

Advanced Measurement Concepts for Mid-infrared Semiconductor Laser based Trace Gas Sensor Technologies: Opportunities & Challenges

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MIOMD 2014

Montpellier,

- New Laser Based Trace Gas Sensor Technology
 - Novel Multipass Gas Absorption Cells & Electronics
 - Quartz Enhanced Photoacoustic Spectroscopy
- Examples of seven Mid-infrared Trace Gas Species
 - NO, CO, SO₂, CH₄, N₂O, H₂O₂ & C₃H₆O
- Future Directions of Laser Based Trace Gas Sensor Technologies and Conclusions

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Wide Range of Trace Gas Sensing Applications

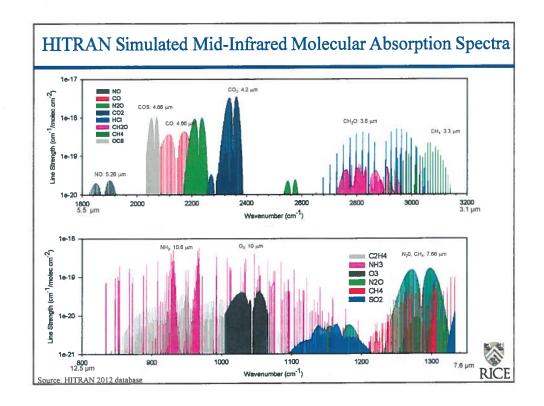
- Urban and Industrial Emission Measurements
 - Industrial Plants
 - Combustion Sources and Processes (e.g. fire detection)
 - Automobile, Truck, Aircraft and Marine Emissions
- Rural Emission Measurements
 - Agriculture & Forestry, Livestock
- Environmental Monitoring
 - Atmospheric Chemistry (e.g isotopologues, climate modeling,...)
 - Volcanic Emissions
- Chemical Analysis and Industrial Process Control
 - Petrochemical, Semiconductor, Pharmaceutical, Metals Processing, Food & Beverage Industries, Nuclear Technology & Safeguards
- Spacecraft and Planetary Surface Monitoring
 - Crew Health Maintenance & Life Support
- Applications in Medical Diagnostics and the Life Sciences
- Technologies for Law Enforcement, Defense and Security
- Fundamental Science and Photochemistry



Laser based Trace Gas Sensing Techniques

- Optimum Molecular Absorbing Transition
 - Overtone or Combination Bands (NIR)
 - Fundamental Absorption Bands (Mid-IR)
- Long Optical Pathlength
 - Multipass Absorption Gas Cell (e.g. White, Herriot, Chernin, Aeris Technologies; Circular Cylindrical: Empa & Loncar)
 - Cavity Enhanced and Cavity Ringdown Spectroscopy
 - Open Path Monitoring (with retro-reflector): Standoff and Remote Detection
 - Fiberoptic Evanescent Wave Spectroscopy
- Spectroscopic Detection Schemes
 - Frequency or Wavelength Modulation
 - Balanced Detection
 - Zero-air Subtraction
 - Photoacoustic & Quartz Enhanced Photoacoustic Spectroscopy (QEPAS)





Mid-IR Source Requirements for Laser Spectroscopy

REQUIREMENTS	IR LASER SOURCE				
Sensitivity (% to pptv)	Optimum Wavelength, Power				
Selectivity (Spectral Resolution)	Stable Single Mode Operation and Narrow Linewidth				
Multi-gas Components, Multiple Absorption Lines and Broadband Absorbers	Mode Hop-free Wavelength Tunability				
Directionality or Cavity Mode Matching	Beam Quality				
Rapid Data Acquisition	Fast Time Response				
Room Temperature Operation	High wall plug efficiency, no cryogenics or cooling water				
Field deployable in harsh environments	Compact & Robust				

Key Characteristics of Mid-IR QCL & ICL Sources - Oct. 2014

Band - structure engineered devices

Emission wavelength is determined by layer thickness – MBE or MOCVD; Type I QCLs operate in the 3 to 24 μm spectral region; Type II and GaSb based ICLs can cover the 3 to 6 μm spectral range.

- Compact, reliable, stable, long lifetime, and commercial availability Fabry-Perot (FP), single mode (DFB) and multi-wavelength devices

Wide spectral tuning ranges in the mid-IR

- 1.5 cm⁻¹ using injection current control for DFB devices
- 10-20 cm⁻¹ using temperature control for DFB devices
- ~100 cm-1 using current and temperature control for QCL DFB Array ~525 cm-1 (22% of c.w.) using an external grating element and FP chips with heterogeneous cascade active region design; also QCL DFB Array

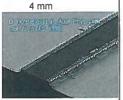
Narrow spectral linewidths

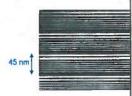
- CW: 0.1 3 MHz & $<\!10kHz$ with frequency stabilization (0.0004 cm $^{-1})$ Pulsed: ~300 MHz

High pulsed and CW powers of QCLs_at TEC/RT temperatures

- Room temperature pulsed power of > 30 W with 44% wall plug
- CW powers of ~ 5 W with 23% wall plug efficiency at 293 °K
- > 600 mW CW DFB @ 285 °K; wall plug efficiency 23% at 4.6 μm







Typical Oil & Gas Production Site near Houston, TX



3-5 km

This figure shows the result of a sequence of four fracking injections obtained by directional drilling which creates horizontal production in target stratum.

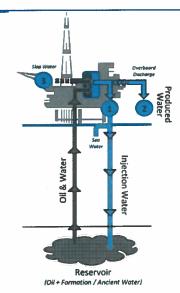


oject f

A proposed DOE-ARPA-E CH_4 detection project at 3.327 μm will start in 2015 at a wellpad of 10 m x10 m with a 1 m spatial resolution.



Oil in Water Detection





- Produced water
 - legislation: < 15 ppm
- Injection water
 - Economic reasons target value: < 5 ppm or lower

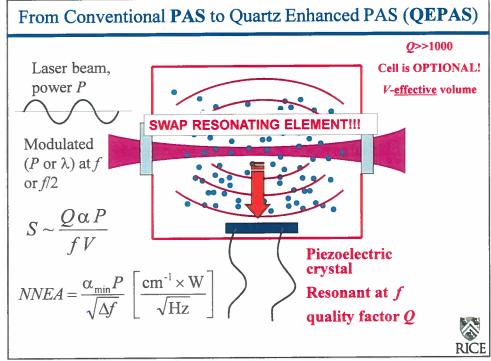
IQCLSW 2014, Policore, Italy: B. Lendl et al, Vienna University of Technology, Austria

Comparison of proposed Rice CH₄ Sensor System and current commercially available CH₄ Platforms

Size	Rice	Picarro	ABB-LGR I	ABB-LGR II	Aerodyne
Opt. Path length and method	MIR TDLAS: ~ 9 m	NIR CRDS: >2000m	NIR OA-ICOS: > 1000m	NIR OA-ICOS: > 2000m	MIR TDLAS: 70-100 m
Sensitivity/sec	< 5-10 ppb	1-2 ppb	5 ppb	2 ppb	<1 ppb
Accuracy (drift)	2 ppb stabilized	2 ppb	20 ppb, temp. stabilized	2 ppb	2 ppb
Cell Volume, cc	60	30	500	2000	2000
Pump Size (10 sec flush time)	~ 1 lpm	~ 0.5 lpm	~ 11 lpm	~45 lpm	~ 45 lpm
Cavity Mirror Reflectance	98.5%-99%	>99.99%	>99.99%	>99.99%	>99,99%
Power Consumption	2-20 W	200 W	70 W	290 W	400 W
Weight	~ 2-4 leg	~20 kg	~ 15 kg	~ 40 lcg	~ 40 kg
Cost	- 20-25K USD	~ 40-50K USD	~ 25K USD	~40K USD	~ 100K USD

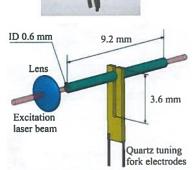
US Department of Energy Advanced Research Project Agency – Energy (ARPA-E), Methane Observation Networks with Innovative Technology to obtain Reductions (MONITOR)





Quartz Tuning Fork as a Resonant Microphone for QEPAS





Unique Properties

- · Extremely low internal losses:
 - Q~10 000 at 1 atm
 - Q~100 000 in vacuum
- Acoustic quadrupole geometry
 - Low sensitivity to external sound
- Large dynamic range (~10⁶) linear from thermal noise to breakdown deformation
 - 300K noise: *x*~10⁻¹¹ cm
 - Breakdown: $x \sim 10^{-2}$ cm
- Wide temperature range: from 1.6K to ~700K

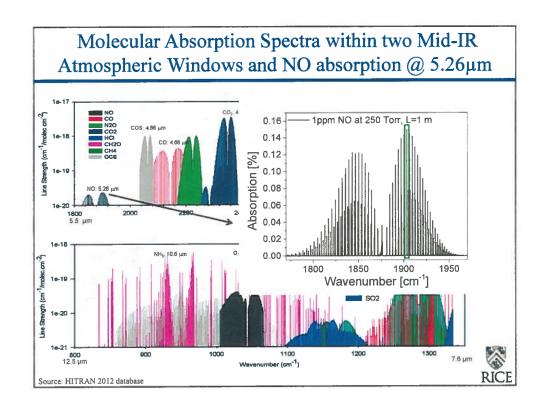
Acoustic Micro-resonator (µR) Tubes

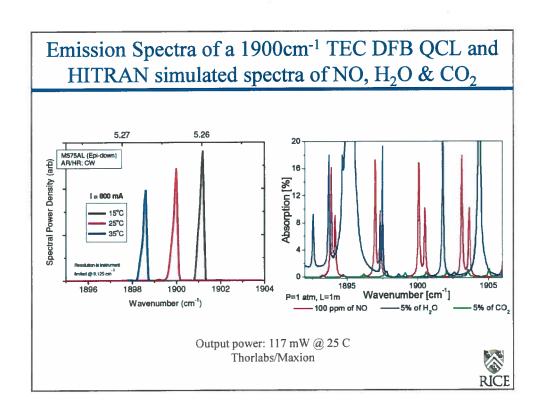
- Optimum inner diameter: 0.6 mm; μR-QTF gap is 25-50 μm
- Optimum mR tubes must be ~ 4.4 mm long (~\(\frac{1}{\sqrt{1}}\)/2 for sound at 32.8 kHz)
- SNR of QTF with µR tubes: <u>×30</u> (depending on gas composition and pressure)

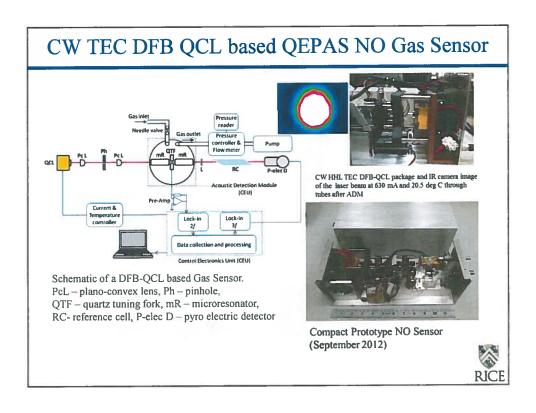
Motivation for Nitric Oxide Detection

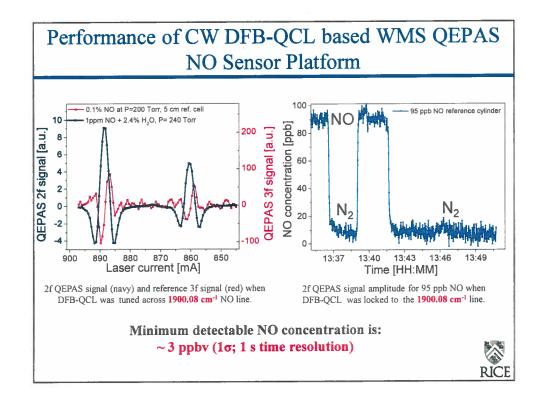
- NO in medicine and biology
 - Important signaling molecule in physiological processes in humans and mammals (1998 Nobel Prize in Physiology/Medicine)
 - Treatment of asthma, chronic obstructive pulmonary disease (COPD) & lung rejection
- · Environmental pollutant gas monitoring
 - Ozone depletion
 - Precursor of smog and acid rain
 - NO_X monitoring from automobile exhaust and power plant emissions
- Atmospheric Chemistry







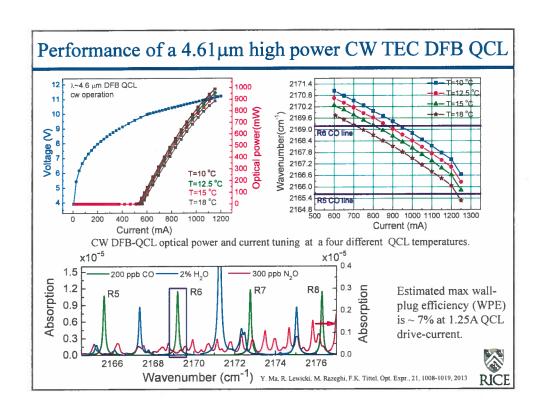


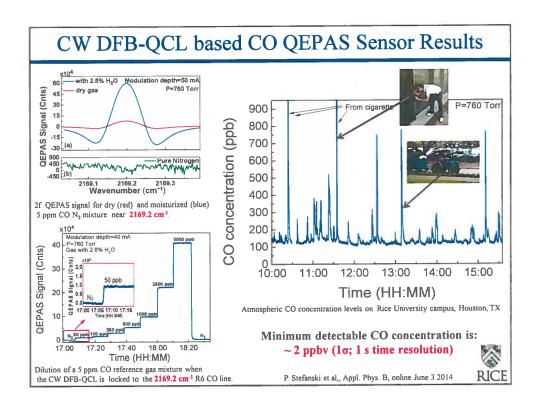


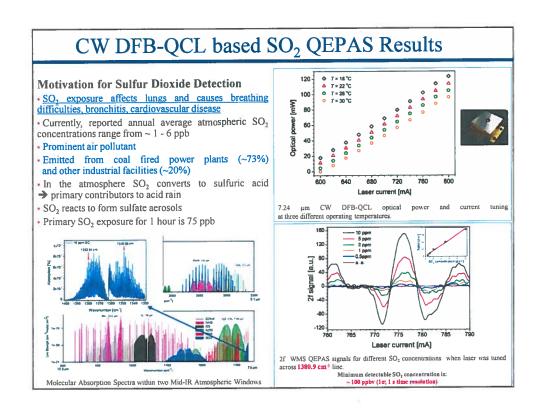
Motivation for Carbon Monoxide Detection

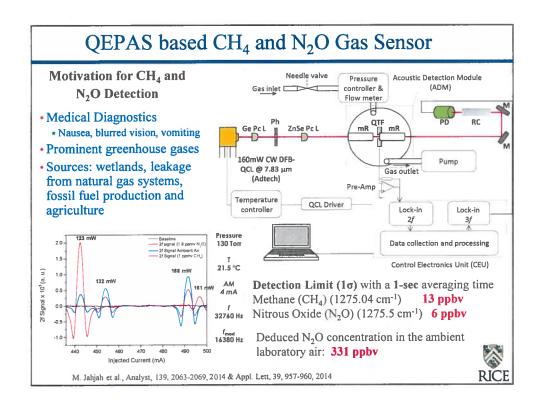
- CO in Medical Diagnostics
 - Hypertension and abnormality in heme metabolism
- Public Health
 - Extremely dangerous to human life even at a low concentrations. CO must be monitored at low concentration levels (<35 ppm).
- Atmospheric Chemistry
 - Incomplete combustion of natural gas, fossil fuel and other carbon containing fuels.
 - Impact on atmospheric chemistry through its reaction with hydroxyl (OH) for troposphere ozone formation and changing the concentration levels of greenhouse gases (e.g. CH₄).

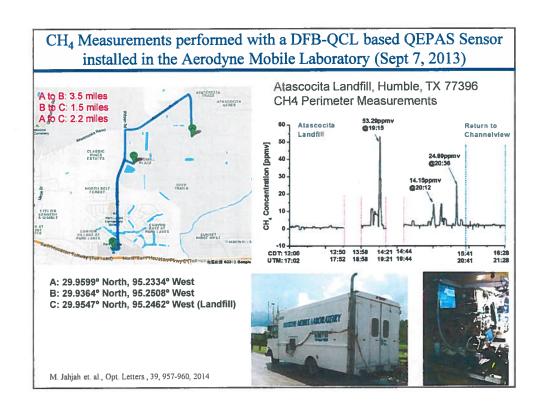






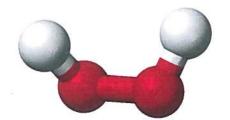






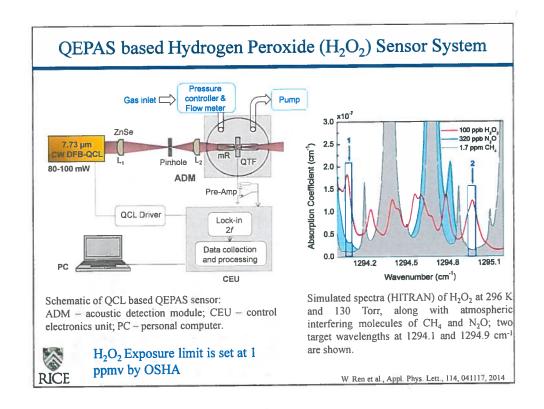
Motivation of H₂O₂ Detection

- Oxidative capacity of atmosphere and balance of HO_x;
- Acid rain formation & In-cloud oxidation of S(IV) to S(VI);
- · Active agent in decontamination and sterilization systems;
- H₂O₂ in breath is a biomarker of oxidative stress;
- H₂O₂ concentration levels in Houston have not been reported despite of atmospheric conditions, such as high humidity, high solar radiation levels, and the presence of the petrochemical industry.

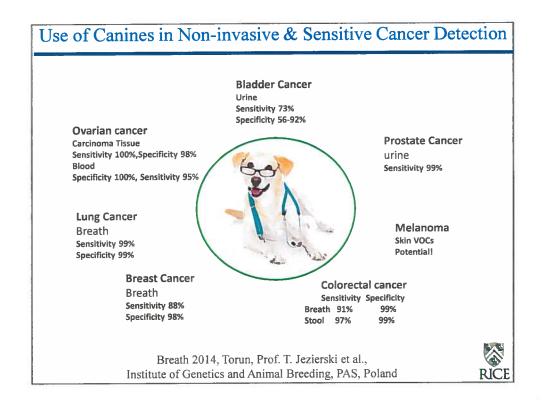


 H_2O_2





	Molecule (Host)	Frequency, cm ⁻¹	Pressure, Torr	NNEA, cm ⁻¹ W/Hz ¹⁶	Power, mW	NEC (t=1s),
vis {	Os (air)	35087.70	700	3.0×10 ⁻⁸	0.8	1,27
	O2 (N2)	13099.30	158	4.74×10-7	1228	13
	C2H2 (N2)*	6523.88	720	4.1×10 ⁻⁹	57	0.03
	NH ₃ (N ₂)*	6528.76	575	3.1×10-9	60	0.06
	C2H4 (N2)*	6177.07	715	5.4×10-8	15	1.7
	CH4 (N2+1.2% H2O)*	6057.09	760	3.7×10-1	16	0.24
NR ≺	N2H4	6470.00	700	4.1×10 ⁻⁹	16	1
	H ₂ S (N ₂)*	6357.63	780	5.6×10-9	45	5
1	HCl (N2 dry)	5739.26	760	5.2×10-4	15	0.7
	CO ₂ (N ₂ +1.5% H2O) *	4991.26	50	1.4×10 ⁻⁸	4.4	18
}	CH2O (N2:75% RH)*	2804.90	75	8.7×10 ⁻⁹	7.2	0.12
	CO (N2 +2.2% H2O)	2176 28	100	1.4×10-7	71	0.002
	CO (propylene)	2196.66	50	7.4×10 ⁻⁸	6.5	0.14
	N2O (air+5%SF4)	2195 63	50	1.5×10 ⁻⁸	19	0.007
Mid-IR	C2H5OH (N2)**	1934.2	770	2.2=10-7	10	90
	NO (N2+H2O)	1900.07	250	7.5×10-9	100	0.003
	H ₂ O ₂	1295.6	150	4.6×10-	100	12
	C2HF5 (N2)***	1208.62	770	7.8×10-9	6.6	0.009
	NH ₃ (N ₂)*	1046.39	110	1.6×10-8	20	0 006
	SF6	948.62	75	2.7x10 ⁻¹⁰	18	5x10 ⁻³ (50 ppt)
	Improved microresonator Improved microresonator an With amplitude modulation NNEA – normalized noise equivalent concentr NEC – noise equivalent concentr	and metal microres	onator fficient.		18 dB oct fil	ter slope.



Advantages & Disadvantages of Canines in Cancer Detection

Advantages

- Non-invasive, safe and easy sample collecting
- Relatively easy training and interpretation of dogs' indications
- Odor samples can be tested several times
- Extremely high detection sensitivity and specificity
- Potential of VOCs are useful in search, rescue and emergency applications

Disadvantages

- To-date a "black-box technology"
- It is a method based on earning a reward, which becomes unreliable after ~ 4 years
- Variation of sensitivity and specificity
- Re-training of dogs is not effective

Breath 2014, Torun, Prof. T. Jezierski et al., Institute of Genetics and Animal Breeding, PAS, Poland

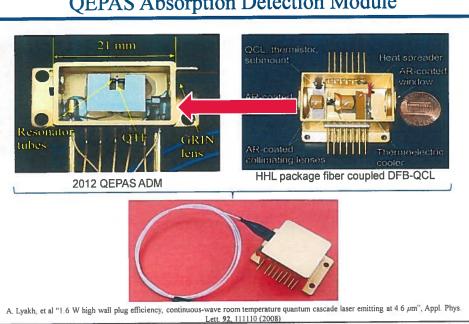


Future Directions and Outlook

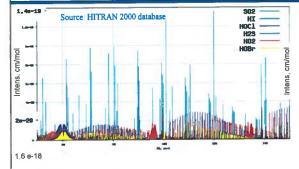
- New target analytes: formaldehyde (CH₂O), ethylene (C₂H₄), ozone (O₃) and nitrate (NO₃
- Ultra-compact, low cost, robust sensors (e.g. CH₄, NO, CO...)
- QCL based ultra-portable atmospheric carbon isotope monitor for ¹²CH₄ & ¹³CH₄
- Monitoring of broadband absorbers: acetone (C₃H₆O):
 MDL of 1.5 ppm with a 7mW ICL & AM, or 20ppb with a 100mW QCL @ 8.23μm; benzene (C₆H₆)...
- Optical power build-up cavity designs (I-QEPAS)
- THz QEPAS based sensors
- Development of trace gas sensor networks



Potential Integration of a CW DFB- QCL and QEPAS Absorption Detection Module



Why is THz based Trace Gas Sensing useful?



Several gas species such as HF, OH, HCN, HCl, HBr, NH₃, H₂O₂, H₂S, H₂O & explosives (in the vapor phase) show strong absorption bands in the THz spectral range.

Mainly rotational levels are involved in THz absorption processes and rotational-translational (R-T) relaxation rates are up to three order of magnitude faster with respect to vibrational-translational (V-T) in the mid-infrared

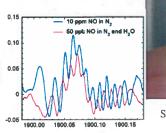
QEPAS signal strongly depends on the energy relaxation rates due to the possibility to operate at low pressure, & thereby taking advantages of the typically very high QTF Q-factors.

Why have QEPAS sensors not been developed in the THz spectral range so far?

Standard QTFs have a very small volume

 $(\sim 0.3 \times 0.3 \times 3 \text{ mm}^3)$

In QEPAS sensor systems, it is critical to avoid laser illumination of the QTF, since the radiation blocked by the QTF prongs results in an undesirable non-zero background as well as a shifting fringe-like interference pattern.

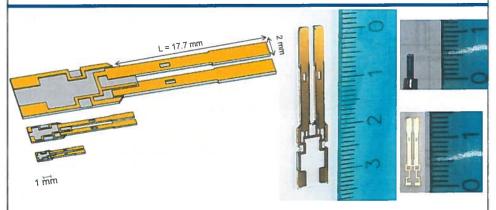




Standard QTF

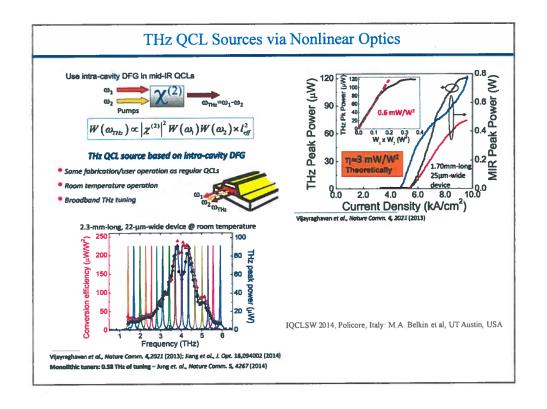
The standard QTF prong separation of 330 μm is comparable with the THz wavelength which prevents the use of a QEPAS sensor architecture in the THz range unless we use large sized QTFs.

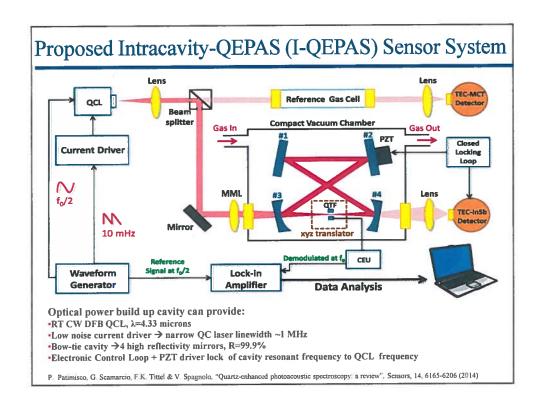
Custom fabricated QTFs scaled in Dimensions that are ~7 & 3 times larger than standard QTFs

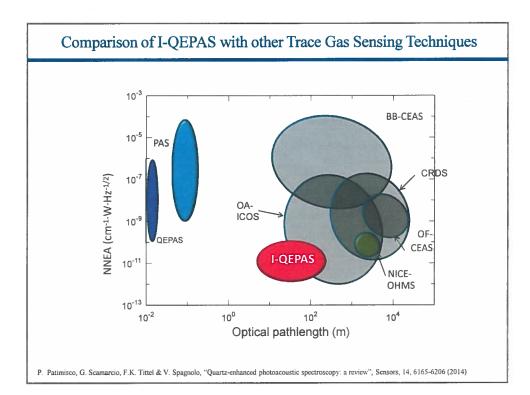


Standard photolithographic techniques were used to etch the custom QTF, starting from a z-cut quartz wafer. Chromium/gold contacts were deposited on both sides of the custom QTF.

Currently verification that the larger QTFs behave similar to a "standard" QTF in terms of vibrational modes and Q factor is in progress







Summary and Conclusions

- Development of robust, compact, sensitive, selective mid-infrared trace gas sensor technology based on room temperature, continuous wave DFB laser diodes and high performance QCLs for environmental monitoring and medical diagnostics.
- Interband cascade and quantum cascade lasers were used in QEPAS and TDLAS based sensor platforms
- Six target trace gas species were detected with a 1 sec sampling time:
 - NO: ~5.26 μm, detection limit of 3 ppbv
 - CO: ~4.61 μm, minimum detection limit of 2 ppbv
 - SO₂: ~7.24 μm, detection limit of 100 ppbv
 - CH₄ and N₂O: \sim 7.28 μ m, detection limits of 13 and 6 ppbv, respectively
 - H_2O_2 : ~7.73 µm, detection limit of 75 ppb
- New target analytes: CH₂O and C₃H₆O

