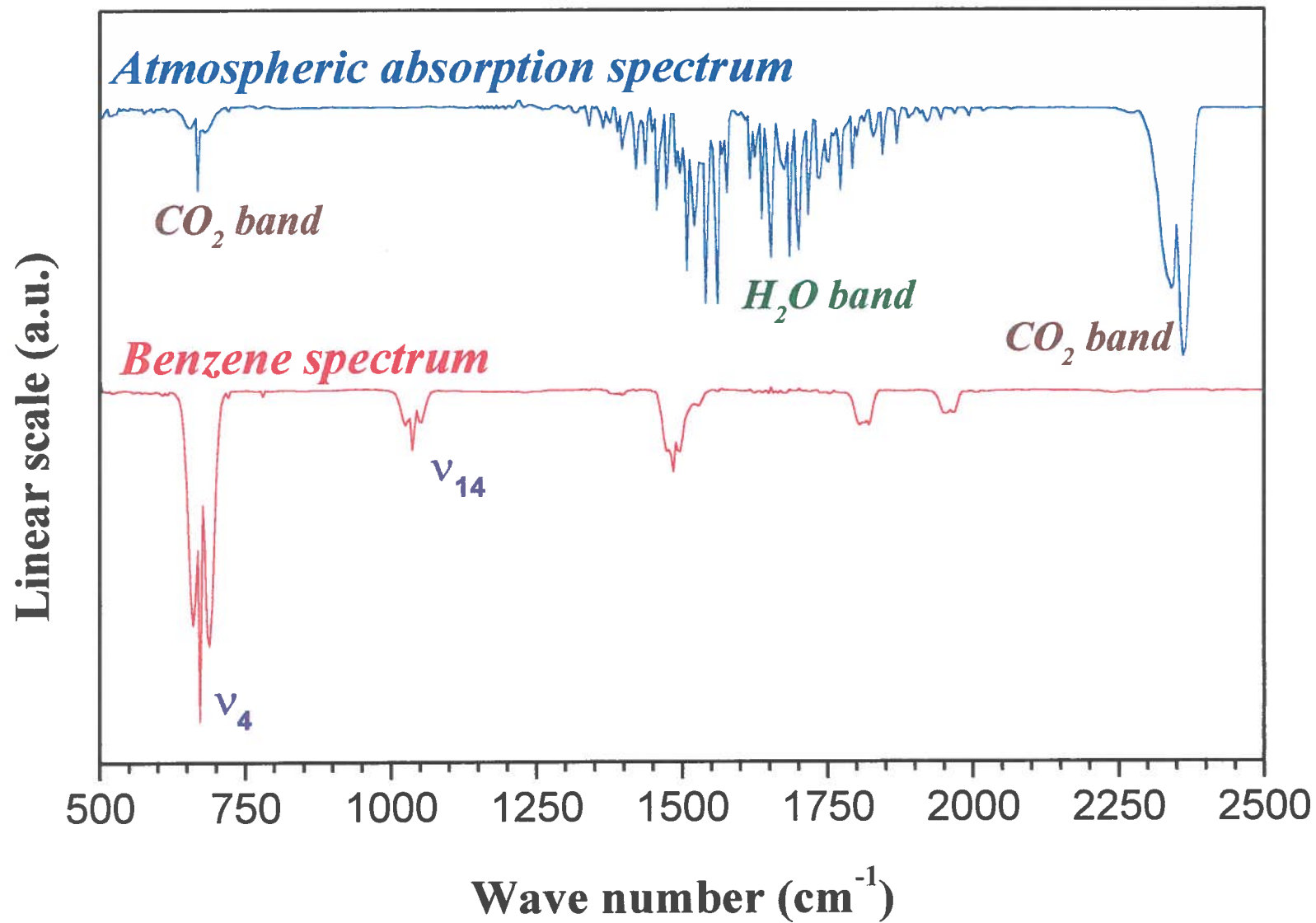
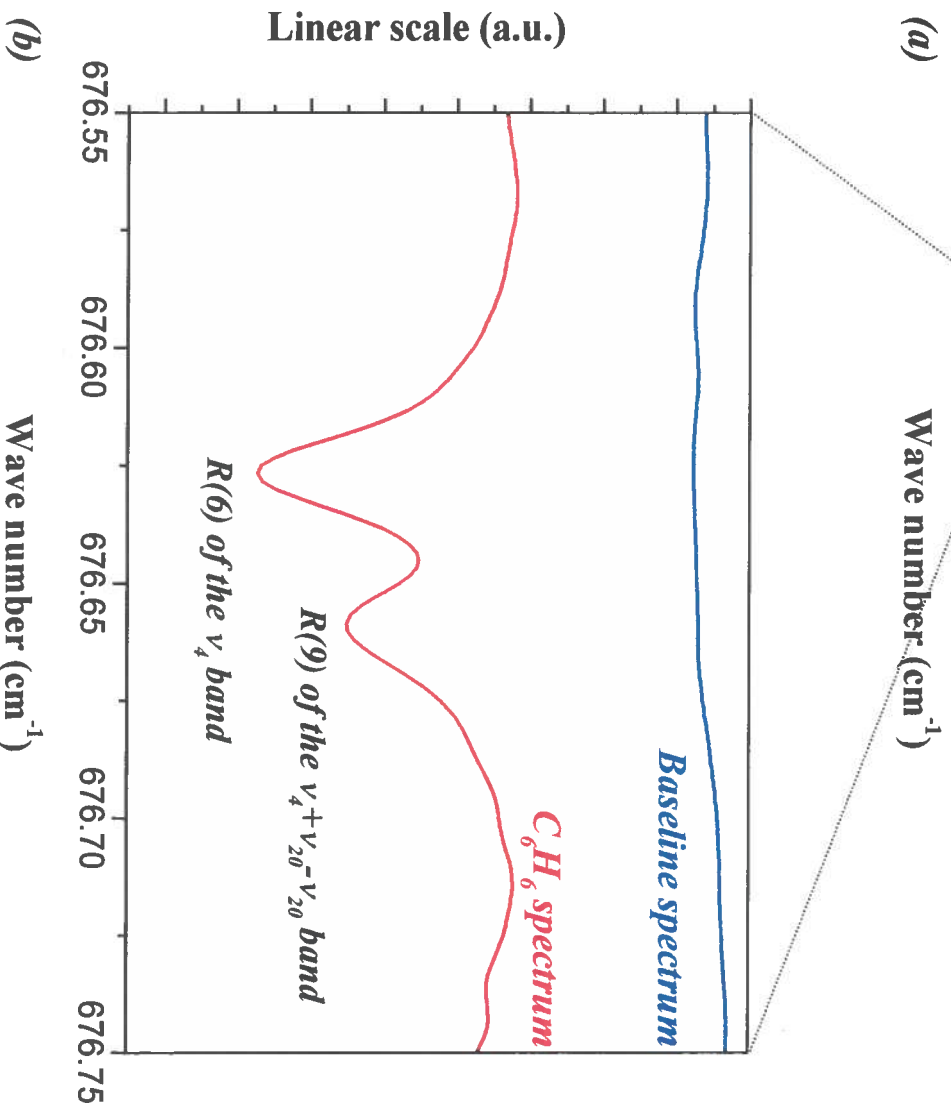
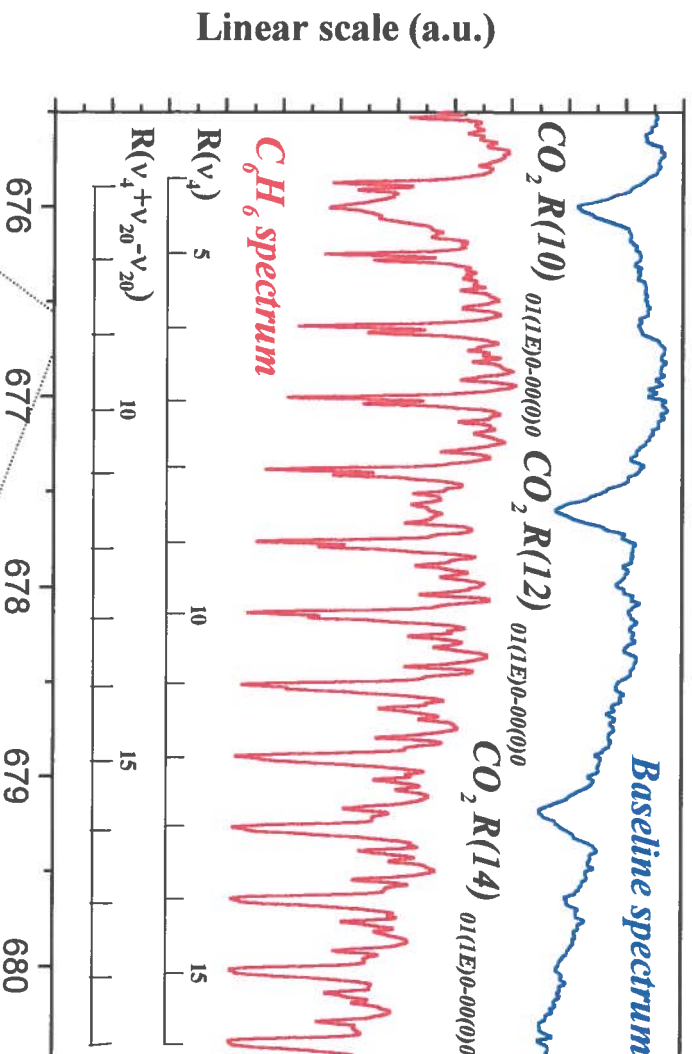


FIGURE 1

DFG spectrum of C_6H_6 in the ν_4 band



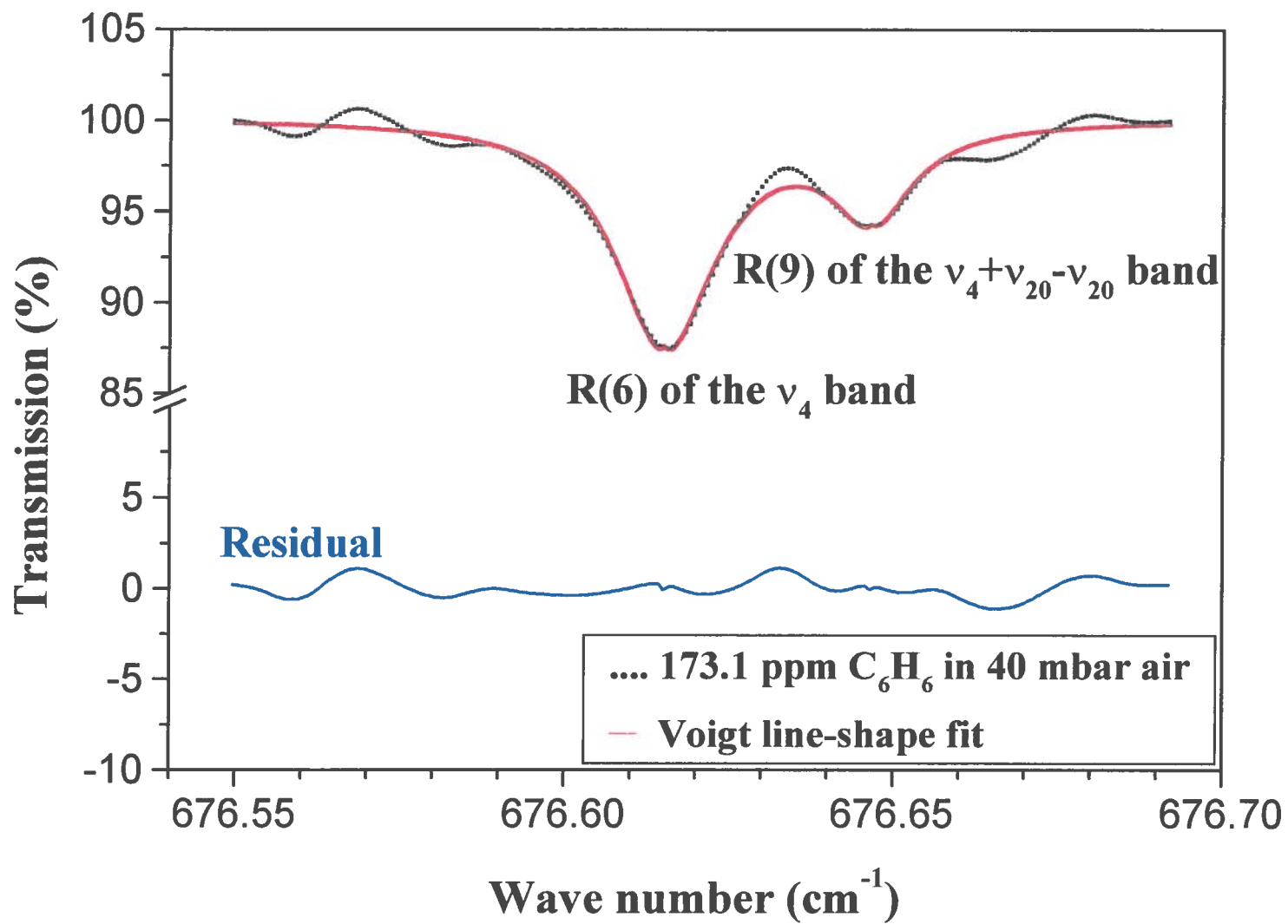
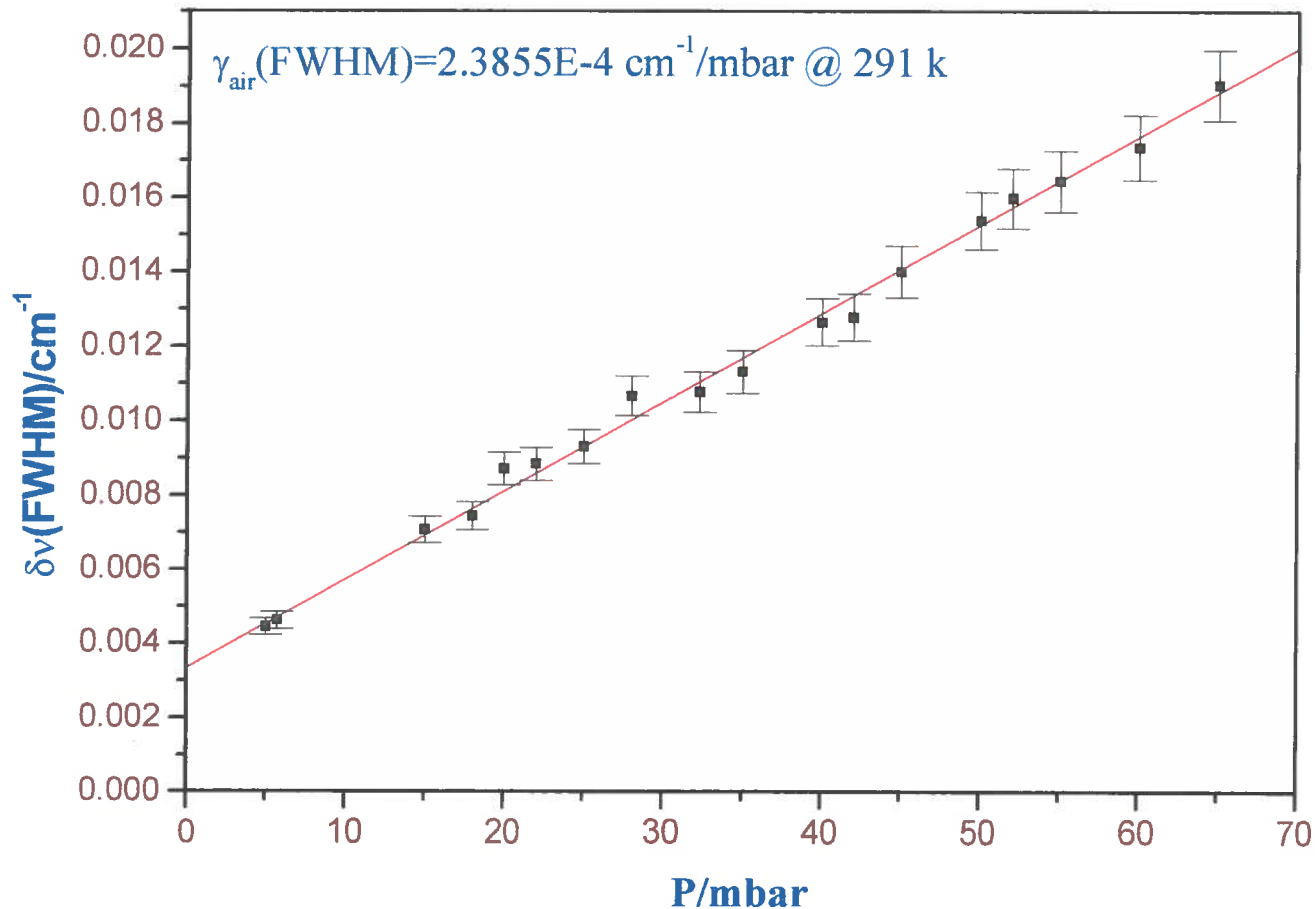


FIGURE 5

Pressure dependence of the linewidth of the C_6H_6 line of the $n4 R(6)$ branch @ 676.6258 cm^{-1}

Preliminary result: $\gamma_{\text{air}}(\text{FWHM})=0.23855\text{ cm}^{-1}/\text{atm @ 291 k}$



SUMMARY

Laser absorption spectroscopy for environmental applications

- ⇒ High sensitivity and selectivity in the fundamental infrared region.
- ⇒ Measurements of trace amounts of various hydrocarbons: acetylene (ν_5 band), benzene (ν_4 band) and ethylene (ν_7 band) with a minimum detectable concentration of <ppm.



OUTLOOK

In situ and real time BTEX detection by means of diode laser based DFG employing QPM-GaAs crystal for the 8-16 μm region.