



Advanced Measurement Concepts for Mid-infrared Semiconductor Laser based Trace Gas Sensor Technologies: Opportunities & Challenges

Frank K. Tittel

Department of Electrical and Computer Engineering,
Rice University, Houston, TX 77005, USA

<http://www.ece.rice.edu/~lasersci/>

OUTLINE

- New Laser Based Trace Gas Sensor Technology
 - Novel Multipass Gas Absorption Cells & Electronics
 - Quartz Enhanced Photoacoustic Spectroscopy
- Examples of nine Mid-infrared Trace Gas Species
 - C_2H_6 , NH_3 , NO , CO , SO_2 , CH_4 , N_2O , H_2O_2 & $\text{C}_3\text{H}_6\text{O}$
- Future Directions of Laser Based Trace Gas Sensor Technologies and Conclusions

CPPTA

Warsaw, Poland

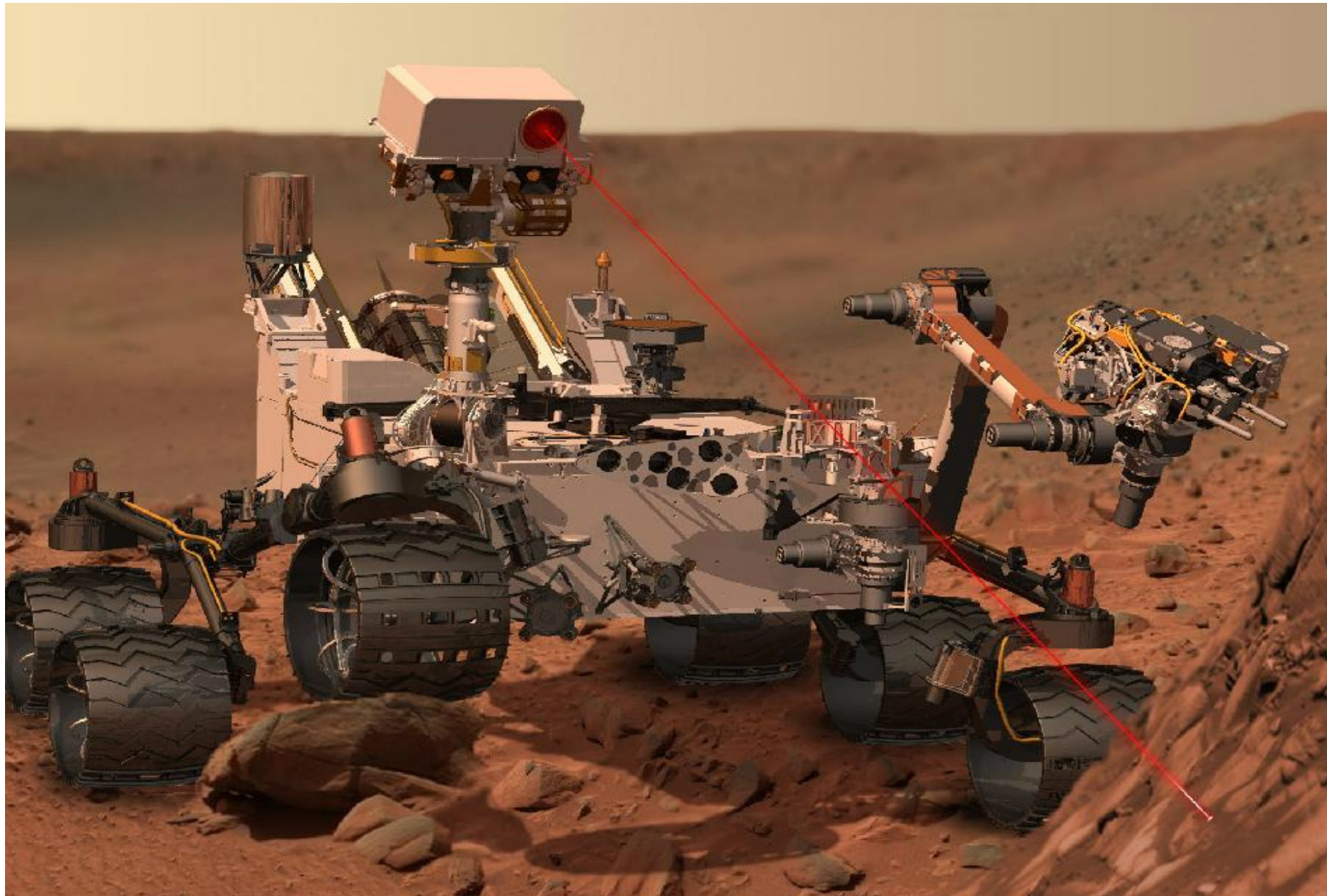
Sept. 23-26,
2014

Research support by NSF ERC MIRTHE, NSF-ANR NexCILAS, the Robert Welch Foundation, and Sentinel Photonics Inc. via an EPA Phase 1 SBIR sub-award is acknowledged

Wide Range of Trace Gas Sensing Applications

- **Urban and Industrial Emission Measurements**
 - Industrial Plants
 - Combustion Sources and Processes (e.g. fire detection)
 - Automobile, Truck, Aircraft and Marine Emissions
- **Rural Emission Measurements**
 - Agriculture & Forestry, Livestock
- **Environmental Monitoring**
 - Atmospheric Chemistry (e.g isotopologues, climate modeling,...)
 - Volcanic Emissions
- **Chemical Analysis and Industrial Process Control**
 - Petrochemical, Semiconductor, Pharmaceutical, Metals Processing, Food & Beverage Industries, Nuclear Technology & Safeguards
- **Spacecraft and Planetary Surface Monitoring**
 - Crew Health Maintenance & Life Support
- **Applications in Medical Diagnostics and the Life Sciences**
- **Technologies for Law Enforcement, Defense and Security**
- **Fundamental Science and Photochemistry**

“Curiosity” Landed on Mars on August 6, 2012



Laser based Trace Gas Sensing Techniques

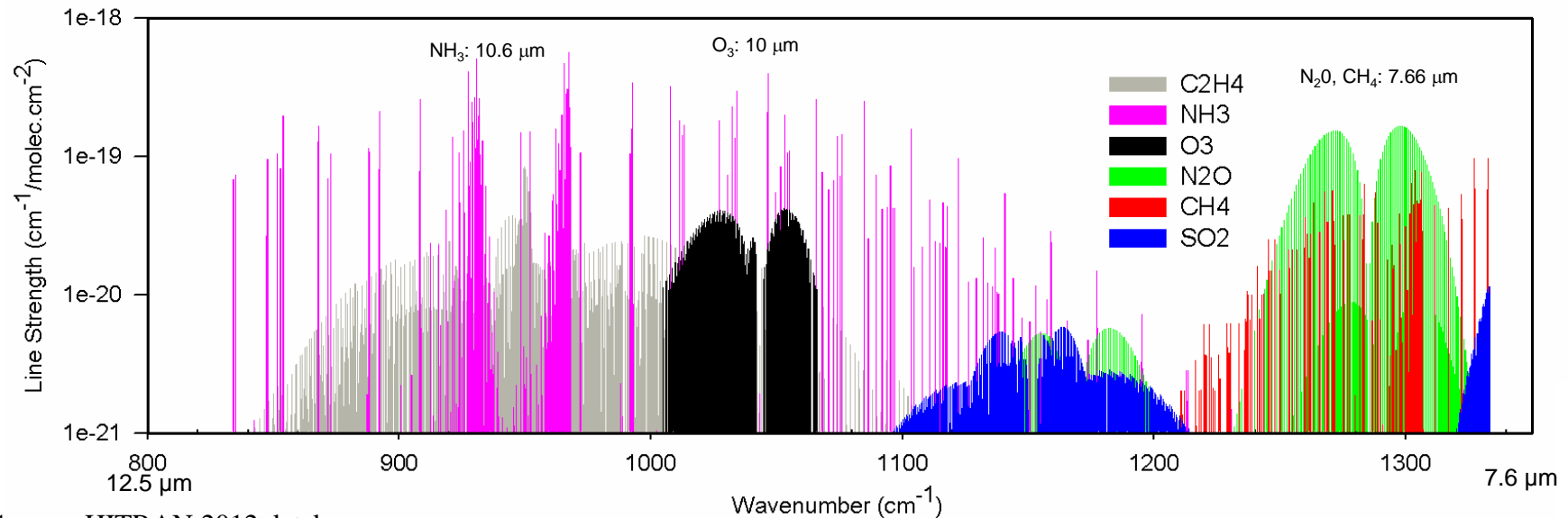
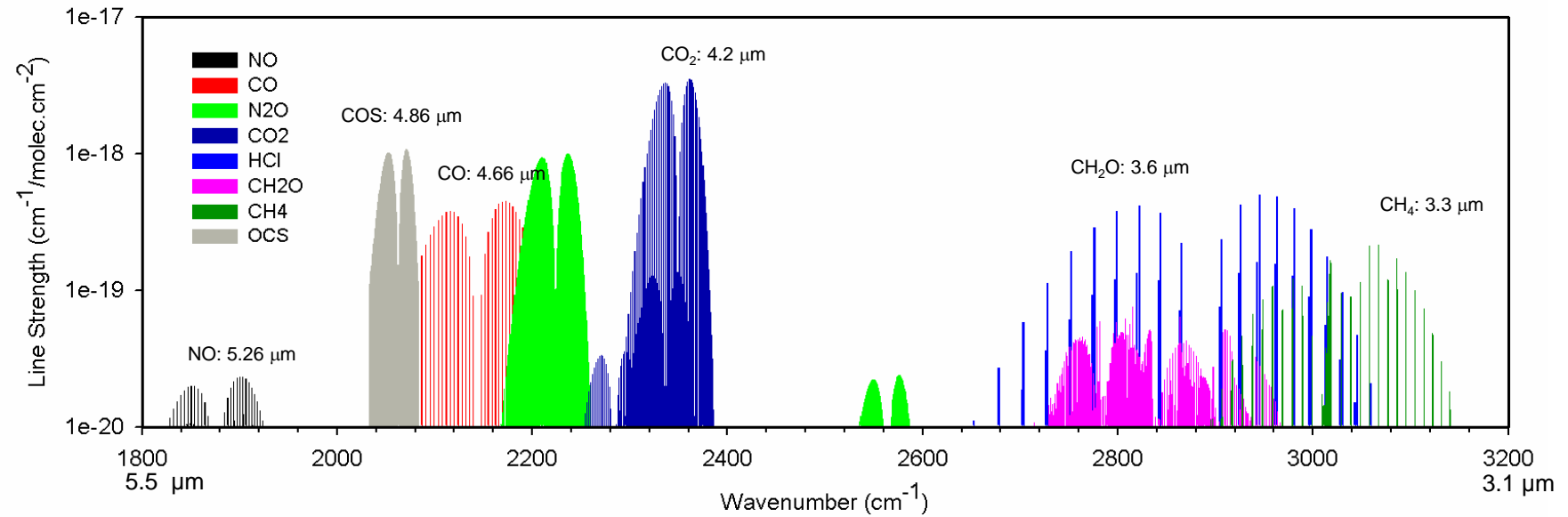
- **Optimum Molecular Absorbing Transition**
 - Overtone or Combination Bands (NIR)
 - Fundamental Absorption Bands (Mid-IR)
- **Long Optical Pathlength**
 - Multipass Absorption Cell (White, Herriot, Chernin, Sentinel Photonics/Aeris Technologies)
 - Cavity Enhanced and Cavity Ringdown Spectroscopy
 - Open Path Monitoring (with retro-reflector): Standoff and Remote Detection
 - Fiberoptic Evanescent Wave Spectroscopy
- **Spectroscopic Detection Schemes**
 - Frequency or Wavelength Modulation
 - Balanced Detection
 - Zero-air Subtraction
 - Photoacoustic & Quartz Enhanced Photoacoustic Spectroscopy (QEPAS)

Demonstration of CH₄ Sensor Performance

- Develop of Protoype Package in Year 1
- Develop of 2nd Generation Sensor System Year 2
- Develop Pre-Production Package



HITRAN Simulated Mid-Infrared Molecular Absorption Spectra

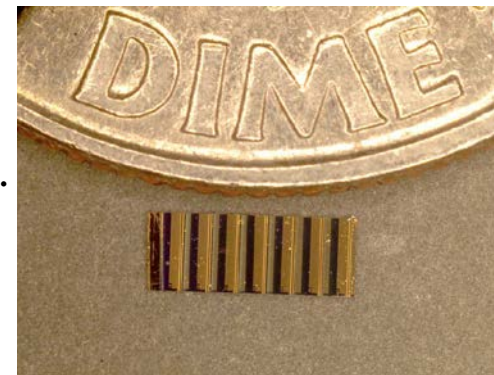


Mid-IR Source Requirements for Laser Spectroscopy

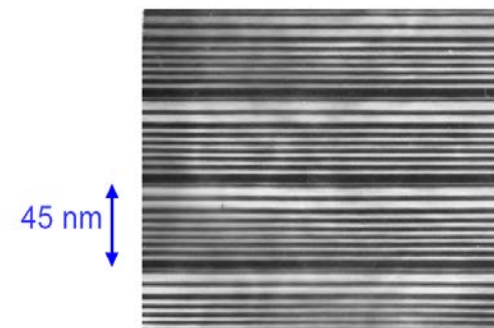
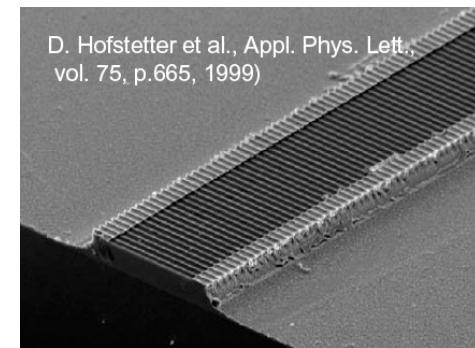
<u>REQUIREMENTS</u>	<u>IR LASER SOURCE</u>
Sensitivity (% to pptv)	Optimum Wavelength, Power
Selectivity (Spectral Resolution)	Stable Single Mode Operation and Narrow Linewidth
Multi-gas Components, Multiple Absorption Lines and Broadband Absorbers	Mode Hop-free Wavelength Tunability
Directionality or Cavity Mode Matching	Beam Quality
Rapid Data Acquisition	Fast Time Response
Room Temperature Operation	High wall plug efficiency, no cryogenics or cooling water
Field deployable in harsh environments	Compact & Robust

Key Characteristics of Mid-IR QCL & ICL Sources – Sept. 2014

- **Band – structure engineered devices**
Emission wavelength is determined by layer thickness – MBE or MOCVD; Type I QCLs operate in the 3 to 24 μm spectral region; Type II and GaSb based ICLs can cover the 3 to 6 μm spectral range.
 - Compact, reliable, stable, long lifetime, and commercial availability
 - Fabry-Perot (FP), single mode (DFB) and multi-wavelength devices
- **Wide spectral tuning ranges in the mid-IR**
 - 1.5 cm^{-1} using injection current control for DFB devices
 - $10\text{--}20\text{ cm}^{-1}$ using temperature control for DFB devices
 - $\sim 100\text{ cm}^{-1}$ using current and temperature control for QCL DFB Array
 - $\sim 525\text{ cm}^{-1}$ (22% of c.w.) using an external grating element and FP chips with heterogeneous cascade active region design; also QCL DFB Array
- **Narrow spectral linewidths**
 - CW: $0.1\text{--}3\text{ MHz}$ & $<10\text{ kHz}$ with frequency stabilization (0.0004 cm^{-1})
 - Pulsed: $\sim 300\text{ MHz}$
- **High pulsed and CW powers of QCLs at TEC/RT temperatures**
 - Room temperature pulsed power of $> 30\text{ W}$ with 44% wall plug efficiency
 - CW powers of $\sim 5\text{ W}$ with 23% wall plug efficiency at 293 °K
 - $> 600\text{ mW}$ CW DFB @ 285 °K ; wall plug efficiency 23% at $4.6\text{ }\mu\text{m}$

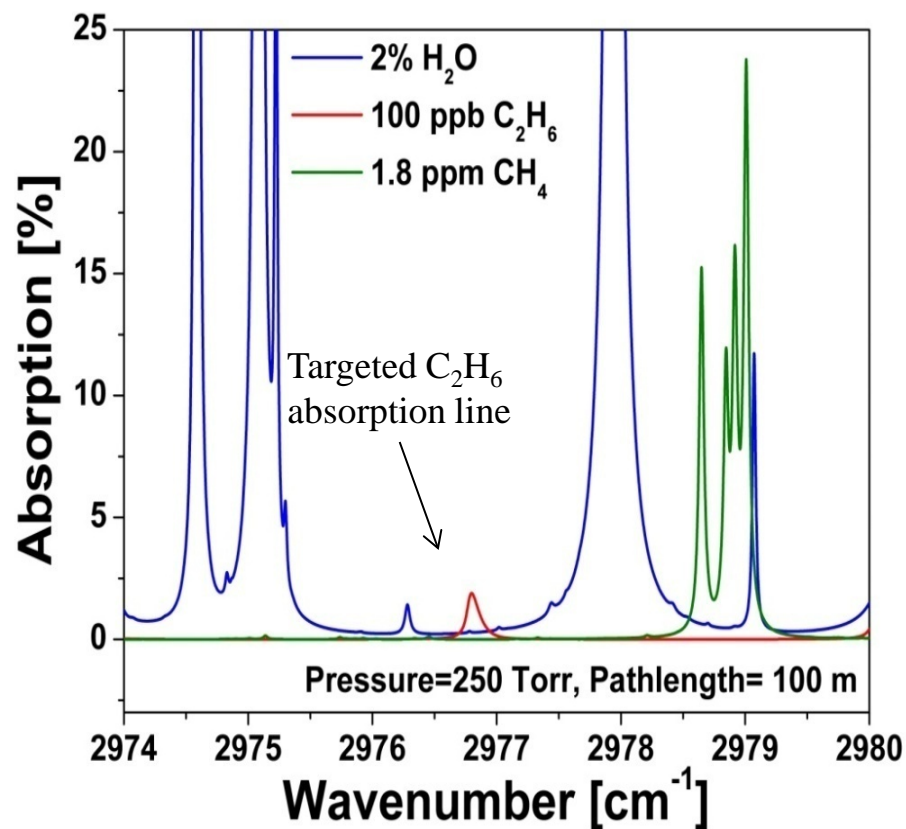


4 mm



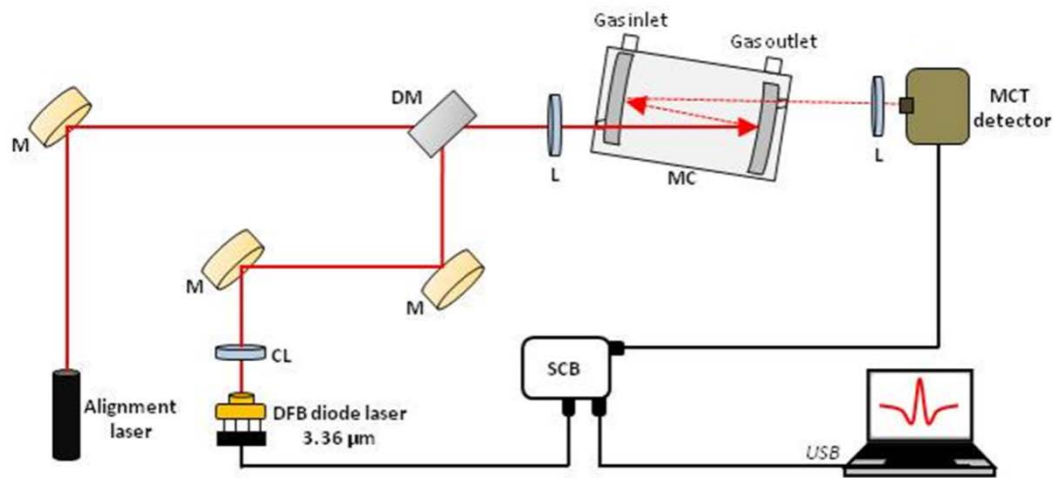
Motivation for Mid-infrared C_2H_6 Detection

- **Applications in medical breath analysis**
 - Asthma
 - Schizophrenia
 - Lung cancer
 - Vitamin E deficiency
- **Applications in atmospheric chemistry and climate R &D**
 - Fossil fuel and biofuel consumption
 - Biomass burning
 - Vegetation/soil
 - Natural gas loss

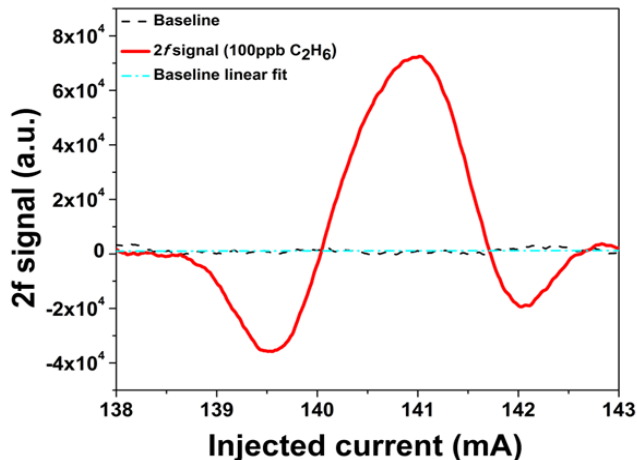


HITRAN absorption spectra of C_2H_6 , CH_4 , and H_2O

C_2H_6 Detection with a 3.36 μm CW DFB LD using a Novel Compact Multipass Absorption Gas Cell and Control Electronics

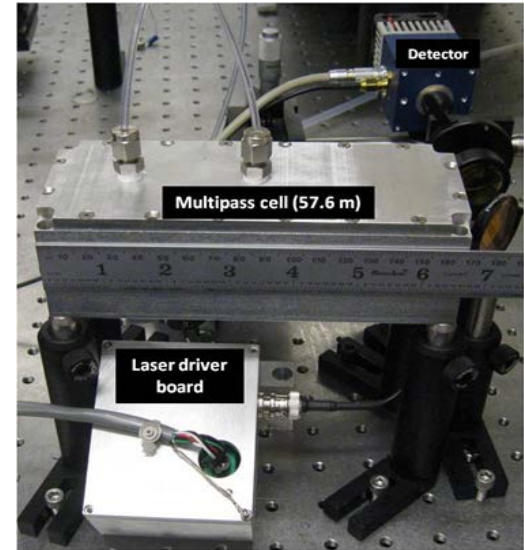


Schematic of a C_2H_6 gas sensor using a Nanoplus 3.36 μm DFB laser diode. M – mirror, CL – collimating lens, DM – dichroic mirror, MC – multipass cell, L – lens, SCB – sensor control board.



2f WMS signal for a C_2H_6 line at 2976.8 cm^{-1} at 200 Torr

Minimum detectable C_2H_6 concentration:
 $\sim 740 \text{ pptv}$ (1σ ; 1 s time resolution)

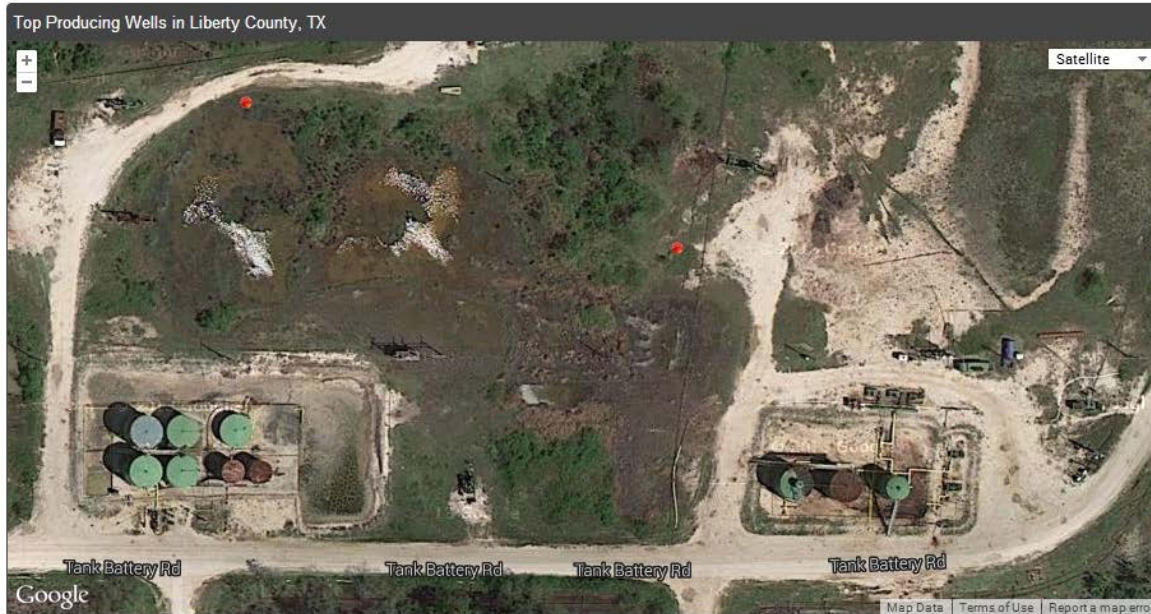


Innovative long path, small volume multipass gas cell: **57.6 m with 459 passes**

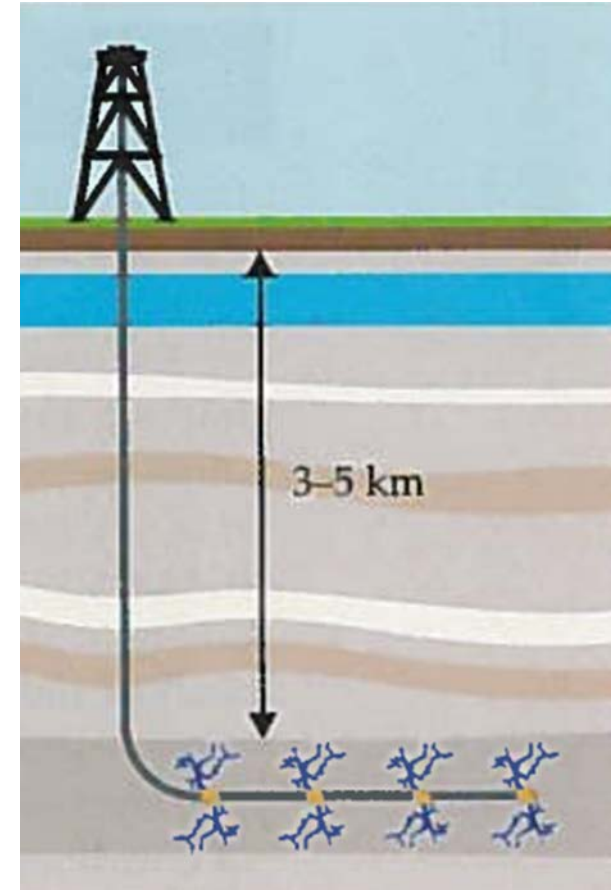


MGC dimensions: **17 x 6.5 x 5.5 (cm)**
Distance between the MGC mirrors: 12.5 cm

Typical Oil & Gas Production Site near Houston, TX



This figure shows the result of a sequence of four fracking injections obtained by directional drilling, which creates horizontal production in target stratum.

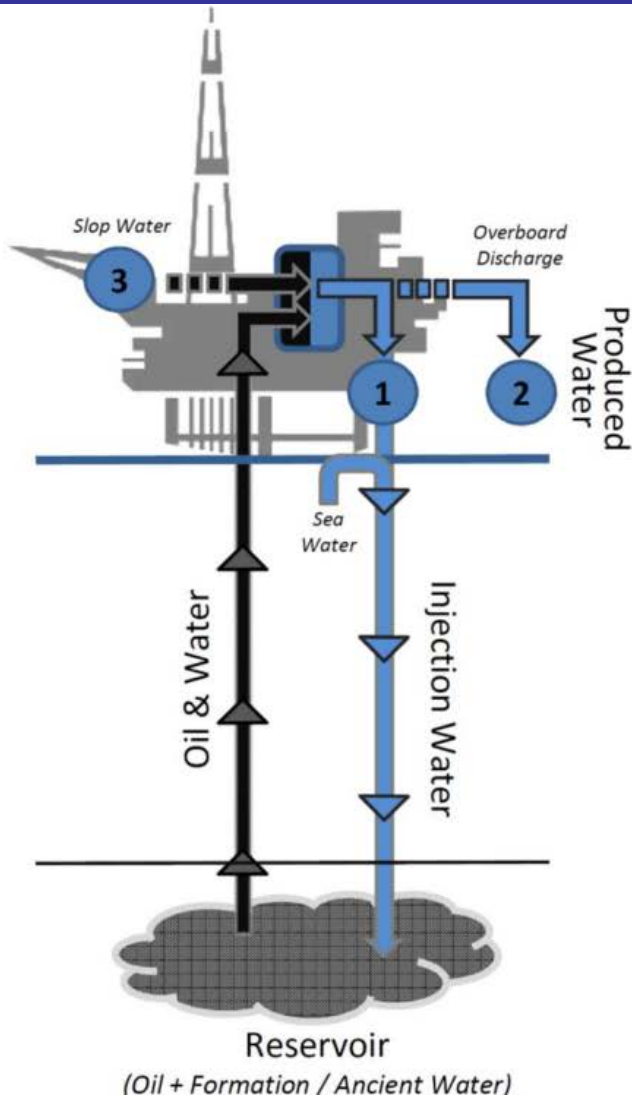


A proposed DOE-ARPA-E CH₄ detection project at 3.327 μm will start in 2015 at a well platform of 10 m x 10 m with a 1 m spatial resolution.



RICE

Oil in Water Detection



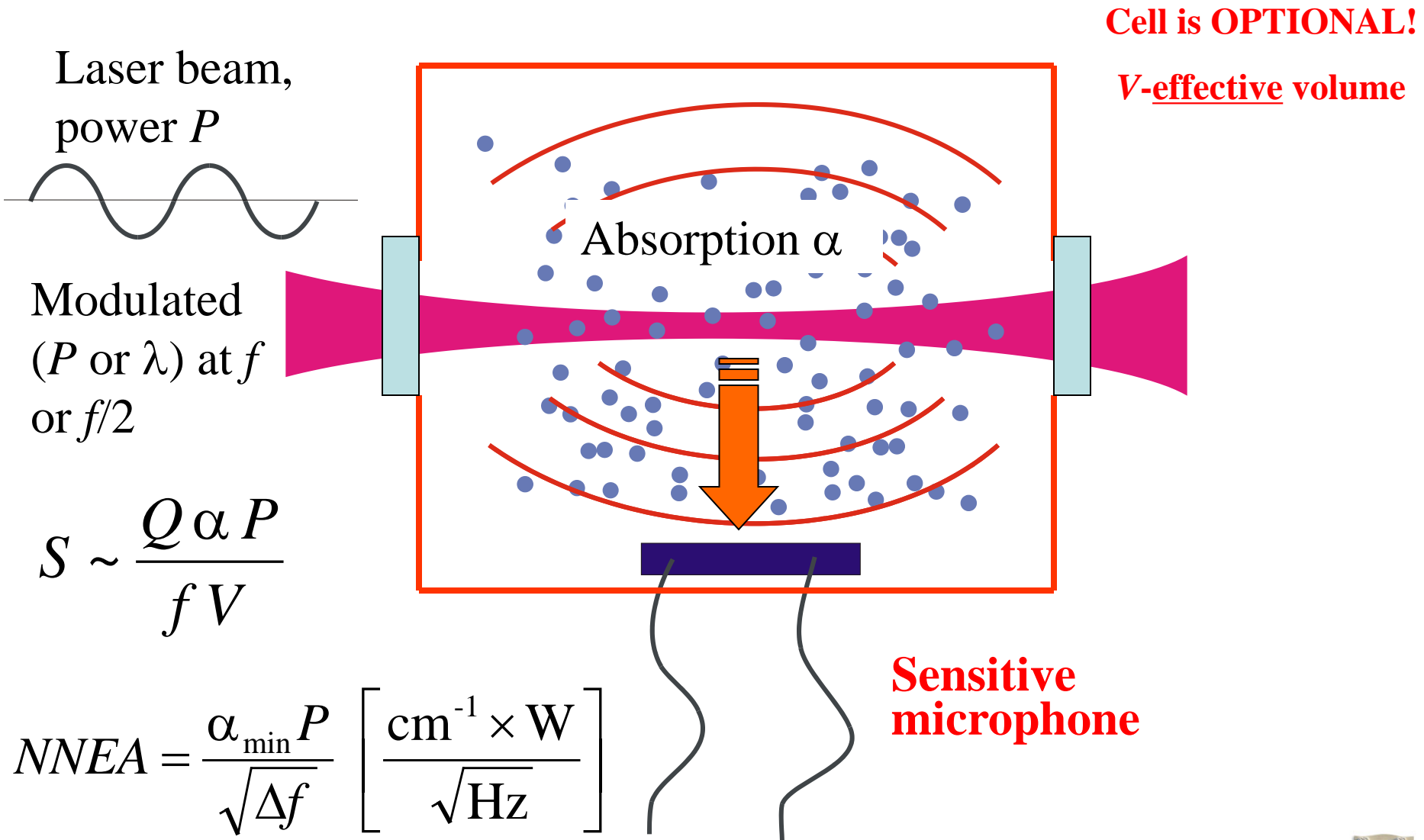
- Produced water
 - legislation: < 15 ppm
- Injection water
 - Economic reasons
target value: < 5 ppm or lower

Motivation for NH_3 Detection

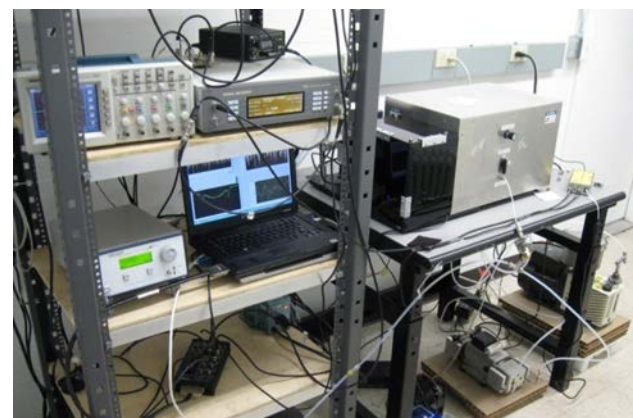
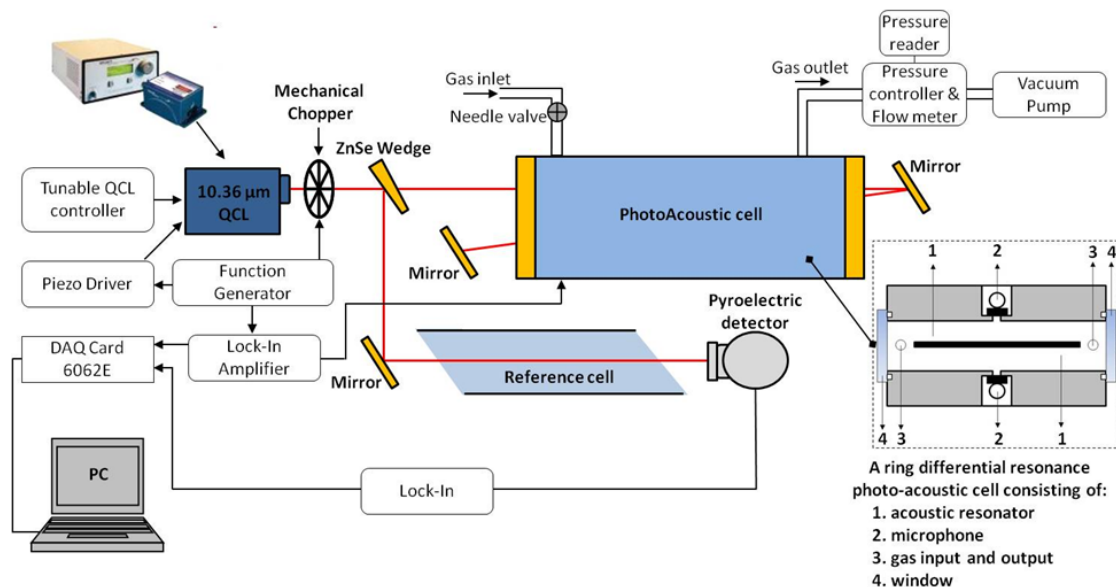
- **Medical diagnostics**
 - Kidney disease
 - Liver failure and Cirrhosis
 - Brain Cells dysfunction
 - Drowsiness and Coma
- **Atmospheric chemistry**
- **Pollutant gases monitoring**
- **Monitoring NH_3 concentrations in the exhaust stream of NO_x removal systems based on selective catalytic reduction (SCR) techniques associated with electric power plants**
- **Spacecraft related trace gas monitoring**



Conventional Photoacoustic Spectroscopy (PAS)

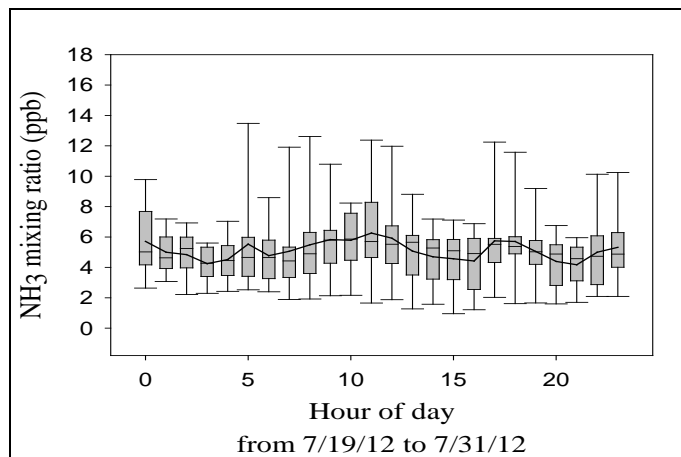


NH₃ Measurements based on an EC-QCL PAS Sensor System

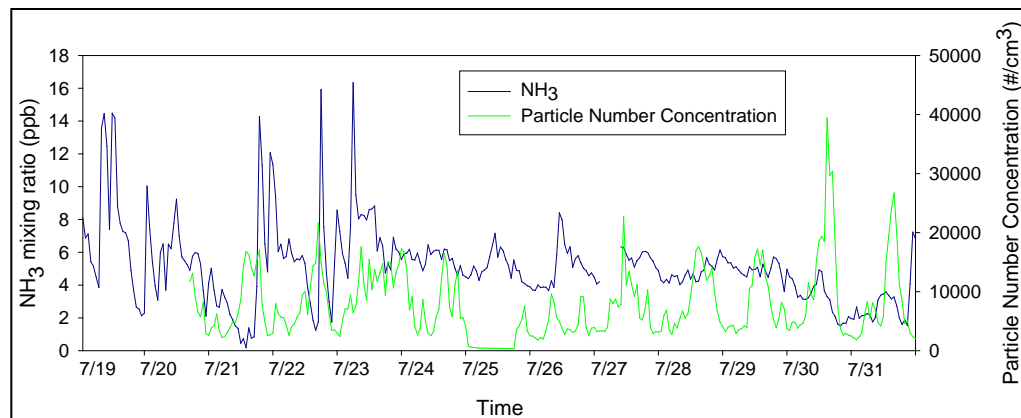


NH₃ sensor deployed at the UH Moody Tower rooftop monitoring site.

Schematic of a Daylight Solutions 10.36 μm CW TEC EC-QCL based PAS NH₃ Sensor.

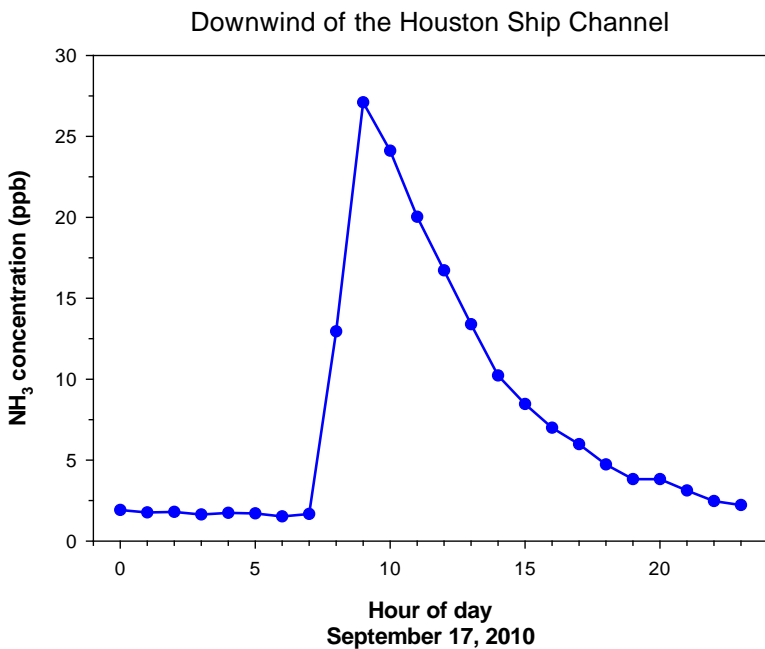
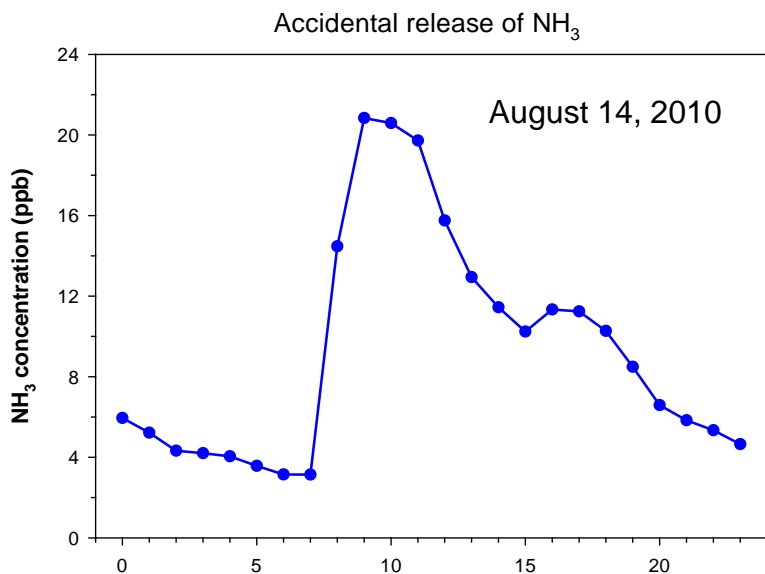


Diurnal profile of atmospheric NH₃ levels in Houston, TX.



Comparison between NH₃ and particle number concentration time series from July 19 to July 31 2012.

Unexpected Remote Detection of NH_3 based on PAS



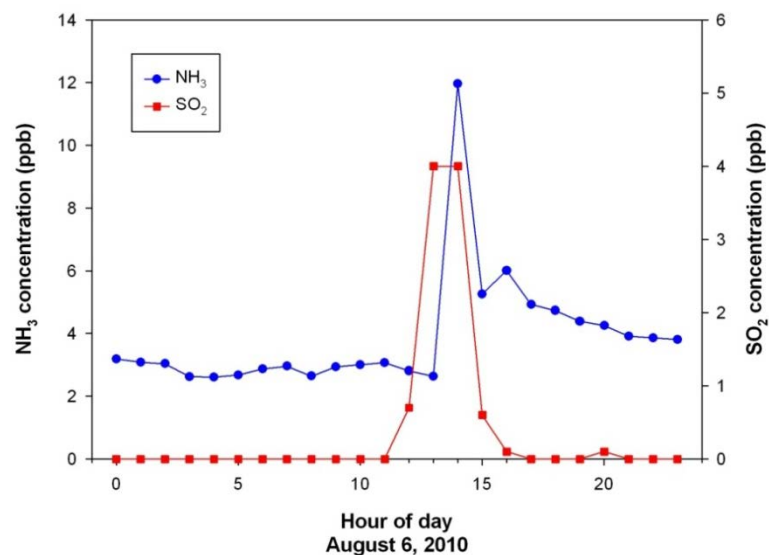
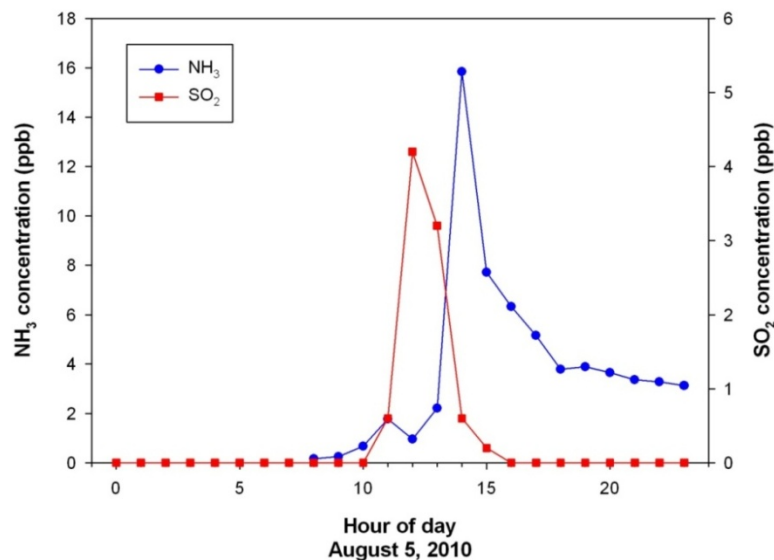
A chemical incident occurred at ~ 6 a.m. after two trucks collided on I-59. Both trucks caught fire. [www.chron.com]



photo: Public Domain / RJN2

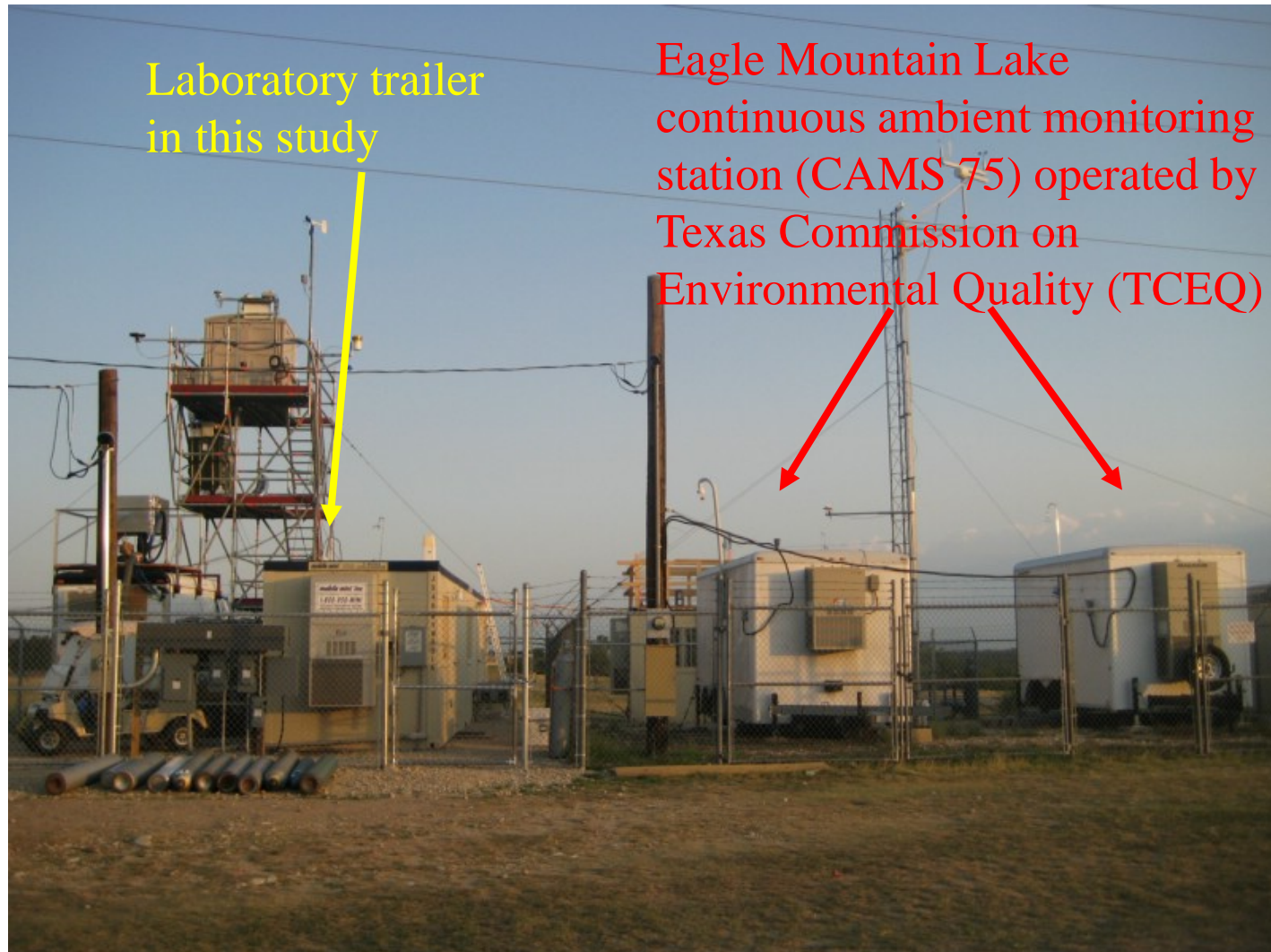
Estimated hourly NH_3 emission from the Houston Ship Channel area is about 0.25 ton. Mellqvist et al., (2007) Final Report, HARC Project H-53.

Remote Detection of Sporadic NH_3 Emissions from the Parish Electric Power Plant, TX



The Parish electric power plant is located near the Brazos River in Fort Bend County, Texas (~27 miles SW from downtown Houston)

Fort-Worth, Dallas(TX) CAMS 75 & TCEQ Monitoring Site

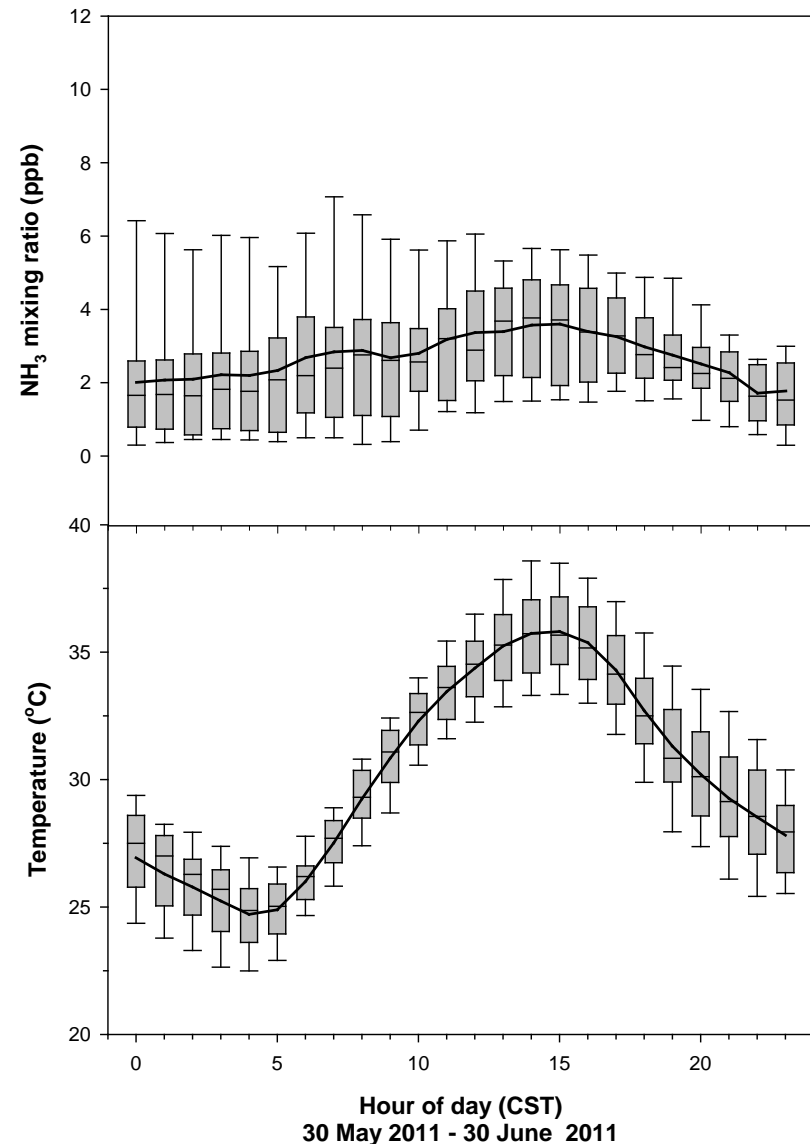


Available Instrumentation at CAMS 75 & TCEQ Monitoring Site

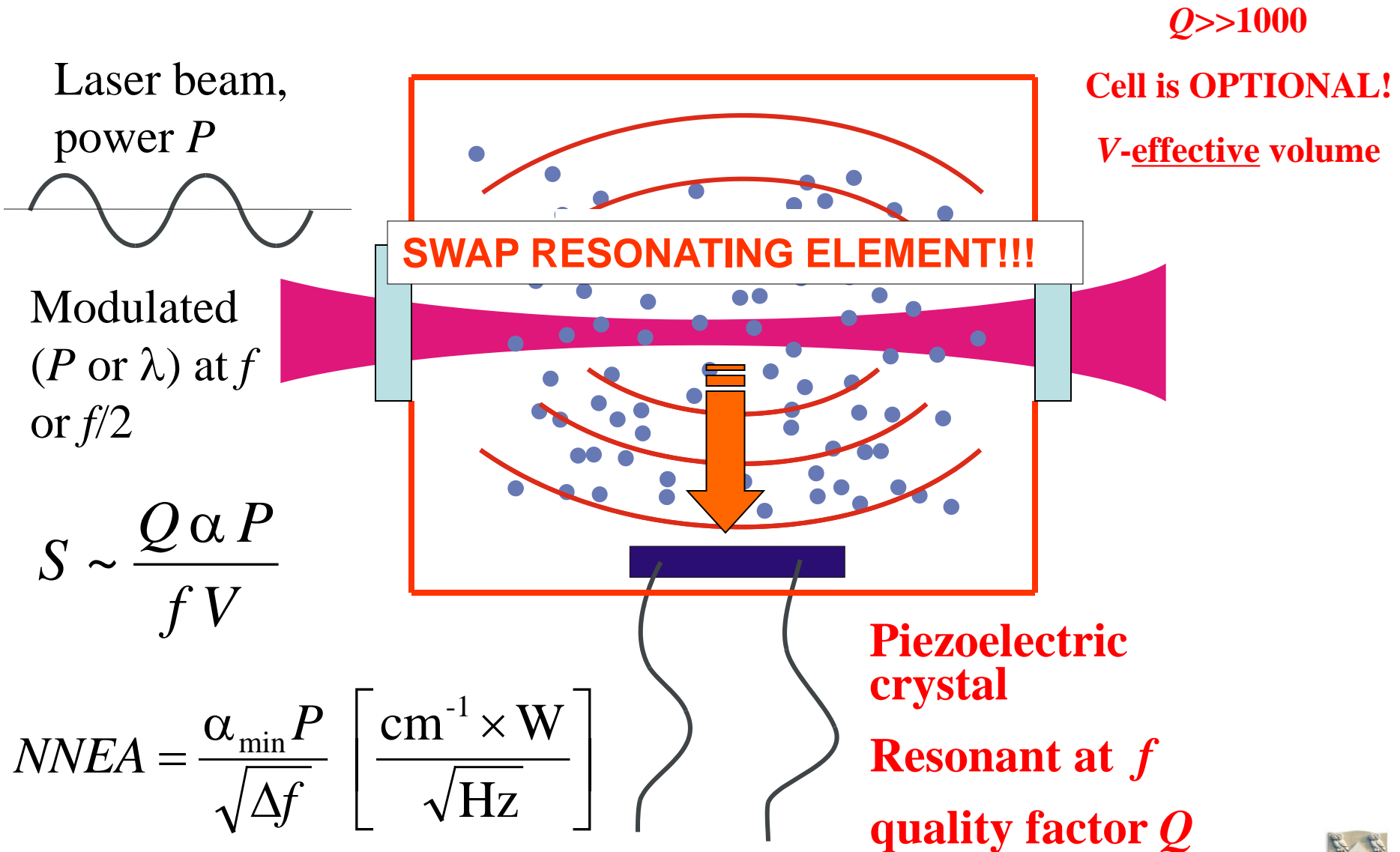
Species/parameter	Measurement technique
NH ₃	Daylight Solutions External Cavity Quantum Cascade Laser (Photo-acoustic Spectroscopy)
CO	Thermo Electron Corp. 48C Trace Level CO Analyzer (Gas Filter Correlation)
SO ₂	Thermo Electron Corp. 43C Trace Level SO ₂ Analyzer (Pulsed Fluorescence)
NO _x	Thermo Electron Corp. 42C Trace Level NO-NO ₂ -NO _x Analyzer (Chemiluminescence)
NO _y	Thermo Electron Corp. 42C-Y NO _y Analyzer (Molybdenum Converter)
HNO ₃	Mist Chamber coupled to Ion Chromatography (Dionex, Model CD20-1)
HCl	Mist Chamber coupled to Ion Chromatography (Dionex, Model CD20-1)
VOC _s	IONICON Analytik Proton Transfer Reaction Mass Spectrometer and TCEQ Automated Gas Chromatograph
PBL height	Vaisala Ceilometer CL31 with updated firmware to work with Vaisala Boundary Layer View software
Temperature	Campbell Scientific HMP45C Platinum Resistance Thermometer
Wind speed	Campbell Scientific 05103 R. M. Young Wind Monitor
Wind direction	Campbell Scientific 05103 R. M. Young Wind Monitor

NH₃ Source Attribution & Temperature Variations

- Emission events from specified point sources (i.e., industrial facilities)
- Estimated NH₃ emissions from cows (1.3 tons/day)
- Estimated NH₃ emissions from soil and vegetation (0.15 tons/day)
- EPA PMF (biogenic:74.1%; light duty vehicles:12.1%; natural gas/industry: 9.4%; heavy duty vehicles:4.4%)
- Livestock might account for approximately 66.4% of total NH₃ emissions
- Increased contribution from industry (→18.9%)



From Conventional PAS to Quartz Enhanced PAS (QEPAS)



Quartz Tuning Fork as a Resonant Microphone for QEPAS

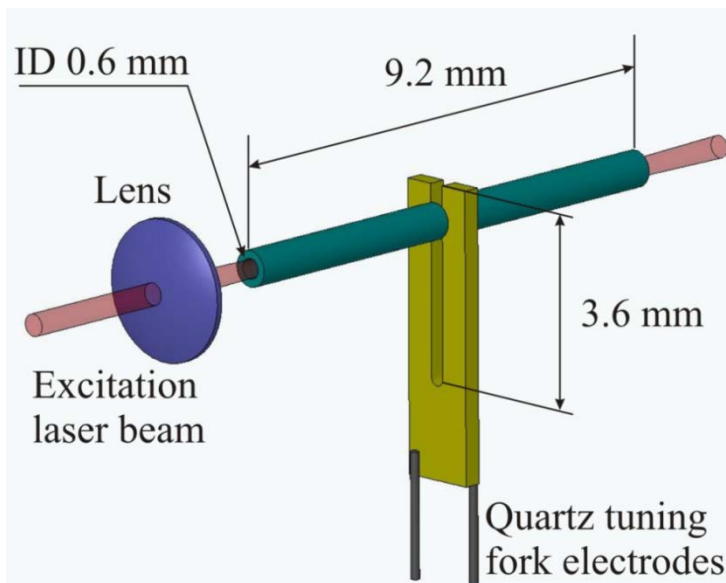


Unique Properties

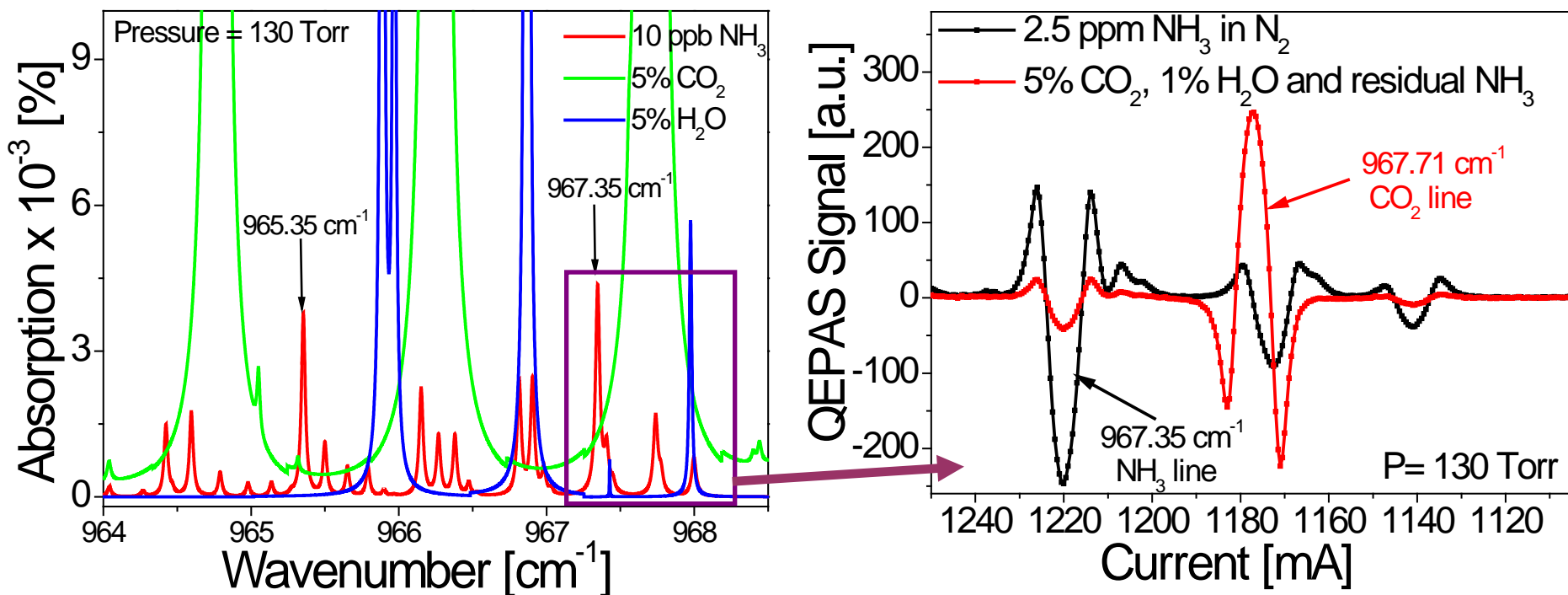
- Extremely low internal losses:
 - $Q \sim 10\,000$ at 1 atm
 - $Q \sim 100\,000$ in vacuum
- Acoustic quadrupole geometry
 - Low sensitivity to external sound
- Large dynamic range ($\sim 10^6$) – linear from thermal noise to breakdown deformation
 - 300K noise: $x \sim 10^{-11}$ cm
 - Breakdown: $x \sim 10^{-2}$ cm
- Wide temperature range: from 1.6K to ~ 700 K

Acoustic Micro-resonator (μ R) Tubes

- Optimum inner diameter: 0.6 mm; μ R-QTF gap is 25-50 μ m
- Optimum mR tubes must be ~ 4.4 mm long ($\sim \lambda/4 < l < \lambda/2$ for sound at 32.8 kHz)
- SNR of QTF with μ R tubes: $\times 30$ (depending on gas composition and pressure)



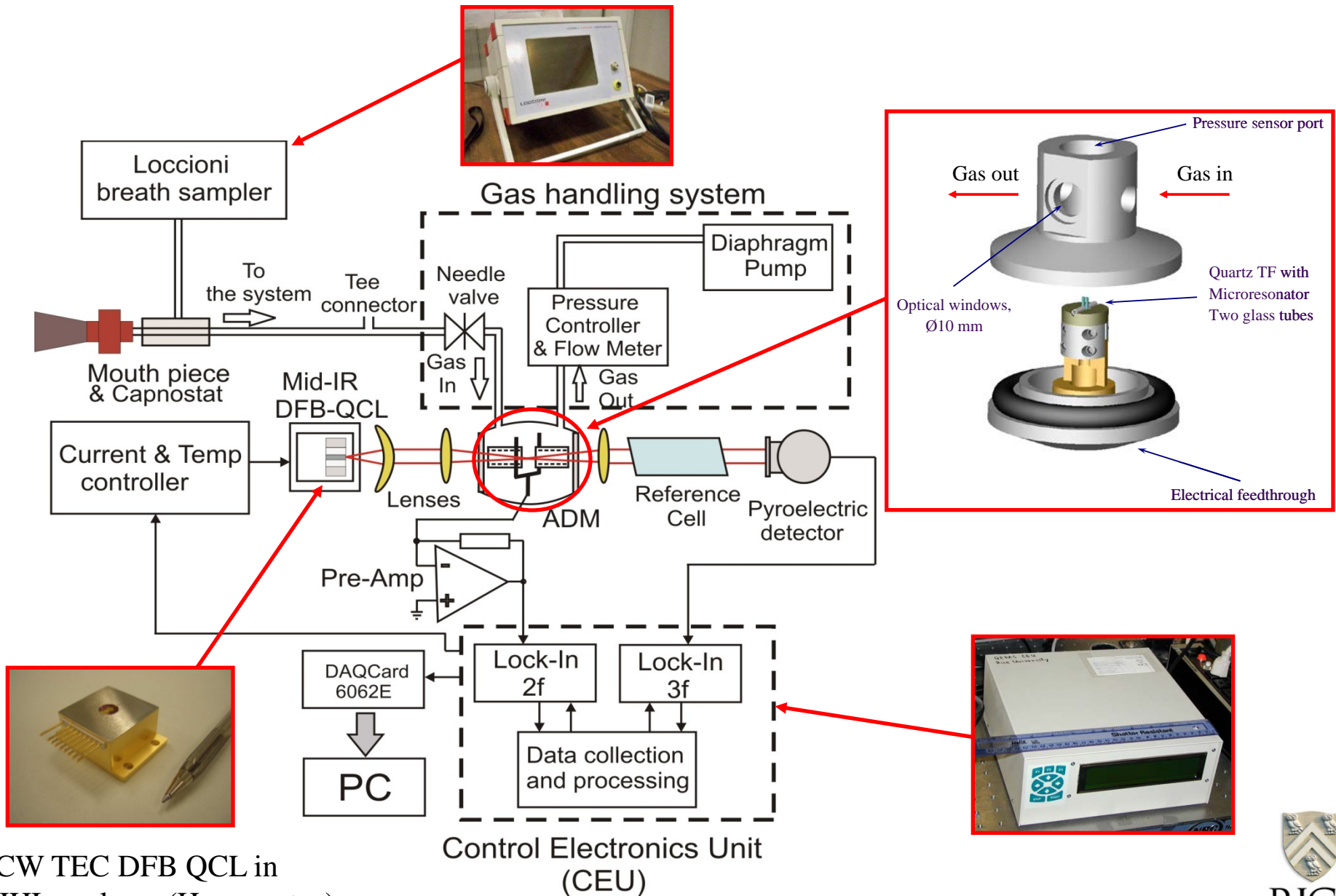
Optimum NH_3 Line Selection for a 10.34 μm CW TEC DFB QCL



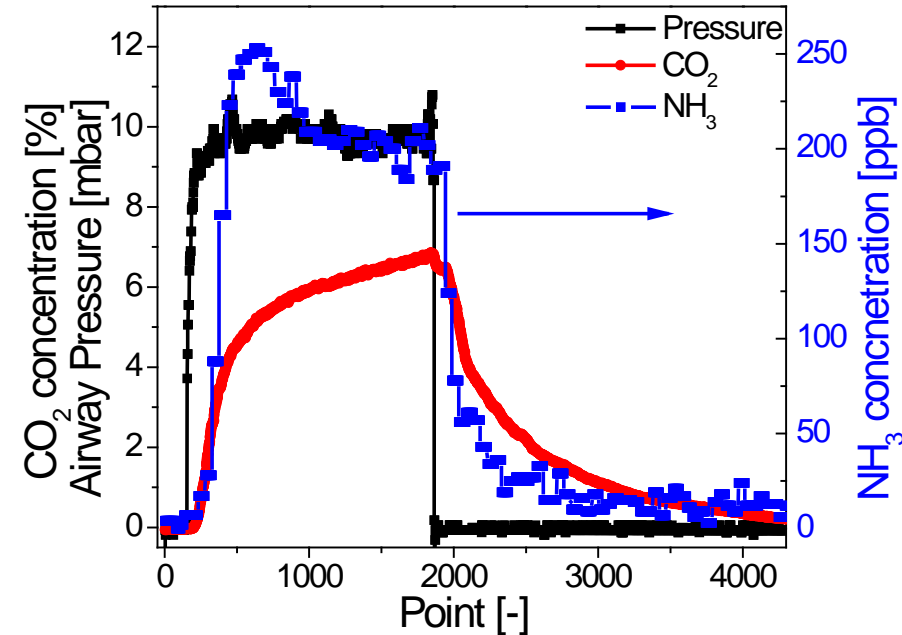
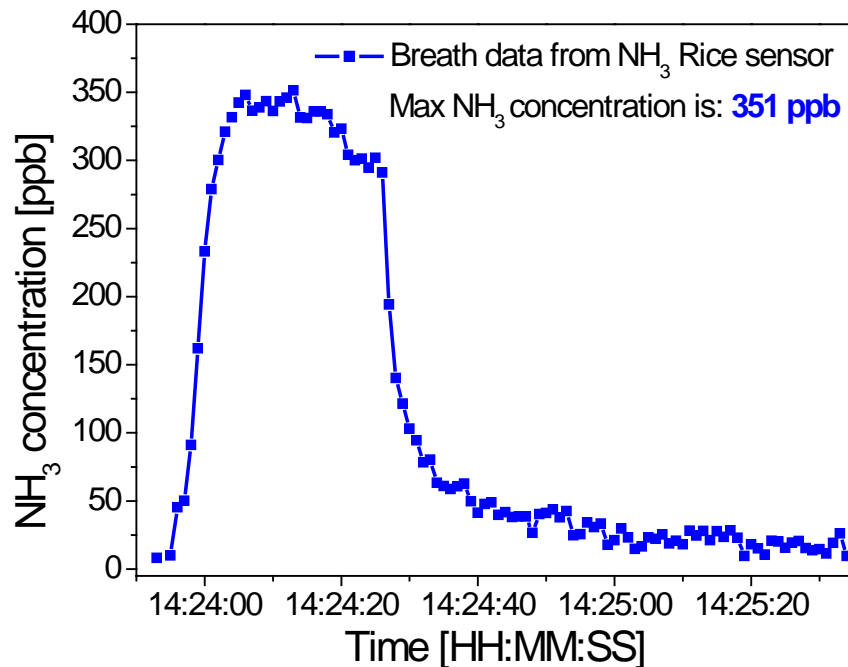
Simulated HITRAN high resolution spectra @ 130 Torr indicating two NH_3 absorption lines of interest

No overlap between NH_3 and CO_2 absorption lines was observed for the selected **967.35 cm⁻¹** NH_3 absorption line in the ν_2 R-band.

QEPAS based NH_3 Gas Sensor Architecture



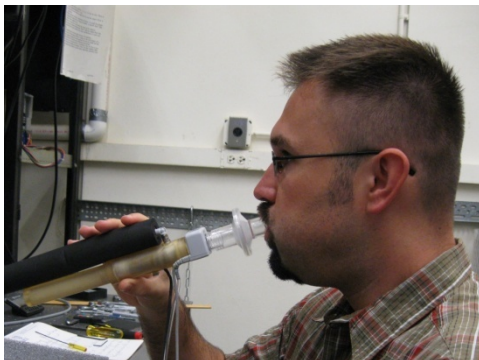
Real-time Exhaled Human NH_3 Breath Measurements



Airway pressure (black), CO₂ (red), and NH₃ (blue) profiles of a single breath exhalation lasting 40sec.

Successful testing of a 2nd generation breath ammonia monitor installed in a clinical environment. (Johns Hopkins, Baltimore, MD and St. Luke's Hospital, Bethlehem, PA)

Minimum detectable concentration of NH_3 is:
~ 6 ppbv at 967.35 cm^{-1} (1 σ ; 1 s time resolution)

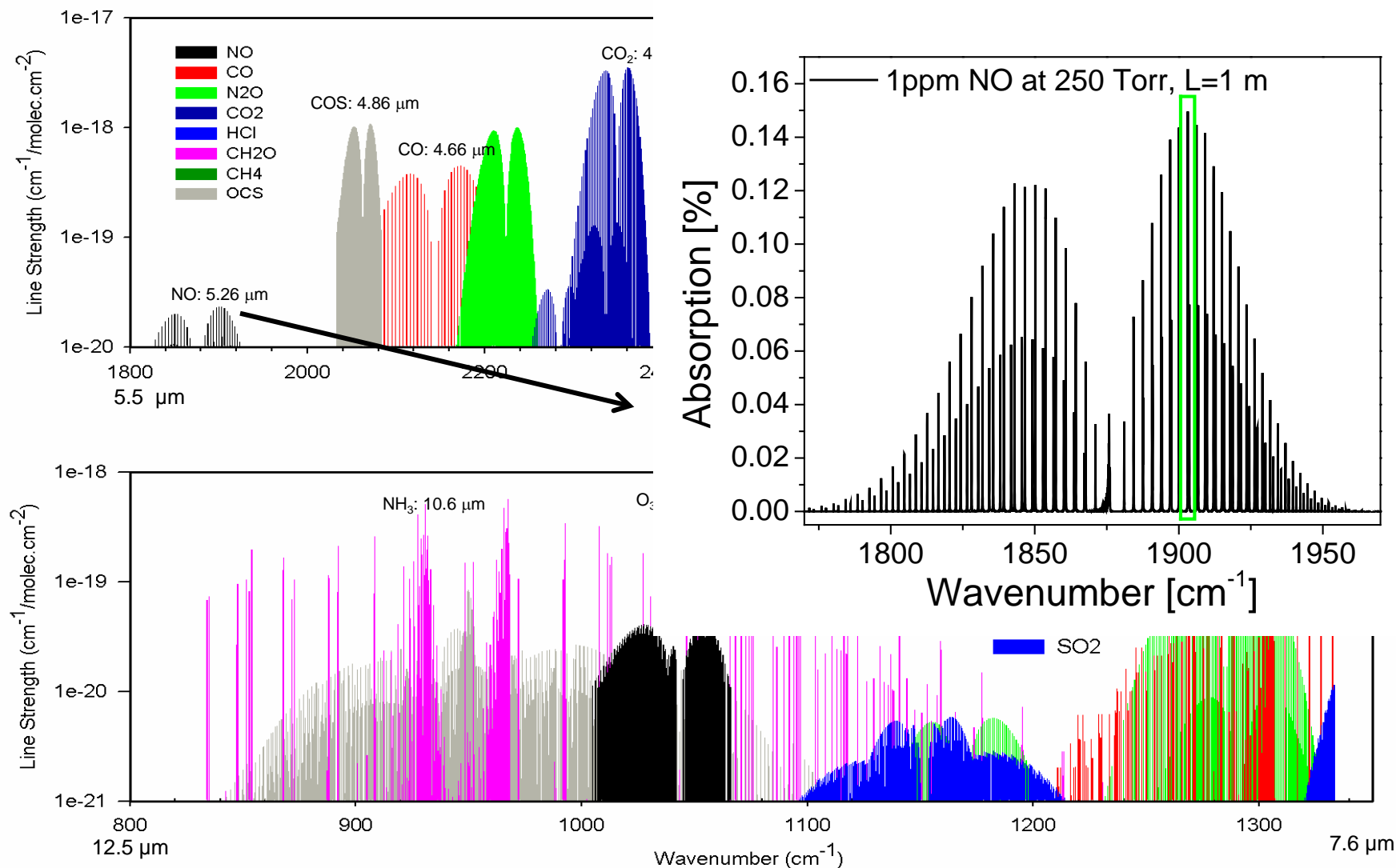


Motivation for Nitric Oxide Detection

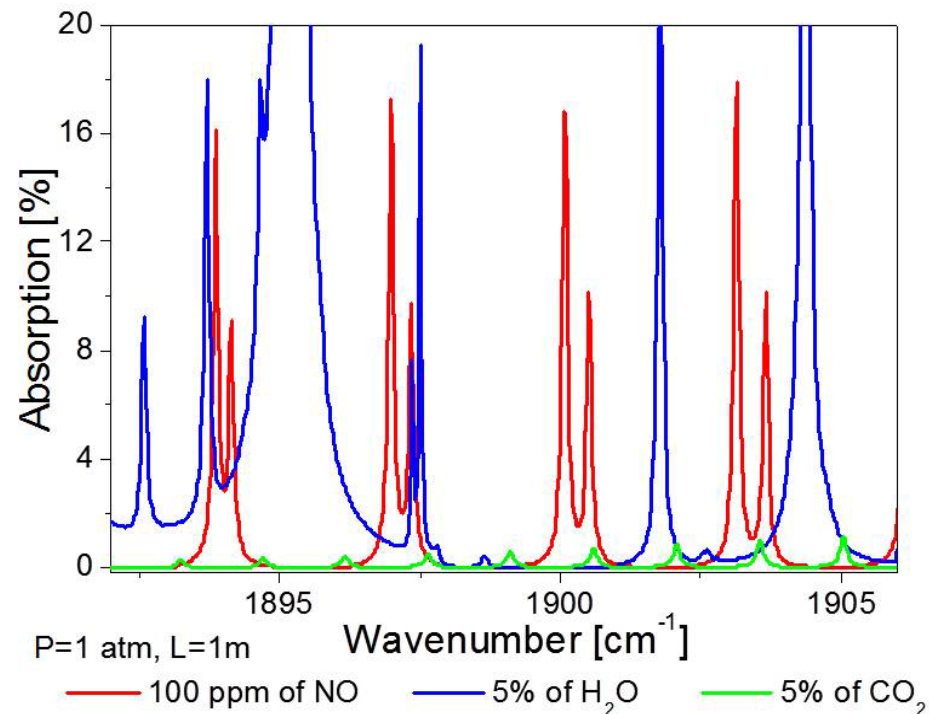
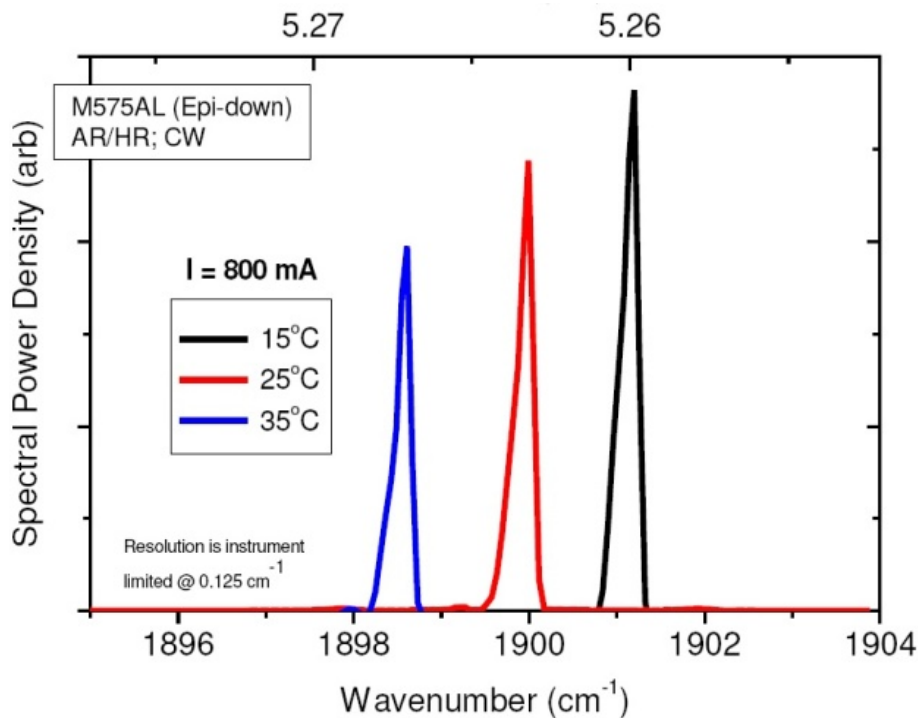
- **NO in medicine and biology**
 - Important signaling molecule in physiological processes in humans and mammals (1998 Nobel Prize in Physiology/Medicine)
 - Treatment of asthma, chronic obstructive pulmonary disease (COPD) & lung rejection
- **Environmental pollutant gas monitoring**
 - Ozone depletion
 - Precursor of smog and acid rain
 - NO_x monitoring from automobile exhaust and power plant emissions
- **Atmospheric Chemistry**



Molecular Absorption Spectra within two Mid-IR Atmospheric Windows and NO absorption @ 5.26 μm

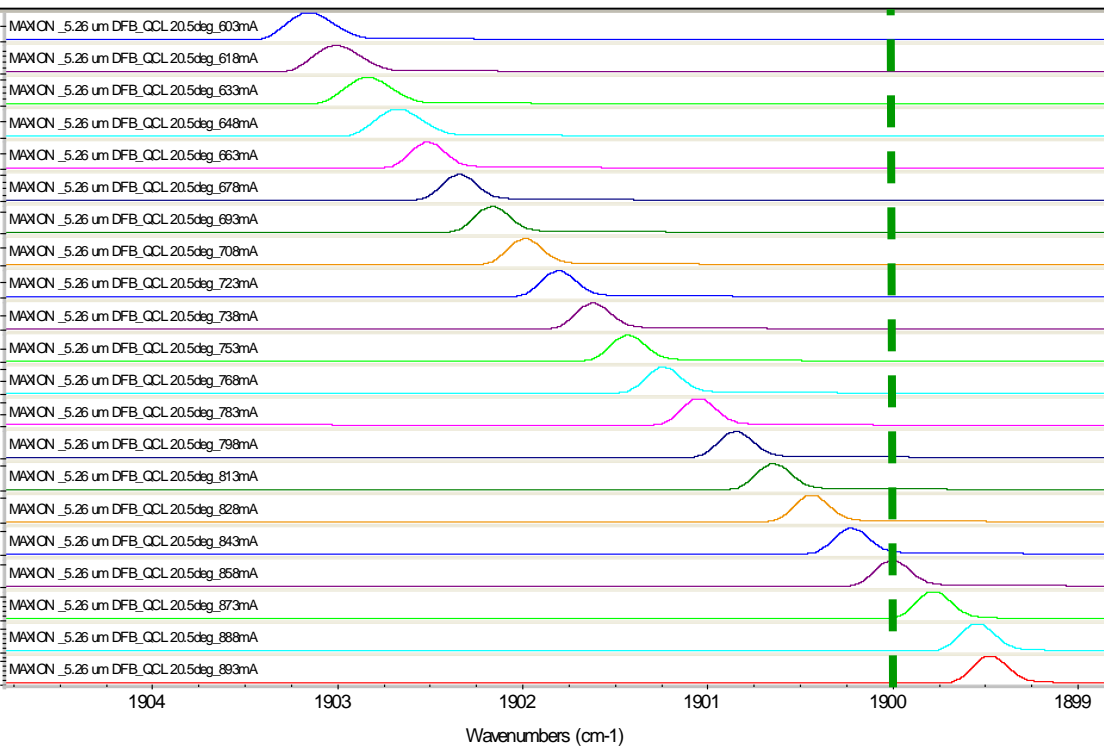


Emission Spectra of a 1900cm^{-1} TEC DFB QCL and HITRAN simulated spectra of NO, H₂O & CO₂

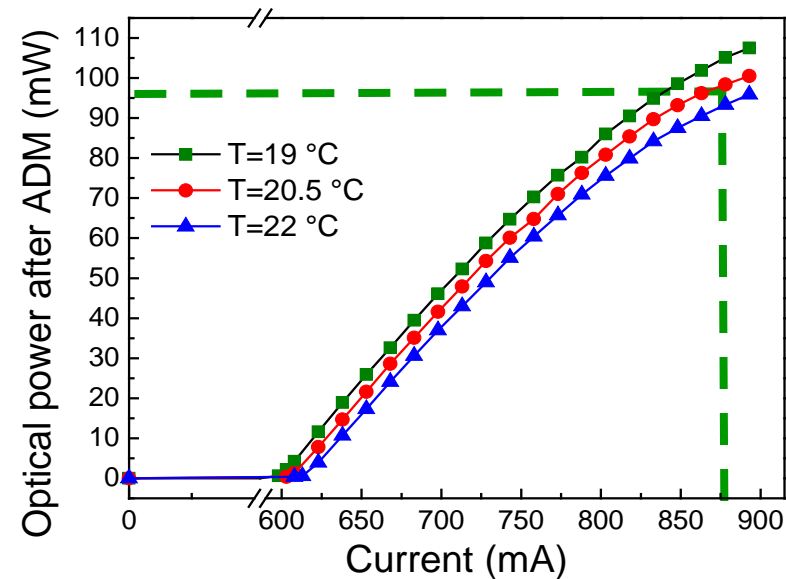


Output power: 117 mW @ 25 C
Thorlabs/Maxion

Performance of a 5.26 μm CW HHL TEC DFB-QCL



Single frequency QCL radiation recorded with FTIR for different laser current values at a QCL temperature of 20.5°C.

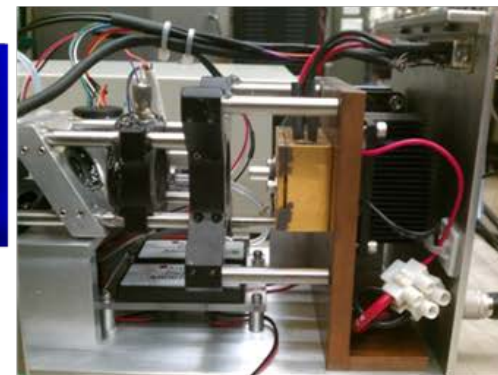
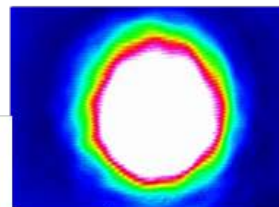
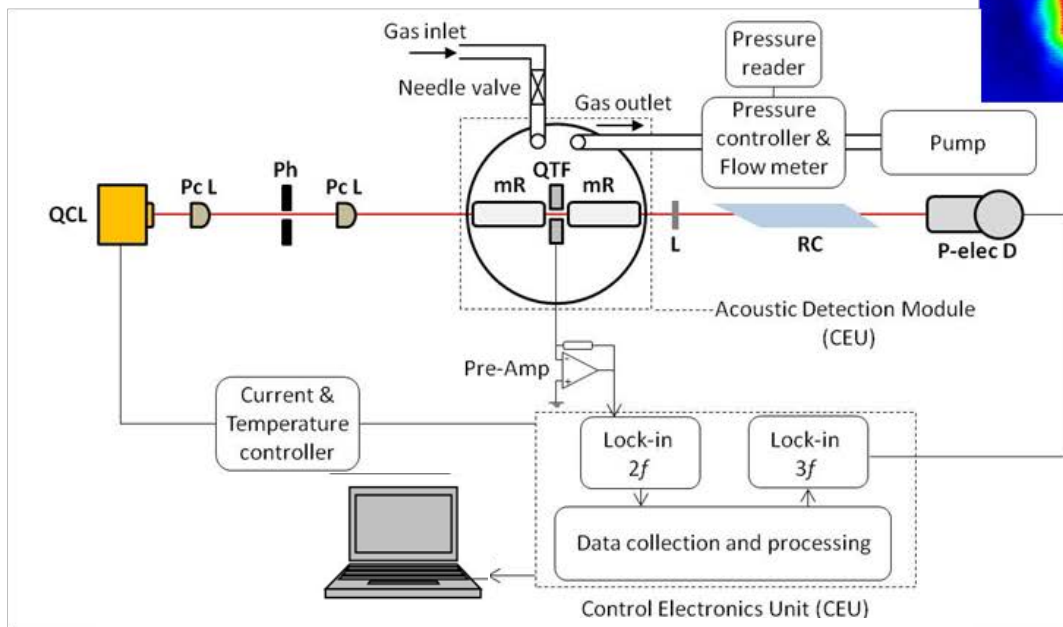


CW DFB-QCL optical power and current tuning at three different temperatures.



RICE

CW TEC DFB QCL based QEPAS NO Gas Sensor



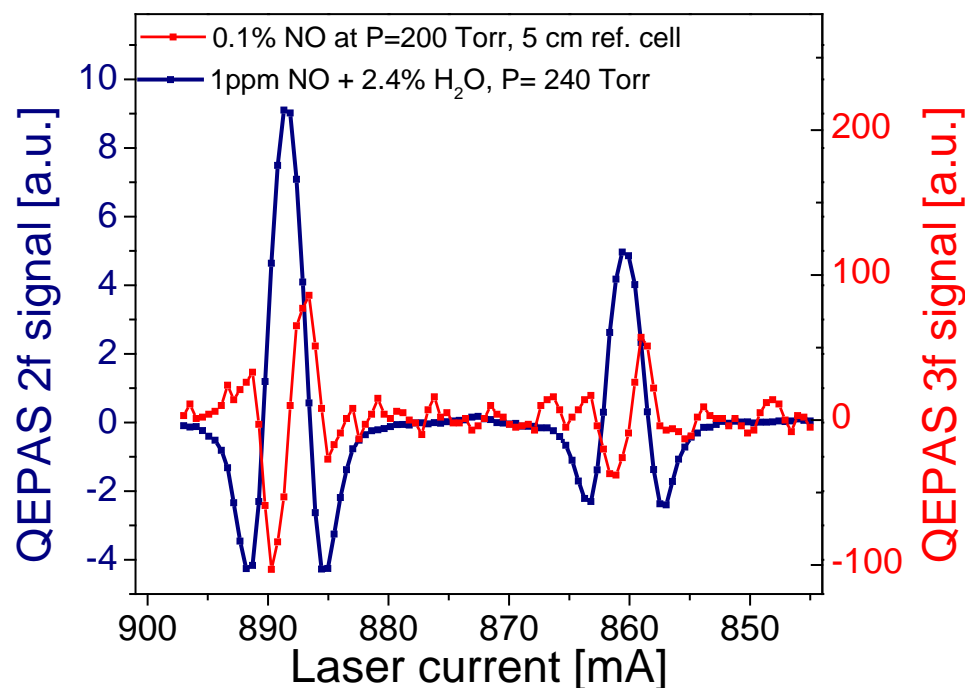
CW HHL TEC DFB-QCL package and IR camera image of the laser beam at 630 mA and 20.5 deg C through tubes after ADM

Schematic of a DFB-QCL based Gas Sensor.
PcL – plano-convex lens, Ph – pinhole,
QTF – quartz tuning fork, mR – microresonator,
RC- reference cell, P-elec D – pyro electric detector

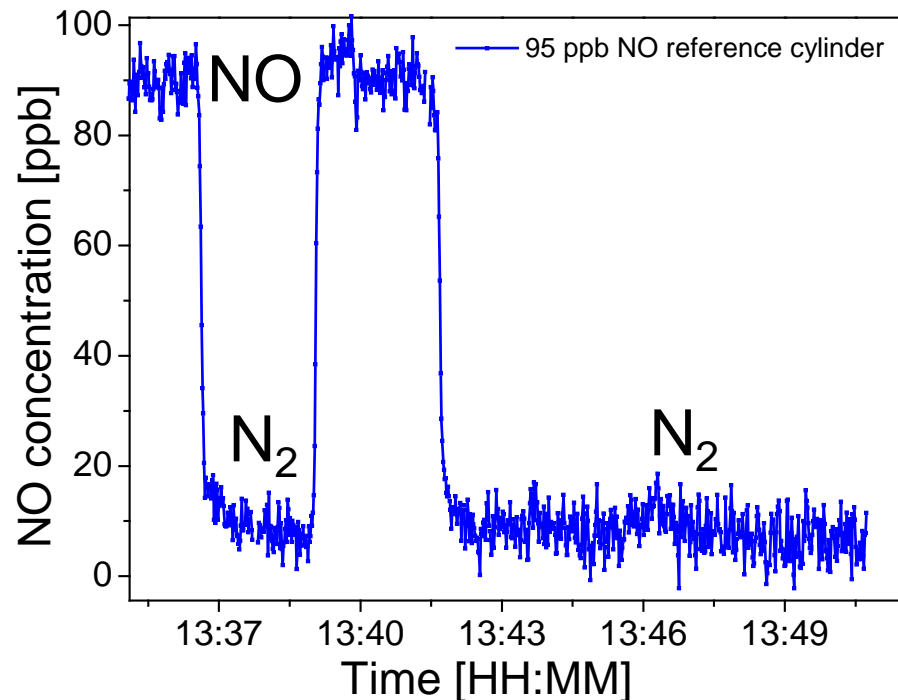


Compact Prototype NO Sensor
(September 2012)

Performance of CW DFB-QCL based WMS QEPAS NO Sensor Platform



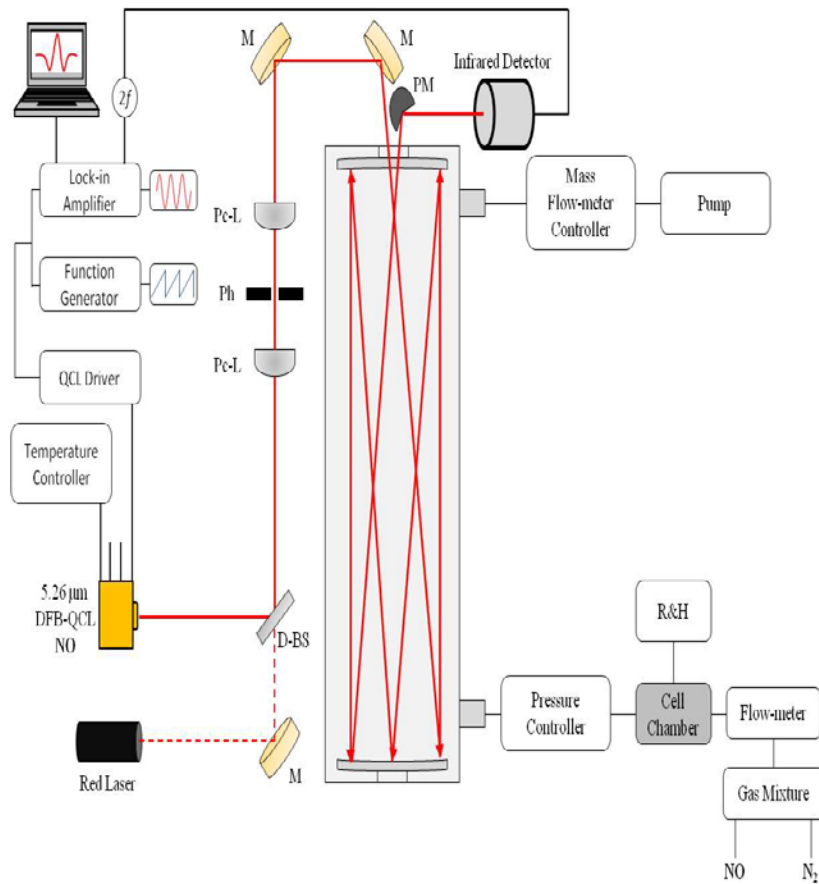
2f QEPAS signal (navy) and reference 3f signal (red) when DFB-QCL was tuned across **1900.08 cm⁻¹** NO line.



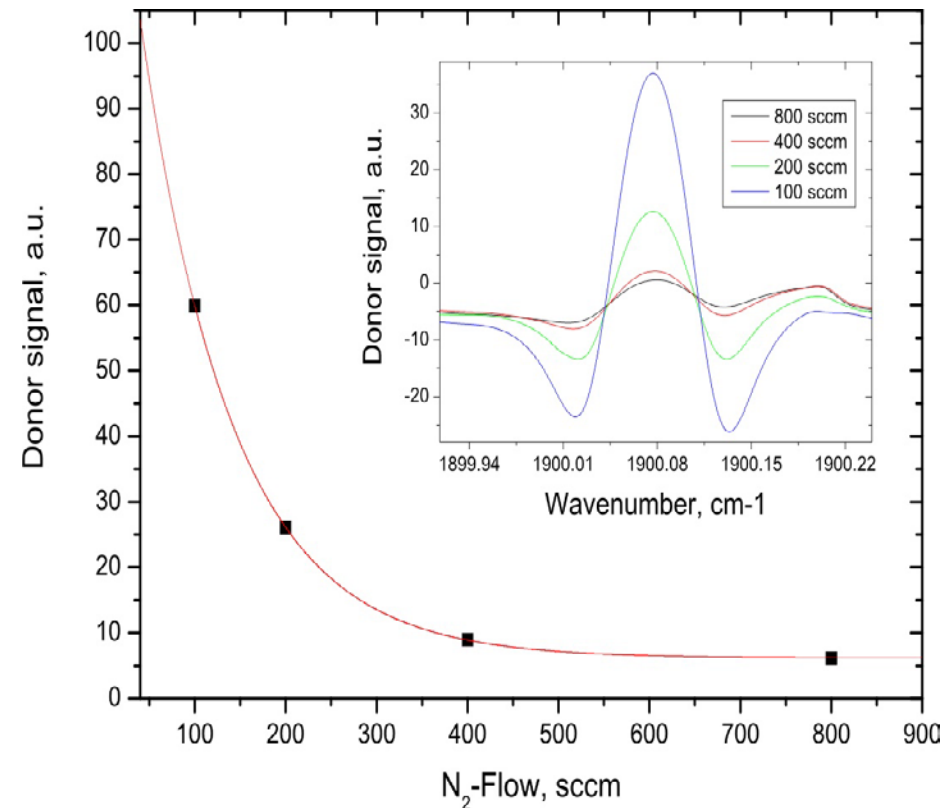
2f QEPAS signal amplitude for 95 ppb NO when DFB-QCL was locked to the **1900.08 cm⁻¹** line.

Minimum detectable NO concentration is:
~ 3 ppbv (1 σ ; 1 s time resolution)

QCL based TDLAS Sensor for Detection of NO Emission from Cancer Cells



Schematic drawing of the sensor setup



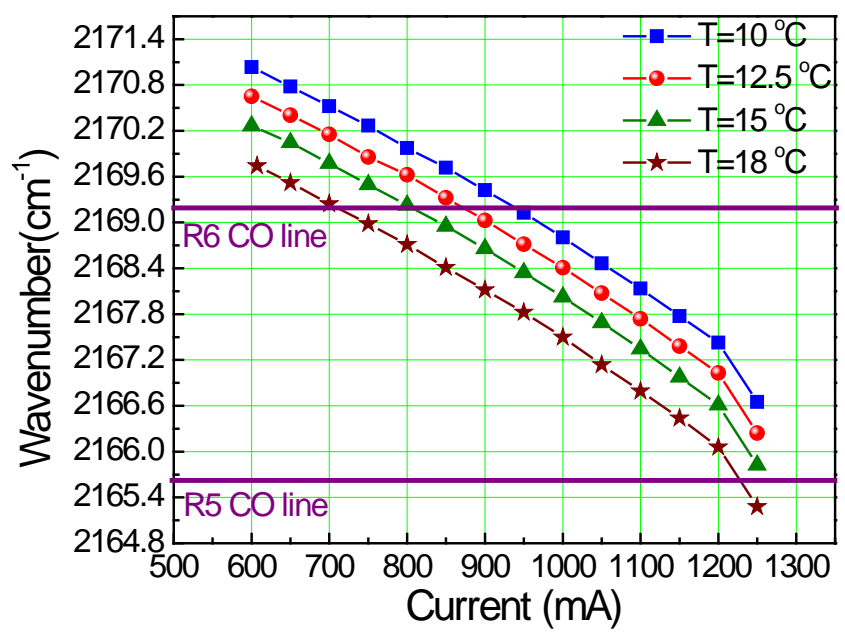
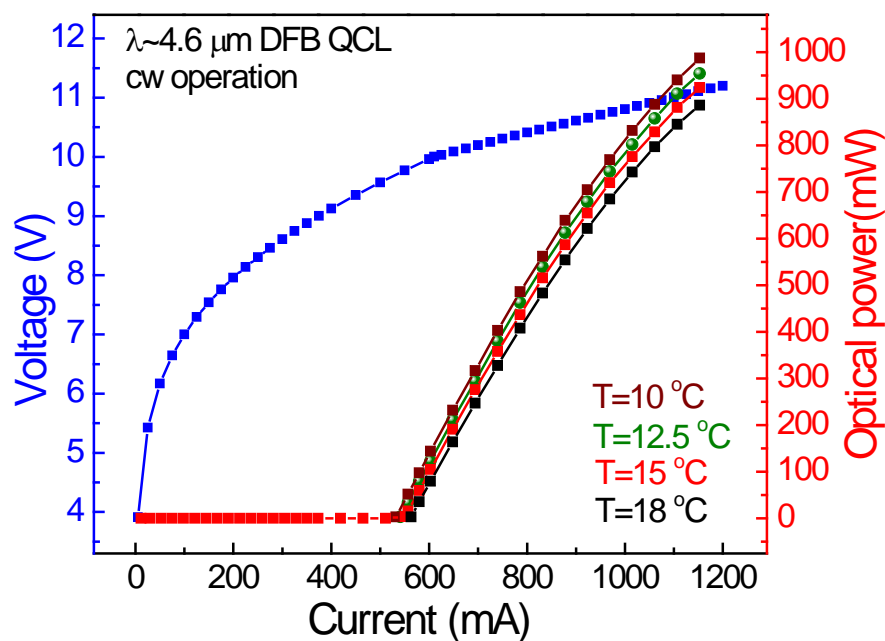
Dependence of the TDLAS sensor signal from biological samples on the gas flow (black squares). The inset shows spectra corresponding to the data points.

Motivation for Carbon Monoxide Detection

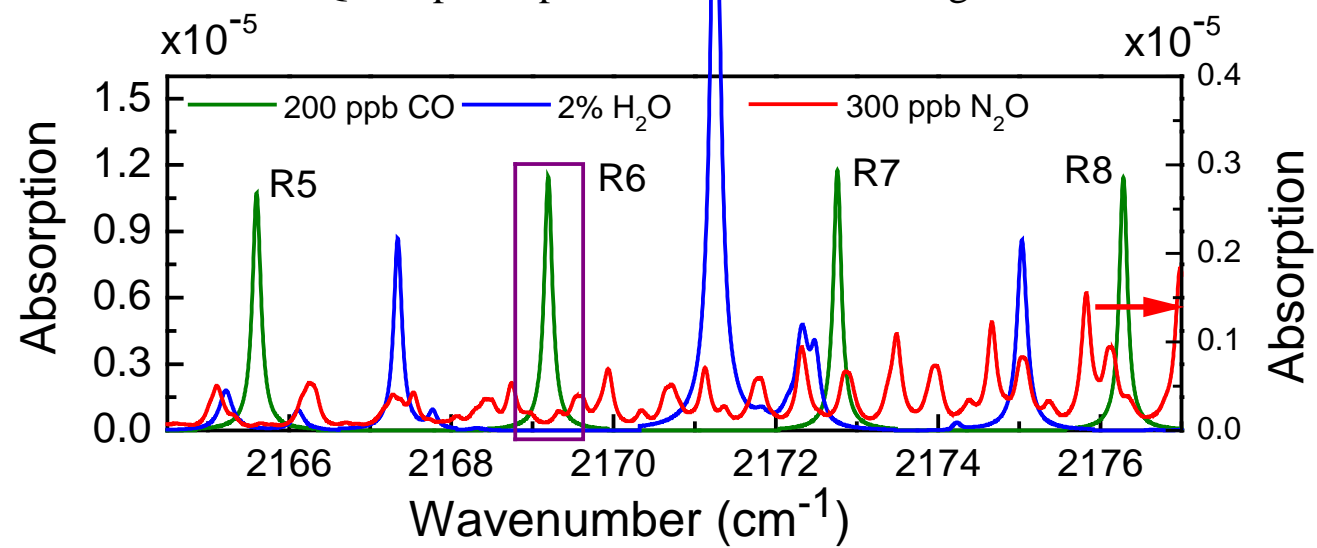
- **CO in Medical Diagnostics**
 - Hypertension and abnormality in heme metabolism
- **Public Health**
 - Extremely dangerous to human life even at a low concentrations. CO must be monitored at low concentration levels (<35 ppm).
- **Atmospheric Chemistry**
 - Incomplete combustion of natural gas, fossil fuel and other carbon containing fuels.
 - Impact on atmospheric chemistry through its reaction with hydroxyl (OH) for troposphere ozone formation and changing the concentration levels of greenhouse gases (e.g. CH₄).



Performance of a 4.61 μm high power CW TEC DFB QCL

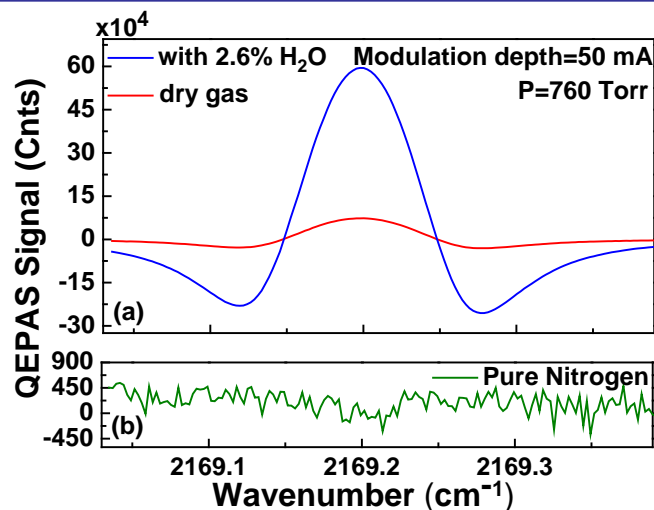


CW DFB-QCL optical power and current tuning at a four different QCL temperatures.

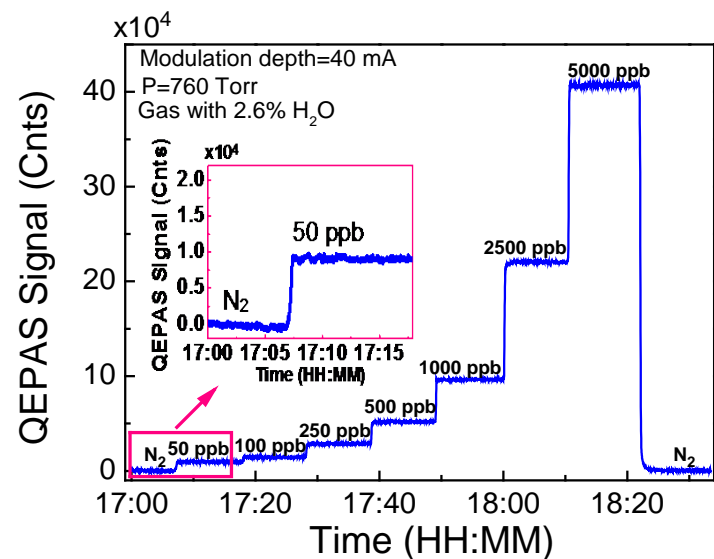


Estimated max wall-plug efficiency (WPE) is $\sim 7\%$ at 1.25A QCL drive-current.

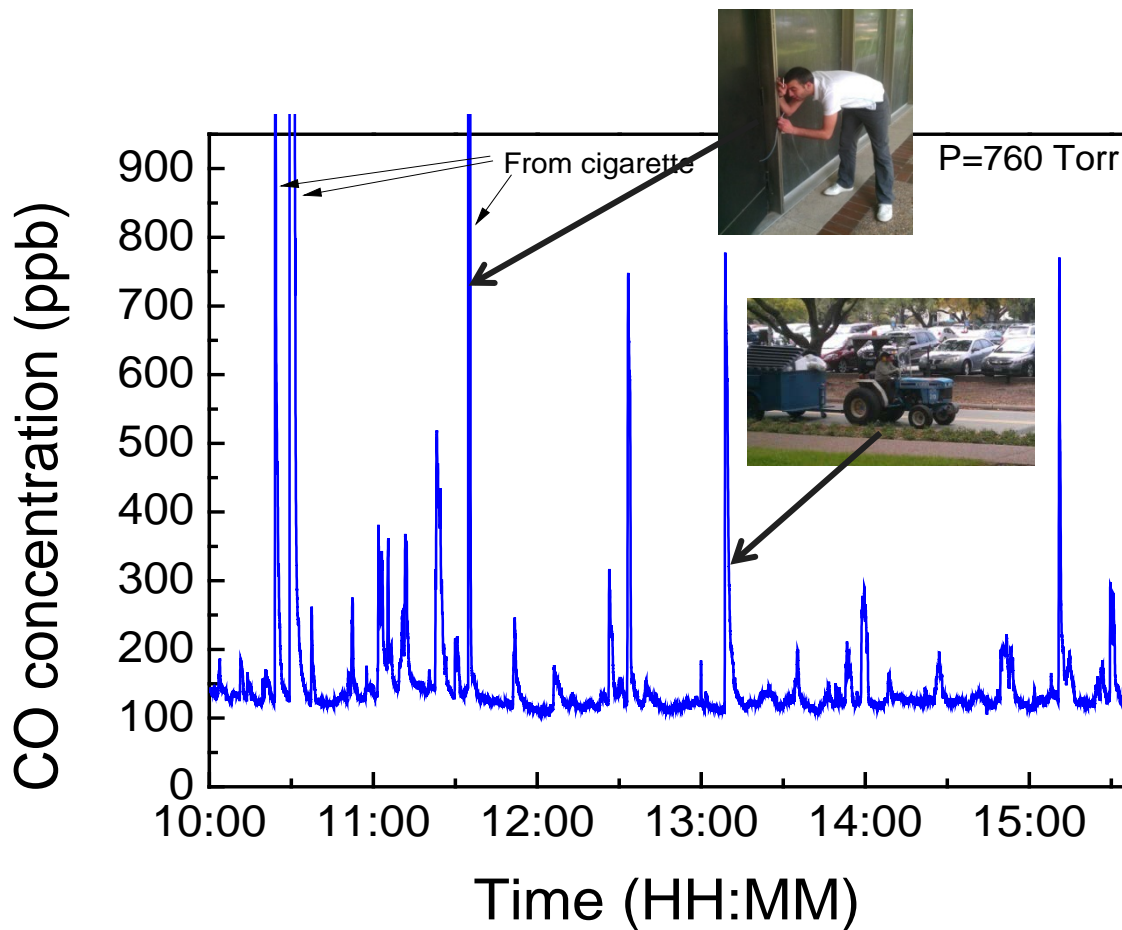
CW DFB-QCL based CO QEPAS Sensor Results



2f QEPAS signal for dry (red) and moisturized (blue) 5 ppm CO:N₂ mixture near **2169.2 cm⁻¹**.



Dilution of a 5 ppm CO reference gas mixture when the CW DFB-QCL is locked to the **2169.2 cm⁻¹** R6 CO line.



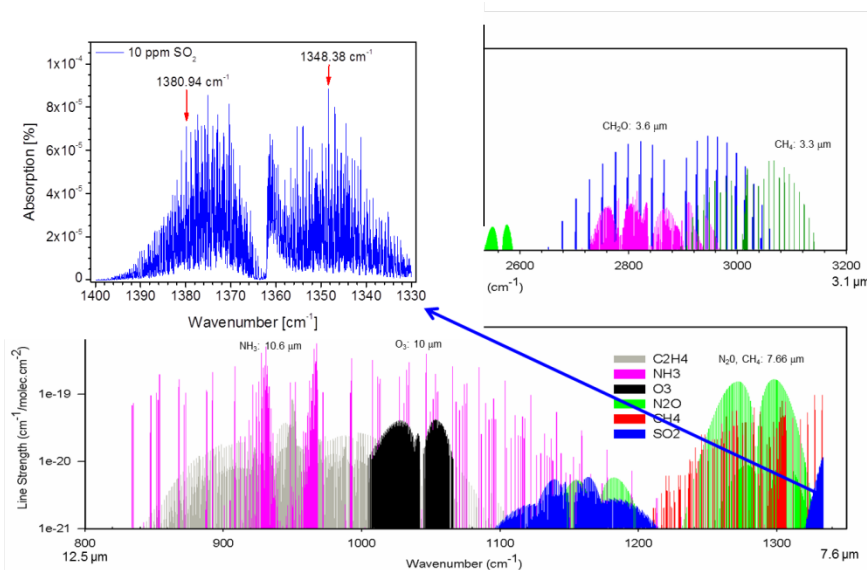
Atmospheric CO concentration levels on Rice University campus, Houston, TX

Minimum detectable CO concentration is:
~ 2 ppbv (1 σ ; 1 s time resolution)

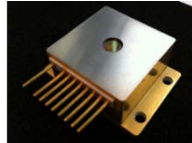
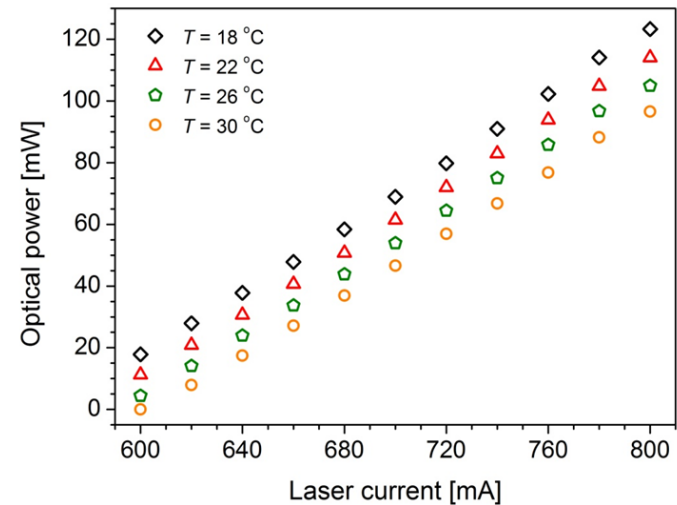
CW DFB-QCL based SO₂ QEPAS Results

Motivation for Sulfur Dioxide Detection

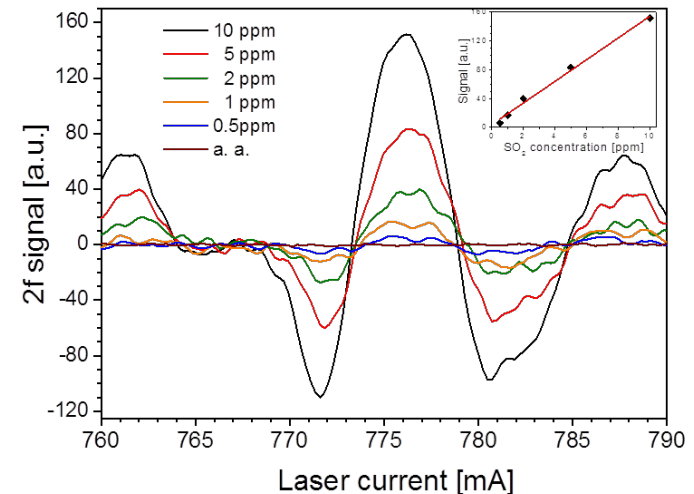
- SO₂ exposure affects lungs and causes breathing difficulties, bronchitis, cardiovascular disease
- Currently, reported annual average atmospheric SO₂ concentrations range from ~ 1 - 6 ppb
- Prominent air pollutant
- Emitted from coal fired power plants (~73%) and other industrial facilities (~20%)
- In the atmosphere SO₂ converts to sulfuric acid
➔ primary contributors to acid rain
- SO₂ reacts to form sulfate aerosols
- Primary SO₂ exposure for 1 hour is 75 ppb



Molecular Absorption Spectra within two Mid-IR Atmospheric Windows



7.24 μm CW DFB-QCL optical power and current tuning at three different operating temperatures.



2f WMS QEPAS signals for different SO₂ concentrations when laser was tuned across **1380.9 cm⁻¹** line.

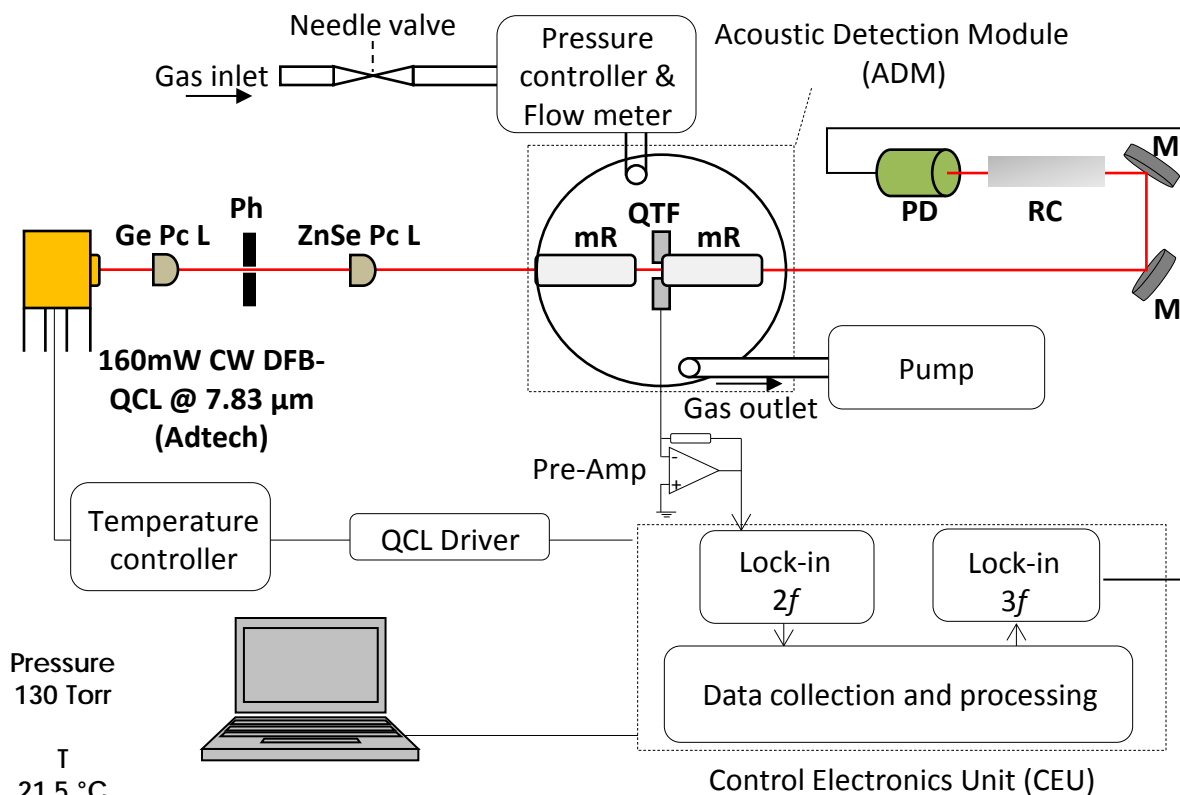
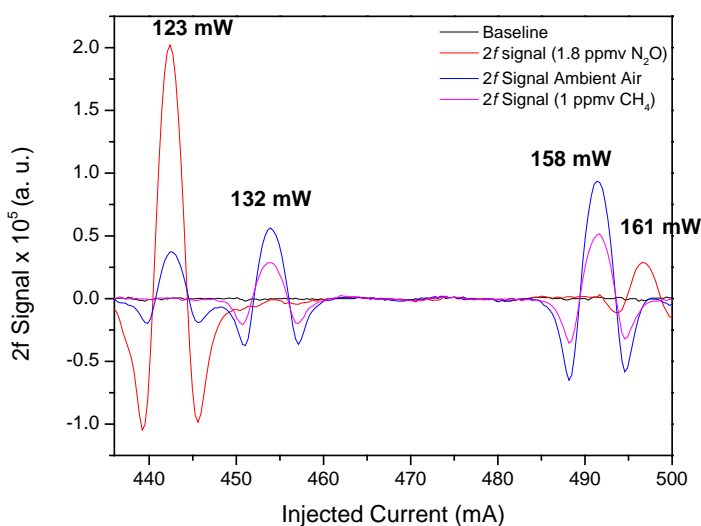
Minimum detectable SO₂ concentration is:

~ 100 ppbv (1 σ ; 1 s time resolution)

QEPAS based CH₄ and N₂O Gas Sensor

Motivation for CH₄ and N₂O Detection

- Medical Diagnostics
 - Nausea, blurred vision, vomiting
- Prominent greenhouse gases
- Sources: wetlands, leakage from natural gas systems, fossil fuel production and agriculture



Pressure
130 Torr

T
21.5 °C

AM
4 mA

f
32760 Hz

f_{mod}
16380 Hz

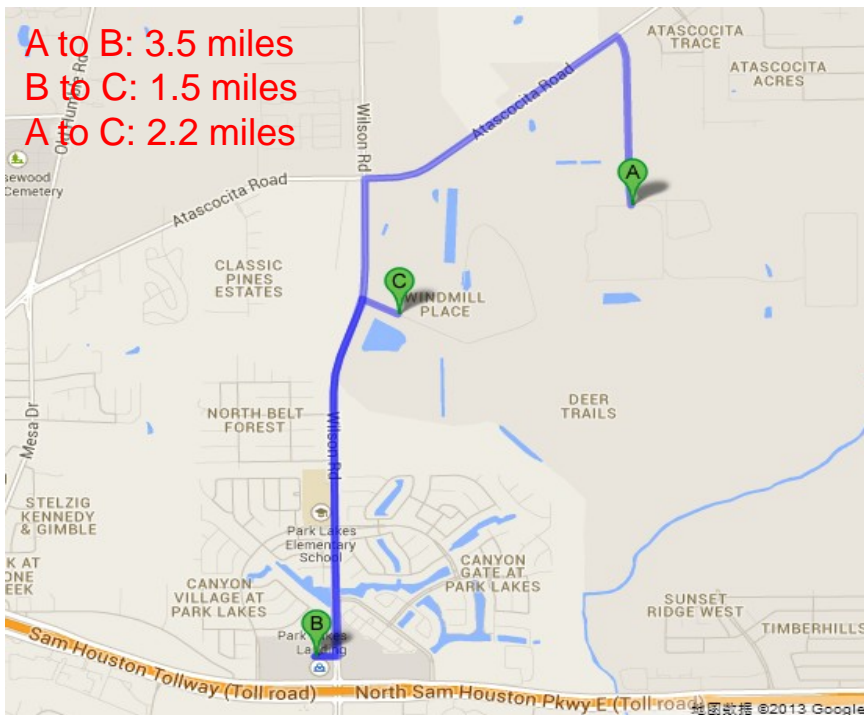
Detection Limit (1σ) with a 1-sec averaging time

Methane (CH₄) (1275.04 cm⁻¹) **13 ppbv**

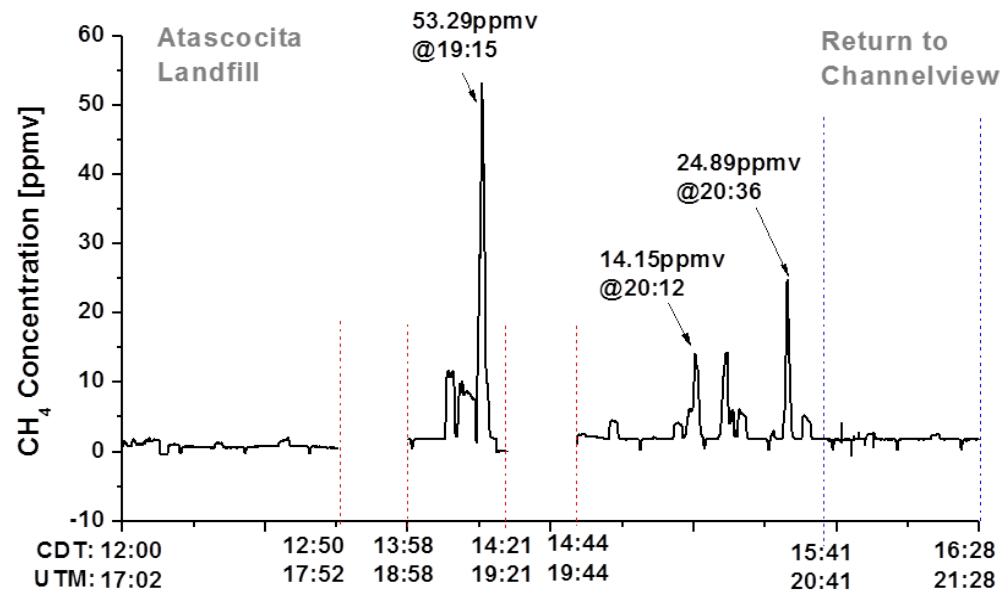
Nitrous Oxide (N₂O) (1275.5 cm⁻¹) **6 ppbv**

N₂O concentration in the ambient laboratory air:
331 ppbv

CH₄ Measurements performed with a DFB-QCL based QEPAS Sensor installed in the Aerodyne Mobile Laboratory (Sept 7, 2013)



Atascocita Landfill, Humble, TX 77396 CH₄ Perimeter Measurements

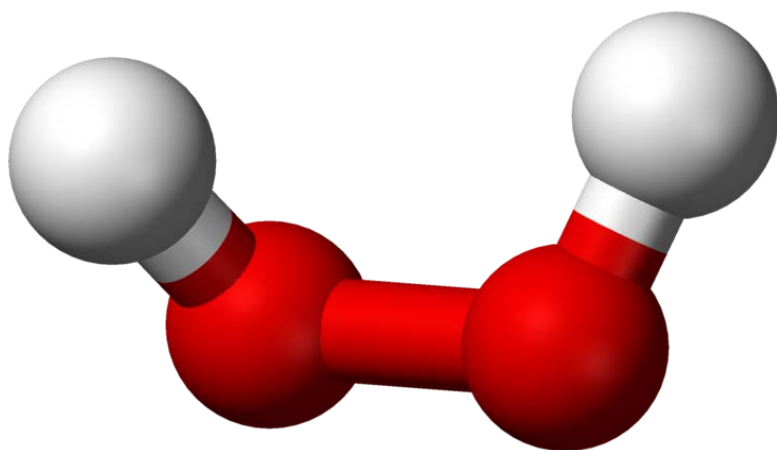


A: 29.9599° North, 95.2334° West
B: 29.9364° North, 95.2508° West
C: 29.9547° North, 95.2462° West (Landfill)

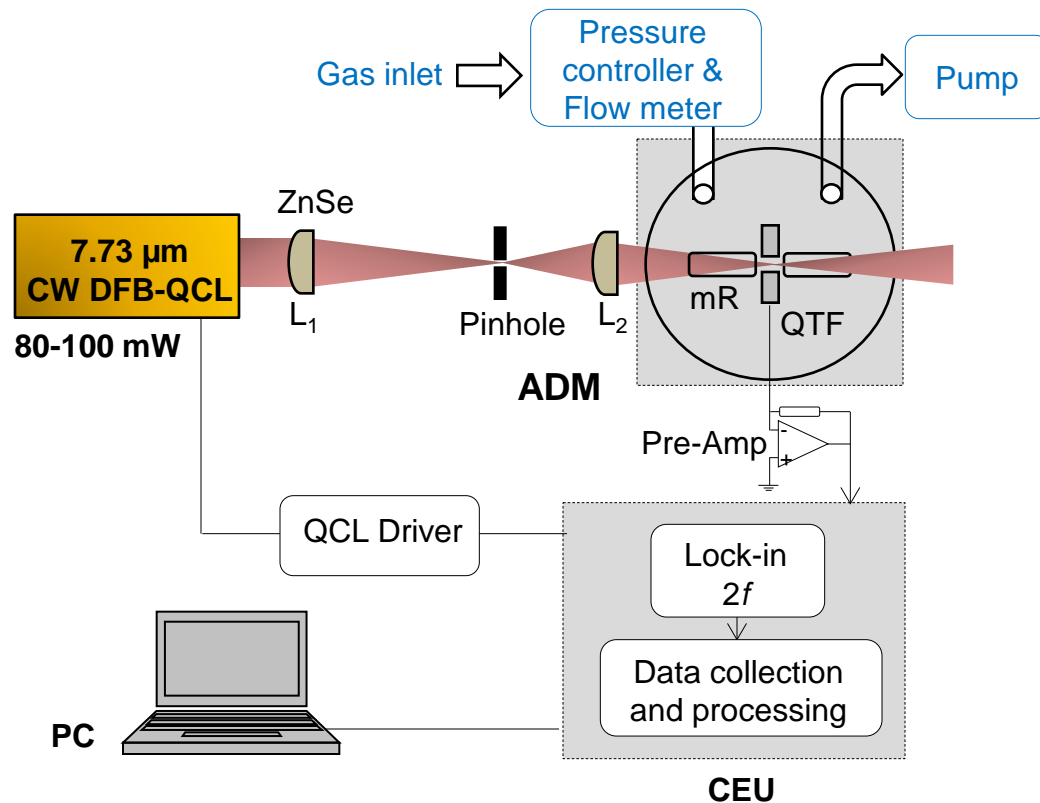


Motivation of H₂O₂ Detection

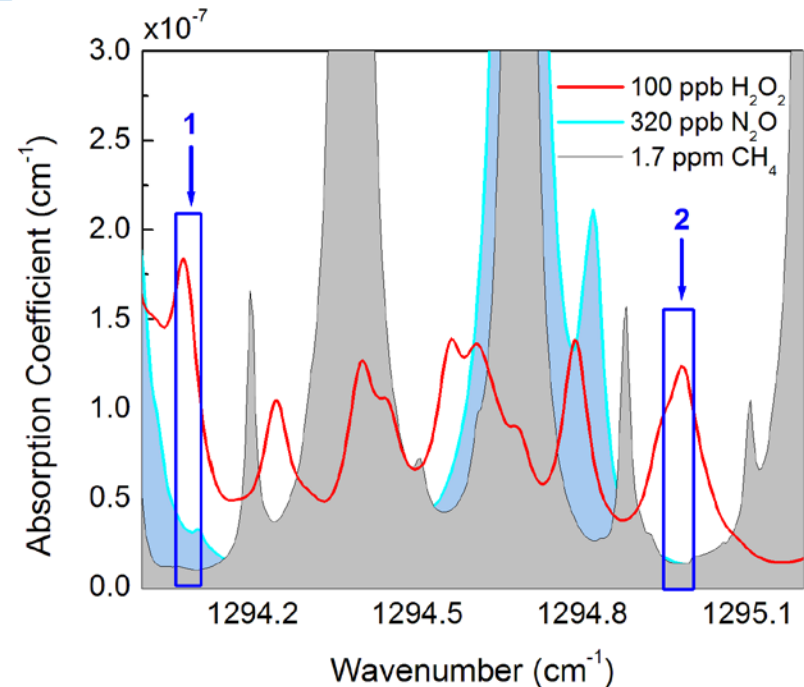
- Oxidative capacity of atmosphere and balance of HO_x;
- Acid rain formation & In-cloud oxidation of S(IV) to S(VI);
- Active agent in decontamination and sterilization systems;
- H₂O₂ in breath is a biomarker of oxidative stress;
- H₂O₂ concentration levels in Houston have not been reported despite of atmospheric conditions, such as high humidity, high solar radiation levels, and the presence of the petrochemical industry.



QEPAS based Hydrogen Peroxide (H_2O_2) Sensor System



Schematic of QCL based QEPAS sensor:
ADM – acoustic detection module; CEU – control electronics unit; PC – personal computer.



Simulated spectra (HITRAN) of H_2O_2 at 296 K and 130 Torr, along with atmospheric interfering molecules of CH_4 and N_2O ; two target wavelengths at 1294.1 and 1294.9 cm^{-1} are shown.

H_2O_2 Exposure limit is set at 1 ppmv by OSHA

QEPAS Performance for Trace Gas Species (September 2014)

	Molecule (Host)	Frequency, cm ⁻¹	Pressure, Torr	NNEA, cm ⁻¹ W/Hz ^{1/2}	Power, mW	NEC (τ=1s), ppmv
VIS	O ₃ (air)	35087.70	700	3.0×10 ⁻⁸	0.8	1.27
	O ₂ (N ₂)	13099.30	158	4.74×10 ⁻⁷	1228	13
	C ₂ H ₂ (N ₂)*	6523.88	720	4.1×10 ⁻⁹	57	0.03
NIR	NH ₃ (N ₂)*	6528.76	575	3.1×10 ⁻⁹	60	0.06
	C ₂ H ₄ (N ₂)*	6177.07	715	5.4×10 ⁻⁹	15	1.7
	CH ₄ (N ₂ +1.2% H ₂ O)*	6057.09	760	3.7×10 ⁻⁹	16	0.24
	N ₂ H ₄	6470.00	700	4.1×10 ⁻⁹	16	1
	H ₂ S (N ₂)*	6357.63	780	5.6×10 ⁻⁹	45	5
Mid-IR	HCl (N ₂ dry)	5739.26	760	5.2×10 ⁻⁸	15	0.7
	CO ₂ (N ₂ +1.5% H ₂ O) *	4991.26	50	1.4×10 ⁻⁸	4.4	18
	CH ₂ O (N ₂ :75% RH)*	2804.90	75	8.7×10 ⁻⁹	7.2	0.12
	CO (N ₂ +2.2% H ₂ O)	2176.28	100	1.4×10 ⁻⁷	71	0.002
	CO (propylene)	2196.66	50	7.4×10 ⁻⁸	6.5	0.14
	N ₂ O (air+5%SF ₆)	2195.63	50	1.5×10 ⁻⁸	19	0.007
	C ₂ H ₅ OH (N ₂)**	1934.2	770	2.2×10 ⁻⁷	10	90
	NO (N ₂ +H ₂ O)	1900.07	250	7.5×10 ⁻⁹	100	0.003
	H ₂ O ₂	1295.6	150	4.6×10 ⁻⁹	100	12
	C ₂ HF ₅ (N ₂)***	1208.62	770	7.8×10 ⁻⁹	6.6	0.009
	NH ₃ (N ₂)*	1046.39	110	1.6×10 ⁻⁸	20	0.006
	SF ₆	948.62	75	2.7×10 ⁻¹⁰	18	5×10 ⁻⁵ (50 ppt)

* - Improved microresonator

** - Improved microresonator and double optical pass through ADM

*** - With amplitude modulation and metal microresonator

NNEA – normalized noise equivalent absorption coefficient.

NEC – noise equivalent concentration for available laser power and τ=1s time constant, 18 dB/oct filter slope.

For comparison: conventional PAS $2.2 \times 10^{-9} \text{ cm}^{-1}\text{W}/\sqrt{\text{Hz}}$ for NH₃

Use of Canines in non-invasive & sensitive Cancer Detection

Bladder Cancer

Urine

Sensitivity 73%

Specificity 56-92%

Prostate Cancer

urine

Sensitivity 99%

Melanoma

Skin VOCs

Potential!



Ovarian cancer

Carcinoma Tissue

Sensitivity 100%, Specificity 98%

Blood

Specificity 100%, Sensitivity 95%

Lung Cancer

Breath

Sensitivity 99%

Specificity 99%

Breast Cancer

Breath

Sensitivity 88%

Specificity 98%

Colorectal cancer

Sensitivity Specificity

Breath 91% 99%

Stool 97% 99%

Breath 2014, Torun, Prof. T. Jezierski et al.,
Institute of Genetics and Animal Breeding, PAS



RICE

Advantages & Disadvantages of Canines in Cancer Detection

- **Advantages**

- Non-invasive, safe and easy sample collecting
- Relatively easy training and interpretation of dogs' indications
- Odor samples can be tested several times
- Extremely high detection sensitivity and specificity
- Potential of VOCs are useful in search, rescue and emergency applications

- **Disadvantages**

- To-date a “black-box technology”
- It is a method based on earning a reward, which becomes unreliable after ~ 4 years
- Variation of sensitivity and specificity
- Re-training of dogs is not effective

Merits of QEPAS based Trace Gas Detection

- Very small sensing module and sample volume (a few mm³ to ~2cm²)
- Extremely low dissipative losses
- Optical detector is not required
- Wide dynamic range
- Frequency and spatial selectivity of acoustic signals
- Rugged transducer – quartz monocrystal; can operate in a wide range of pressures and temperatures
- Immune to environmental acoustic noise, sensitivity is limited by the fundamental thermal TF noise: $k_B T$ energy in the TF symmetric mode
- Absence of low-frequency noise: SNR scales as \sqrt{t} , up to $t=3$ hours as experimentally verified

QEPAS: some challenges

- Cost of Spectrophone assembly
- Sensitivity scales with laser power
- Effect of H₂O
- Responsivity depends on the speed of sound and molecular energy transfer processes
- Cross sensitivity issues

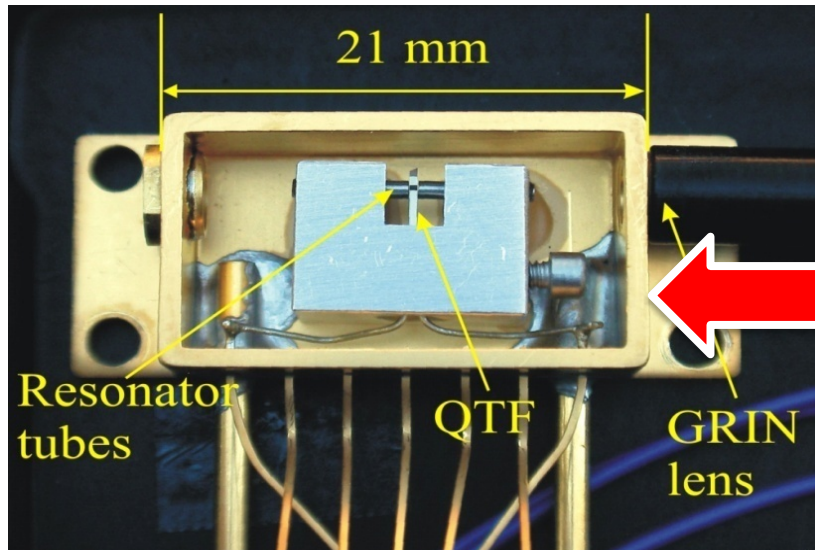


Future Directions and Outlook

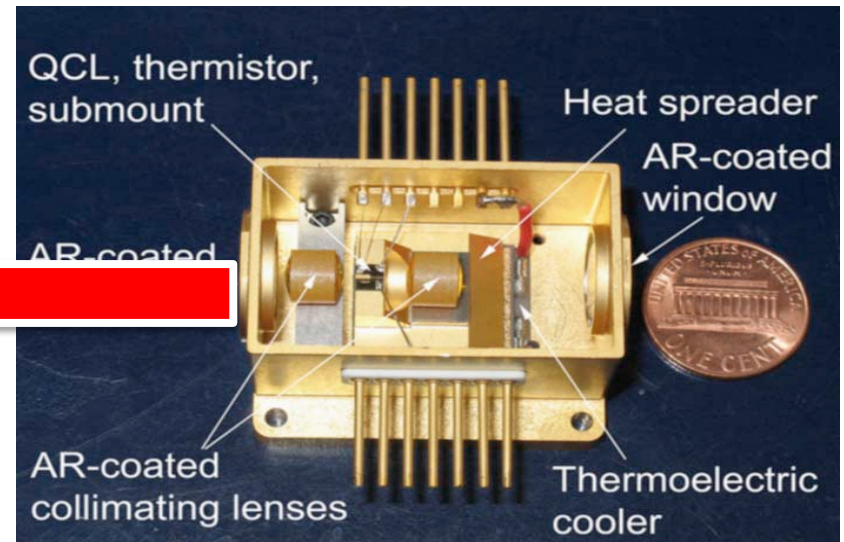
- New target analytes: formaldehyde (CH_2O), ethylene (C_2H_4), ozone (O_3) and nitrate (NO_3)
- Ultra-compact, low cost, robust sensors (e.g. CH_4 , NO, CO...)
- QCL based ultra-portable atmospheric carbon isotope monitor for $^{12}\text{CH}_4$ & $^{13}\text{CH}_4$
- Monitoring of broadband absorbers: acetone ($\text{C}_3\text{H}_6\text{O}$): MDL of 1.5 ppm with a 7mW ICL & AM, or 20ppb with a 100mW QCL @ $8.23\mu\text{m}$; benzene (C_6H_6)...
- Optical power build-up cavity designs (I-QEPAS)
- THz QEPAS based sensors
- Development of trace gas sensor networks



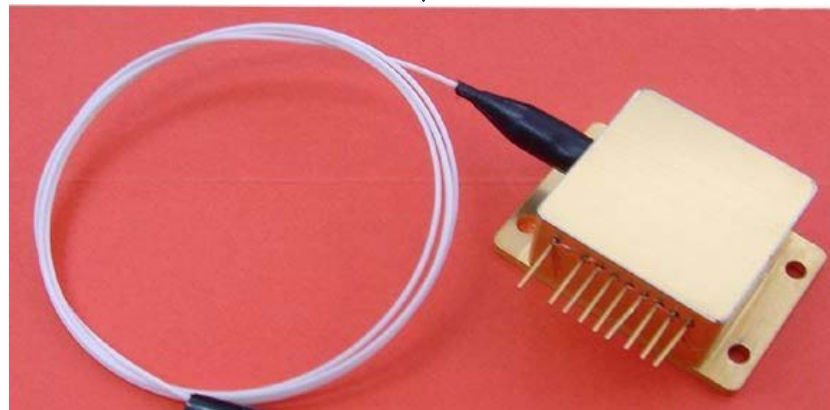
Potential Integration of a CW DFB- QCL and QEPAS Absorption Detection Module



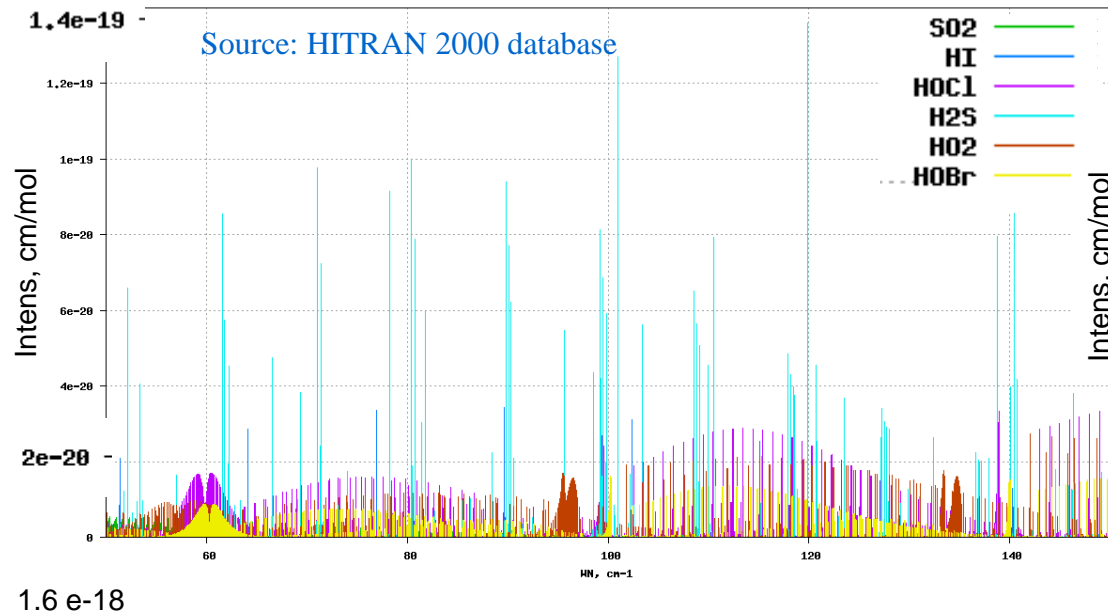
2012 QEPAS ADM



HHL package fiber coupled DFB-QCL



Why is THz based Trace Gas Sensing useful ?



Several gas species such as HF, OH, HCN, HCl, HBr, NH₃, H₂O₂, H₂S, H₂O & explosives (in the vapor phase) show strong absorption bands in the THz spectral range.

Mainly rotational levels are involved in THz absorption processes and rotational-translational (R-T) relaxation rates are **up to three order of magnitude faster** with respect to vibrational-translational (V-T) in the mid-infrared

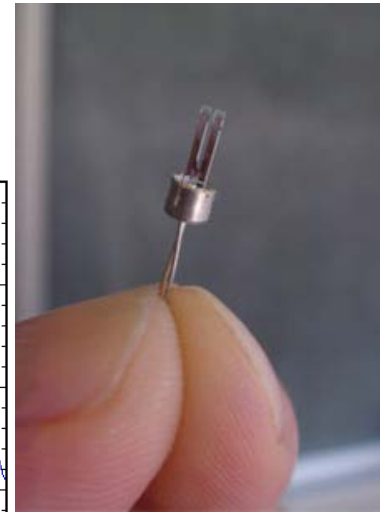
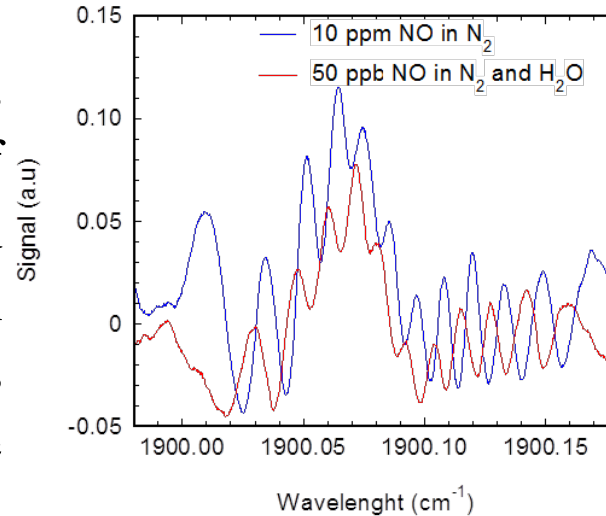


QEPAS signal strongly depends on the energy relaxation rates due to the possibility to **operate at low pressure**, & thereby taking advantages of the typically **very high QTF Q-factors**.

Why have QEPAS sensors not been developed in the THz spectral range so far?

Standard QTFs have a very small volume
($\sim 0.3 \times 0.3 \times 3 \text{ mm}^3$)

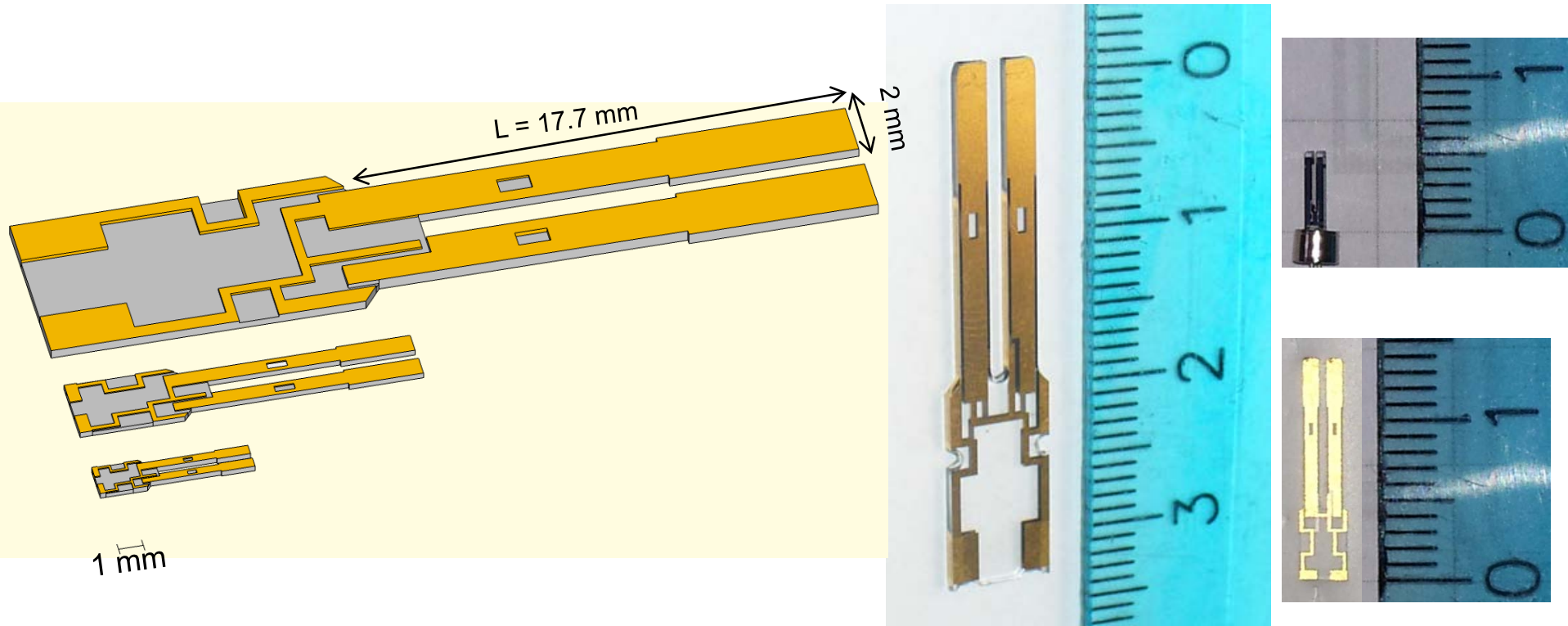
In QEPAS sensor systems, it is critical to avoid laser illumination of the QTF, since the radiation blocked by the QTF prongs results in an undesirable non-zero background as well as a shifting fringe-like interference pattern.



Standard QTF

The standard QTF prong separation of 330 μm is comparable with the THz wavelength which prevents the use of a QEPAS sensor architecture in the THz range unless we use large sized QTFs.

Custom fabricated QTFs scaled in Dimensions (~7 & 3 times larger) with respect to a standard QTF



Standard photolithographic techniques were used to etch the custom QTF, starting from a z-cut quartz wafer. Chromium/gold contacts were deposited on both sides of the custom QTF.

Currently verification that the larger QTFs behave similar to a “standard” QTF in terms of vibrational modes and Q factor is in progress

THz QCL Sources via Nonlinear Optics

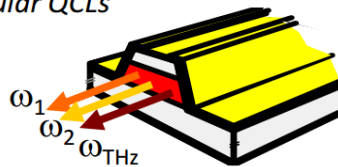
Use intra-cavity DFG in mid-IR QCLs



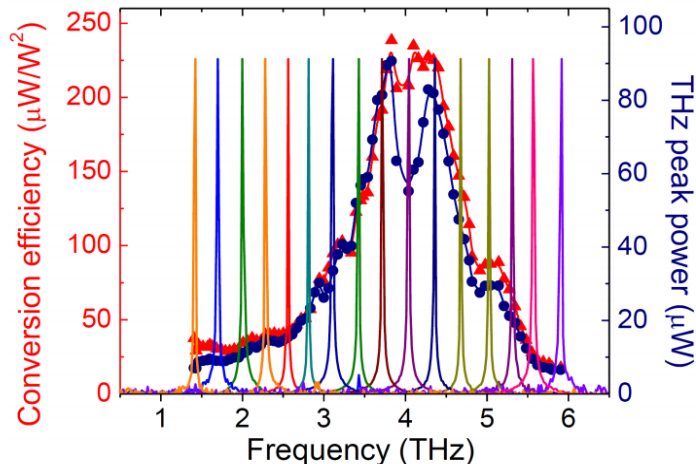
$$W(\omega_{THz}) \propto |\chi^{(2)}|^2 W(\omega_1) W(\omega_2) \times l_{eff}^2$$

THz QCL source based on intra-cavity DFG

- Same fabrication/user operation as regular QCLs
- Room temperature operation
- Broadband THz tuning

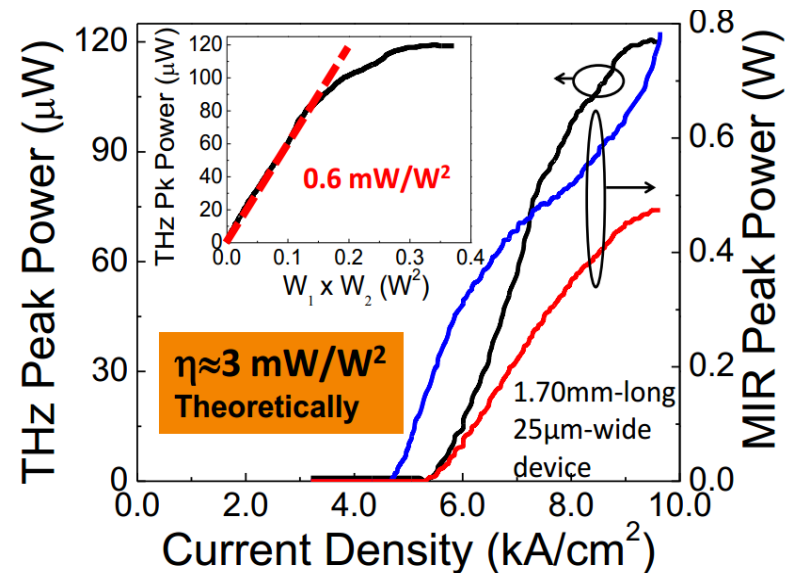


2.3-mm-long, 22- μ m-wide device @ room temperature



Vijayraghavan et al., *Nature Comm.* **4**, 2021 (2013); Jiang et al., *J. Opt.* **16**, 094002 (2014)

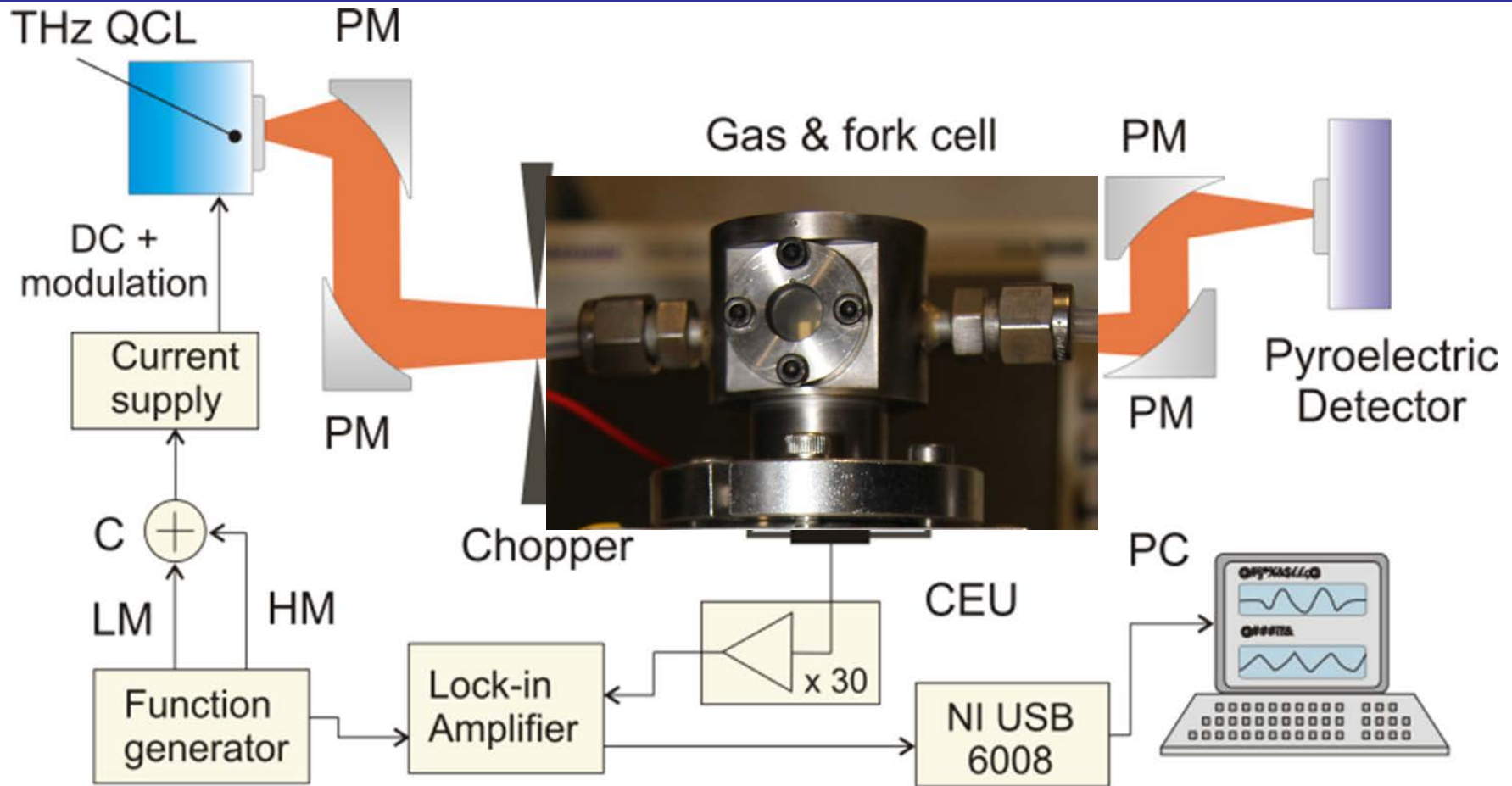
Monolithic tuners: 0.58 THz of tuning – Jung et. al., *Nature Comm.* **5**, 4267 (2014)



Vijayraghavan et al., *Nature Comm.* **4**, 2021 (2013)

IQCLSW 2014, Policore, Italy: M.A. Belkin et al, UT Austin, USA

THz QEPAS Sensor System for Methanol (CH_3OH) Detection



Legend

PM – parabolic mirror

QTF – quartz tuning fork

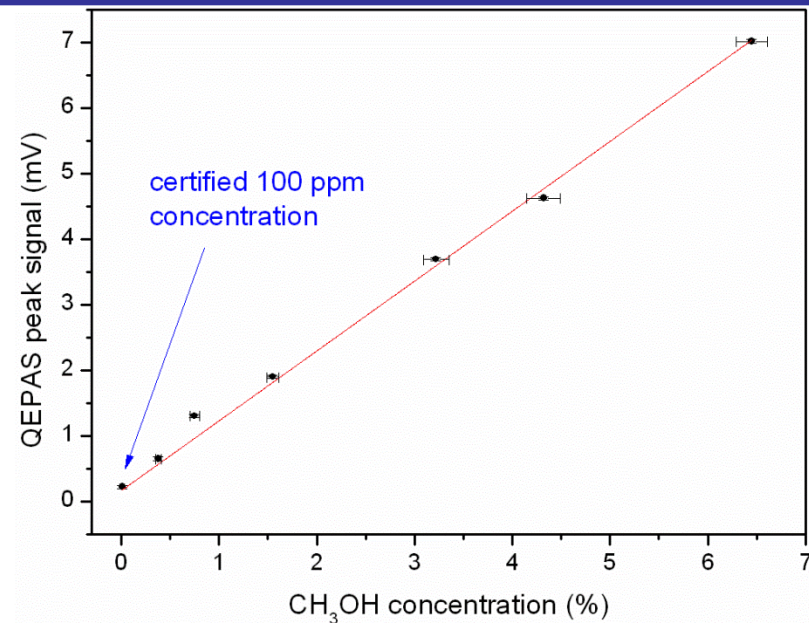
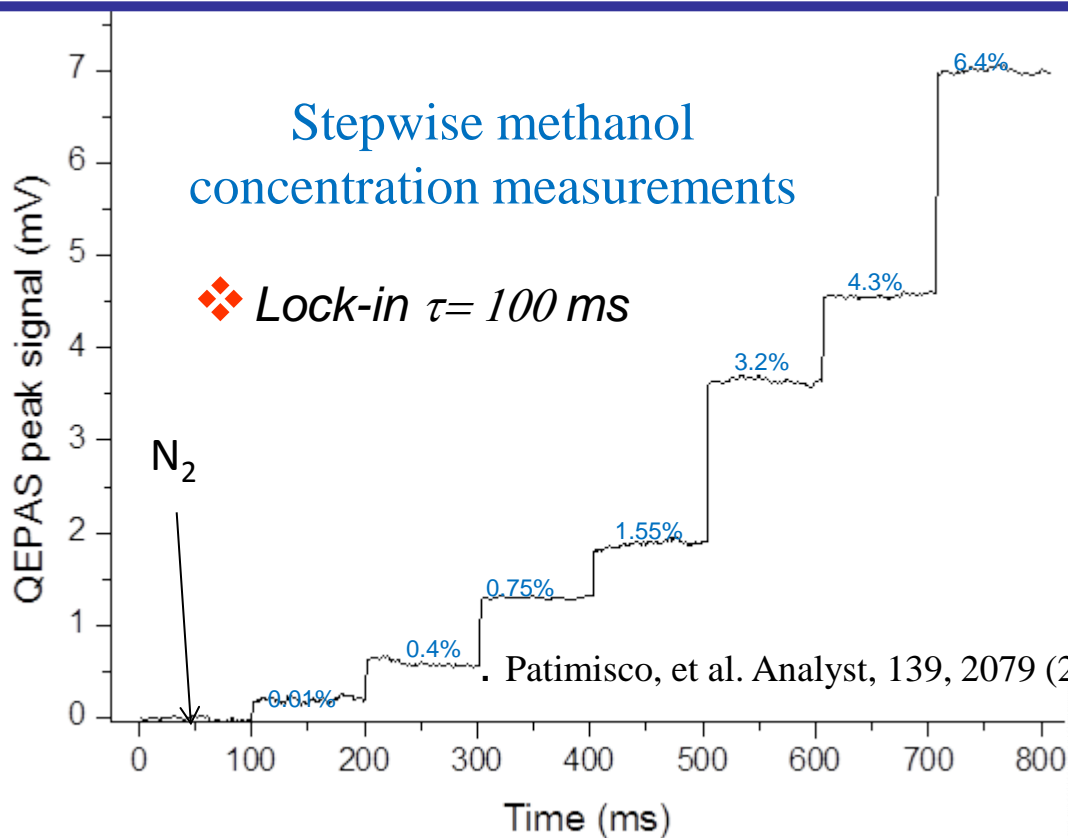
C – power combiner

LM - low-frequency modulation (ramp)

HM -high-frequency modulation

CEU- control electronics unit

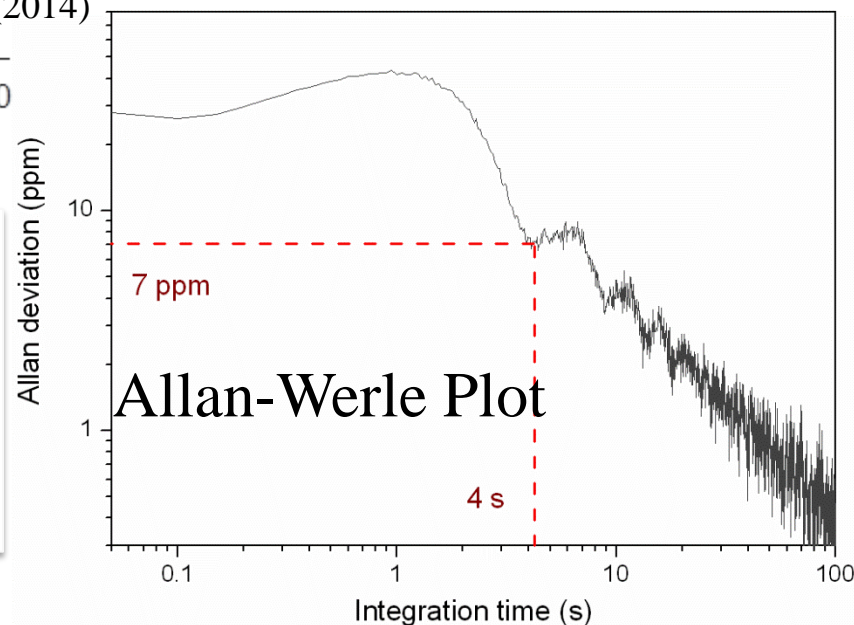
THz QEPAS Performance in Locked Mode and Long-term Stability



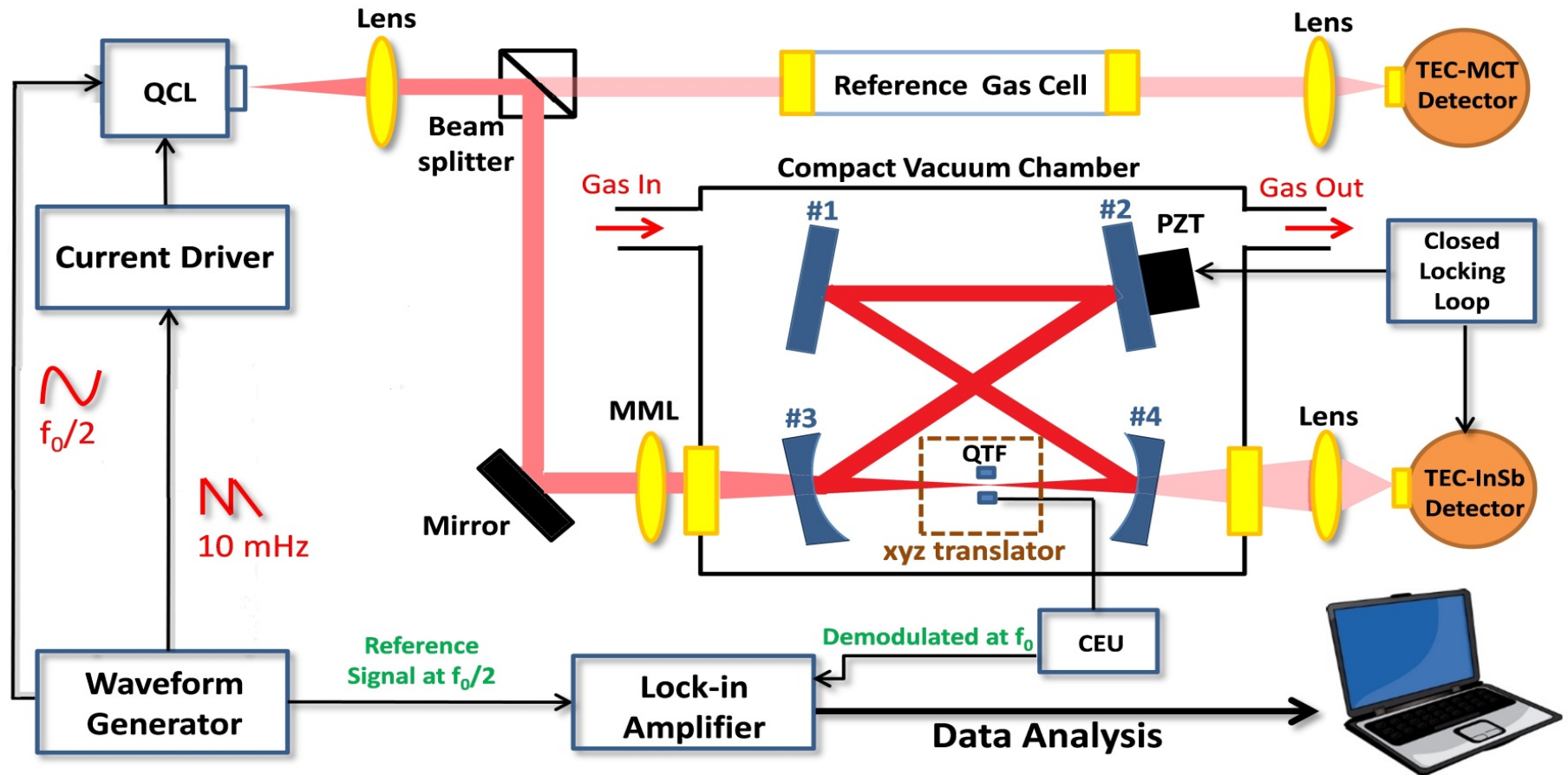
NEC = 7 ppm (@ a laser power of 40 μ W & a 4sec lock-in constant)

Absorption coefficient normalized to detection bandwidth and optical power:

$$NNEA = 2.0 \times 10^{-10} \text{ cm}^{-1} \text{ W}(\text{Hz})^{-1/2}$$



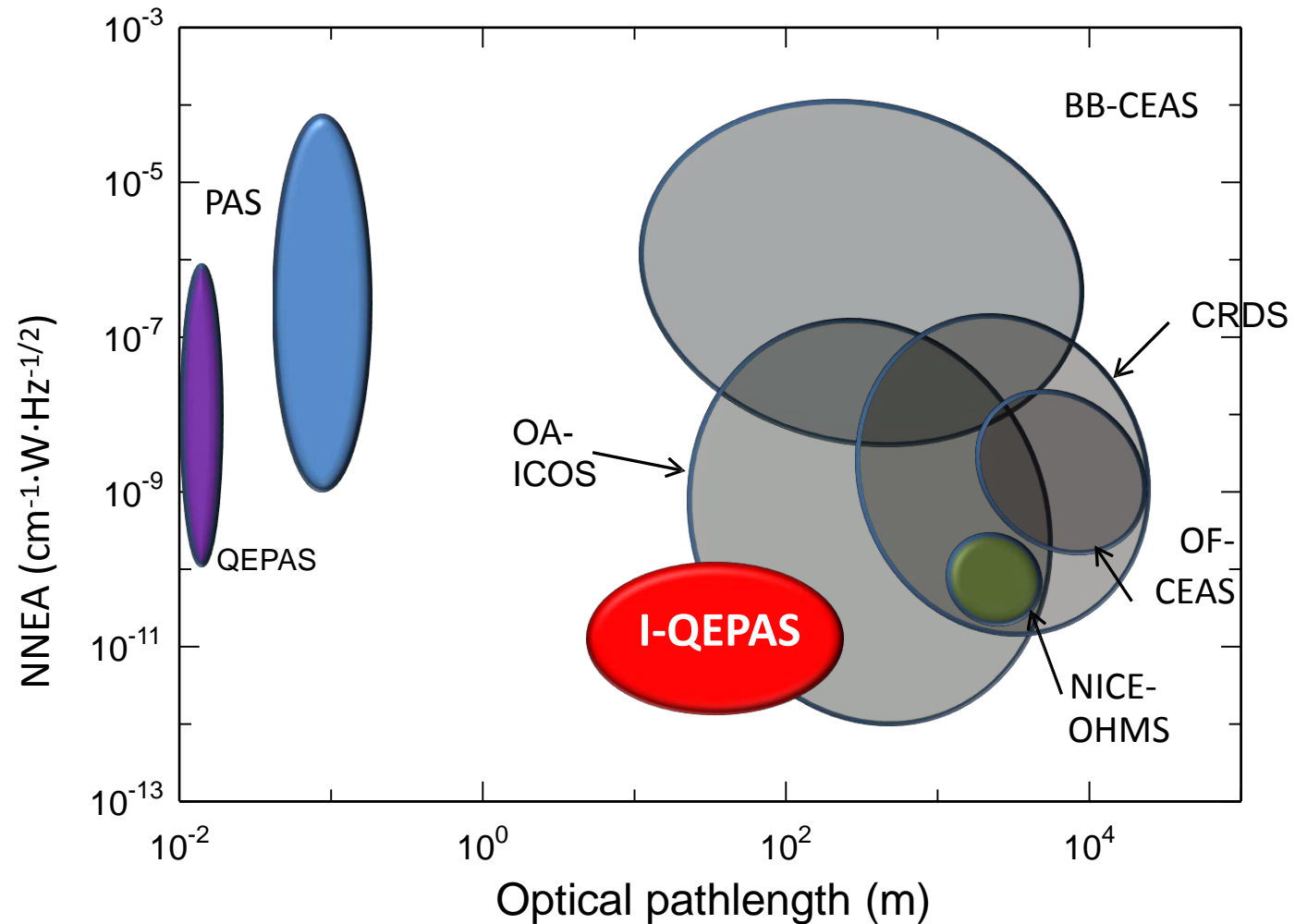
Proposed Intracavity-QEPAS (I-QEPAS) Sensor System



Optical power build up cavity can provide:

- RT CW DFB QCL, $\lambda=4.33\text{ microns}$
- Low noise current driver \rightarrow narrow QC laser linewidth $\sim 1\text{ MHz}$
- Bow-tie cavity \rightarrow 4 high reflectivity mirrors, $R=99.9\%$
- Electronic Control Loop + PZT driver lock of cavity resonant frequency to QCL frequency

Comparison of I-QEPAS with Other Trace Gas Sensing Techniques

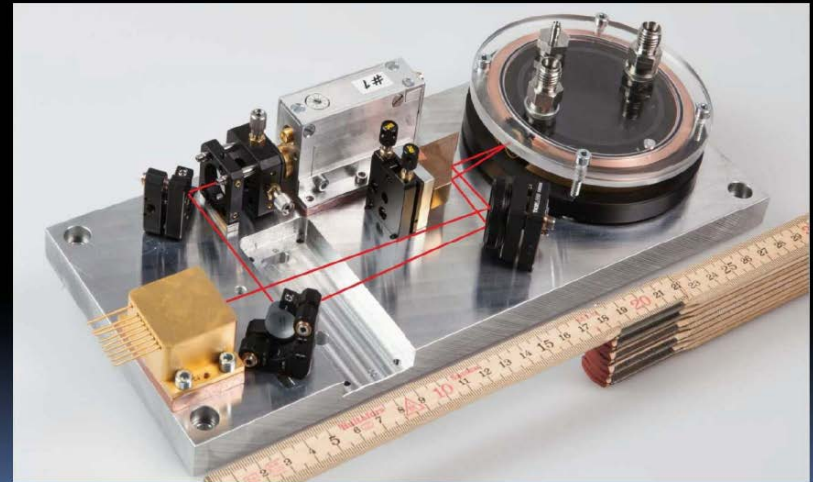


Cylindrical Multi-pass Trace Gas Absorption Cell

Absorption Cell

- No movable parts
- Toroidal design corrects optical aberrations
- Flexibility for alignment
- High path-to-volume ratio

Shoe-box size instrumentation

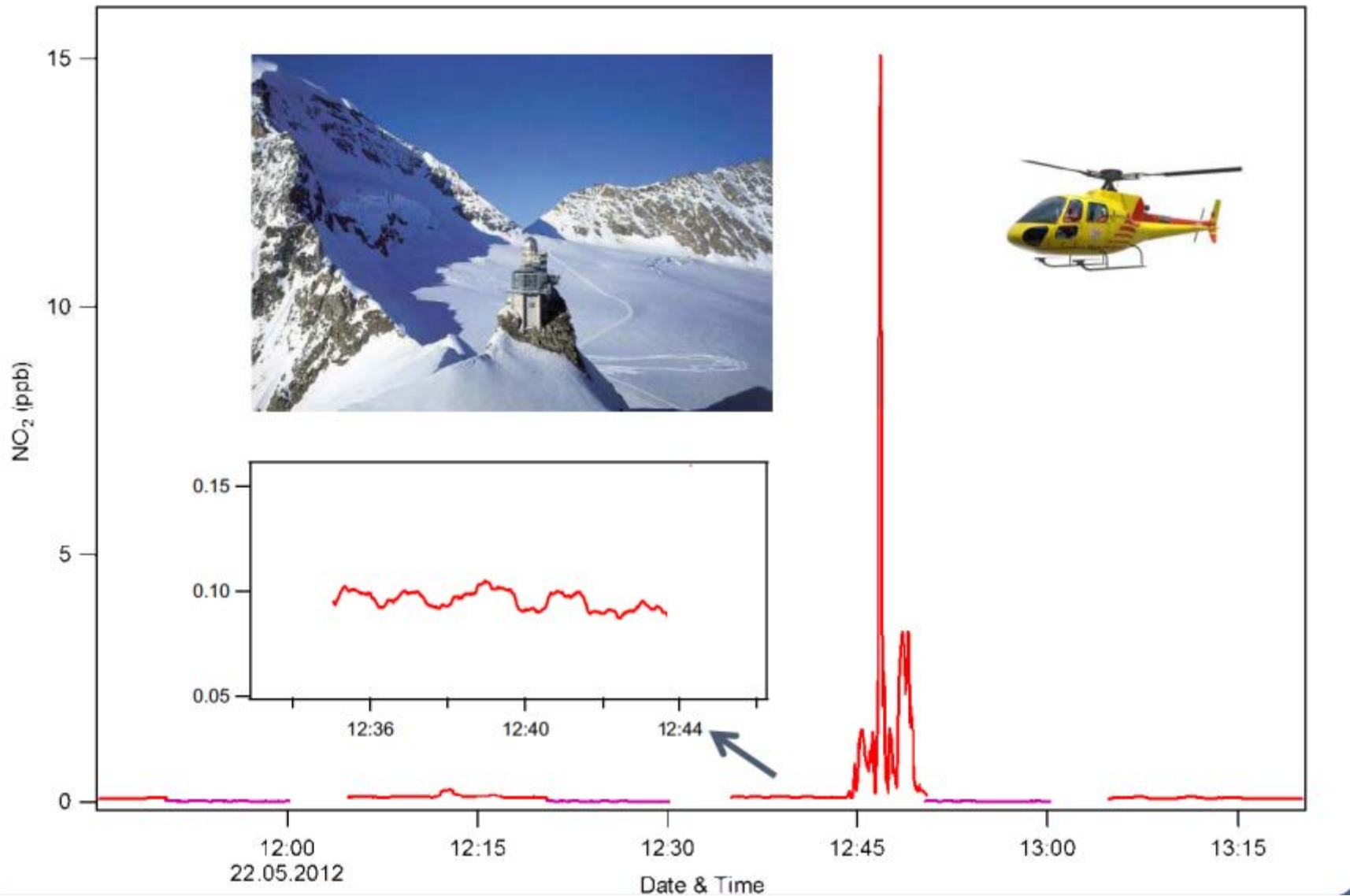


Jouy et al., Analyst, (2014).

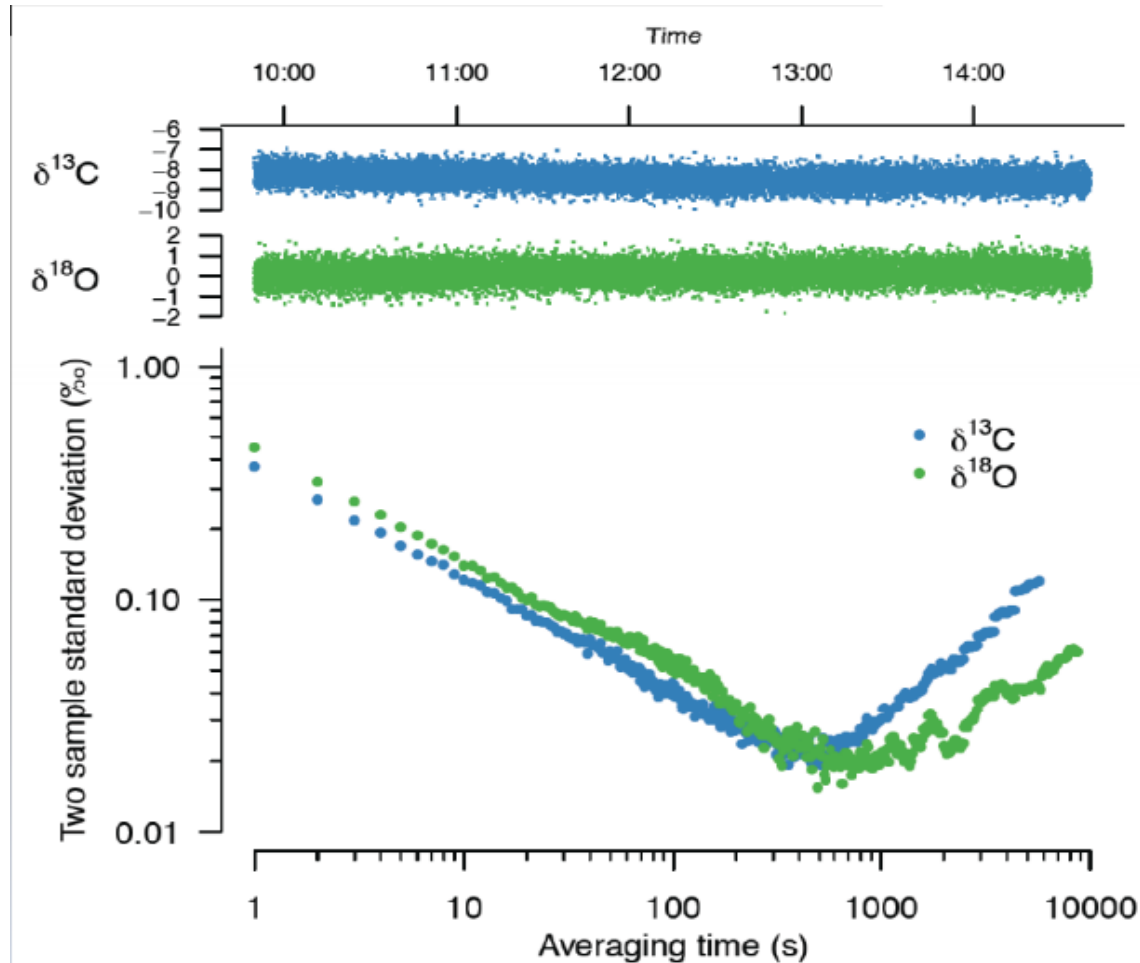
Small and portable



NO₂ Sensitivity Test at Jungfrauoch (3850m asl), Switzerland



Isotopic Ratio Measurements of $\delta^{18}\text{O}$ (‰) & $\delta^{13}\text{C}$ (‰) performed at the Jungfrauoch (3850m asl), Switzerland



Precision (1σ)	
1 s:	0.5 ‰
60 s	< 0.07 ‰
600 s	< 0.03 ‰

Year	$1\sigma_{\min}$
2006	0.16 ‰
2009	0.046 ‰
2012	0.025 ‰

Allan-Werle variance plot

Summary and Conclusions

- Development of robust, compact, sensitive, selective mid-infrared trace gas sensor technology based on room temperature, continuous wave DFB laser diodes and high performance QCLs for [environmental monitoring and medical diagnostics](#).
- Interband cascade and quantum cascade lasers were used in [TDLAS, PAS and QEPAS based sensor platforms](#)
- Eight target trace gas species were detected with a 1 sec sampling time:
 - C_2H_6 : $\sim 3.36\text{ }\mu\text{m}$, detection sensitivity of 740 pptv using TDLAS
 - NH_3 : $\sim 10.4\text{ }\mu\text{m}$, detection sensitivity of ~ 1 ppbv (200 sec averaging time)
 - NO : $\sim 5.26\text{ }\mu\text{m}$, detection limit of 3 ppbv
 - CO : $\sim 4.61\text{ }\mu\text{m}$, minimum detection limit of 2 ppbv
 - SO_2 : $\sim 7.24\text{ }\mu\text{m}$, detection limit of 100 ppbv
 - CH_4 and N_2O : $\sim 7.28\text{ }\mu\text{m}$, detection limits of 13 and 6 ppbv, respectively
 - H_2O_2 : $\sim 7.73\text{ }\mu\text{m}$, detection limit of 75 ppb
- New target analytes: CH_2O and $\text{C}_2\text{H}_6\text{O}$

