



Advanced Infrared Semiconductor Laser based Chemical Sensing Technologies: Opportunities and Challenges

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OUTLINE

IEEE
LEOS

Texas A & M
College
Station

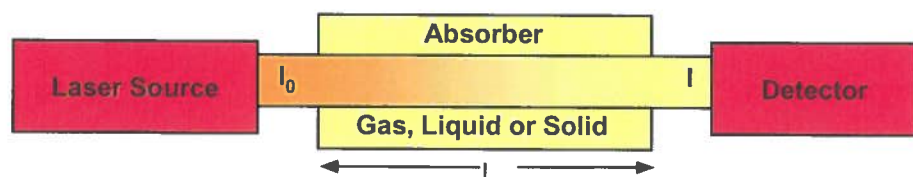
April 8, 2008

- Motivation: Wide Range of Chemical Sensing
- Fundamentals of Laser Absorption Spectroscopy
- New laser sources and sensing technologies
- Selected Applications of Trace Gas Detection
 - Quartz Enhanced L-PAS (ammonia, Freon 125 and acetone)
 - Nitric Oxide Detection (Faraday Rotation & Remote Sensing)
- Future Directions and Conclusions

Wide Range of Trace Gas Sensing Applications

- **Urban and Industrial Emission Measurements**
 - Industrial Plants
 - Combustion Sources and Processes (e.g. fire detection)
 - Automobile, Truck, Aircraft and Marine Emissions
- **Rural Emission Measurements**
 - Agriculture & Forestry, Livestock
- **Environmental Monitoring**
 - Atmospheric Chemistry
 - Volcanic Emissions
- **Chemical Analysis and Industrial Process Control**
 - Petrochemical, Semiconductor, Nuclear Safeguards, Pharmaceutical, Metals Processing, Food & Beverage Industries
- **Spacecraft and Planetary Surface Monitoring**
 - Crew Health Maintenance & Life Support
- **Applications in Health and Life Sciences**
- **Technologies for Law Enforcement and National Security**
- **Fundamental Science and Photochemistry**

Fundamentals of Laser Absorption Spectroscopy

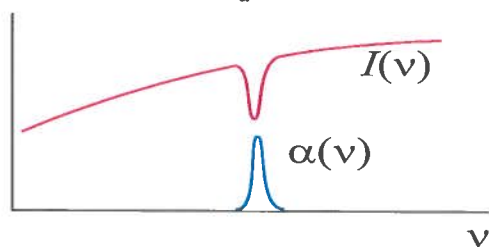


Beer-Lambert's Law of Linear Absorption

$$I(\nu) = I_0 e^{-\alpha(\nu) P_a L}$$

$\alpha(\nu)$ - absorption coefficient [$\text{cm}^{-1} \text{atm}^{-1}$]; L - path length [cm]

ν - frequency [cm^{-1}]; P_a - partial pressure [atm]



$$\alpha(\nu) = C \cdot S(T) \cdot g(\nu - \nu_0)$$

C - total number of molecules of absorbing gas/atm/ cm^3 [$\text{molecule} \cdot \text{cm}^{-3} \cdot \text{atm}^{-1}$]

S - molecular line intensity [$\text{cm} \cdot \text{molecule}^{-1}$]

$g(\nu - \nu_0)$ - normalized spectral lineshape function [cm], (Gaussian, Lorentzian, Voigt)

Key Requirements: Sensitivity, specificity, rapid data acquisition and multi-species detection

Optimum Molecular Absorbing Transition

- Overtone or Combination Bands (NIR)
- Fundamental Absorption Bands (MID-IR)

Long Optical Pathlengths

- Multipass Absorption Cell
- Cavity Enhanced, Cavity Ringdown & Intracavity Spectroscopy
- Open Path Monitoring (with retro-reflector)

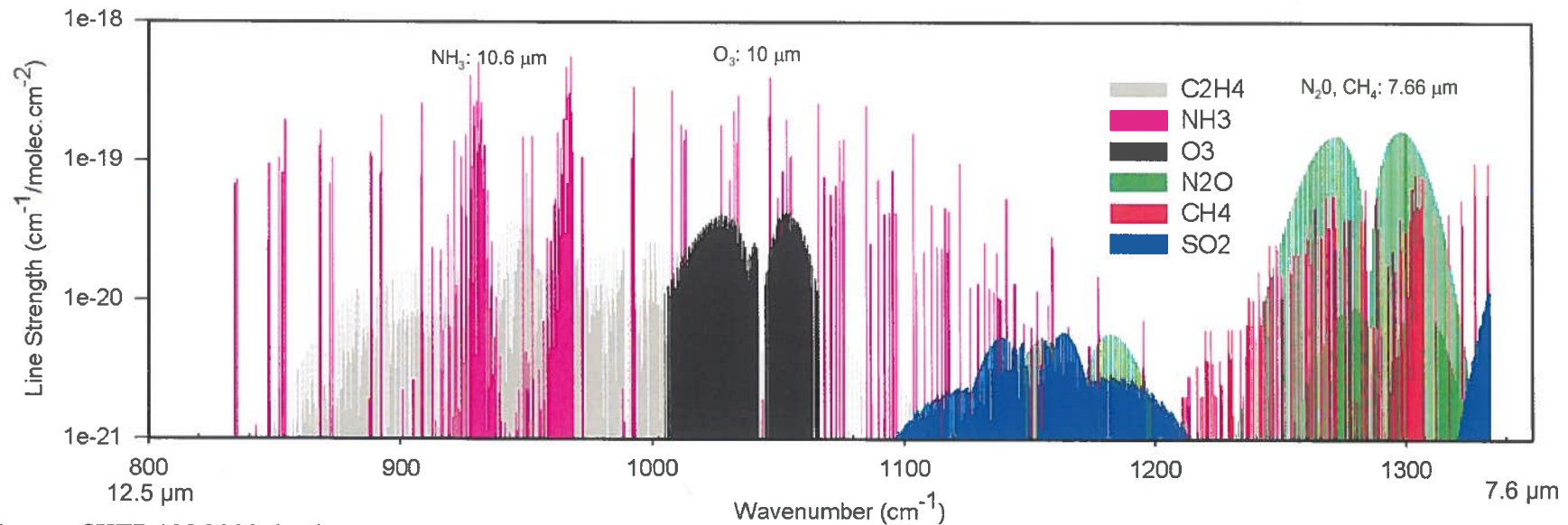
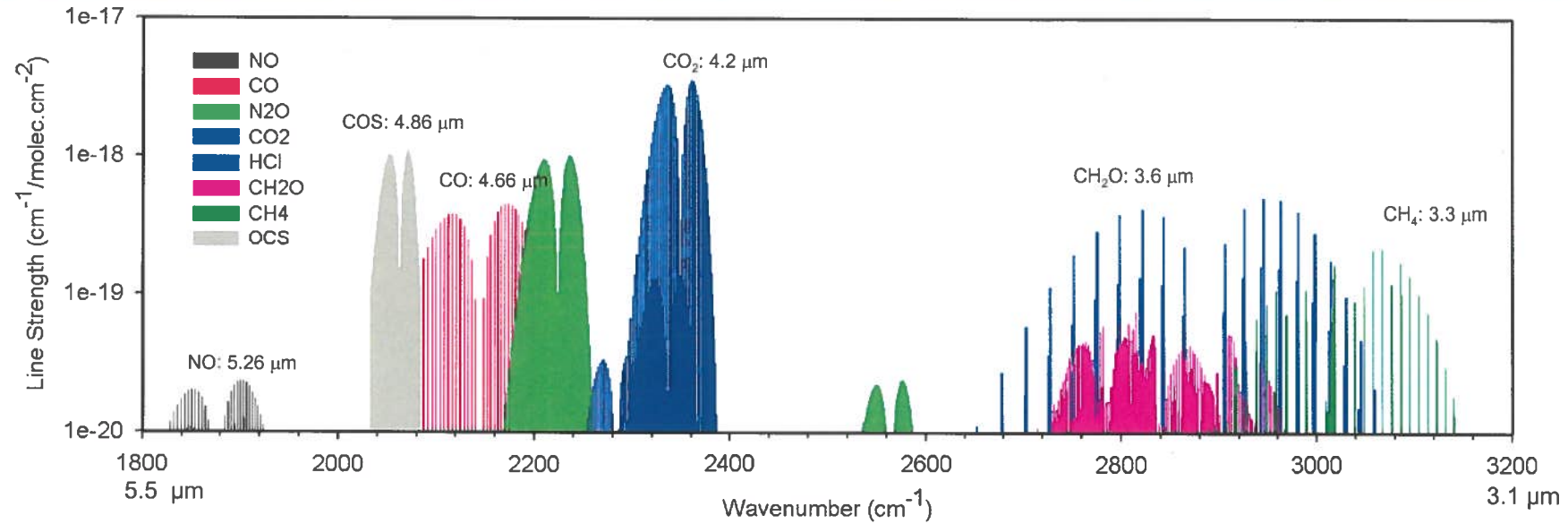
Spectroscopic Detection Schemes

- Wavelength & Frequency Modulation
- Balanced Detection
- Zero-air Subtraction
- Photoacoustic Spectroscopy

Mid-IR Source Requirements for Laser Spectroscopy

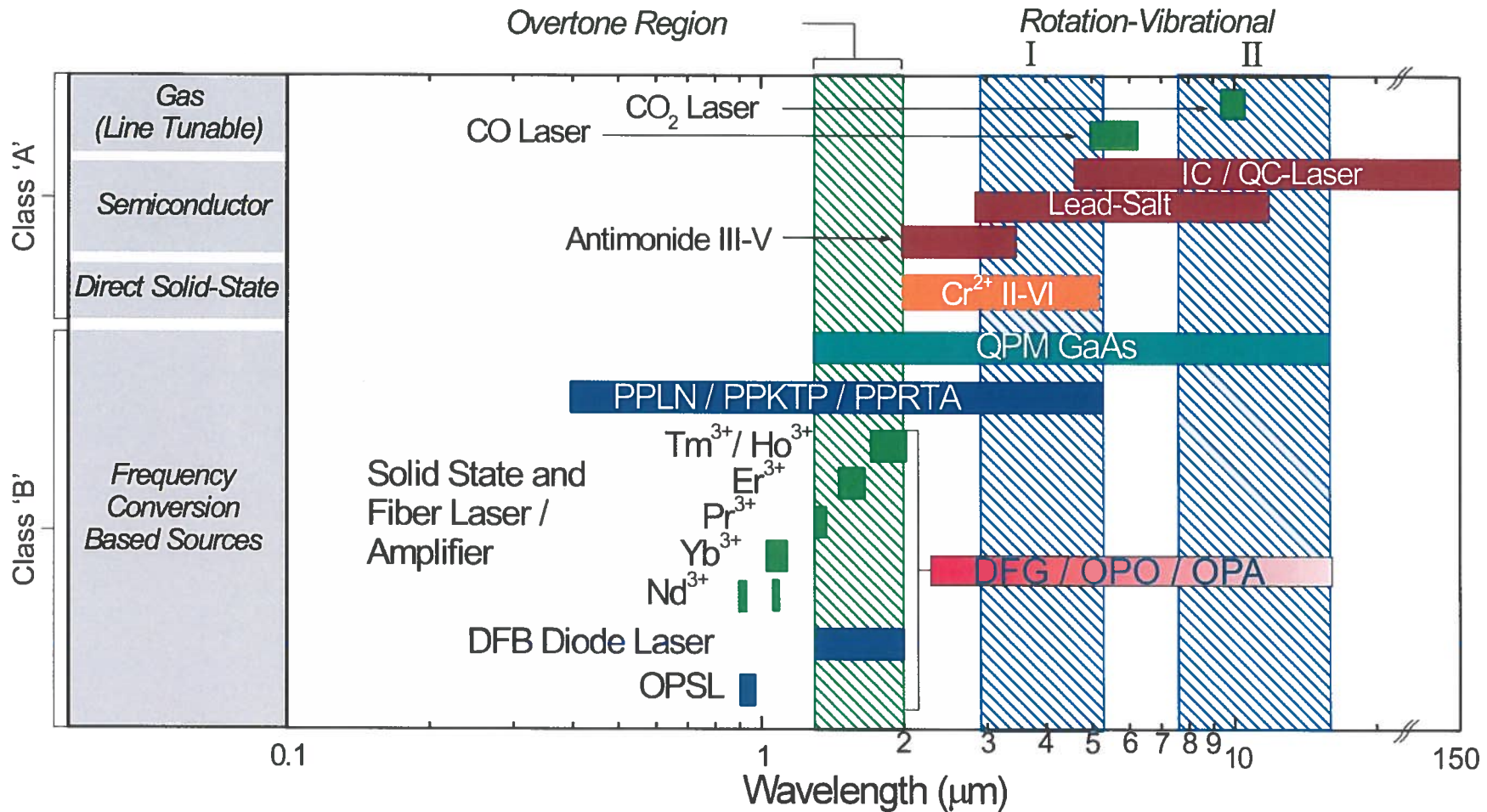
<u>REQUIREMENTS</u>	<u>IR LASER SOURCE</u>
Sensitivity (% to ppt)	Wavelength, Power
Selectivity (Spectral Resolution)	Single Mode Operation and Narrow Linewidth
Multi-gas Components, Multiple Absorption Lines and Broadband Absorbers	Tunable Wavelength
Directionality or Cavity Mode Matching	Beam Quality
Rapid Data Acquisition	Fast Time Response
Room Temperature Operation	No Consumables
Field deployable	Compact & Robust

Molecular Absorption Spectra within two Mid-IR Atmospheric Windows



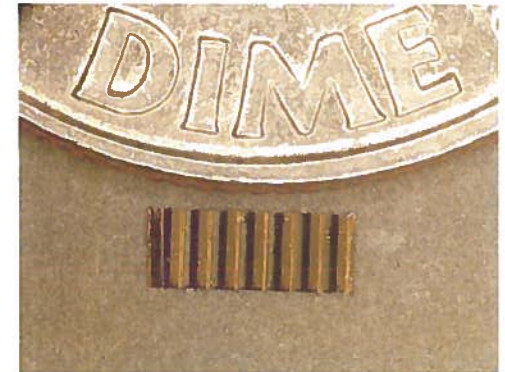
Source: HITRAN 2000 database

IR Laser Sources and Wavelength Coverage

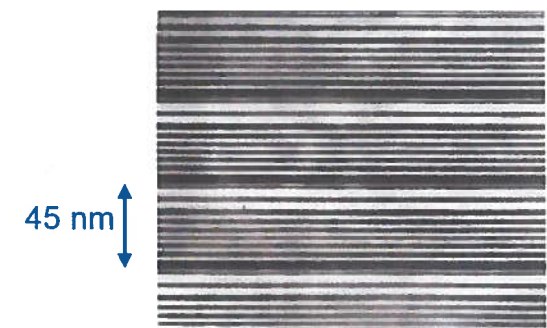


Key Characteristics of mid-IR QCLs and ICL Sources

- **Band – structure engineered devices**
(emission wavelength is determined by layer thickness – MBE or MOCVD);
mid-infrared QCLs operate from 3 to 24 μm
- Compact, reliable, stable, long lifetime, and commercial availability
- Fabry-Perot (FP), single mode (DFB) and multi-wavelength
- **Spectral tuning range in the mid-IR**
(4-24 μm for QCLs and 3-5 μm for ICLs)
 - 1.5 cm^{-1} using injection current control
 - 10-20 cm^{-1} using temperature control
 - > 265 cm^{-1} using an external grating element and with heterogeneous cascade active region design
- **Narrow spectral linewidth**
cw: 0.1 - 3 MHz & <10Khz with frequency stabilization (0.0004 cm^{-1});
pulsed: ~ 300 MHz (chirp from heating)
- **High pulsed and cw powers at TEC/RT temperatures**
 - Pulsed peak powers of 1.6 W; high temperature operation ~425K
 - Average power levels: 1-600 mW (current wall plug $\eta \sim 4\%$)
 - ~ 50 mW, TEC CW DFB @ 5 and 10 μm Alpes; Princeton,
 - Adtech Optics, Maxison Technologies, Hamamatsu, Daylight
~ 300 mW @ 8.3 μm (Agilent Technologies & Harvard)
 - > 600 mW (CW FP) @RT & a wall plug efficiency of >9.3%;
>150 mW (CW DFB) at 298 K (Northwestern)

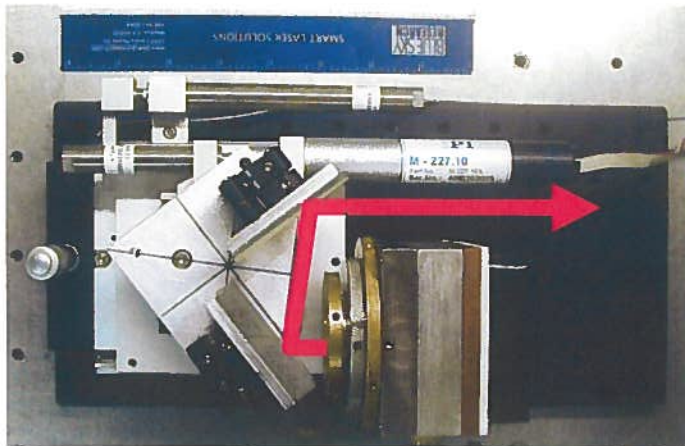
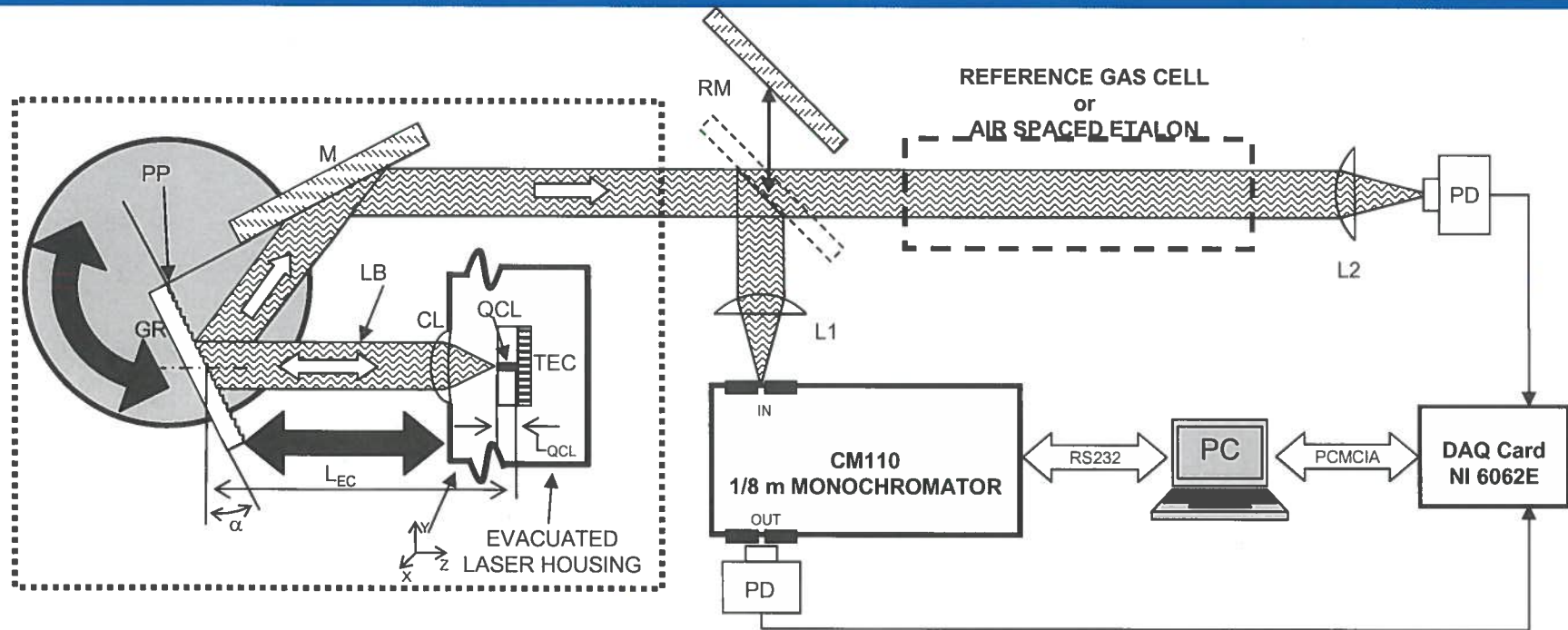


4 mm



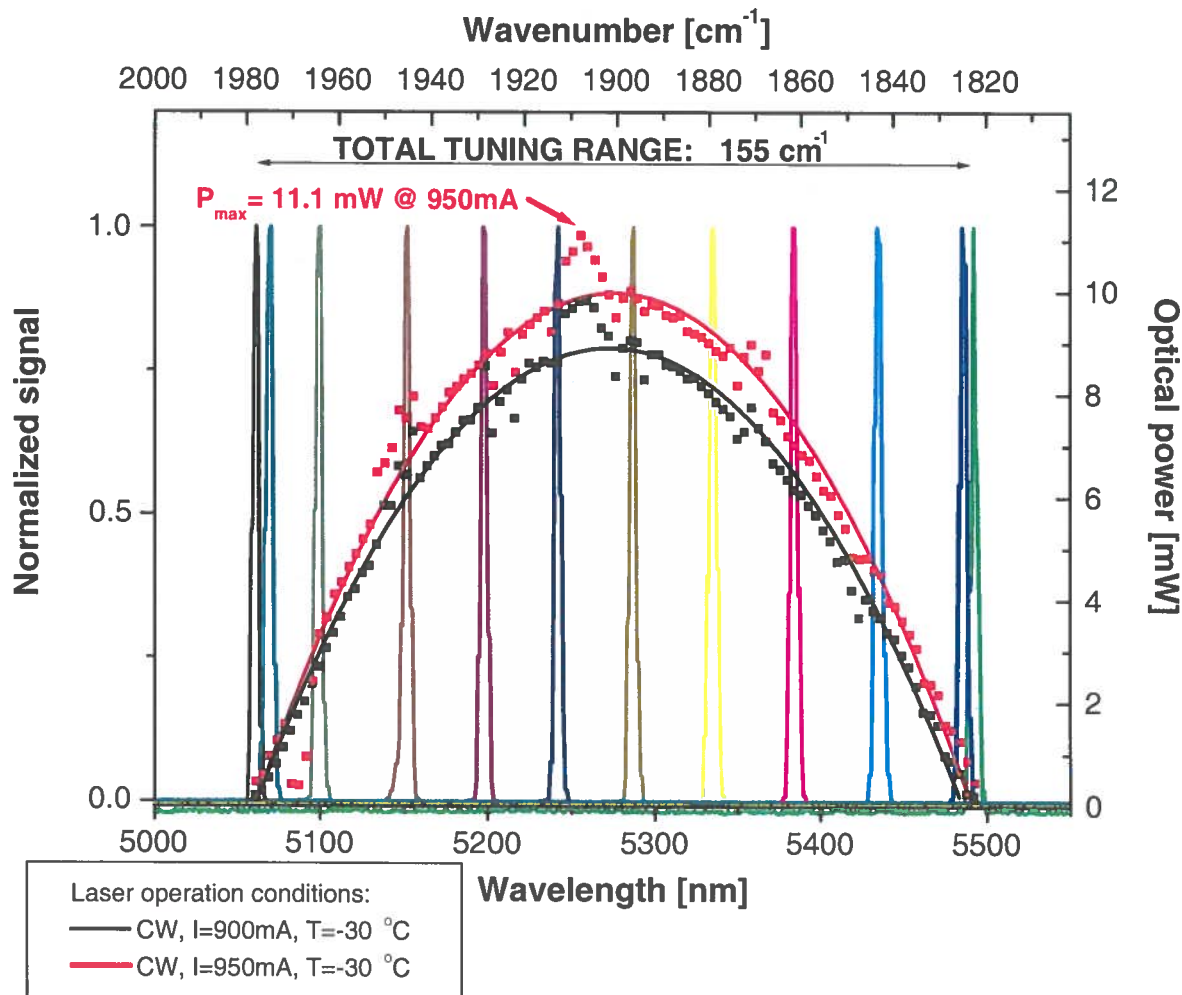
Widely Tunable, CW, TEC
Quantum Cascade Lasers

Tunable external cavity QCL based spectrometer

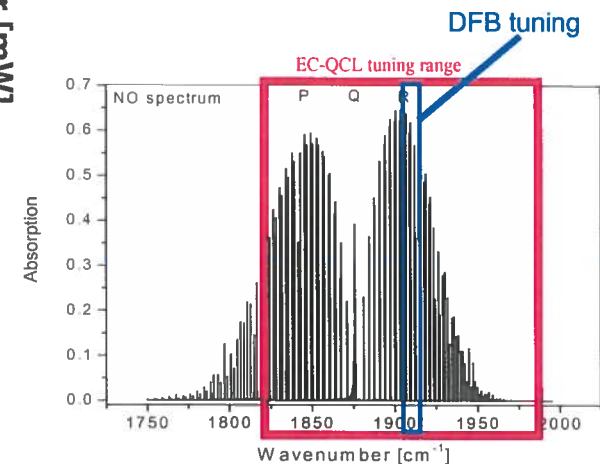


- Fine wavelength tuning
 - PZT controlled EC-length
 - PZT controlled grating angle
 - QCL current control
- Motorized coarse grating angle tuning
- Vacuum tight QCL enclosure with build-in 3D lens positioner (TEC laser cooling + optional chilled water cooling)

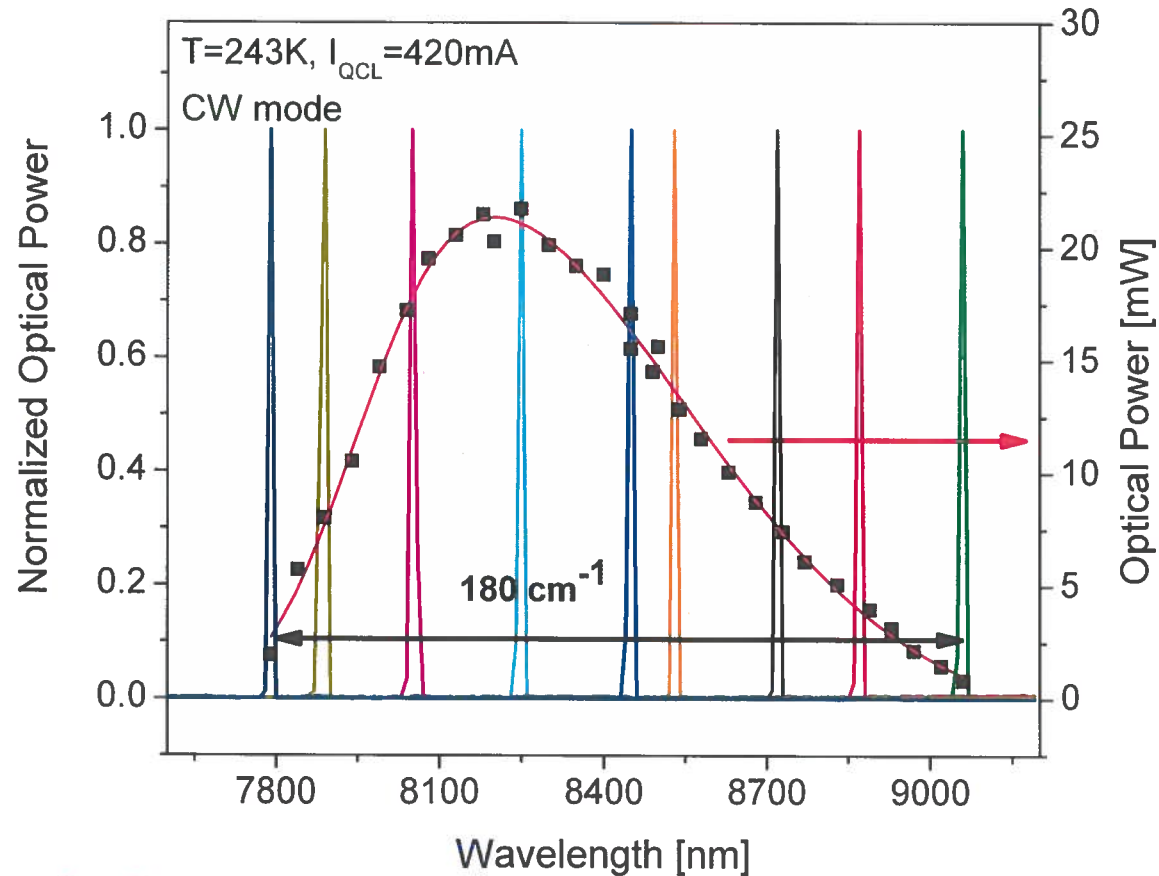
Wide Wavelength Tuning of a 5.3 μm EC-QCL



- Coarse wavelength tuning of **155 cm^{-1}** is performed by varying diffraction grating angle
- Power output is **$\sim 11 \text{ mW}$**
- Access to Q(3/2) transition of NO at 1875.8 cm^{-1} for LMR spectroscopy



Performance of 8.4 μm EC-QCL Spectroscopic Source



Tunability **180 cm^{-1}** @ $8.4 \mu\text{m}$ (1100 to 1280 cm^{-1})

AR coating:

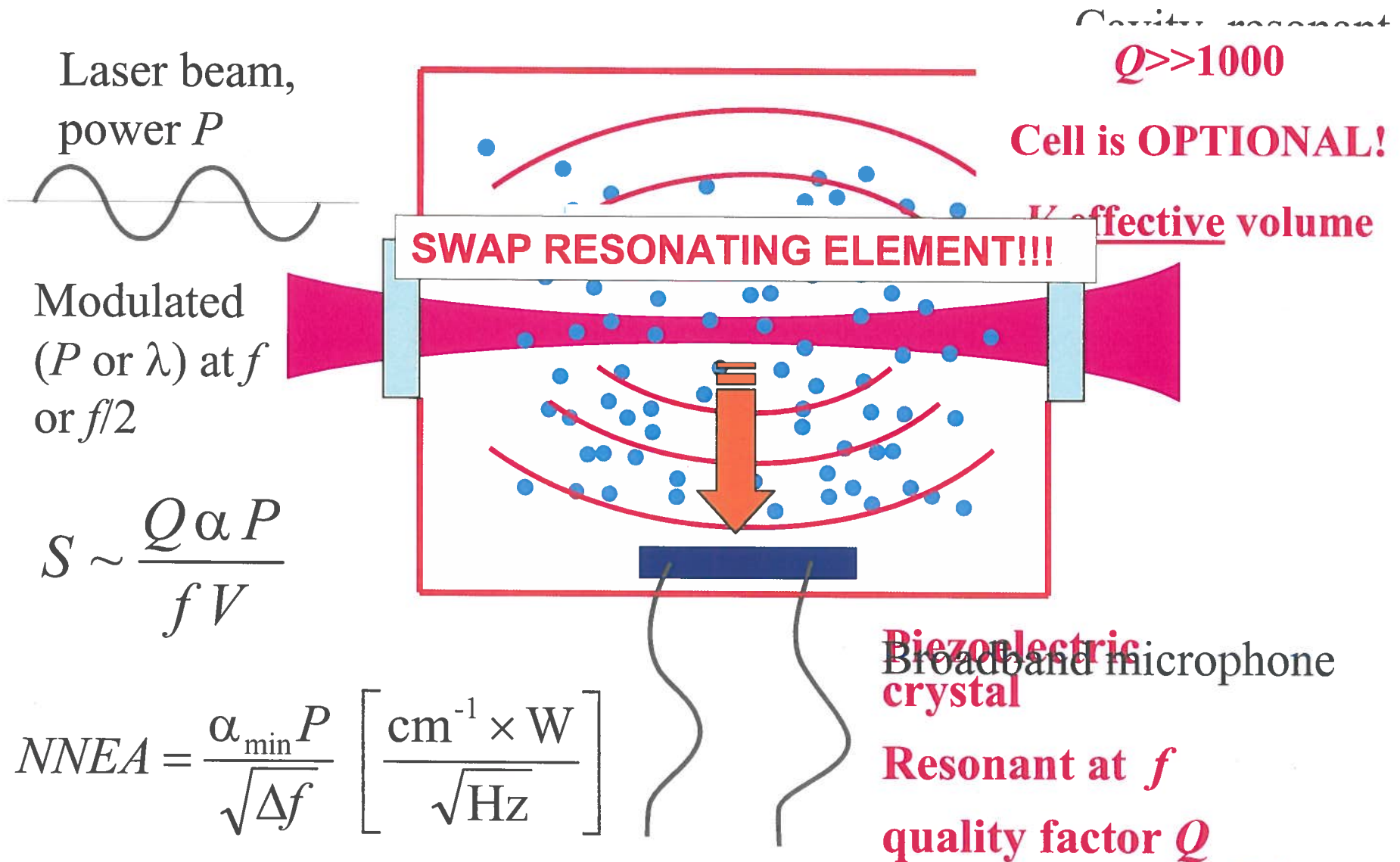
$$R_{\text{AR}} \approx 2 \times 10^{-4}$$

$P_{\text{EC-opt}}$ up to **50 mW (cw)**
($I_{\text{QCL}} = 680 \text{ mA} \rightarrow P = 44 \text{ mW}$)



Quartz Enhanced Photoacoustic Spectroscopy

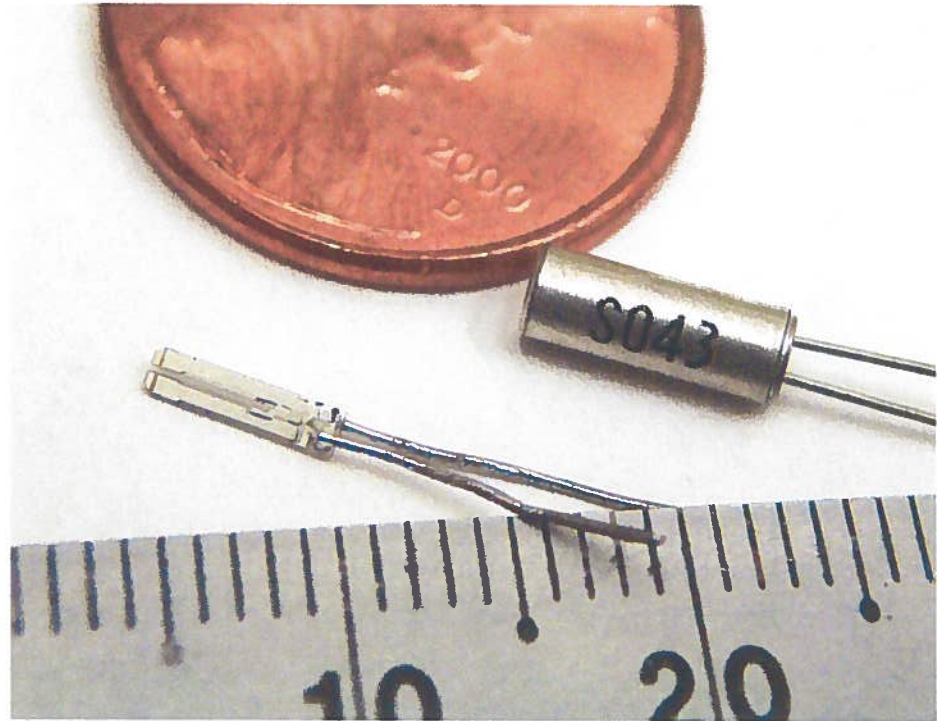
From conventional PAS to QEPAS



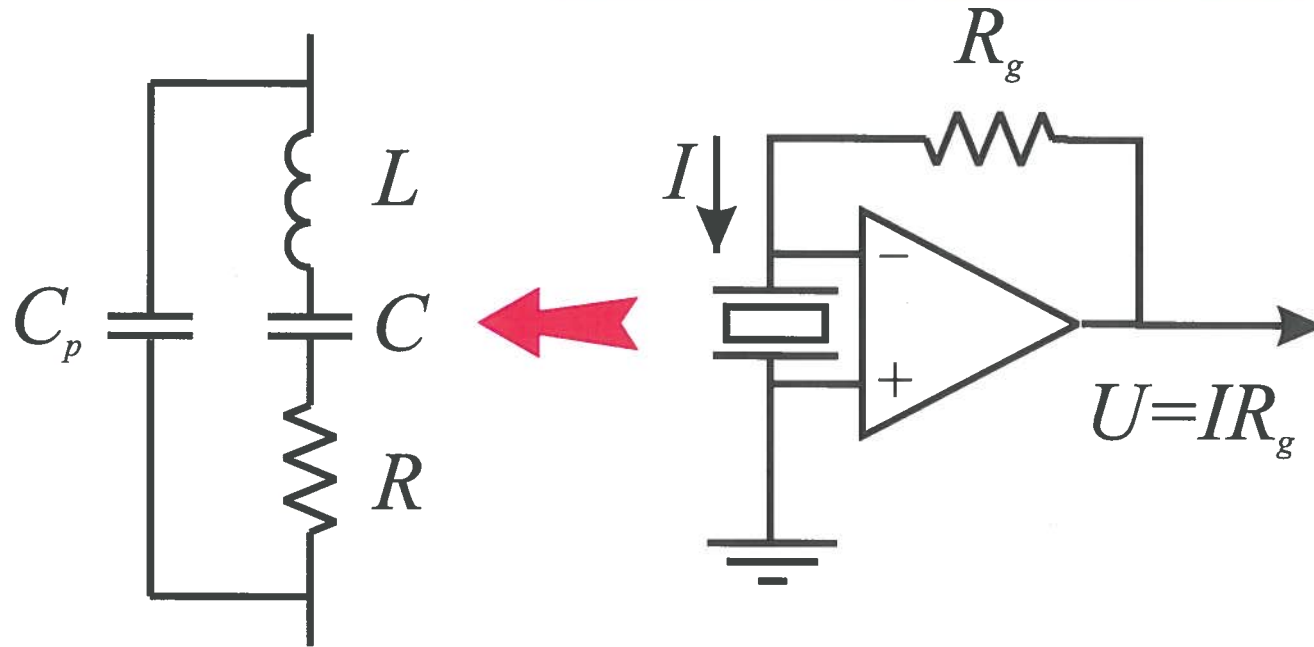
Quartz Tuning Fork (TF) as a Resonant Microphone



- Resonant frequency $f=32.8$ kHz
- Intrinsically high Q factor: $Q_{\text{vacuum}} \sim 125\,000$, $Q_{\text{air}} \sim 10\,000$ at ambient conditions;
- Piezoelectric: requires no transducer
- Miniature size
- Mass produced for clocks – low cost



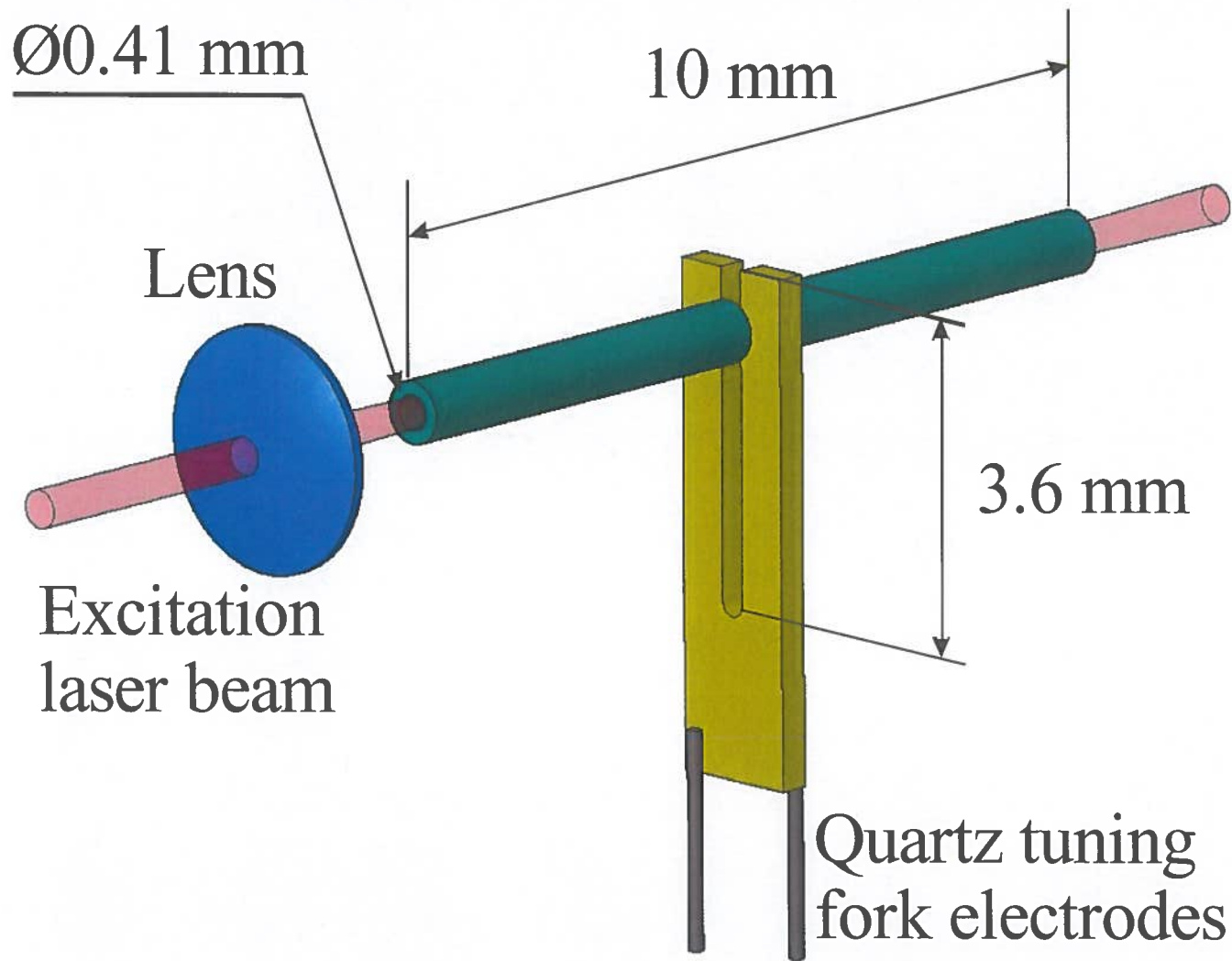
QEPAS Signal Detection



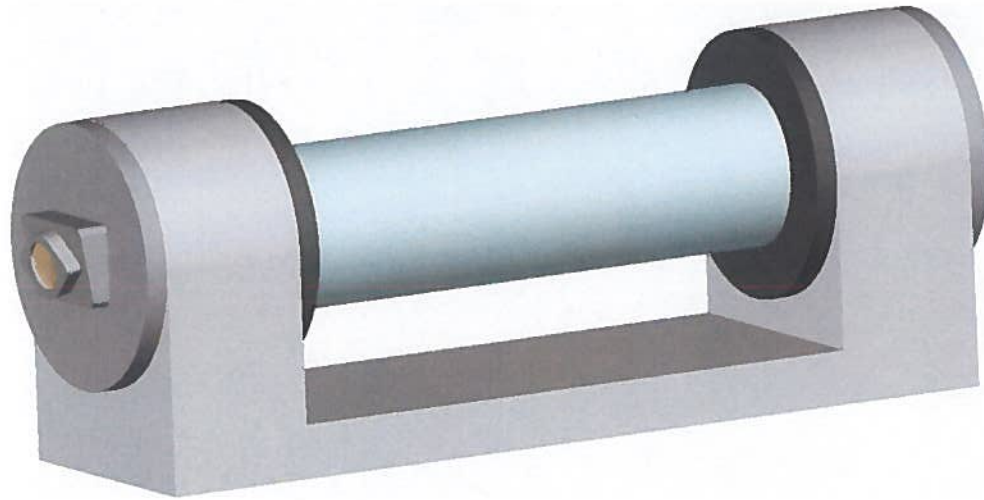
Amplifier:

- Fast
- Low noise
- High impedance
- Low 1/f noise

Absorption Detection Module for QEPAS based Gas Sensor

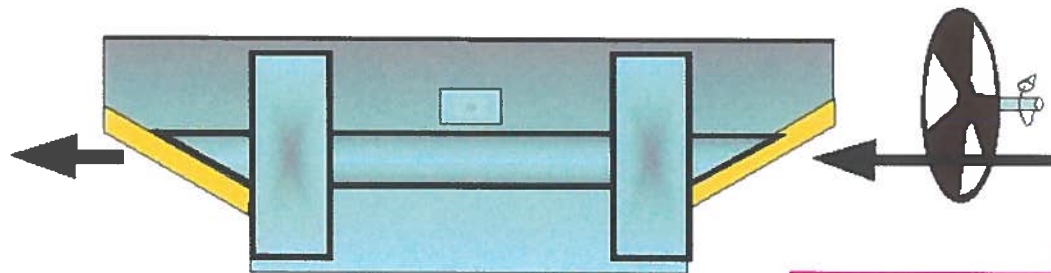


Comparative Size of Absorption Detection Modules (ADM)



- Resonant frequency $f = 32.8$ kHz
- Intrinsically high Q factor: $Q_{\text{vacuum}} \sim 125\,000$
 $Q_{\text{air}} \sim 10\,000$ for ambient conditions
- Piezoelectric: requires no transducer
- Miniature size
- Mass produced for watches & clocks – low cost

Optical multipass cell (100 m):
 $l \sim 70$ cm, $V \sim 3000$ cm³

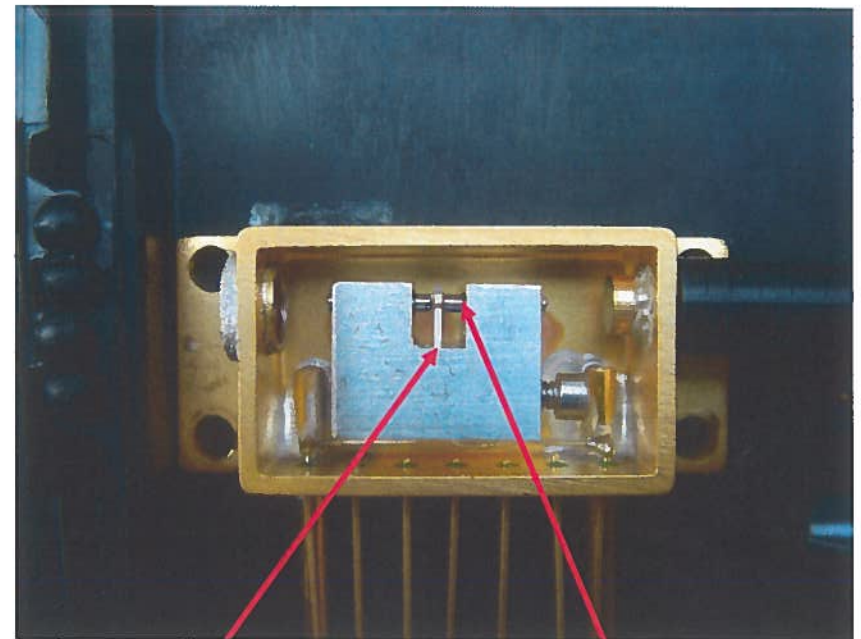
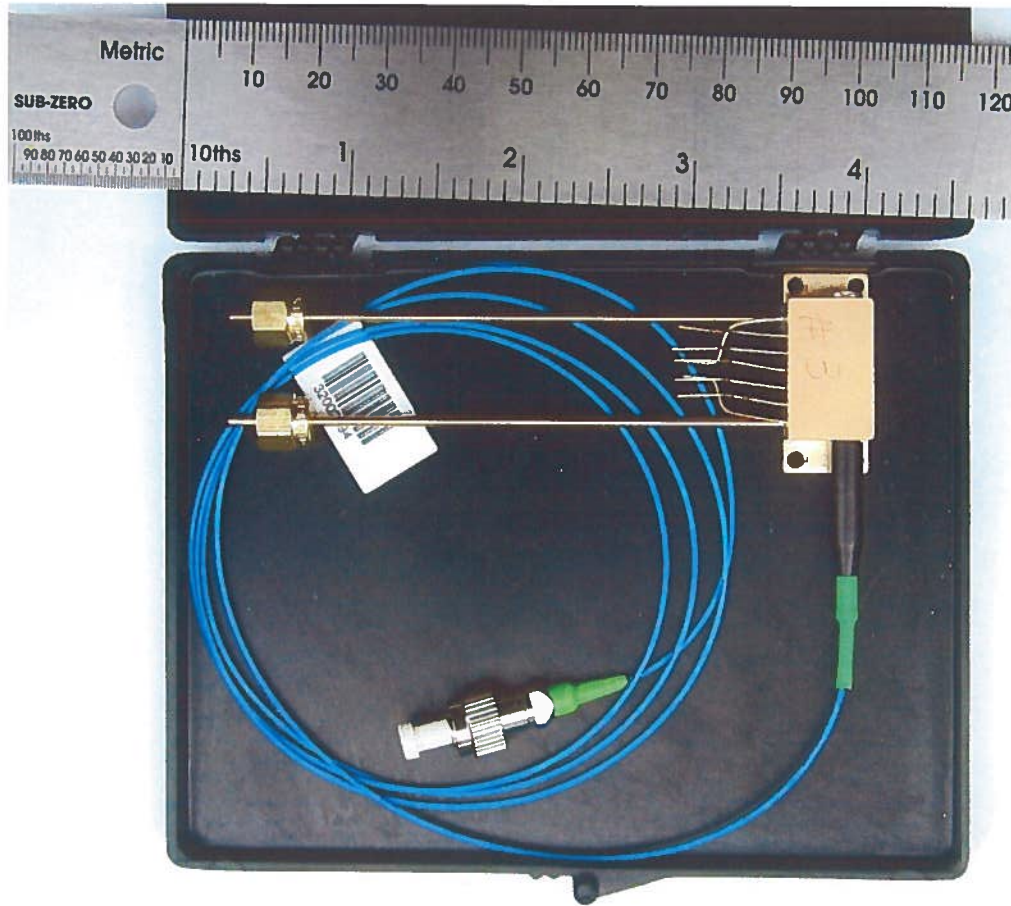


Resonant photoacoustic cell (1000 Hz):
 $l \sim 60$ cm, $V \sim 50$ cm³



QEPAS spectrophone:
 $l \sim 1$ cm, $V \sim 0.05$ cm³

Alignment-free QEPAS Absorption Detection Module



Quartz Tuning Fork

Acoustic Micro Resonator

Merits of QE Laser-PAS based Trace Gas Detection

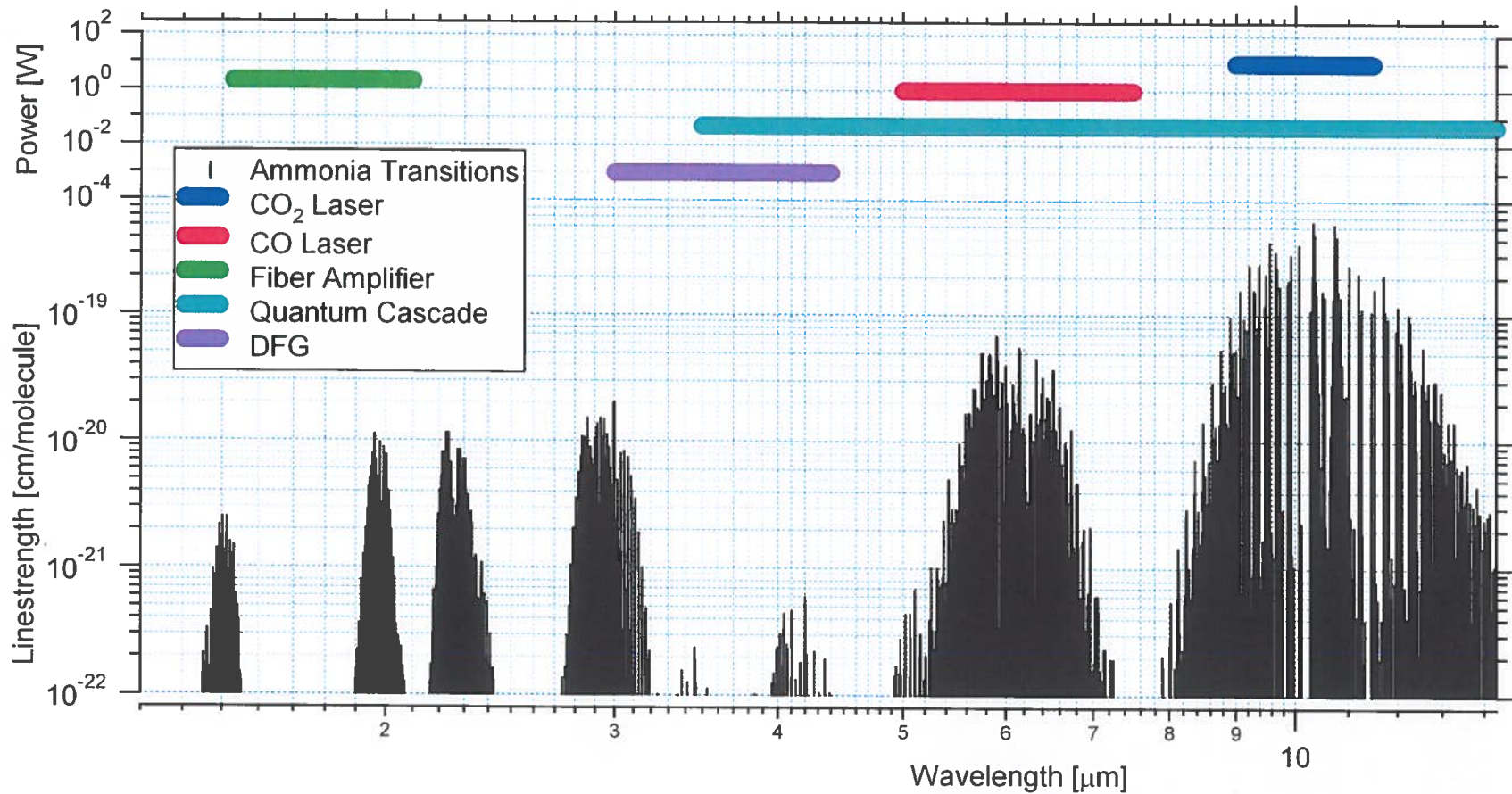
- High sensitivity (ppm to ppb gas concentration levels) and excellent dynamic range
- Immune to ambient and flow acoustic noise, laser noise and etalon effects
- Significant reduction of sample volume ($< 1 \text{ mm}^3$)
- Applicable over a wide range of pressures
- Temperature, pressure and humidity insensitive
- Rugged and low cost (compared to other optical sensor architectures)

Trace Gas Sensing Examples

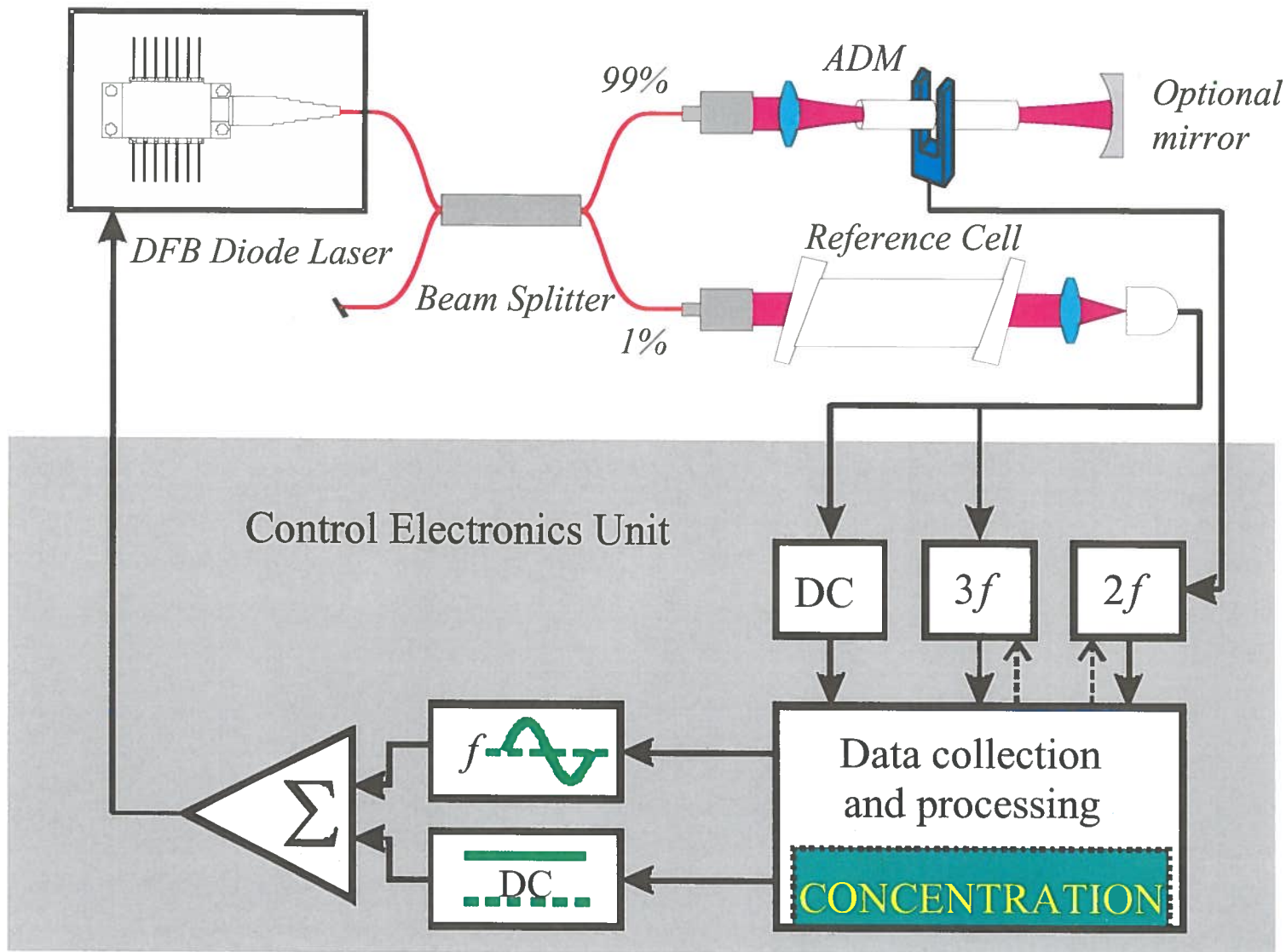
Motivation for NH₃ Detection

- Monitoring of gas separation processes
- Detection of ammonium-nitrate explosives
- Spacecraft related gas monitoring
- Monitoring NH₃ concentrations in the exhaust stream of NO_x removal systems based on selective catalytic reduction (SCR) techniques
- Semiconductor process monitoring & control
- Monitoring of industrial refrigeration facilities
- Pollutant gas monitoring
- Atmospheric chemistry
- Medical diagnostics (kidney & liver dysfunctions)

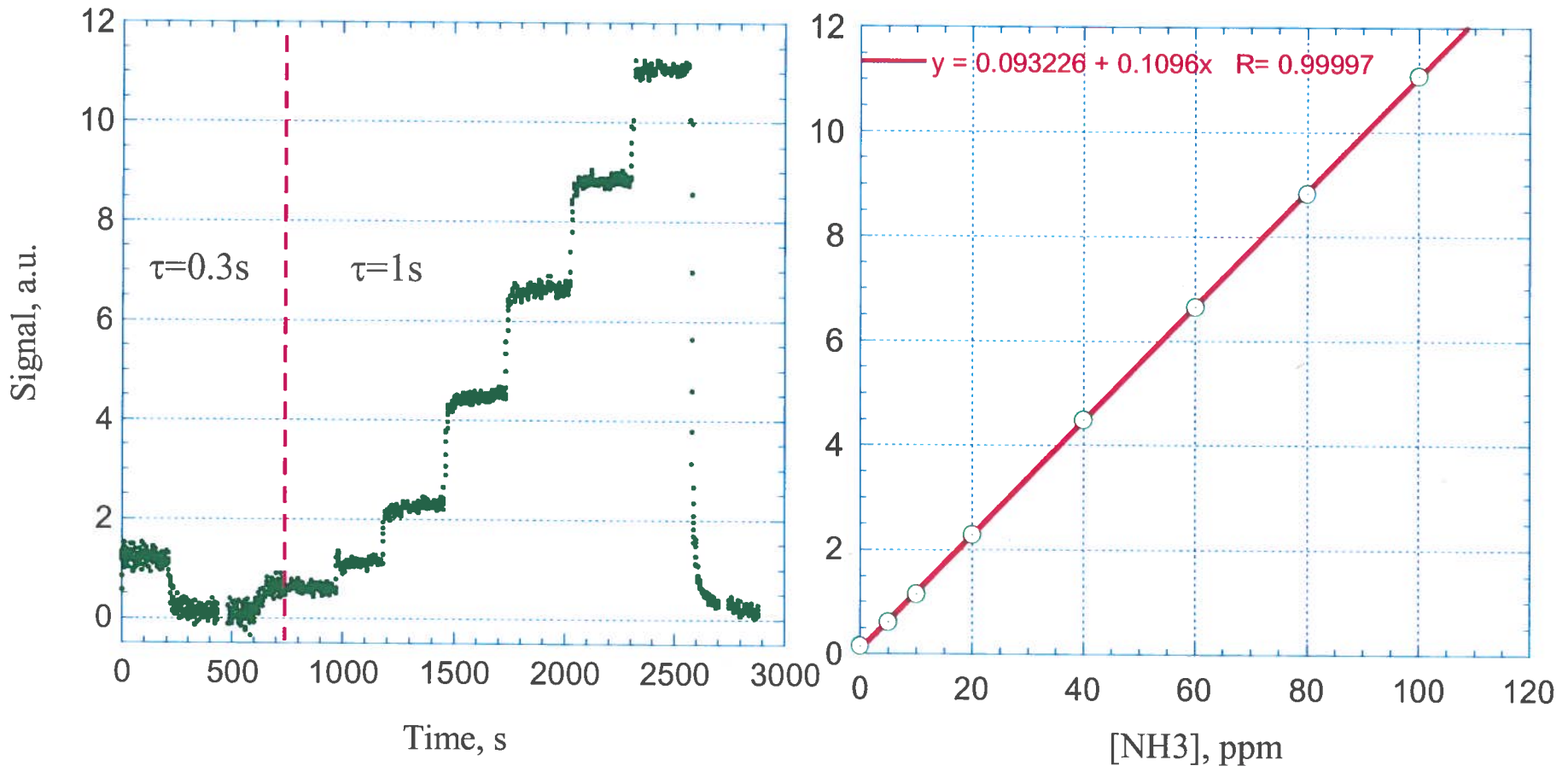
Infrared NH₃ Absorption Spectra



QEPAS based Gas Sensor Architecture



Calibration and Linearity of a 1.53 μm QEPAS based NH_3 Sensor



Noise-equivalent concentration (NEC).
for $t=1\text{s}$ time constant is 0.06 ppm for
60mW excitation power at 6528.76 cm^{-1}

90 last points of each step averaged

Noise-equivalent absorption (NEA) coefficient $k=3.1 \times 10^{-9}\text{ cm}^{-1}\text{W}/\text{Hz}^{1/2}$

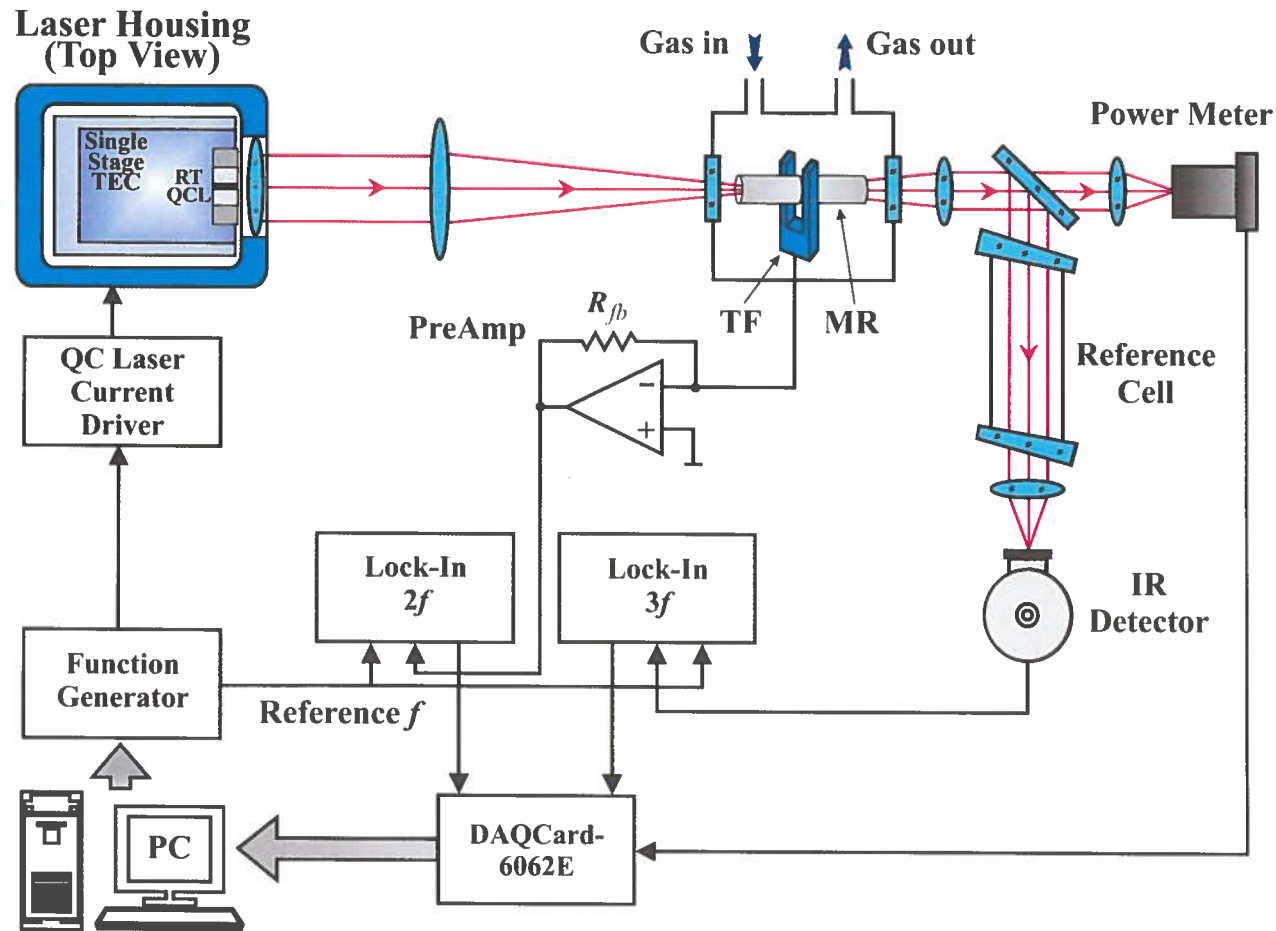
Biomarkers Present in Exhaled Human Breath

More than 400 different molecules in breath;
many with well defined biochemical pathways

BROADBAND ABSORBERS

	Compound	Concentration	Physiological basis/Pathology Indication
→	Acetaldehyde	ppb	Ethanol metabolism
→	Acetone	ppm	Decarboxylation of acetoacetate, diabetes
	Ammonia	ppb	protein metabolism, liver and renal disease
	Carbon dioxide	%	Product of respiration, <i>Helicobacter pylori</i>
	Carbon disulfide	ppb	Gut bacteria, schizophrenia
	Carbon monoxide	ppm	Production catalyzed by <i>heme oxygenase</i>
	Carbonyl sulfide	ppb	Gut bacteria, liver disease
	Ethane	ppb	Lipid peroxidation and oxidative stress
→	Ethanol	ppb	Gut bacteria
	Ethylene	ppb	Lipid peroxidation, oxidative stress, cancer
→	Hydrocarbons	ppb	Lipid peroxidation/metabolism
	Hydrogen	ppm	Gut bacteria
→	Isoprene	ppb	Cholesterol biosynthesis
	Methane	ppm	Gut bacteria
→	Methanethiol	ppb	Methionine metabolism
	Methanol	ppb	Metabolism of fruit
→	Methylamine	ppb	Protein metabolism
	Nitric oxide	ppb	Production catalyzed by <i>nitric oxide synthase</i>
	Oxygen	%	Required for normal respiration
→	Pentane	ppb	Lipid peroxidation, oxidative stress
	Water	%	Product of respiration

9.56 μm CW DFB QCL based QEPAS Ammonia Sensor



Noise-equivalent concentration (NEC)
for $t=1\text{s}$ time constant is 0.006 ppm for 20mW
excitation power at 1046.4 cm^{-1} (110 Torr)

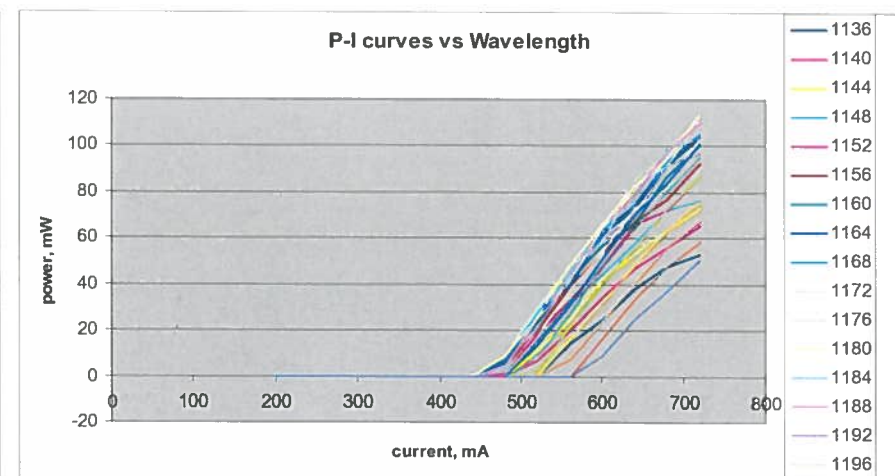
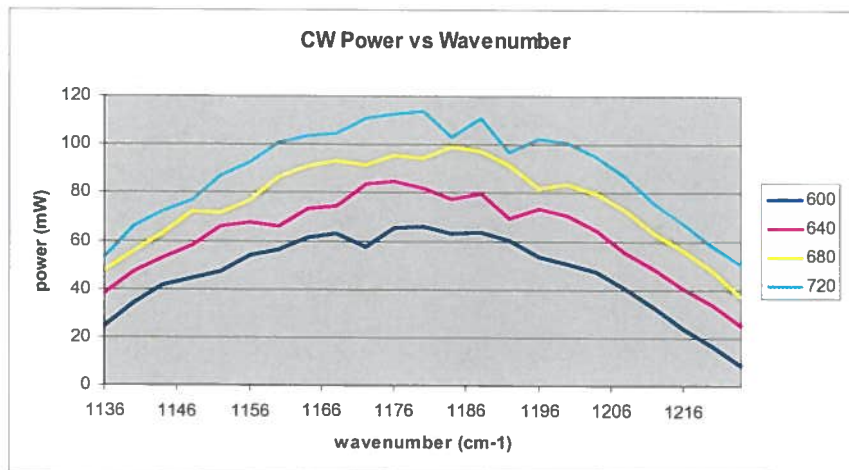
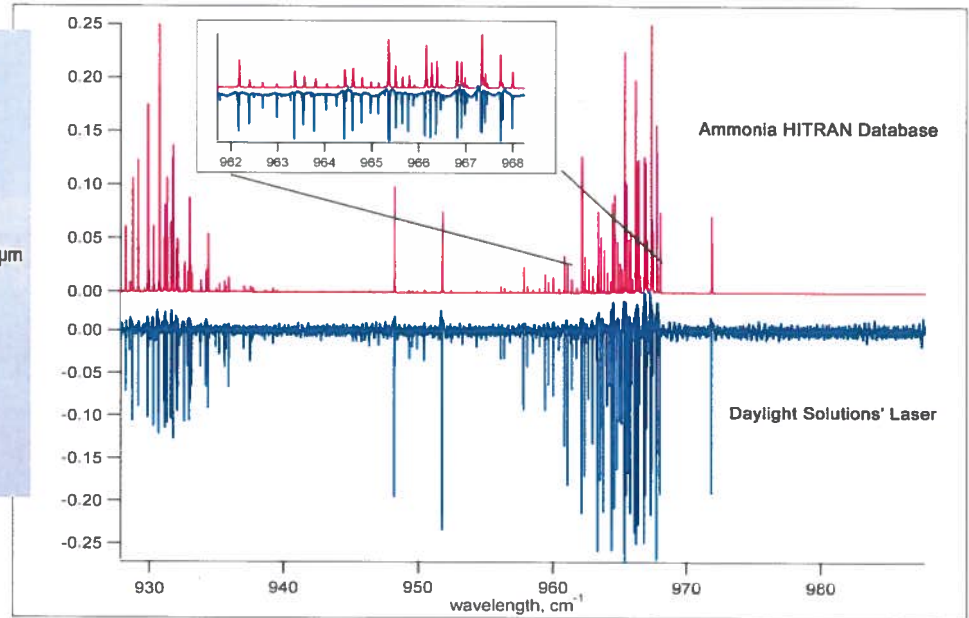


Commercial widely tunable cw EC-QCL

Mid-IR Lasers From Daylight Solutions



- CW, Mode-Hop Free
- Linewidth <math>< 0.001 \text{ cm}^{-1}</math>
- Center wavelengths from 4 to 12 μm
- Broad tuning range up to 10%
- No cryogenic cooling
- Average power up to 50 mW
- Tuning speed: full range <math>< 2\text{s}</math>
- Superb wavelength accuracy
- Shipping in September 2007

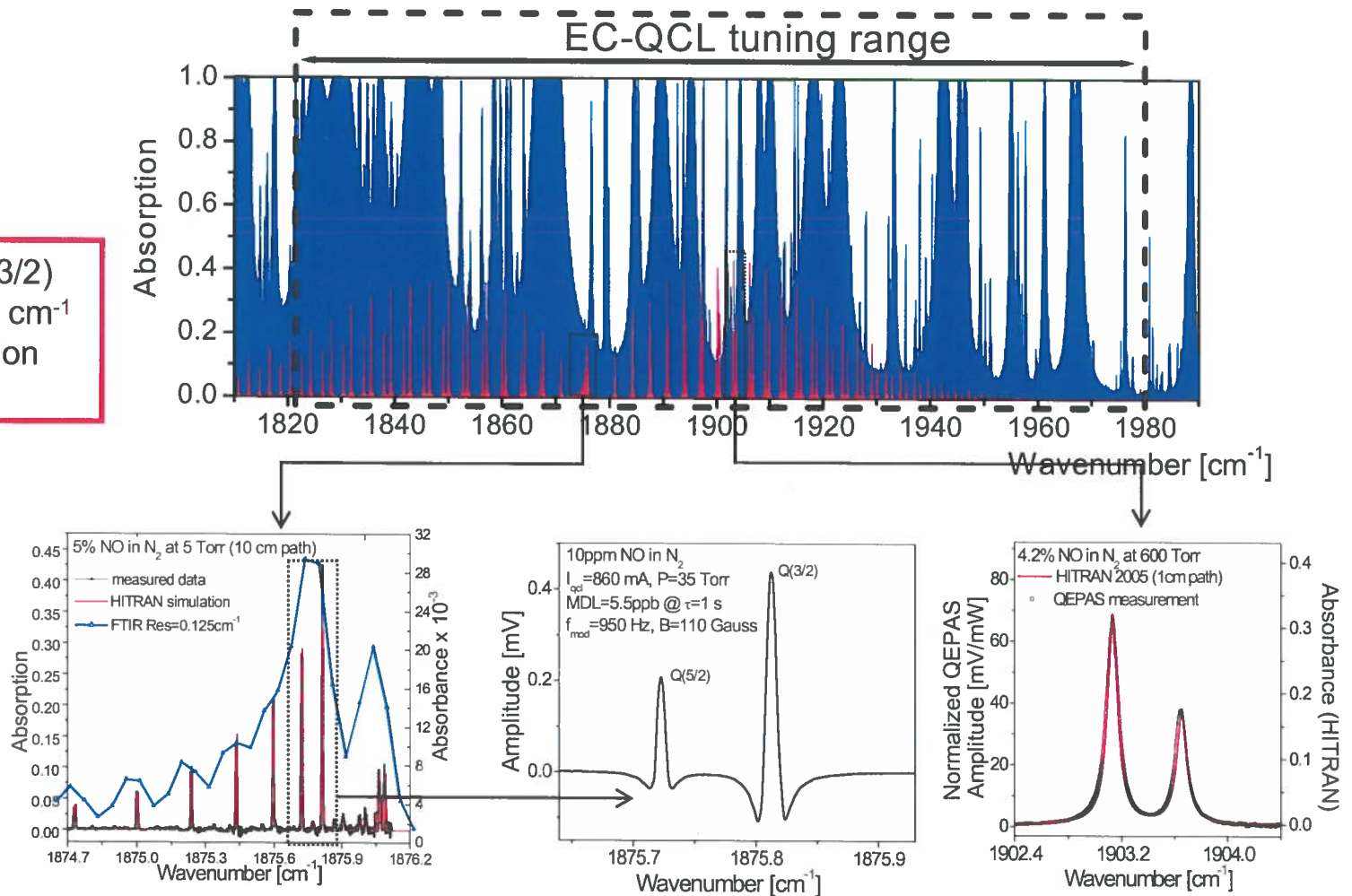


Motivation for Nitric Oxide Detection

- Atmospheric Chemistry
- Environmental pollutant gas monitoring
 - NO_x monitoring from automobile exhaust and power plant emissions
 - Precursor of smog and acid rain
- Industrial process control
 - Formation of oxynitride gates in CMOS Devices
- NO in medicine and biology
 - Important signaling molecule in physiological processes in humans and mammals (1998 Nobel Prize in Physiology/Medicine)
 - Treatment of asthma, COPD, acute lung rejection

High resolution spectroscopy with a 5.3 μm EC-QCL

Access to NO Q(3/2) transition at 1875.8 cm^{-1} for Faraday rotation spectroscopy

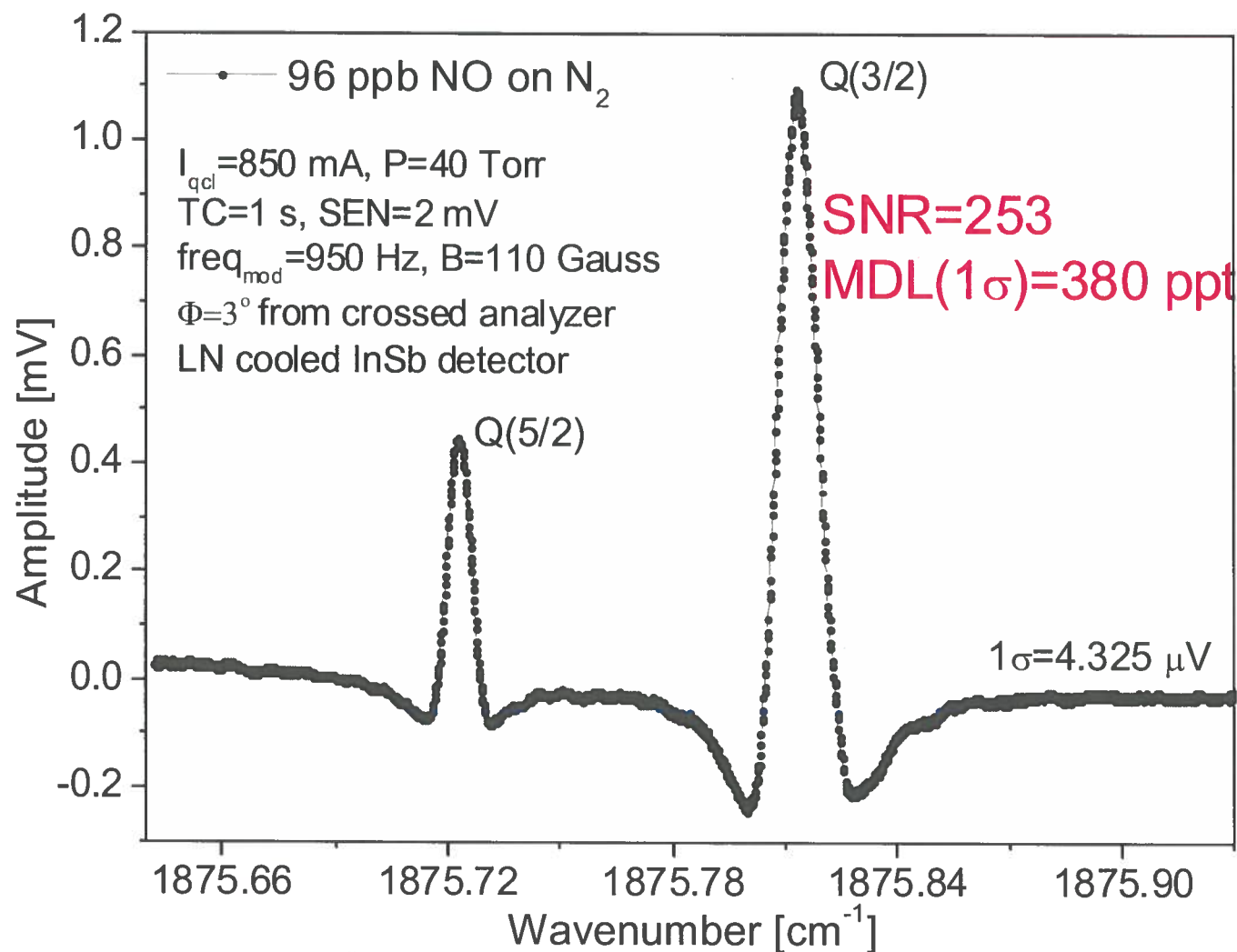


- Mode hop free scan of up to $\sim 2.5 \text{ cm}^{-1}$ with a resolution $< 0.001 \text{ cm}^{-1}$ (30MHz) can be performed anywhere within the tuning range

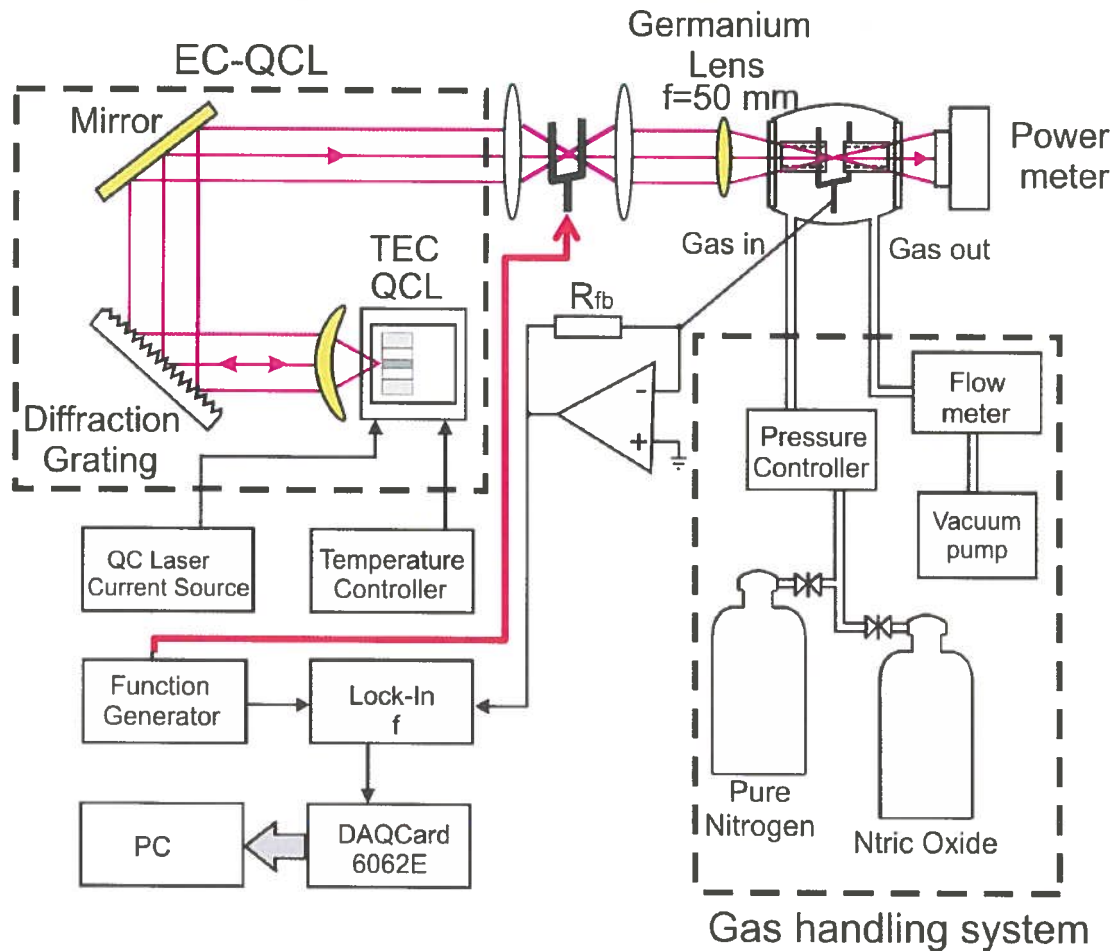
In collaboration with:



Magnetic Rotation Spectroscopy of Nitric Oxide



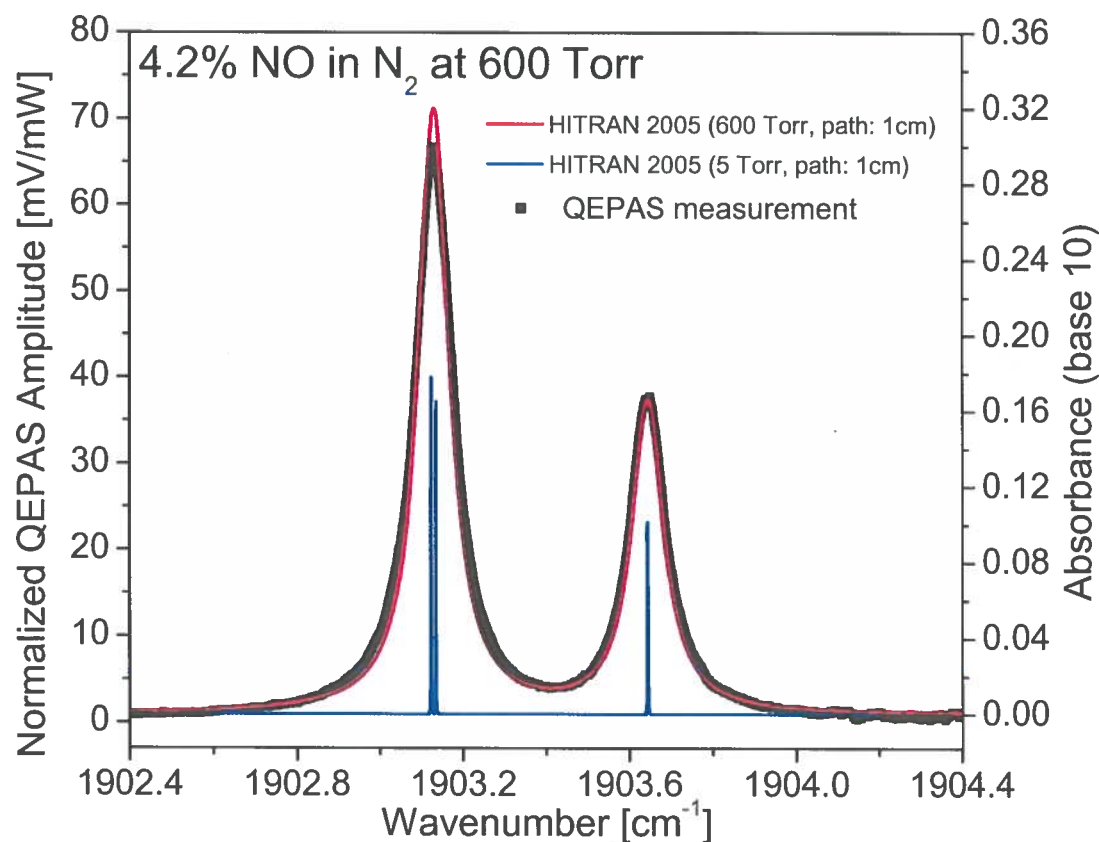
QCL based Quartz-Enhanced Photoacoustic Gas Sensor



QEPAS characteristics:

- High sensitivity (ppm to ppb)
- Excellent dynamic range
- Immune to environmental noise
- Ultra-small sample volume ($< 1\text{ mm}^3$)
- Sensitivity is limited by the fundamental thermal TF noise
- Compact, rugged and low cost
- Potential for trace gas sensor networks

High resolution EC-QCL based QEPAS

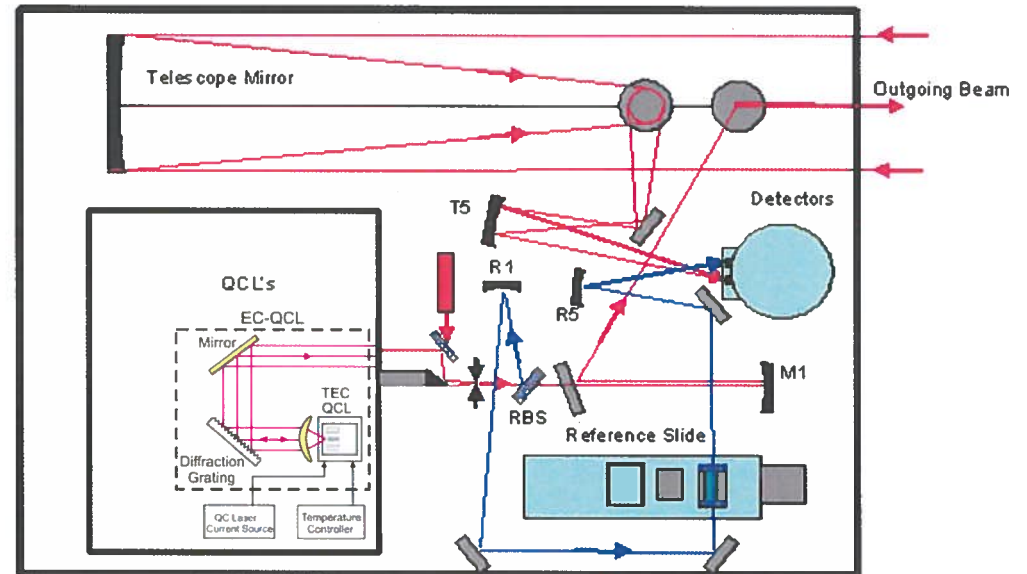
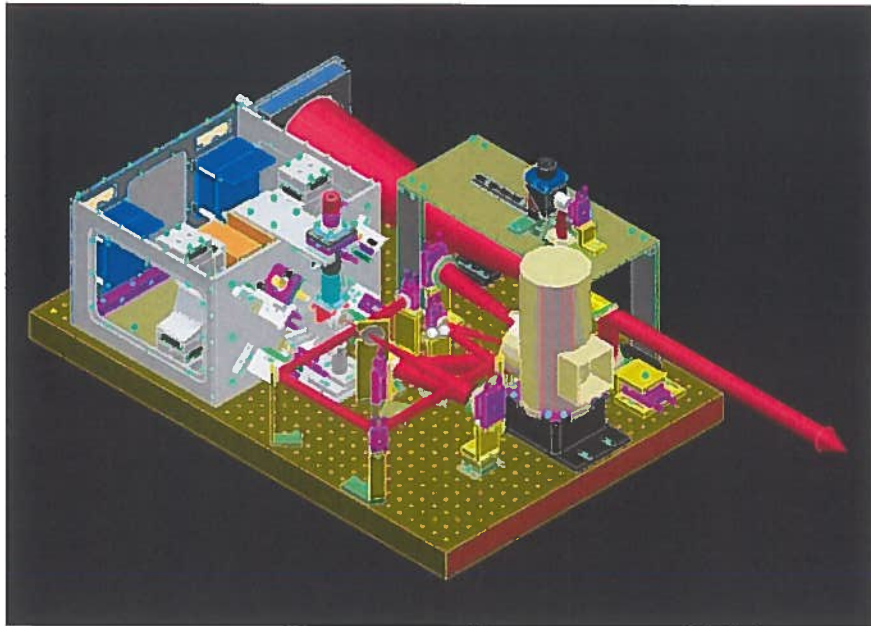


External Amplitude Modulation:

- QTF is used as a mechanical chopper at $f \sim 32\text{kHz}$
- No chirp associated with the laser current modulation
- High resolution mode-hop-free tuning is possible

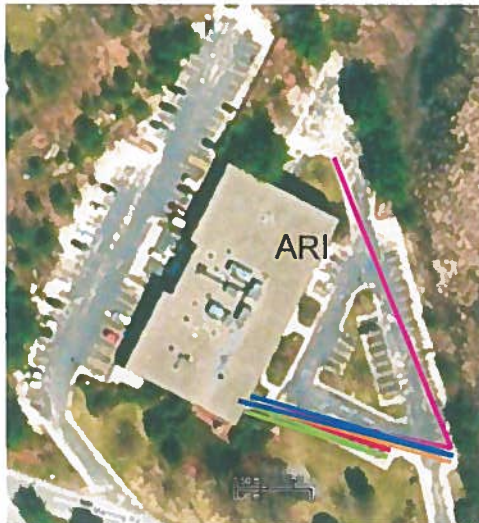
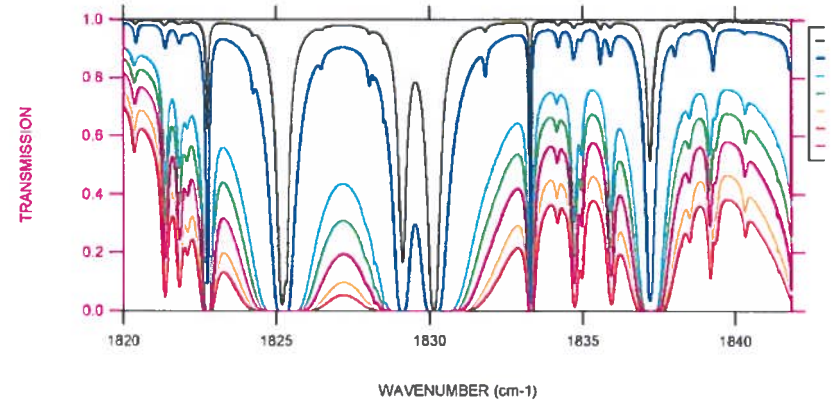


Design of an EC-QCL Based Remote Sensing System



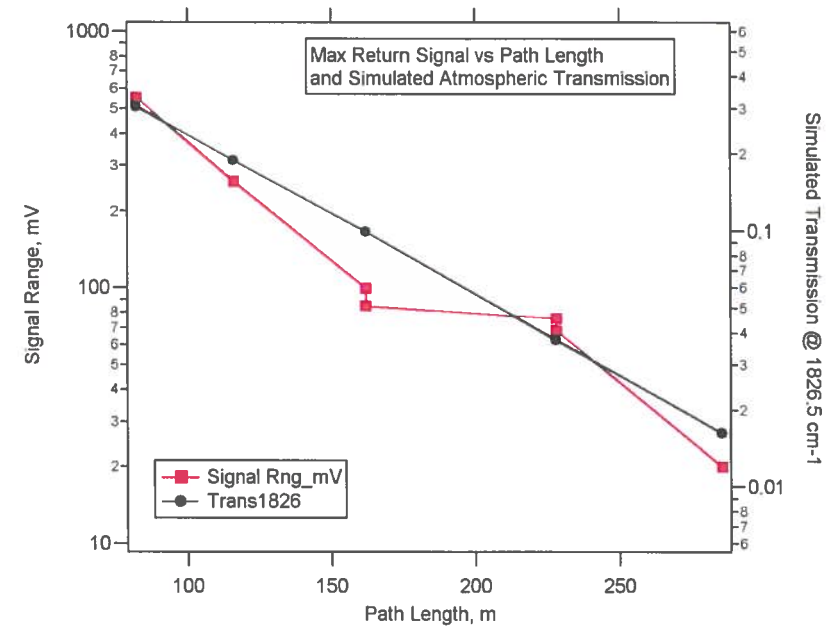
- An upgraded version of a four-laser pulsed QCL system
- The optical set-up, electronics and control software modified for CW-QCL operation
- First tests performed with a DFB CW-QCL operating at $\sim 5.5\mu\text{m}$

Outdoor Open Path Measurements (Influence of Atmospheric Transmission)



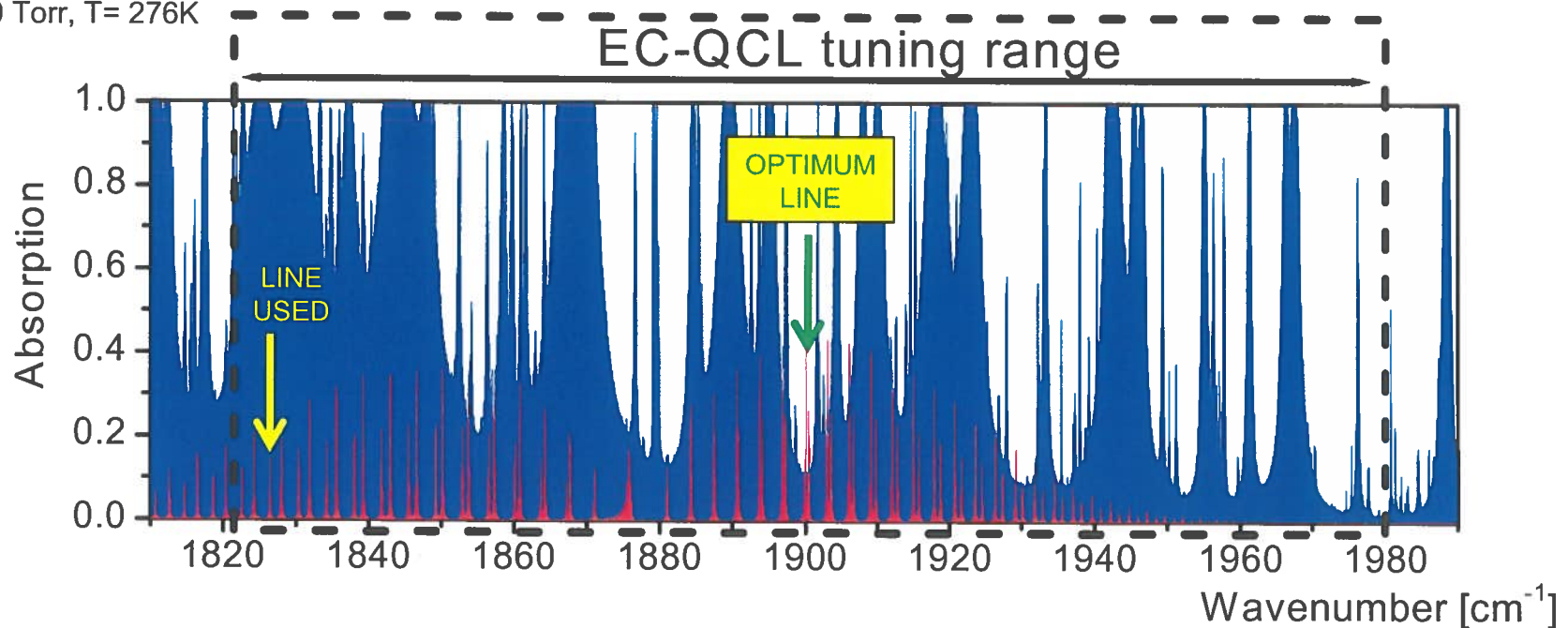
Open Path
Measurements
CW QCL
1826 cm^{-1}

Ranges
(1/2
total)
#1, 41m
#2, 58m
#3, 81m
#4, 114m
#5, 143m



High resolution spectroscopy with a 5.3 μm EC-QCL

286m open path
H₂O mixing ratio: 0.006
CO₂ mixing ratio: 380 ppm
P=760 Torr, T= 276K



EC-QCL allows selection of an absorption line with:

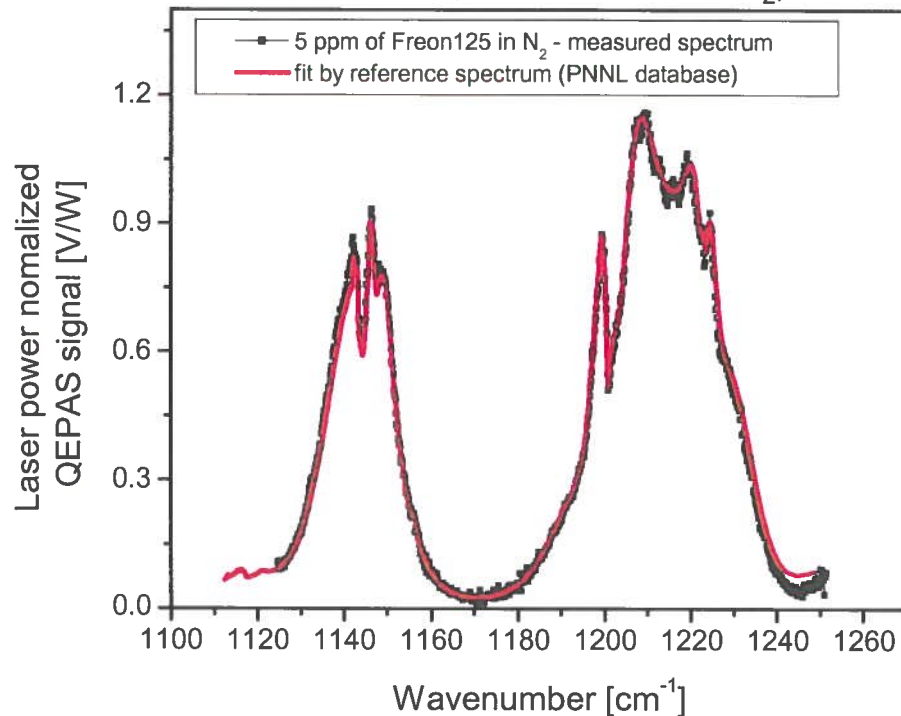
- Higher Line Intensity
- Lower Spectral Interference
- Higher Atmospheric Transmission

Monitoring of broadband absorbers

- Freon 125 (C_2HF_5)
 - Refrigerant (leak detection)
 - Safe simulant for toxic chemicals, e.g. chemical warfare agents
- Acetone (CH_3COCH_3)
 - Recognized biomarker for diabetes

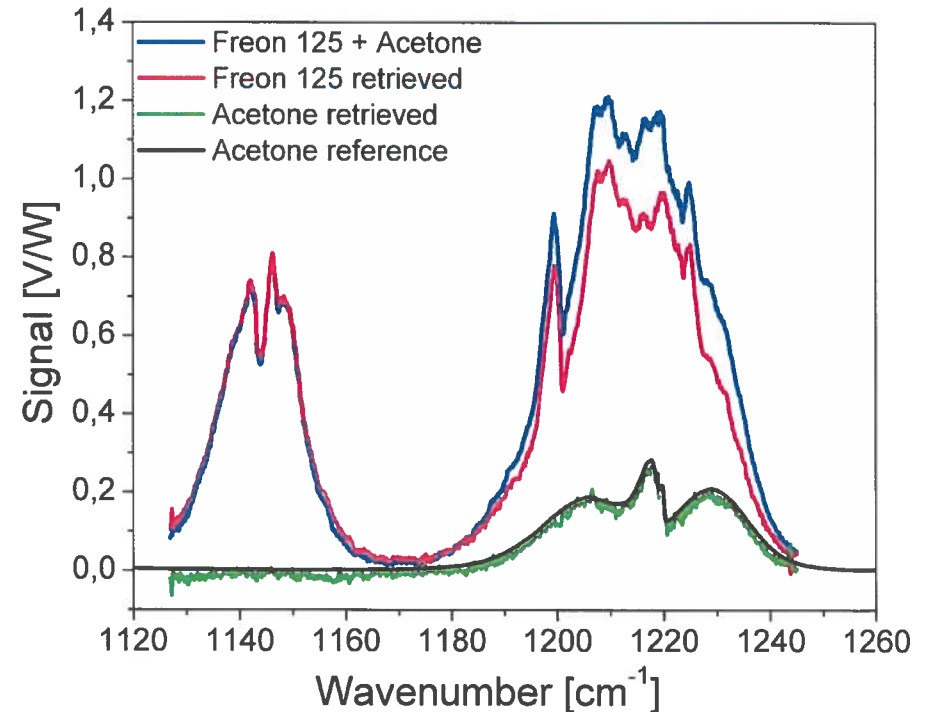
QEPAS based Freon 125 and Acetone concentration measurements with a tunable 8.4 μm CW EC-QCL

QEPAS concentration measurement of Freon 125 (5ppm mixture in N_2)



- Minimum detection limit (1σ) of **~4.5 ppb** was obtained for Freon 125 with an average laser power of 6.6 mW

QEPAS concentration measurement of a Freon 125 and acetone mixture



- Wide tunability enables excellent molecular selectivity for broad band absorbers



QEPAS Performance for 12 Trace Gas Species (April '08)

Molecule (Host)	Frequency, cm^{-1}	Pressure, Torr	NNEA, $\text{cm}^{-1}\text{W}/\text{Hz}^{1/2}$	Power, mW	NEC ($\tau=1\text{s}$), ppmv
H_2O (N_2)**	7306.75	60	1.9×10^{-9}	9.5	0.09
HCN (air: 50% RH)*	6539.11	60	$< 4.3 \times 10^{-9}$	50	0.16
C_2H_2 (N_2)*	6523.88	720	4.1×10^{-9}	57	0.03
NH_3 (N_2)*	6528.76	575	3.1×10^{-9}	60	0.06
C_2H_4 (N_2)*	6177.07	715	5.4×10^{-9}	15	1.7
CH_4 (N_2)*	6057.09	950	2.9×10^{-8}	13.7	2.1
CO_2 (breath ~100% RH)	6361.25	150	8.2×10^{-9}	45	40
H_2S (N_2)*	6357.63	780	5.6×10^{-9}	45	0.20
CO_2 ($\text{N}_2+1.5\% \text{H}_2\text{O}$) *	4991.26	50	1.4×10^{-8}	4.4	18
CH_2O ($\text{N}_2:75\% \text{RH}$)*	2804.90	75	8.7×10^{-9}	7.2	0.12
CO (N_2)	2196.66	50	5.3×10^{-7}	13	0.5
CO (propylene)	2196.66	50	7.4×10^{-8}	6.5	0.14
N_2O (air+5% SF_6)	2195.63	50	1.5×10^{-8}	19	0.007
$\text{C}_2\text{H}_5\text{OH}$ (N_2)**	1934.2	770	2.2×10^{-7}	10	90
C_2HF_5 (N_2 ***)	1208.62	770	7.8×10^{-9}	6.6	0.009
NH_3 (N_2)*	1046.39	110	1.6×10^{-8}	20	0.006

* - Improved microresonator

** - Improved microresonator and double optical pass through ADM

*** - With amplitude modulation and metal microresonator

NNEA – normalized noise equivalent absorption coefficient.

NEC – noise equivalent concentration for available laser power and $\tau=1\text{s}$ time constant, 18 dB/oct filter slope.

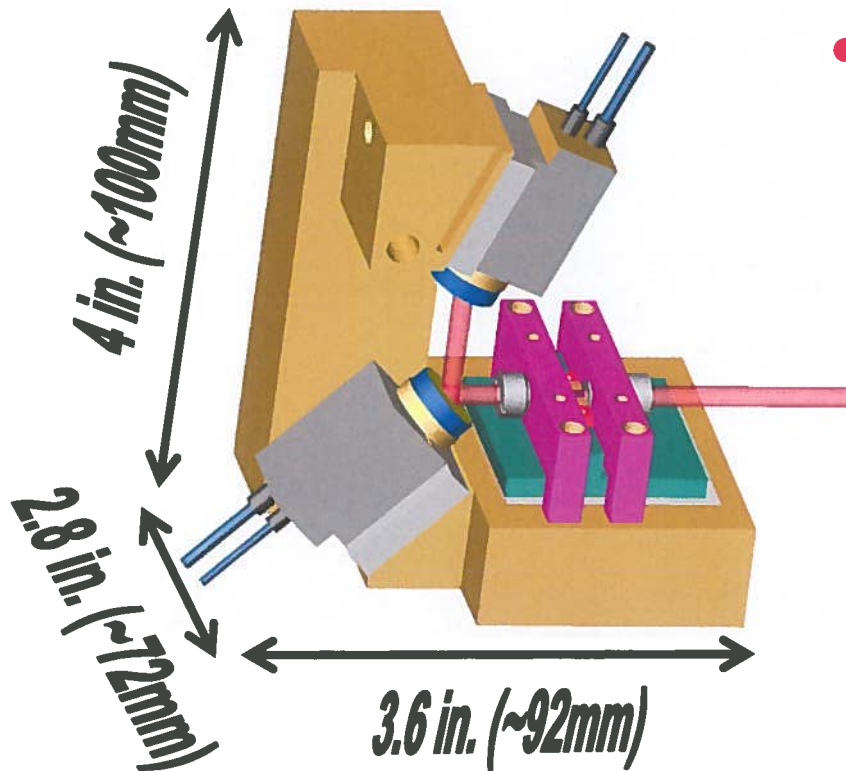
For comparison: conventional PAS 2.2 $(2.6) \times 10^{-9} \text{ cm}^{-1}\text{W}/\sqrt{\text{Hz}}$ (1,800; 10,300 Hz) for NH_3 *, ()**

* M. E. Webber et al, Appl. Opt. 42, 2119-2126 (2003); ** J. S. Pilgrim et al, SAE Intl. ICES 2007-01-3152



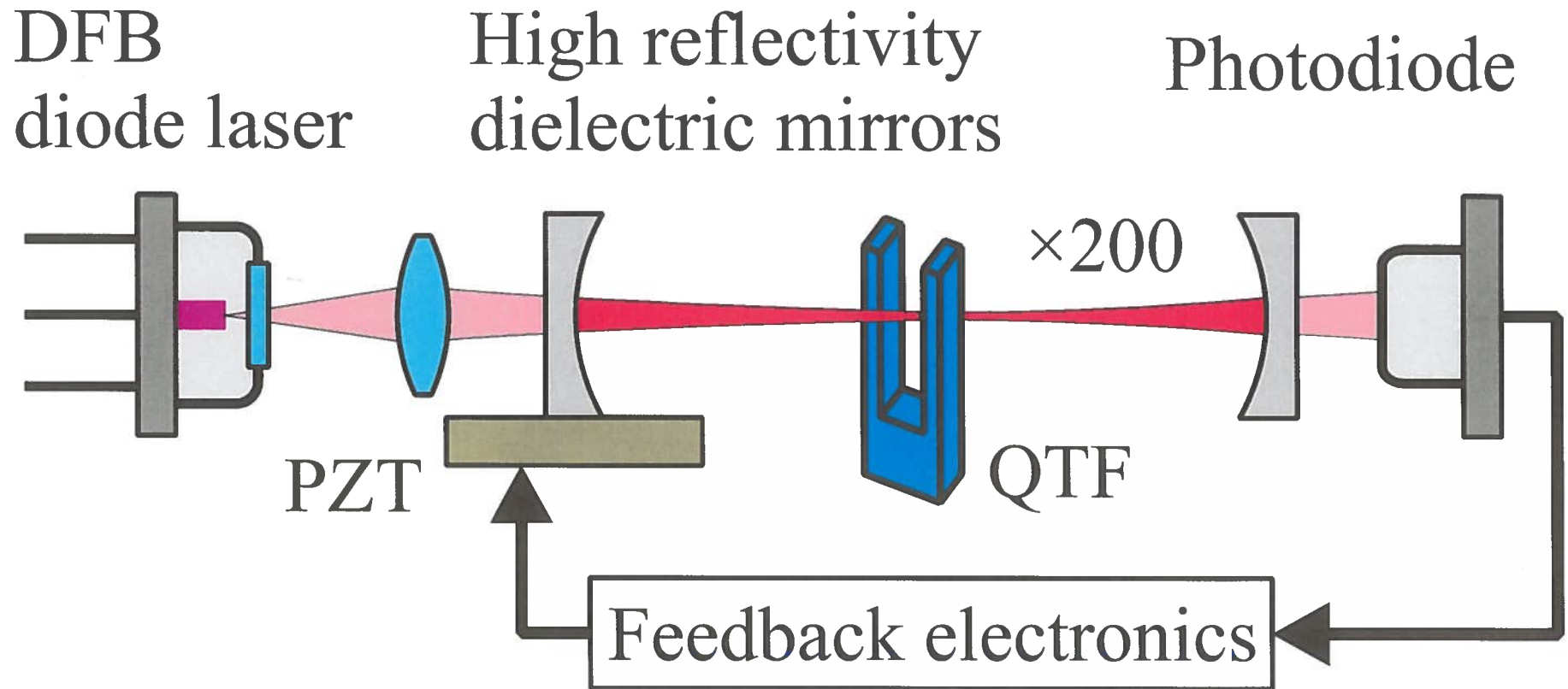
Future of Chemical Trace Gas Sensing

New design of fast broadly tunable EC-QCLs (2008)

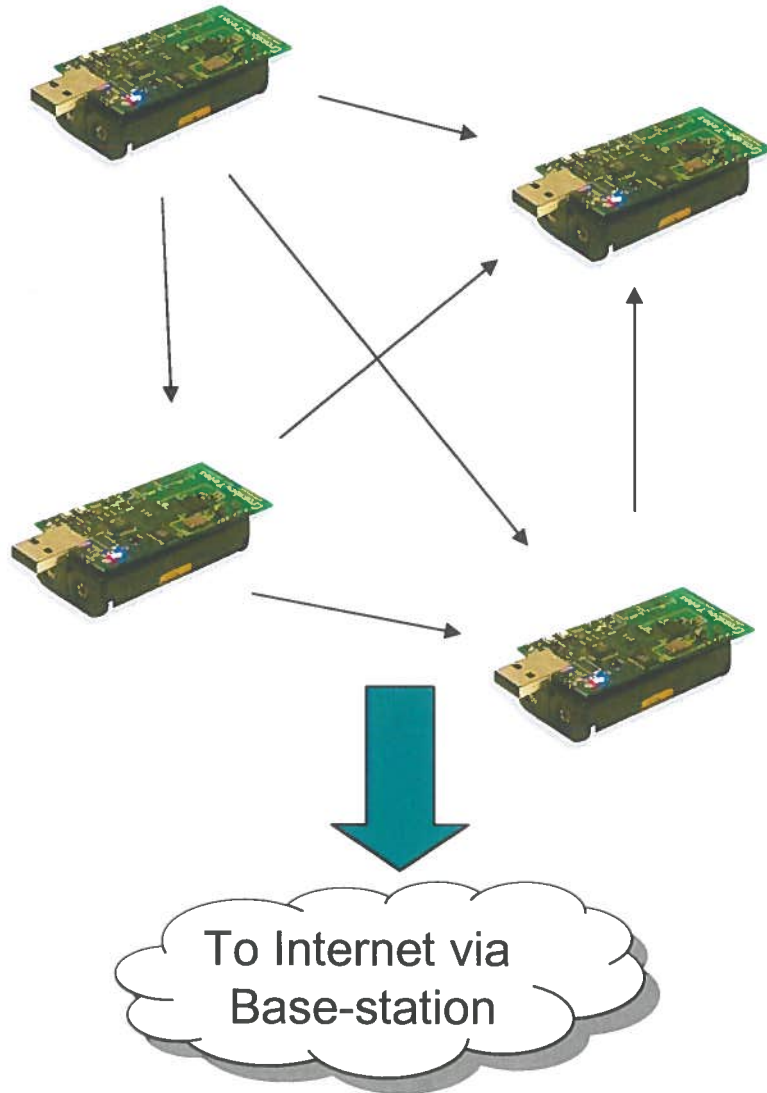


- New optical configuration
Folded cavity (configuration #1)
- Fast tuning capabilities:
 - Coarse Broadband Scanning
($\sim 55 \text{ cm}^{-1}$ @ $5 \mu\text{m}$) **up to 5 KHz**
(compared to available technologies $< 10 \text{ Hz}$)
 - High resolution mode-hop free tuning ($\sim 3.2 \text{ cm}^{-1}$ @ $5 \mu\text{m}$)
up to 5 KHz
(compared to available technology 100-200 Hz)

Proposed QEPAS-OPBC Sensor Configuration

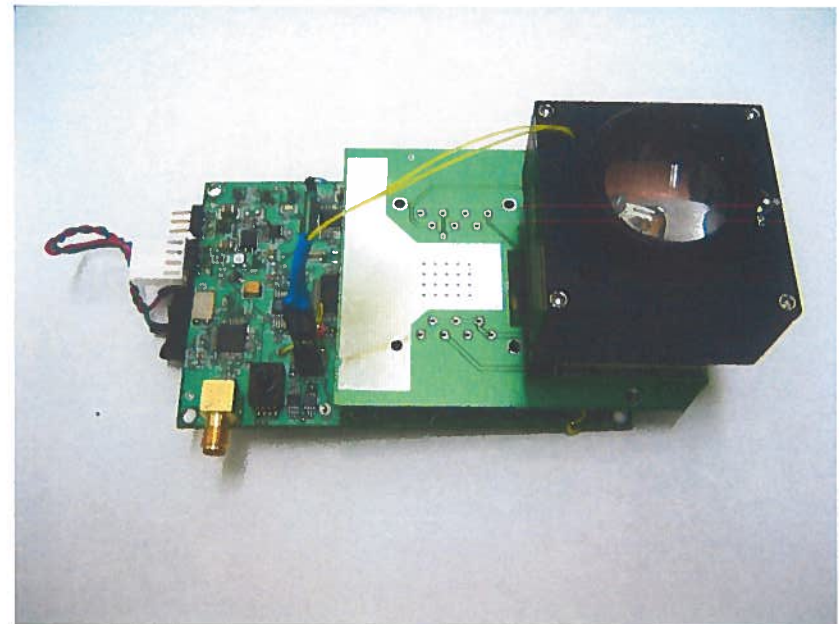
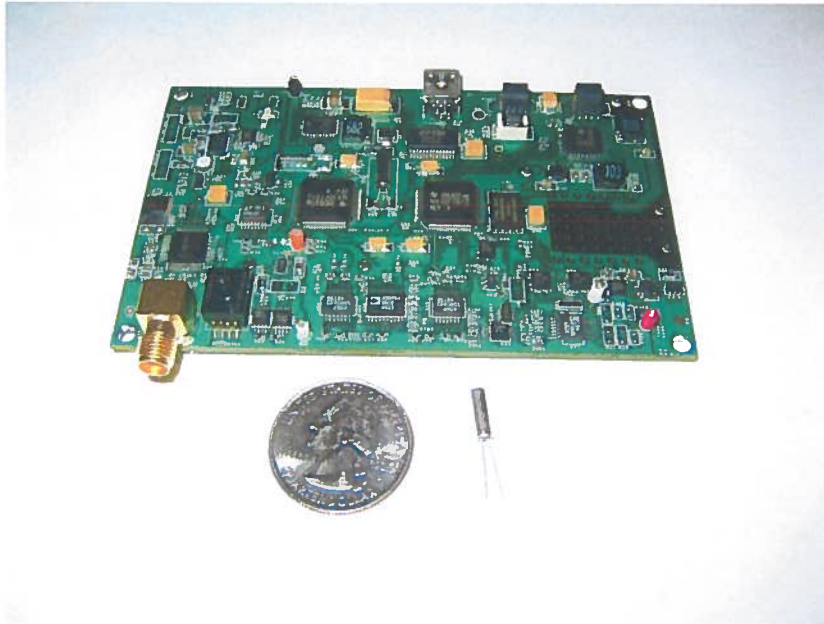


Wireless Sensor Networks for Gas Sensing



- Each point called “mote”
- Advantages?
 - Spatial resolution
 - Measure fluxes
- What is needed?
 - Low power
 - Low cost
 - Ultra miniature
 - Replicable
 - Autonomy

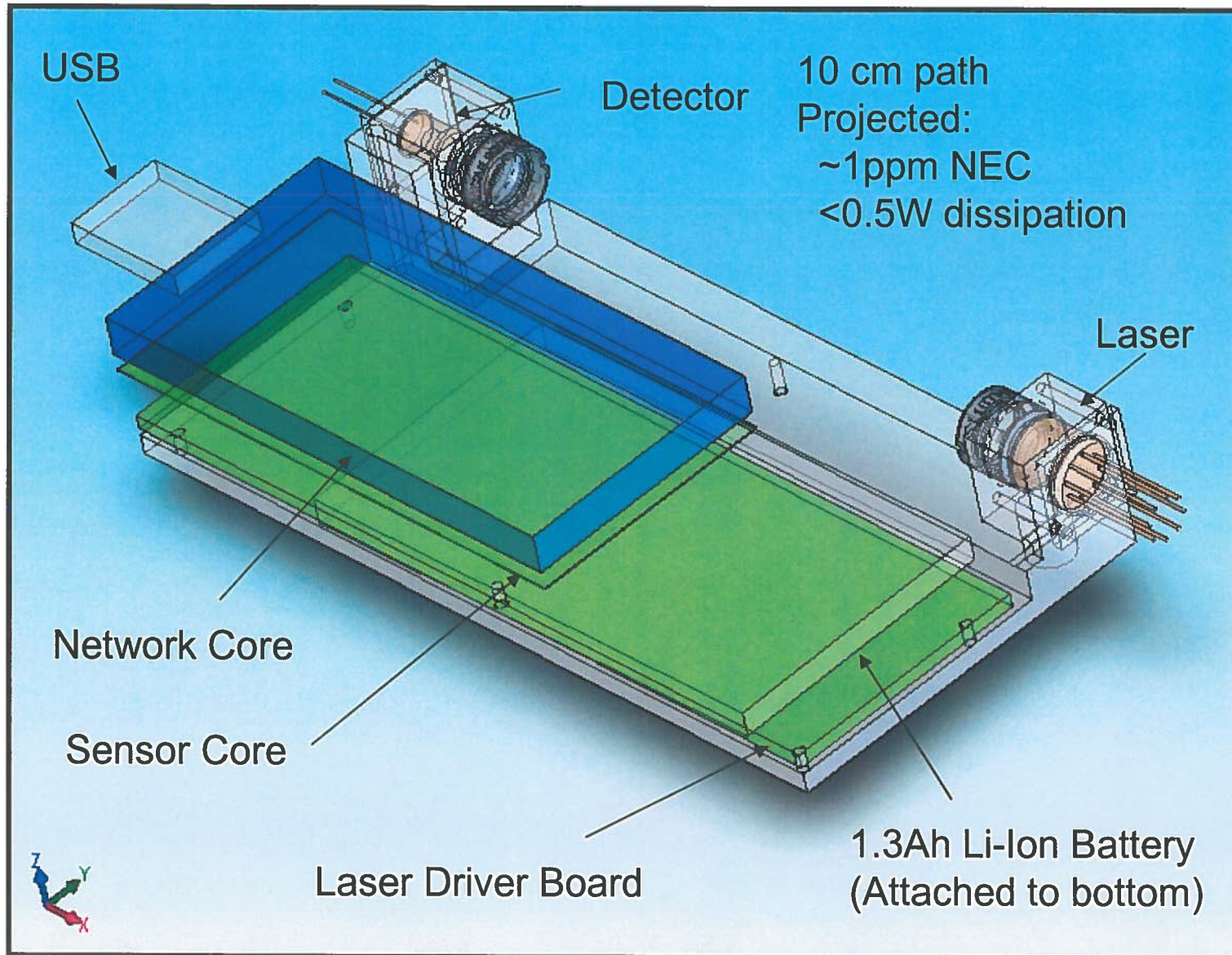
Miniature QEPAS CO₂ sensor ($\lambda=2\mu\text{m}$) v2.0 boards



- Small size
- Relatively low cost
- High efficiency switching power supplies
- PWM Peltier cooler driver
- 0.2W control system power consumption
- Detection sensitivity* of CO₂ 110 ppm with 1sec. lock-in TC
- Over 10³ improvement in sensitivity is possible @4.2 μm

*G. Wysocki, A. A. Kosterev, and F. K. Tittel "Influence of Molecular Relaxation Dynamics on Quartz-Enhanced Photoacoustic Detection of CO₂ at $\lambda = 2 \mu\text{m}$ ", Applied Physics B 85, 301-306 (2006)

Miniature LAS CO₂ sensor ($\lambda=2.7\mu\text{m}$) boards



Summary & Future Directions of QCL based Gas Sensor Technology

- **Quantum and Interband Cascade Laser based Trace Gas Sensors**
 - Compact, tunable, and robust
 - High sensitivity ($<10^{-4}$) and selectivity (3 to 500 MHz)
 - Capable of fast data acquisition and analysis
 - Detected 13 trace gases to date: NH_3 , CH_4 , N_2O , CO_2 , CO , NO , H_2O , COS , C_2H_4 , H_2CO , SO_2 , $\text{C}_2\text{H}_5\text{OH}$, C_2HF_5 and several isotopic species of C, O, N and H.
- **New Applications of Trace Gas Detection**
 - Environmental Monitoring (urban quality - H_2CO and, isotopic ratio measurements of CO_2 and CH_4 , fire detection and quantification of engine exhausts)
 - Industrial process control and chemical analysis (NO , NH_3 , H_2O , and H_2S)
 - Medical & biomedical diagnostics (NO , NH_3 , N_2O , H_2CO and CH_3COCH_3)
 - Hand-held sensors and sensor network technologies (CO_2)
- **Future Directions and Collaborations**
 - Improvements of the existing sensing technologies using novel, thermoelectrically cooled, cw, high power, and broadly wavelength tunable mid-IR interband and intersubband quantum cascade lasers
 - New applications enabled by novel broadly wavelength tunable quantum cascade lasers based on heterogeneous EC-QCL (i.e sensitive concentration measurements of broadband absorbers, in particular VOCs, HCs and multi-species detection)
 - Development of optically gas sensor networks based on QEPAS and LAS

