

OUTLINE

Mid-infrared laser based trace gas sensor technologies: recent advances and applications-I

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GE Global Research

Houston, TX

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Wide Range of Trace Gas Sensing Applications

- Urban and Industrial Emission Measurements
 - Industrial Plants
 - Combustion Sources and Processes (e.g. fire detection)
 - Automobile, Truck, Aircraft and Marine Emissions

Rural Emission Measurements

• Agriculture & Forestry, Livestock

Environmental Monitoring

- Atmospheric Chemistry (e.g isotopologues, climate modeling,...)
- Volcanic Emissions

Chemical Analysis and Industrial Process Control

- Petrochemical, Semiconductor, Pharmaceutical, Metals Processing, Food & Beverage Industries, Nuclear Technology & Safeguards
- Spacecraft and Planetary Surface Monitoring
 - Crew Health Maintenance & Life Support
- Applications in Medical Diagnostics and the Life Sciences
- Technologies for Law Enforcement, Defense and Security
- Fundamental Science and Photochemistry



"Curiosity" Landed on Mars on August 6, 2012



Laser-Based Trace Gas Sensing Techniques

- Optimum Molecular Absorbing Transition
 - Overtone or Combination Bands (NIR)
 - Fundamental Absorption Bands (Mid-IR)
- Long Optical Pathlength
 - Multipass Absorption Gas Cell (e.g., White, Herriot, Chernin, Aeris Technologies, and Circular Cylindrical Multipass Cell
 - Cavity Enhanced and Cavity Ringdown Spectroscopy
 - Open Path Monitoring (with retro-reflector or back scattering from topographic target): Standoff and Remote Detection
 - Fiberoptic & Wave-guide Evanescent Wave Spectroscopy
- Spectroscopic Detection Schemes
 - Frequency or Wavelength Modulation
 - Balanced Detection
 - Zero-air Subtraction



- Faraday Rotation Spectroscopy (limited to paramagnetic chemical species)
- Differential Optical Dispersion Spectroscopy (DODiS)
- Noise Immune Cavity Enhanced-Optical Heterodyne Molecular Spectroscopy (NICE-OHMS)
- Frequency Comb Spectroscopy
- Laser Induced Breakdown Spectroscopy (LIBS)



HITRAN Simulated Mid-Infrared Molecular Absorption Spectra



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Selection of Absorption lines in the mid-IR Spectral Range (3 – 5 µm)

- $2.5 \ \mu m < l < 5 \ \mu m (4000 \ cm^{-1} 1900 \ cm^{-1})$
- Access to molecular fundamental rotational-vibrational states
- Atmospheric window (3.5-4.8 µm)



Mid-IR Source Requirements for Laser Spectroscopy

REQUIREMENTS	IR LASER SOURCE
Sensitivity (% to pptv)	Optimum Wavelength and Power
Selectivity (Spectral Resolution) or Specificity	Stable Single Mode Operation and Narrow Linewidth
Multi-gas Components, Multiple Absorption Lines, and Broadband Absorbers	Mode Hop-Free Wavelength Tunability
Directionality or Cavity Mode Matching	Beam Quality
Rapid Data Acquisition	Fast Time Response Time
Room Temperature Operation	High Wall Plug Efficiency, No Cryogenics or Cooling Water
Field Deployable in Harsh Environments	Compact and Robust

Key Characteristics of Mid-IR QCL & ICL Sources – March 2016

Band – structure engineered devices

Emission wavelength is determined by layer thickness – MBE or MOCVD; QCLs operate in the 3 to 24 μ m spectral region and ICLs can cover the 3 to $5 \,\mu\text{m}$ spectral range.

- Compact, reliable, stable, long lived, and commercially available
- Fabry-Perot (FP), single mode (DFB) and multi-wavelength devices

<u>Wide spectral tuning ranges in the mid-IR</u>

- 1.5 cm⁻¹ using injection current control for DFB devices
- 10-20 cm⁻¹ using temperature control for DFB devices
- ~100 cm⁻¹ using current and temperature control for QCLs DFB Array
- ~ 525 cm⁻¹ (22% of c.w.) using an external grating element and FP chips with heterogeneous cascade active region design; also QCL DFB array & Optical Frequency Combs (OFCs) > 100 to $<450 \text{ cm}^{-1}$ with kHz to subkHz resolution and a comb spacing of > 10 GHz

Narrow spectral linewidths

- CW: 0.1 3 MHz & <10kHz with frequency stabilization Pulsed: ~ 300 MHz

High pulsed and CW powers of QCLs and ICLs at **TEC/RT temperatures**

- Room temperature pulsed peak power of ~203 W with 10% wall plug efficiency for QCLs
- ~1 W CW DFB TEC/RT QCL; wall plug efficiency 23% at 4.6 µm
- ~ 5-10mW CW DFB ICLs at TEC/RT in the 3 to 4 μ m spectral range



4 mm





ALPES Low-dissipation DFB devices at 4.5 μ m



08.09.2014 21

A miniaturized External Cavity QCL with MEMS Technology





wavenumber [cm⁻¹]



© Fraunhofer

Methane Detection

Methane is one of the major atmospheric greenhouse gases contributing to global warming and climate change.

ARPA-E Monitor Aeris Rechnologies, Inc and Maxion Technologies, Inc (Thorlabs, Inc

- Global warming potential (GWP) of 25 compared to GWP of 1 for CO₂ for a 100-year period)
- Short lifetime in the atmosphere (~12 yrs) compared to CO_2 and N_2O
- Atmospheric background concentration: ~1.8 ppm





Typical Texas Oil & Gas Production Site near Houston



The result of a sequence of four fracking injections obtained by directional drilling which creates horizontal production in the target stratum is depicted in a figure as published in Physics Today 2016. A DOE-ARPA-E funded methane detection project at $3.33 \mu m$ was started in 2015. Texas located wellpad sites (Texas located well pad sites typically measure 10-30 m with a 1 m spatial resolution)





CH₄ and N₂O Measurements performed with a DFB-QCL based QEPAS Sensor installed in a Mobile Laboratory (operated by Aerodyne, Inc.)

QEPAS Sensor Platform



Comparison of proposed Rice CH₄ Sensor System and current commercially available CH₄ Sensor Platforms

	Rice	Picarro	ABB-LGR I	ABB-LGR II	Aerodyne
Opt. Path length and method	MIR TDLAS: ~ 9 m	NIR CRDS: >2000m	NIR OA-ICOS: > 1000m	NIR OA-ICOS: > 2000m	MIR TDLAS: 70- 100 m
Sensitivity/sec	< 5-10 ppb	1-2 ppb	5 ppb	2 ppb	<1 ppb
Accuracy (drift)	2 ppb stabilized	2 ppb	stabilized	2 ppb	2 ppb
Cell Volume, cc	60	30	500	2000	2000
Pump Size (10 sec flush time) Cavity Mirror	~ 1 lpm	~ 0.5 lpm	~ 11 lpm	~ 45 lpm	~ 45 lpm
Reflectance	98.5%-99%	>99.99%	>99.99%	>99.99%	>99.99%
Power Consumption	2-20 W	200 W	70 W	200 W	400 W
Weight	~ 2-4 kg	~ 20 kg	~ 15 kg	~ 40 kg	~ 40 kg
Cost	~ 20-25K USD	~ 40-50K USD	~ 25K USD	~ 40K USD	~ 100K USD

US Department of Energy Advanced Research Project Agency – Energy (ARPA-E), Methane Observation Networks with Innovative Technology to obtain Reductions (MONITOR)



Spectroscopy Fundamentals

Beer's Law:
$$\frac{I_t}{I_o} = \exp(-k_v L)$$

where: I_t is transmitted light intensity I_0 is incident light intensity k_v is absorption coefficient L is optical path

Multipass Gas Cell (MPGC):

The minimum detection limit can be improved by increasing the <u>effective optical path</u> without increasing the <u>physical length</u>.

> 3D Rice LSG simulation of a multipass cell Based on RLSG custom software (L=100mm R=100mm D=15mm)

$$k_{v} = S(T) P x_{i} \phi_{v}$$

CH₄ Absorption Line Selection



The fundamental v₁ and v₃ CH₄ bands are located at ~7.7 μm and ~3.3 μm, respectively

- A high detection sensitivity for methane measurements using quantum cascade lasers (QCLs) at 7.7 µm was previously reported
- Compact, TEC, CW, DFB ICLs emitting at 3-4 µm wavelengths became recently commercially available
- An interference-free CH₄ absorption line located at 3038.5 cm⁻¹ was selected as the optimum target absorption line
- The $3.3\mu m CH_4$ absorption line can be used at atmospheric pressure 17

ICL Characterization & Performance Evaluation

Nanoplus ICL, 3.291 µm center-wavelength



Performance evaluation for a 3.291- μ m CW RT ICL at different operating temperatures and injection currents. (a) ICL output power response curves; (b) Emission wavenumber curves.

Current turning rate: -0.232308 cm⁻¹/mA; Temperature turning rate: -0.23994 cm⁻¹/°C

Schematic of current CH₄ Gas Sensor System





ICL source (Nanoplus)

- Current: 42 mA
- Temperature: 30 °C
- Power: 1.5 mW
- Multipass gas cell (Sentinel Photonics/Aeris Technologies, Inc)
 - 54.6 meter, 435-passes, sealed
 - Sampling volume: 220 mL
 - Dimensions: $16.9 \times 6.6 \times 5.3 \text{ cm}^3$

Sensor system platform

- Two-floor design with folded optical path
- Low power consumption: 6 W
- Dimensions: $32 \times 20 \times 17 \text{ cm}^3$



Current Electronics Controller for CH₄ Sensor System



Control unit

- Laptop+NI DAQ+OEM laser driver
- Direct absorption spectroscopy
- DAQ: acquiring data & scanning the ICL wavelength

OEM laser driver for ICL

- Neo Monitors, Oslo, Norway
- Size: 10 x 8 cm²
- Low noise characteristic: ≤1 nA/√Hz
- On-board TEC driver: ±3 A, 15 V
- Single voltage power supply 12-24V



Data Processing for CH₄ Detection



A 4-step algorithm for CH₄ detection

- > 150 spectra were averaged
- Baseline of the spectral scan was fitted and eliminated
- Linearized spectrum using fringe spacing of a germanium etalon
- Lorentzian line shape fitting to retrieve concentration information

Interference-free absorption line of CH_4 at 3038.5 cm⁻¹ obtained from laboratory air at atmospheric pressure together with a fitted baseline and a transmission signal from a germanium etalon.

Allan-Werle Deviation Analysis



An Allan-Werle deviation plot was acquired in a time period of ~ 1.5 hours using a certified 2 ppm CH_4 cylinder with a 1 Hz sampling rate

1-s measurement precision is σ =10.53 ppb 60-s measurement precision is σ =1.43 ppb



Stationary Laboratory Measurements





CH₄ concentrations measured over a 7-day period in ambient air on the Rice University campus during May 2015.



Laboratory Stationary Measurements



Diurnal variations of CH_4 mixing ratio. Bottom whisker, bottom box line, top box line and top whisker indicate the 5th, 25th, 75th and 95th percentile, respectively. Line inside the boxes and continuous solid line represent the hourly median and mean of the data respectively

The diurnal profile of the methane concentration shows an increase in concentration during the early morning with a subsequent gradual decrease after \sim 8:00 am CDT to its typical background level of \sim 1.87 ppm in the Greater Houston area in May 2015.

Recent mobile Field Tests: December 2015







 CH_4 concentrations measured for a sampling period of ~ 10 minutes at a Clean Energy CNG O'Rourke Natural Gas Station in Houston, TX.



CH₄ Sensor System Summary (2015-2016)

- A 3.291 µm CW room-temperature ICL based absorption sensor was developed for methane detection using a 54.6 m optical path length multipass gas cell.
- A two-floor mechanical design with a folded optical path resulted in a sensor system dimension of 32 x 20 x 17 cm³
- Good electrical power management resulted in a **low power consumption** of the CH_4 sensor system: 6 W.
- A minimum detectable Limit (MDL) of 10.5 ppb for CH₄ with a 1 sec integration time was achieved.
- Laboratory measurements and mobile-mode field tests were conducted and results demonstrate the suitability of the sensor system to generate CH₄ spatial distributions in a typical U.S. urban area and at an oil and gas storage facility in Houston, TX.

Portable three Line Methane Sampling System for Laboratory and Field Deployment





Motivation for mid-infrared Ethane (C_2H_6) Detection



HITRAN absorption spectra of C_2H_6 , CH_4 , and H_2O



C₂H₆ Detection with a 3.36 µm CW DFB Diode Laser using a novel compact Multipass Absorption Cell and Control Electronics



Schematic of a C_2H_6 gas sensor using a Nanoplus 3.36 μ m DFB laser diode. M – mirror, CL – collimating lens, DM – dichroic mirror, MC – multipass cell, L – lens, SCB – sensor control board.



2f WMS signal for a C_2H_6 line at 2976.8 cm⁻¹ at 200 Torr

Minimum detectable C_2H_6 concentration: ~ 740 pptv (1 σ ; 1 s time resolution)



Innovative long path, small volume multipass gas cell: **57.6 m with 459 passes**



MGC dimensions: **17 x 6.5 x 5.5 (cm)** Distance between the MGC mirrors: 12.5 cm

Motivation for mid-infrared Formaldehyde Detection

• Atmospheric chemistry and climate

- Important volatile organic compound (VOC) present in all regions of the atmosphere which reacts in the presence of sunlight to yield ozone.
- Primary H₂CO sources are vehicle exhaust and fugitive industrial emissions
- Secondary H₂CO sources orginate from the breakdown of primary VOCs via photochemical oxidation

Industrial Applications

- Textile industry
- Automobile industry
- Adhesive resins for use in carpeting & plywood



HITRAN absorption spectra of H_2CO



Formaldehyde Line Selection in the 3-4 μ m Spectral Region



Sensor Configuration

Laser source Nanoplus ICL, 3.6 μm Injection current: 50 mA Output power: 3mW

Compact multipass cell Sentinel Inc. 7.6 cm mutipass cell length 32 ml sampling volume 3.7 m effective optical length

Dichroic mirror Current controller Alignment laser Function Iris generator Gas inlet (459 passes) MGC 7.6cm Lock-in amplifier Outlet [**Beam pattern** Parabolic DAQ-PC mirror MCT

 $\overline{\mathbf{O}}$

 λ -modulation scheme

Temperature controller

Representative H₂CO Sensor Calibration Results

• H₂CO gas standard: Kin-Tek gas standard generator



H₂CO Detection Sensitivity

• Minimum detection concentration: 1.5 ppb with a 140 sec averaging time



Noise Limitations

• Zero air measurements: 1s sampling rate



H₂CO Sensor System Summary (2014-2015)

- Development of laser-based absorption sensors for H₂CO detection using an <u>interband cascade laser</u> & a <u>compact xx m multipass absorption cell.</u>
- A minimum detection concentration of 1.5 ppb with 140 sec averaging time was achieved.
- Future work is planned to further improve the sensor detectivity to sub-ppb concentration level by using a multipass cell with an increased effective optical path length. Preliminary results show that a minimum detection concentration of 1 ppb with 10 sec averaging time can be achieved.

Conclusions and Future Developments

- Development of robust, compact, sensitive, selective mid-IR trace gas sensor technology based on RT, CW high performance DFB ICLs & QCLs for detection of explosives and TICs as well as environmental monitoring and medical diagnostics.
- Interband cascade and quantum cascade lasers were used in <u>QEPAS and TDLAS based</u> <u>sensor platforms</u>
- Performance evaluation of seven target trace gas species were reported. The minimum detection limit (MDL) with a 1 sec sampling time were :
 - C_2H_6 : MDL of .24 ppbv at ~3.36 µm; CH₄: MDL of 13 ppbv at ~7.28 µm; N₂O: MDL of 6 ppbv at ~7.28 µm.
- Development of Trace Gas Sensors for the monitoring of broadband absorbers: $acetone(C_3H_6O)$, propane (C_3H_8) , benzene (C_6H_6)

Hydrogen Peroxide (H₂O₂)

- Strong oxidant species in the atmosphere
- Associated with the formation of acid rain an atmospheric aerosols
- Employed in the synthesis of multiple chemical products & as bleaching agent in the pulp and paper industry
- Used for decontamination and sterilization of medical and pharmaceutical facilities
- Biomarker of lung and respiratory system diseases in exhaled breath



Vapor-Phase Hydrogen Peroxide (VPHP)

- VPHP is used for:
 - Decontamination of health-care and pharmaceutical facilities
 - Sterilization of medical equipment and packing materials in the food industry
 - VPHP units: gas-phase H_2O_2 generated from concentrated liquid H_2O_2 solutions
 - H₂O₂ concentrations between 200-1200 ppm are produced in the gas-phase and maintained for ~10 min
- After decontamination procedures, ambient H_2O_2 concentrations need to be monitored





Source: Bioquell UK Ltd



Techniques for H_2O_2 Detection

- Wet-chemistry methods based on fluorescence spectroscopy, colorimetric analysis and chemiluminescence
 - Transfer from gas to liquid phase required for subsequent analysis
 - Interference from other species and formation of sampling artifacts
- Mid-infrared laser based spectroscopy
 - Direct detection in the gas-phase
 - Real-time detection
 - High sensitivity and specificity



H₂O₂ Absorption in the mid-infrared spectral Region



Fundamental v_6 H₂O₂ band located at ~7.5-8.3 µm



Previous Employed Absorption Lines

Previous mid-IR sensor systems developed for H_2O_2 detection suffer from significant interferences from other gas species, particularly N₂O and H₂O vapor



1295.55 cm⁻¹ QEPAS-based sensor system (*Rei, et al., APL, 2014*)

1296.2 cm⁻¹ Multipass absorption-based sensor system (*Cao, et al., Adv. Opt. Tech* 2014)

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Selection of optimum Absorption Line

- A comprehensive spectral study was conducted
- Potential interferences from H_2O vapor, N_2O , CH_4 and CO_2 were considered
- An interference-free absorption line at 1234.05 cm⁻¹ was selected for H₂O₂ detection





EC-QCL Operating Characteristics

- CW EC-QCL (Model 21080-MHF, Daylight Solutions)
- Tuning range:1175-1300 cm⁻¹
- Mode-hop-free range: 1225-1285 cm⁻¹
- Power output: < 200 mW





Sensor Architecture





Photo of H₂O₂ Sensor Configuration

EC-QCL 21080-MHF (Daylight Solutions) **Optical components** Gas **Gas** inlet outlet Multipass cell AMAC-76 (Aerodyne Inc.) **Vigo Detector** (PVMI-3TE-8) Flow and pressure controllers



Parameter Optimization

• Pressure and modulation amplitude levels were optimized for improved SNR



- EC-QCL current & relative temperature: 300 mA & 0 °C
- Laser power output: ~ 66 mW



Sensor System Response





Calibration Results

- Different gas-phase H₂O₂ concentrations were generated by flowing air over aqueous solutions of different strengths
- Gas-phase concentrations were determined by fitting the direct absorption signals using the HITRAN database



Calibration curve and sensor response at H_2O_2 concentrations between 3 and 21 ppm



Sensor System Sensitivity

Allan-Werle deviation analysis



Minimum detection limit (MDL): 25 ppb @ 280-sec integration time



H₂O₂ Sensor System Summary

- Selected absorption line at 1234. 05 cm⁻¹ effectively alleviates interference issues identified previously for H_2O_2 detection
- MDL and ability to operate with no interference from water make our sensor system suitable for the monitoring of H_2O_2 in:
 - Industrial sites to establish possible exceedances of OSHA permissible exposure levels (PELs)
 - Decontamination/sterilization locations using VPHP
 - Exhaled breath as biomarker of lung-related diseases
- Further improvement of the MDL is necessary for application in other fields such as atmospheric monitoring



Broadband THz QCLs

• Multistack QC laser with four quantum well active regions



Broadband Spectrum



M. Rösch, et al., Nature Photonics, 9, 42-47, 2015



Experimental setup

• Mix a DFB QCL with a source operating by DFG (widely tunable)



P. Friedli, et al., Appl Phys Lett, 102, 222104, 2013.



Use a Fabry-Perot as a spectroscopy tool

- Can be used to measure the spectrum
- S/N limited by stability



For an optimum build-up of pulses inside a sensing cavity resonator, three conditions must be met:

 Sensing cavity has to support equidistant frequency eigenmodes.

Intracavity dispersion compensation

 Separation of the cavity eigenmodes and the separation of the frequency comb modes must be equal.

Matching both cavity length

 ntire comb must be shifted to overlap with the cavity eigenfrequencies.

Alignment of the eigenfrequencies





Cavity design for ICL-comb source

HR mirrors radius of curvature calculation

Cavity stability criterion:

Specs for cavity mirrors design:



 R_1 : radius of curvature of mirror #1 R_2 : radius of curvature of mirror #2 H. Kogelnik, Proc. IEEE 54,1312-1329, 1966 N. Hodgson & H. Weber, "Laser Resonators & Beam Propagation" Chapter 1, Springer 2005



Cavity design for ICL-comb source

Optical Cavity Performances:

✓ For effective length:

 $L_{eff} = 200 \mathrm{m}$

Number of half round-trip:

 $N = \frac{L_{eff}}{L_{cav}} = 6667$

Fraction of residual optical intra-cavity power after N half round-trips

 $R_{mirror}^{N} \approx 26\%$

✓ Finesse \Im :

$$\zeta \approx \frac{\pi}{1 - R_{mirror}} \approx 15700$$

 \checkmark Width of the cavity mode:

 $\Delta v = \frac{FSR}{\zeta} = 637 \, KHz$

✓ Intra-cavity power enhancement factor:

$$G = \frac{\zeta}{\pi} = 5000$$



Cavity design for an ICL-comb source

Cavity length calculation





The possibility to tune the comb modes spacing allows us to choose a cavity length of 30mm, maximizing the effective absorption length

Sensing cavity design for ICL- OFC source

Specifications for cavity mirror material

- Transmittance within the 3-4 μm spectral range
- Transparent in the visible (red diode laser for pump beam alignment)
- High surface quality (for HR and AR coatings to be applied)



LohnStar Optics was contacted on Jan.14, 2016 for a quotation of AR & HR coatings on sapphire optics from Mellor Optics. Inc [http://www.lohnstaroptics.com; www.melleroptics.com]



Proposed Sensing Cavity Locking to JPL-NRL ICL

Cavity locking system



RICE

Schematic of CE-OFC sensor system

CE-OFC-SENSOR

TARGET: detection and identification of toxic industrial chemicals with strong absorption features in the 3-4 μ m range.



Two Sensor Configurations:

Sensor system will offer:

- High detection sensitivity
- High spectral stability and purity
- High selectivity
- Possibility of multi-gas detection



Fast Mid-infrared detector

• Use of an intersubband Quantum Well Infrared Photoconductor (QWIP) detector









Is it a comb?

 Beat note measurement of the photocurrent (at 7.5GHz)

– For modes of amplitude E_k , the photocurrent at $\Delta \omega$





Frequency noise of a comb

- Use a (matched) optical cavity as an optical discriminator
- Measures all the modes at once



 $N\sqrt{N} \approx 300 \times$



Frequency noise power spectral density

- Single mode versus comb
- Above noise floor
- No significant limit





Beatnote spectrum

- The very narrow width confirms the correlations between modes
 - Uncorrelated lines could not be narrower than Schawlow-Townes (100's Hz)
 - However the signal is only about 2% of the c.w. photocurrent







Summary, Conclusions and Future Developments

- Development of robust, compact, sensitive, selective mid-IR trace gas sensor technology based on RT, CW high performance DFB ICLs & QCLs for detection of explosives and TICs as well as environmental monitoring and medical diagnostics.
- Interband cascade and quantum cascade lasers were used in <u>QEPAS and TDLAS based</u> <u>sensor platforms</u>
- Performance evaluation of seven target trace gas species were reported. The minimum detection limit (MDL) with a 1 sec sampling time were :
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- Development of Trace Gas Sensors for the monitoring of broadband absorbers: $acetone(C_3H_6O)$, propane (C_3H_8) , benzene (C_6H_6)
- Development of Mid-IR Electrically pumped Interband Cascade Optical Frequency Combs (OFCs) with JPL, Pasadena, CA, NRL, Washington, and the U. of Bari (Italy).

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