



Advanced Infrared Semiconductor Laser based Chemical Sensing Technologies: Opportunities and Challenges

F.K. Tittel, Y. Bakirkin, R.F. Curl, A.A. Kosterev, R. Lewicki, S. So and G. Wysocki

Rice Quantum Institute, Rice University, Houston, TX, USA
<http://ecc.rice.edu/lasersci/>

OUTLINE

Zhejiang University
Hangzhou, China

July 7, 2008

- Motivation: Wide Range of Chemical Sensing
- Fundamentals of Laser Absorption Spectroscopy
- New Laser Sources and Sensing Technologies
- Selected Applications of Trace Gas Detection
 - Quartz Enhanced L-PAS (NH₃, Freon 125, Acetone and TATP)
 - Nitric Oxide Detection (Faraday Rotation Spectroscopy, Remote Sensing & Cavity Enhanced Spectroscopy)
- Future Directions and Conclusions

Work supported by NSF, NASA, DOE and Robert Welch Foundation

Rice University, Houston



Rice University, Houston



Wide Range of Trace Gas Sensing Applications

- **Urban and Industrial Emission Measurements**
 - Industrial Plants
 - Combustion Sources and Processes (e.g. fire detection)
 - Automobile, Truck, Aircraft and Marine Emissions
- **Rural Emission Measurements**
 - Agriculture & Forestry, Livestock
- **Environmental Monitoring**
 - Atmospheric Chemistry
 - Volcanic Emissions
- **Chemical Analysis and Industrial Process Control**
 - Petrochemical, Semiconductor, Nuclear Safeguards, Pharmaceutical, Metals Processing, Food & Beverage Industries
- **Spacecraft and Planetary Surface Monitoring**
 - Crew Health Maintenance & Life Support
- **Applications in Health and Life Sciences**
- **Technologies for Law Enforcement and National Security**
- **Fundamental Science and Photochemistry**



Air Composition

Main Components

- Nitrogen 78%
- Oxygen 21%
- Water 0.8%
- CO₂ 0.03 %

Trace Components

- Methane 1.7 ppm
- CO 0.4 ppm
- N₂O 0.3 ppm
- O₃ 0.03 ppm
- H₂CO 0.001 ppm
- ...



Worldwide Megadirty Megacities

	Population, m		Sulphur dioxide	Particulate matter	Lead	Carbon monoxide	Nitrogen dioxide	Ozone
	1990 est.	2000 proj.						
Bangkok	7 16	10 28	0	●	●	0	0	0
Beijing	9 74	11 47	●	●	0	0	0	●
Bombay	11 13	15 43	0	●	0	0	0	0
Buenos Aires	11 58	13 05	0	0	0	0	0	0
Castro	9 08	11 77	0	●	0	0	0	0
Calcutta	11 83	15 94	0	●	0	0	0	0
Delhi	8 62	12 77	0	●	0	0	0	0
Jakarta	9 42	13 23	0	●	0	0	0	0
Karachi	7 67	11 57	0	●	0	0	0	0
London	10 57	10 79	0	0	0	0	0	0
Los Angeles	10 47	10 91	0	0	0	●	●	●
Manila	8 40	11 48	0	●	0	0	0	0
Mexico City	19 37	24 44	●	●	●	●	0	●
Moscow	9 39	10 11	0	0	0	0	0	0
New York	15 85	16 10	0	0	0	0	0	0
Rio de Janeiro	11 12	13 00	0	0	0	0	0	0
Sao Paulo	18 42	23 60	0	●	0	0	0	0
Seoul	11 33	12 97	●	●	0	0	0	●
Shanghai	13 30	14 69	0	0	0	0	0	0
Tokyo	20 52	21 32	0	0	0	0	0	0

Source: United Nations ● high pollution ○ moderate to heavy pollution ○ low pollution ○ no data available



Megacity Air Pollution: Houston, TX



Monitoring Methane in Rice - Based Agroecosystems



International Space Station



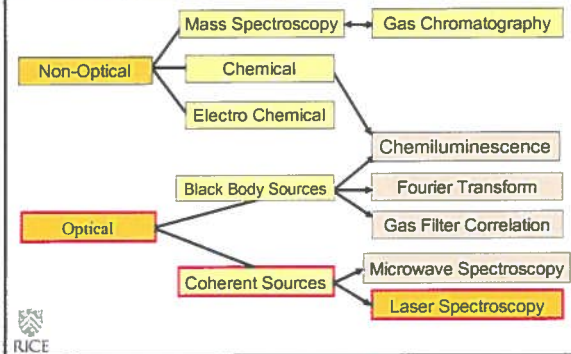
Mars NASA Pathfinder Climate Monitoring



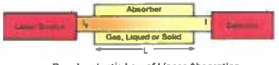
Atmospheric Chemistry of Volcanic Plumes



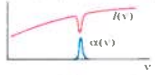
Existing Methods for Trace Gas Detection



Fundamentals of Laser Absorption Spectroscopy



Beer-Lambert's Law of Linear Absorption
 $I(\nu) = I_0 e^{-\alpha(\nu) P_s L}$
 $\alpha(\nu)$ - absorption coefficient [$\text{cm}^{-1} \text{atm}^{-1}$], L - path length (cm)
 ν - frequency [cm^{-1}], P_s - partial pressure (atm)



$\alpha(\nu) = C \cdot S(\nu) \cdot g(\nu - \nu_0)$
 C - total number of molecules of absorbing gas (atm cm^{-3}) [molecule $\text{cm}^{-3} \text{atm}^{-1}$]
 S - molecular line intensity [$\text{cm}^2 \text{molecule}^{-1}$]
 $g(\nu - \nu_0)$ - normalized spectral lineshape function (cm), (Gaussian, Lorentzian, Voigt)

Key Requirements: Sensitivity, specificity, rapid data acquisition and multi-species detection

Optimum Molecular Absorbing Transition

- Overtone or Combination Bands (NIR)
- Fundamental Absorption Bands (MID-IR)

Long Optical Pathlengths

- Multipass Absorption Cell
- Cavity Enhanced, Cavity Ringdown & Intracavity Spectroscopy
- Open Path Monitoring (with retro-reflector)

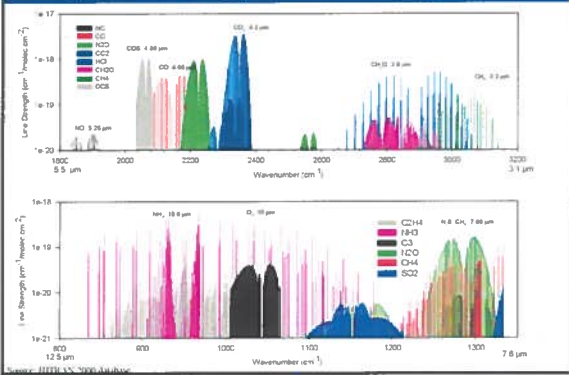
Spectroscopic Detection Schemes

- Wavelength & Frequency Modulation
- Balanced Detection
- Zero-air Subtraction
- Photoacoustic Spectroscopy

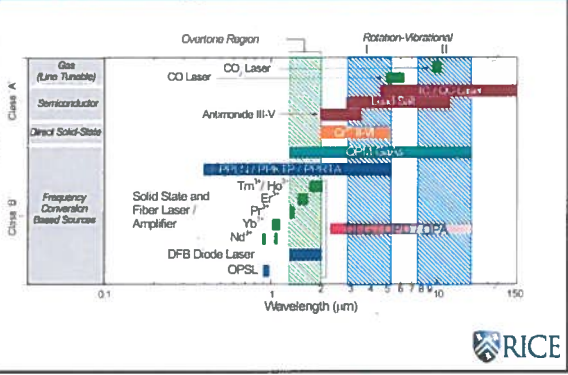
Mid-IR Source Requirements for Laser Spectroscopy

REQUIREMENTS	IR LASER SOURCE
Sensitivity (% to ppt)	Wavelength, Power
Selectivity (Spectral Resolution)	Single Mode Operation and Narrow Linewidth
Multi-gas Components, Multiple Absorption Lines and Broadband Absorbers	Tunable Wavelength
Directionality or Cavity Mode Matching	Beam Quality
Rapid Data Acquisition	Fast Time Response
Room Temperature Operation	No Consumables
Field deployable	Compact & Robust

Molecular Absorption Spectra within the two Mid-IR Atmospheric Windows

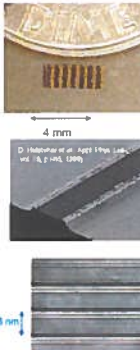


IR Laser Sources and Wavelength Coverage



Key Characteristics of mid-IR QCLs and ICL Sources-2008

- **Band - structure engineered devices** (emission wavelength is determined by layer thickness - MBE or MOCVD); mid-infrared QCLs operate from 3 to 24 μm
- Compact, reliable, stable, long lifetime, and commercial availability
- Fabry-Pérot (FP), single mode (DFB) and multi-wavelength
- **Spectral tuning range in the mid-IR** (4-24 μm for QCLs and 3-5 μm for ICLs)
 - 1.5 cm^{-1} using injection current control
 - 10-20 cm^{-1} using temperature control
 - ~300 cm^{-1} using an external grating element and heterogeneous cascade active region design; also QCL laser array (Harvard)
- **Narrow spectral linewidth**
 - CW: 1-2 MHz & ~10 MHz with frequency stabilization @ 0.004 cm^{-1} , pulsed: ~300 MHz (sharp from heating)
- **High pulsed and cw powers of QCLs at TEC/RT temperatures**
 - Pulsed and CW powers of ~1.5 W, high temperature operation ~300 K
 - Average power levels: 1-600 mW (current wall plug ~9-9%)
 - ~50 mW, TEC CW DFB @ 5 and 10 μm (DLS, Alpes, Princeton)
 - ~300 mW @ 8.3 μm (Harvard, Hamamatsu, Alpes, Northwestern)
 - > 600 mW (CW FP) @ RT & a wall plug efficiency of ~9%, Adtech Optics, Harvard, Pranalytica
 - > 150 mW (CW DFB) at 298 K (Northwestern)



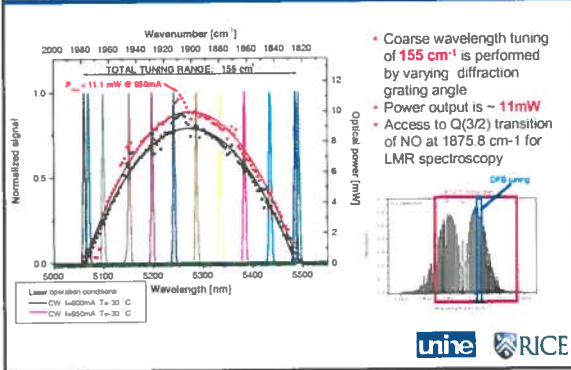
Widely Tunable, CW, TEC Quantum Cascade Lasers

Tunable external cavity QCL based spectrometer

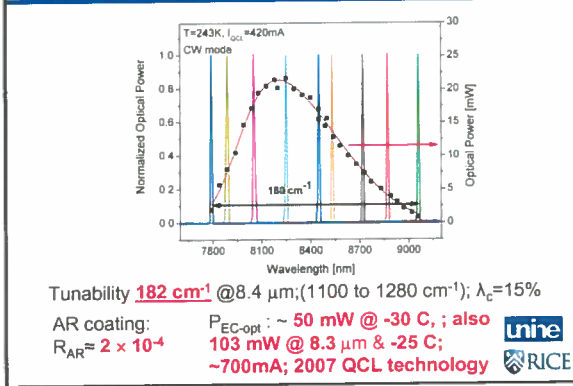
- Fine wavelength tuning
 - PZT controlled EC-length
 - PZT controlled grating angle
 - QCL current control
- Motorized coarse grating angle tuning
- Vacuum tight QCL enclosure with build-in 3D lens positioner (TEC laser cooling + optional chilled water cooling)

RICE

Wide Wavelength Tuning of a 5.3μm EC-QCL



Performance of 8.4 μm cw EC-QCL Spectroscopic Source



Quartz Enhanced
Photoacoustic Spectroscopy

From conventional PAS to QEPAS

Laser beam, power P

Modulated (P or λ) at f or $f/2$

$S \sim \frac{Q \alpha P}{f V}$

$NNEA = \frac{\alpha_{min} P}{\sqrt{\Delta f}} \left[\frac{cm^{-1} \times W}{\sqrt{Hz}} \right]$

SWAP RESONATING ELEMENT!!!

Cell is OPTIONAL!

Effective volume

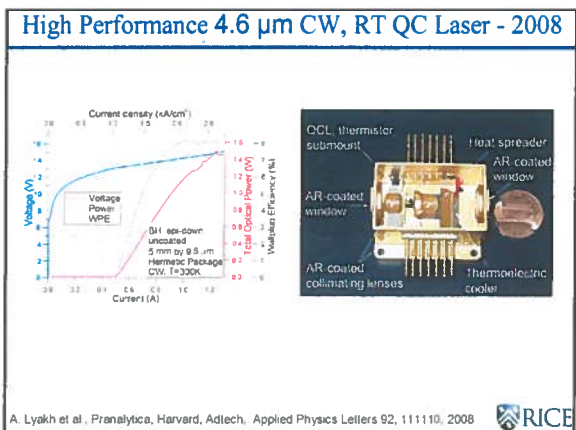
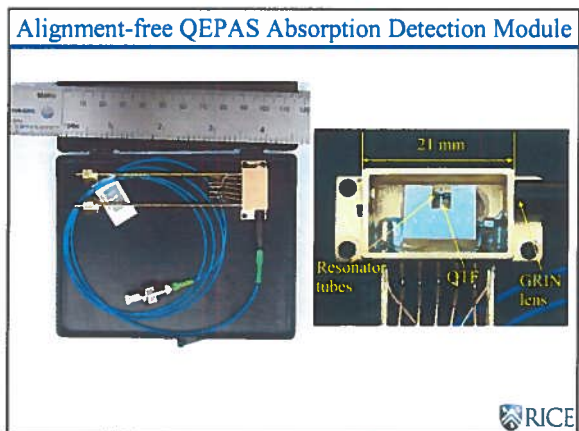
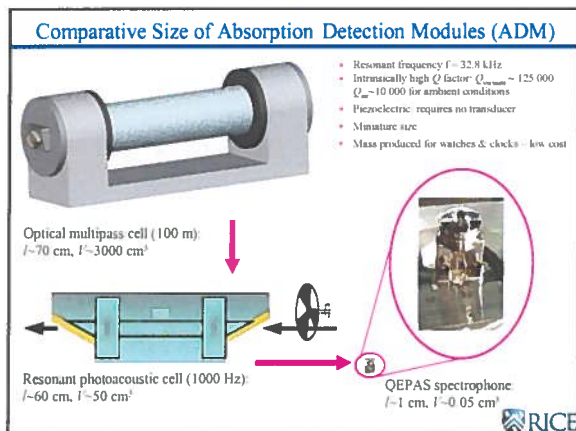
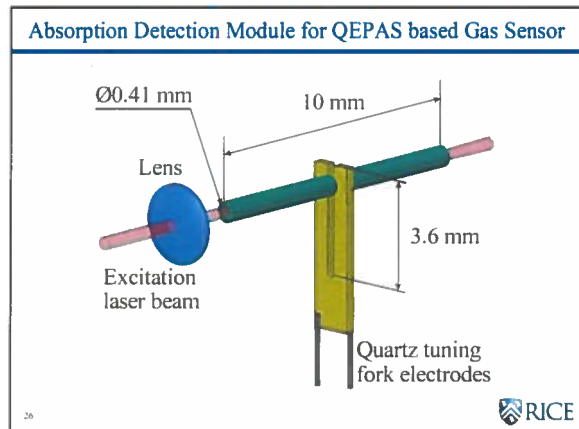
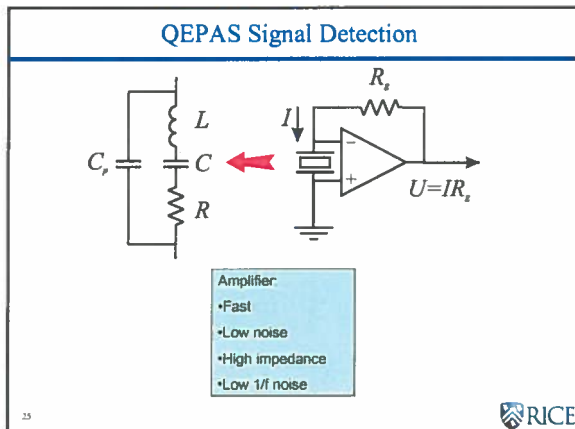
Piezoelectric microphone crystal

Resonant at f quality factor Q

RICE

Quartz Tuning Fork (TF) as a Resonant Microphone

- Resonant frequency $f=32.8$ kHz
- Intrinsically high Q factor: $Q_{vacuum} \sim 125,000$, $Q_{air} \sim 10,000$ at ambient conditions.
- Piezoelectric: requires no transducer
- Miniature size
- Mass produced for clocks – low cost



- ### Merits of QE Laser-PAS based Trace Gas Detection
- High sensitivity (ppm to ppb gas concentration levels) and excellent dynamic range
 - Immune to ambient and flow acoustic noise, laser noise and etalon effects
 - Significant reduction of sample volume ($< 1 \text{ mm}^3$)
 - Applicable over a wide range of pressures
 - Temperature, pressure and humidity insensitive
 - Ultra-compact, rugged and low cost (compared to other laser based sensor architectures)
- RICE

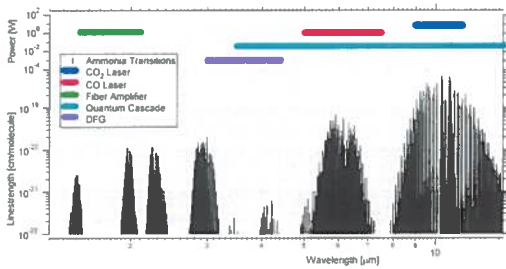
Trace Gas Sensing Examples

Motivation for NH₃ Detection

- Monitoring of gas separation processes
- Detection of ammonium-nitrate explosives
- Spacecraft related gas monitoring
- Monitoring NH₃ concentrations in the exhaust stream of NO_x removal systems based on selective catalytic reduction (SCR) techniques
- Semiconductor process monitoring & control
- Monitoring of industrial refrigeration facilities
- Pollutant gas monitoring
- Atmospheric chemistry
- Medical diagnostics (kidney & liver diseases)

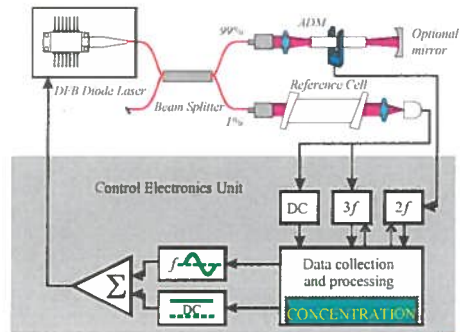


Infrared NH₃ Absorption Spectra



M. Wehber et al. 2004. Prostaglandin

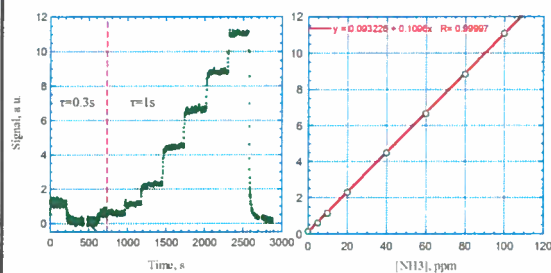
QEPAS based Gas Sensor Architecture



34



Calibration and Linearity of a 1.53 μm QEPAS based NH₃ Sensor

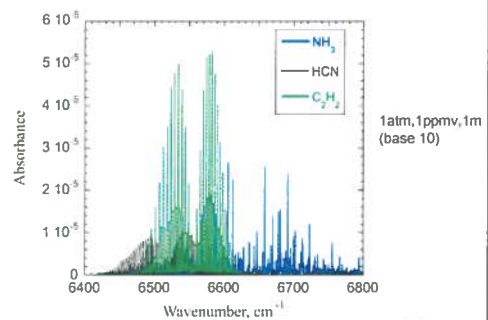


Noise-equivalent concentration (NEC) for $\tau=1s$ time constant is 0.06 ppm for 60mW excitation power at 6528.76 cm⁻¹

90 last points of each step averaged

Noise-equivalent absorption (NEA) coefficient $k=3.1 \times 10^{-9} \text{ cm}^{-1} \text{ W/Hz}^{1/2}$

Infrared NH₃, HCN and C₂H₂ Absorption Spectra



PNNL FTIR Database



Commercial widely tunable cw EC-QCL

Mid-IR Lasers From Daylight Solutions

- CW Wavelength range: 1.5 - 2.5 μm
- Power: 100 mW
- Current consumption: 100 mA @ 12 V
- Small form factor: 100 mm x 100 mm x 100 mm
- No frequency tuning
- Average power: 100 mW
- Single-pass: Full range: 100 mW
- Output: 100 mW @ 1.5 μm
- Wavelength: 1.5 μm

Motivation for Nitric Oxide Detection

- Atmospheric Chemistry
- Environmental pollutant gas monitoring
 - NO_x monitoring from automobile exhaust and power plant emissions
 - Precursor of smog and acid rain
- Industrial process control
 - Formation of oxynitride gates in CMOS Devices
- NO in medicine and biology
 - Important signaling molecule in physiological processes in humans and mammals (1998 Nobel Prize in Physiology/Medicine)
 - Treatment of asthma, COPD, acute lung rejection
- Photofragmentation of nitro-based explosives (TNT)

Laser-based ICOS Nitric Oxide Sensor

Online NO concentration measurements at 3l/min exhalation.

Intercomparison of ICOS and commercial chemiluminescence sensor (Sievers - solid line)

NO-N₂ mixture @ 100 Torr
Effective L = 700 m

A 1 σ deviation of the amplitude corresponds to a 6.7 ppb detection limit (1 sec.)

M. R. McCurdy, Y. Bashkun, G. Wysocki, P. K. Tjoelke. Performance of an enhanced nitric oxide and carbon dioxide sensor using quantum cascade laser based retrograde cavity optical spectroscopy to measure in Journal of Biomedical Optics, 2017

High resolution spectroscopy with a 5.3 μm EC-QCL

Access to NO Q(3/2) transition at 1875.8 cm⁻¹ for Faraday rotation spectroscopy

Mode hop free scan of up to $\sim 2.5 \text{ cm}^{-1}$ with a resolution $< 0.001 \text{ cm}^{-1}$ (30MHz) can be performed anywhere within the tuning range

In collaboration with **unine**

QCL based Quartz-Enhanced Photoacoustic Gas Sensor

QEPAS characteristics:

- High sensitivity (ppm to ppb)
- Excellent dynamic range
- Immune to environmental noise
- Ultra-small sample volume ($< 1 \text{ mm}^3$)
- Sensitivity is limited by the fundamental thermal TF noise
- Compact, rugged and low cost
- Potential for trace gas sensor networks

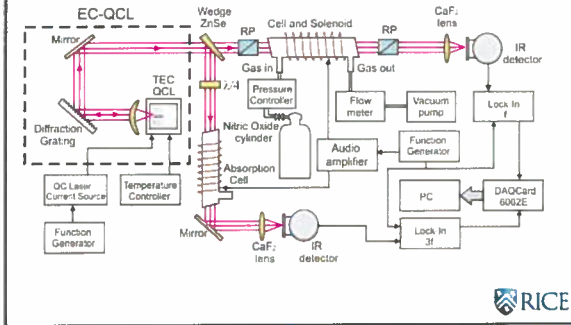
High resolution EC-QCL based QEPAS

4.2% NO in N₂ at 600 Torr

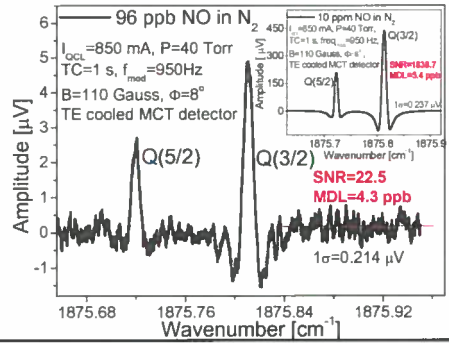
External Amplitude Modulation:

- QTF is used as a mechanical chopper at $f = 32 \text{ kHz}$
- No chirp associated with the laser current modulation
- High resolution mode-hop-free tuning is possible

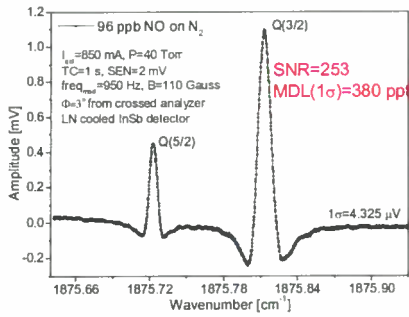
EC-QCL based Magnetic Rotation Spectroscopy



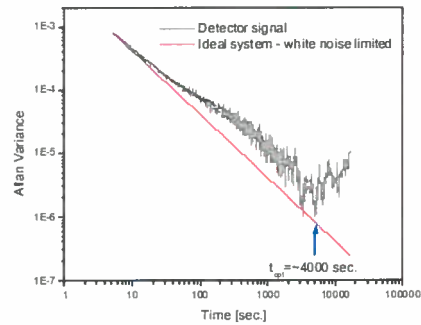
Magnetic rotation spectrum of Q(3/2) and Q(5/2) transitions of nitric oxide



Magnetic Rotation Spectroscopy of Nitric Oxide



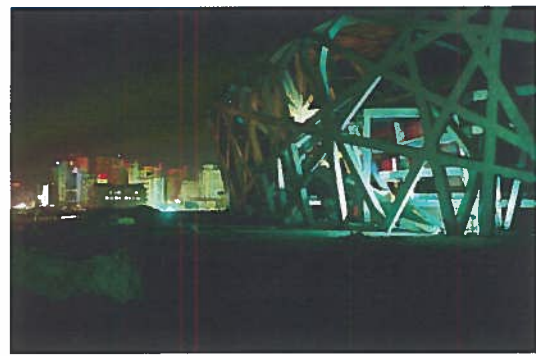
Allan Plot of EC-QCL based MRS NO Sensor



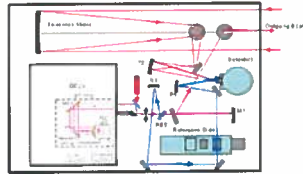
EC-QCLbased MRS NO Sensor for IAP, Beijing



National Stadium, Beijing, July-Sept. 2008

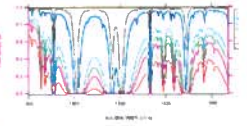


Design of an EC-QCL Based Remote Sensing System

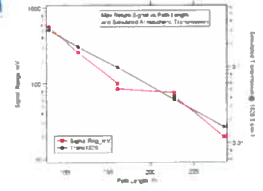


- An upgraded version of a four-laser pulsed QCL system
- The optical set-up, electronics and control software modified for CW-QCL operation
- First tests performed with a DFB CW-QCL operating at $\sim 5.5\mu\text{m}$

Outdoor Open Path Measurements (Influence of Atmospheric Transmission)



- Open Path Measurement CW QCL 1628 cm
- Range: 125 m
- 1000 Hz
- 1000 Hz
- 1000 Hz
- 1000 Hz

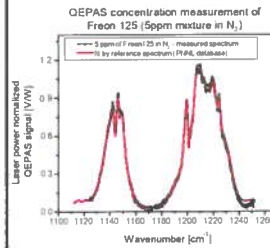


Monitoring of broadband absorbers

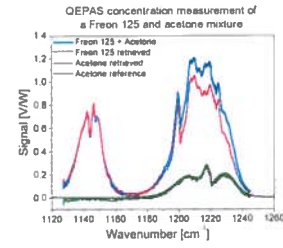
- Freon 125 (C_2HF_5)
 - Refrigerant (leak detection)
 - Safe simulant for toxic chemicals, e.g. chemical warfare agents
- Acetone (CH_3COCH_3)
 - Recognized biomarker for diabetes
- TATP, Acetone Peroxide ($\text{C}_6\text{H}_{12}\text{O}_4$)
 - Highly Explosive



QEPAS based Freon 125 and Acetone concentration measurements with a tunable $8.4\mu\text{m}$ CW EC-QCL



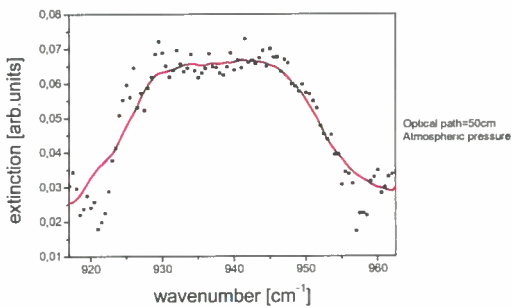
- Minimum detection limit (1 σ) of **-4.5 ppb** was obtained for Freon 125 with an average laser power of 6.6 mW



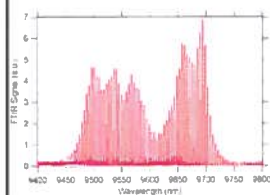
- Wide tunability enables excellent molecular selectivity for broad band absorbers



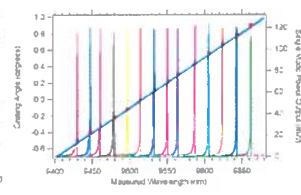
TATP Absorption Spectrum



Applications of Efficient Mid-Infrared $9.6\mu\text{m}$ QCL



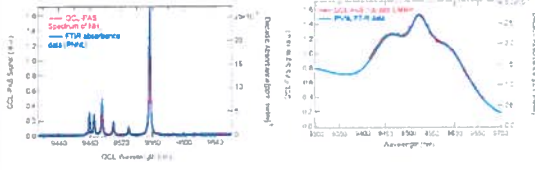
FTIR analysis of the broadband output from FP QCL (68mW per facet from uncoated chip)



Wavelength Tuning Characteristics and FTIR spectrum



Applications of Efficient Mid-Infrared 9.6μm QCLs



Measured QCL-PAS spectrum of NH_3 comparison with FTR spectrum of NH_3

Demonstrated QCL-PAS spectrum of 100 ppb DMMP in CDA superimposed on FTR spectrum

A. Mukherjee et al. *Practicality, Inc and Adtech, Inc, Appl. Optics*, 47: 1543-1548, 2008

QEPAS Performance for 12 Trace Gas Species (July '08)

Molecule (Host)	Frequency, cm ⁻¹	Pressure, Torr	NSEA, cm ² /Watt/Hz	Power, mW	NEC (ppm)
H_2O (N_2) ^{††}	7308.75	60	1.9×10^6	9.3	0.99
HCN (air+50% RH) ^{††}	6539.11	60	4.3×10^6	50	0.16
C_2H_2 (N_2) ^{††}	6523.88	730	4.1×10^6	57	0.03
NH_3 (N_2) ^{††}	6528.76	375	3.1×10^6	60	0.06
C_2H_4 (N_2) ^{††}	6177.07	715	5.4×10^6	15	1.7
C_2H_6 (N_2) ^{††}	6057.09	950	2.0×10^6	13.7	2.1
CO_2 (breath ~100% RH)	6161.25	130	8.2×10^6	45	40
H_2S (N_2) ^{††}	6337.63	780	5.6×10^6	45	0.20
CO_2 (N_2 +1.5% H_2O) ^{††}	4791.26	50	1.4×10^6	4.4	18
CH_2O (N_2 +35% RH) ^{††}	2804.90	75	8.7×10^6	7.2	0.12
CO (N_2)	2196.66	50	5.3×10^6	1.1	3.5
CO (propylene)	2196.66	50	7.4×10^6	0.5	0.14
N_2O (air+5% N_2)	2195.63	50	1.5×10^6	19	0.007
$\text{C}_2\text{H}_2\text{OH}$ (N_2) ^{†††}	1934.2	770	2.2×10^6	10	50
$\text{C}_2\text{H}_4\text{O}$ (N_2) ^{†††}	1208.62	770	7.8×10^6	6.6	0.009
NH_3 (N_2) ^{††}	1046.39	110	1.6×10^6	20	0.006

^{††} Improved microcavities
^{†††} Improved microcavities and double optical pass through ADM
 †††† With amplitude modulation and optical microcavities
 NSEA - normalized noise equivalent absorption coefficient
 NEC - noise equivalent concentration for available laser power and ††††† time constant, 10 dB cut filter delay

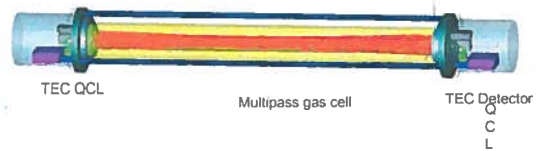
For comparison, conventional PAS 2.2 (2.6) $\times 10^7$ cm²/W/Hz (1.800) 10,300 Hz for NH_3 (1991)

* M. E. Wubbler et al. *Appl. Opt.* 42, 2119-2126 (2003); ** J. S. Fajman et al. *S&E Intl.* ICES 2002-06, 3152



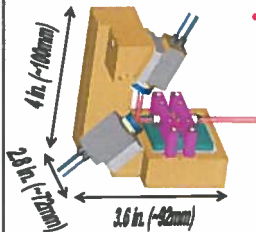
Future of Chemical Trace Gas Sensing

Conceptual Design of Ultra-compact QCL Trace Gas Sensor



Aerodyne, Inc. Barry McManus et al., 2008

New design of fast broadly tunable EC-QCLs (2008)

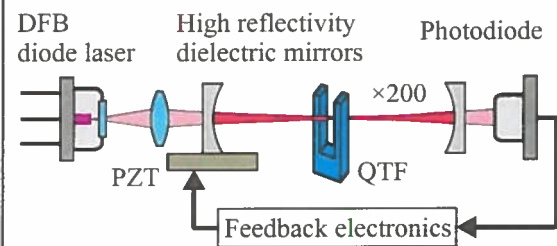


- New optical configuration
Folded cavity (configuration #1)
- Fast tuning capabilities:
 - Coarse Broadband Scanning (~55 cm^{-1} @ 5 μm) **up to 5 KHz** (compared to available technologies <10Hz)
 - High resolution mode-hop free tuning (~3.2 cm^{-1} @ 5 μm) **up to 5 KHz** (compared to available technology 100-200 Hz)

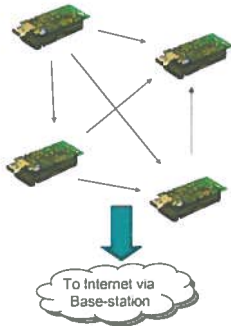
Patent pending, G. Wysocki, F. K. Tittel, 2007



Proposed QEPAS-OPBC Sensor Configuration



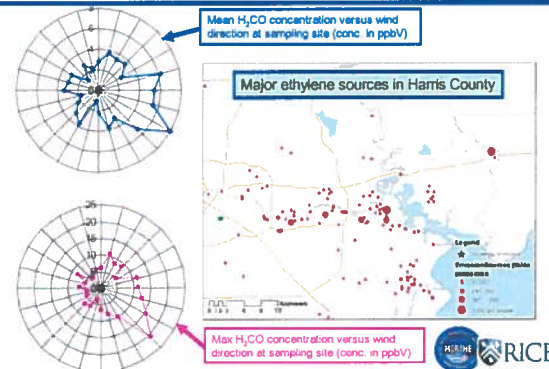
Wireless Sensor Networks for Trace Gas Sensing



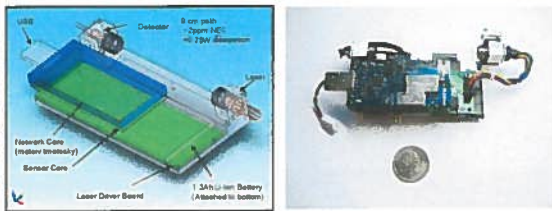
- Each point called “mote”
- Advantages?
 - Spatial resolution
 - Measure fluxes
- What is needed?
 - Low power
 - Low cost
 - Ultra miniature
 - Replicable
 - Autonomy



H₂CO Concentration (ppb) Versus Wind Direction



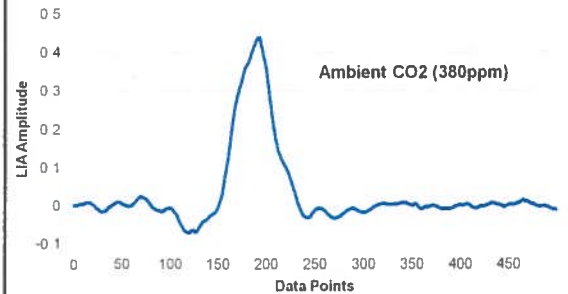
PHOTONS v4.0 - 2.7 μ m CO₂ Direct Absorption Based Sensor



- Small size
- Relatively low cost
- High efficiency switching power supplies
- PWM Peltier cooler driver
- 0.2W control system power consumption
- Detection sensitivity of CO₂ 1 ppm with 1sec. lock-in time constant
- Over 100x improvement in sensitivity is possible @ 4.2 μ m



LAS based CO₂ Spectrum at 2.7 μ m



Summary & Future Directions of QCL based Gas Sensor Technology

- Quantum and Interband Cascade Laser based Trace Gas Sensors
 - Compact, tunable, and robust
 - High sensitivity ($<10^{-4}$) and selectivity (3 to 500 MHz)
 - Capable of fast data acquisition and analysis
 - Detected 13 trace gases to date: NH₃, CH₄, N₂O, CO₂, CO, NO, H₂O, COS, C₂H₄, H₂CO, SO₂, C₂H₅OH, C₂HF₃ and several isotopic species of C, O, N and H.
- New Applications of Trace Gas Detection
 - Environmental Monitoring (urban quality - H₂CO and isotopic ratio measurements of CO₂ and CH₄, fire detection and quantification of engine exhausts)
 - Industrial process control and chemical analysis (NO, NH₃, H₂O, and H₂S)
 - Medical & biomedical diagnostics (NO, NH₃, N₂O, H₂CO and CH₃COCH₃)
 - Hand-held sensors and sensor network technologies (CO₂)
- Future Directions and Collaborations
 - Improvements of the existing sensing technologies using novel, thermoelectrically cooled, cw, high power, and broadly wavelength tunable mid-IR interband and intersubband quantum cascade lasers
 - New applications enabled by novel broadly wavelength tunable quantum cascade lasers based on heterogeneous EC-QCL (i.e. sensitive concentration measurements of broadband absorbers, in particular VOCs, HCs and multi-species detection)
 - Development of optically gas sensor networks based on QEPAS and LAS

