



Trace gas detection in ambient air using a 7.9 μm QC-DFB laser

LACEA
2000
Santa Fe,
NM

Anatoliy A. Kosterev, Robert F. Curl, and Frank K. Tittel

Rice Quantum Institute, Rice University, Houston, TX 77251-1892

C. Gmachl, F. Capasso, D.L. Sivco, J.N. Baillargeon, A.L. Hutchinson,
and A.Y. Cho

*Bell Laboratories, Lucent Technologies, 700 Mountain Avenue, Murray
Hill, NJ 07974*

OUTLINE

- ❖ Background and Motivation
- ❖ Detection of simple molecules in air, cw laser operation
- ❖ Detection of ethanol and linear regression data analysis
- ❖ Pulsed operation based laser spectrometer
- ❖ Summary and future outlook

Spectral Coverage by Diode/QC Lasers

N_2O



H_2S



NO_2



NO



C_6H_6



CH_3OH



NH_3



H_2CO



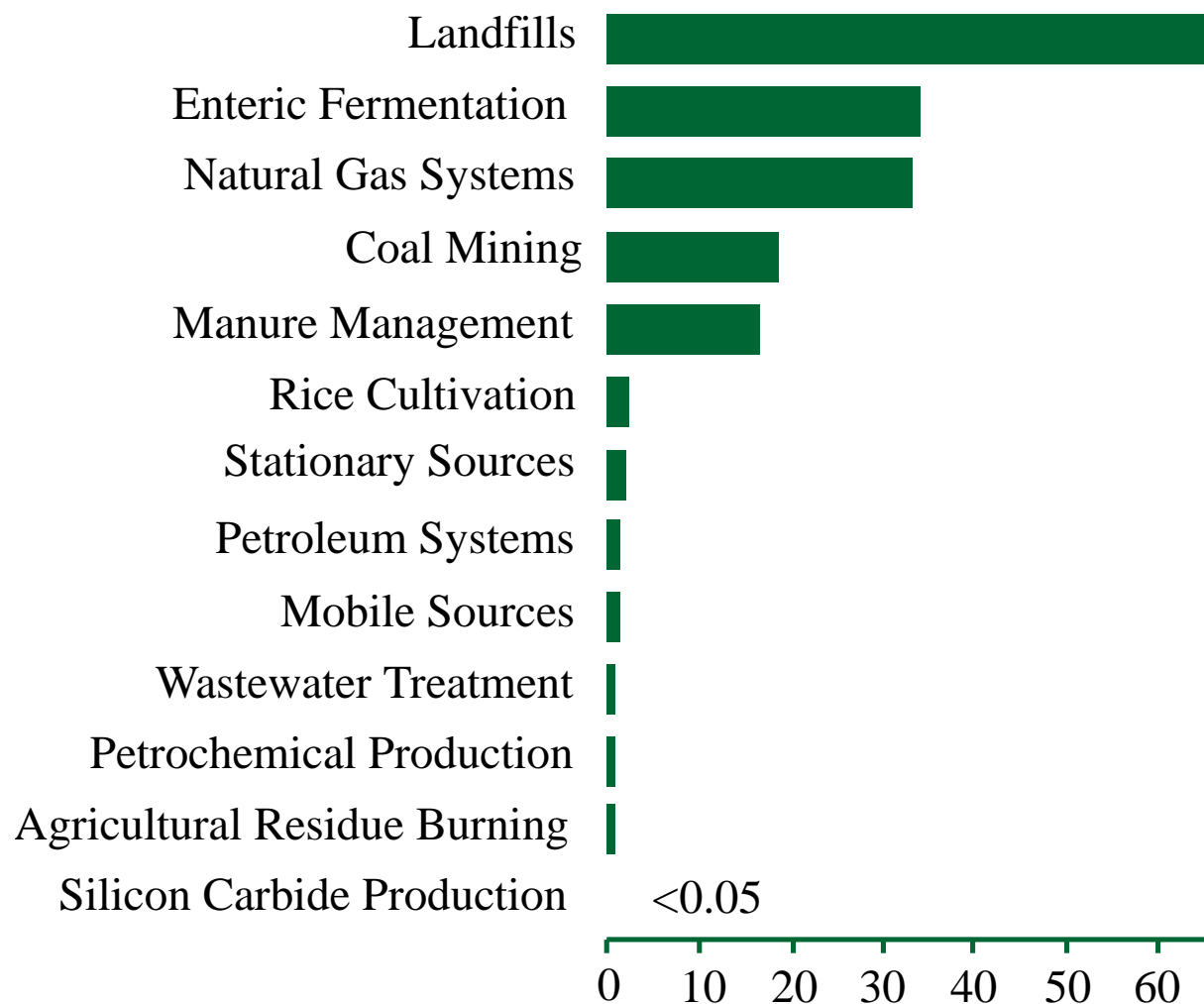
CH_4



Motivation for CH₄ and N₂O Detection

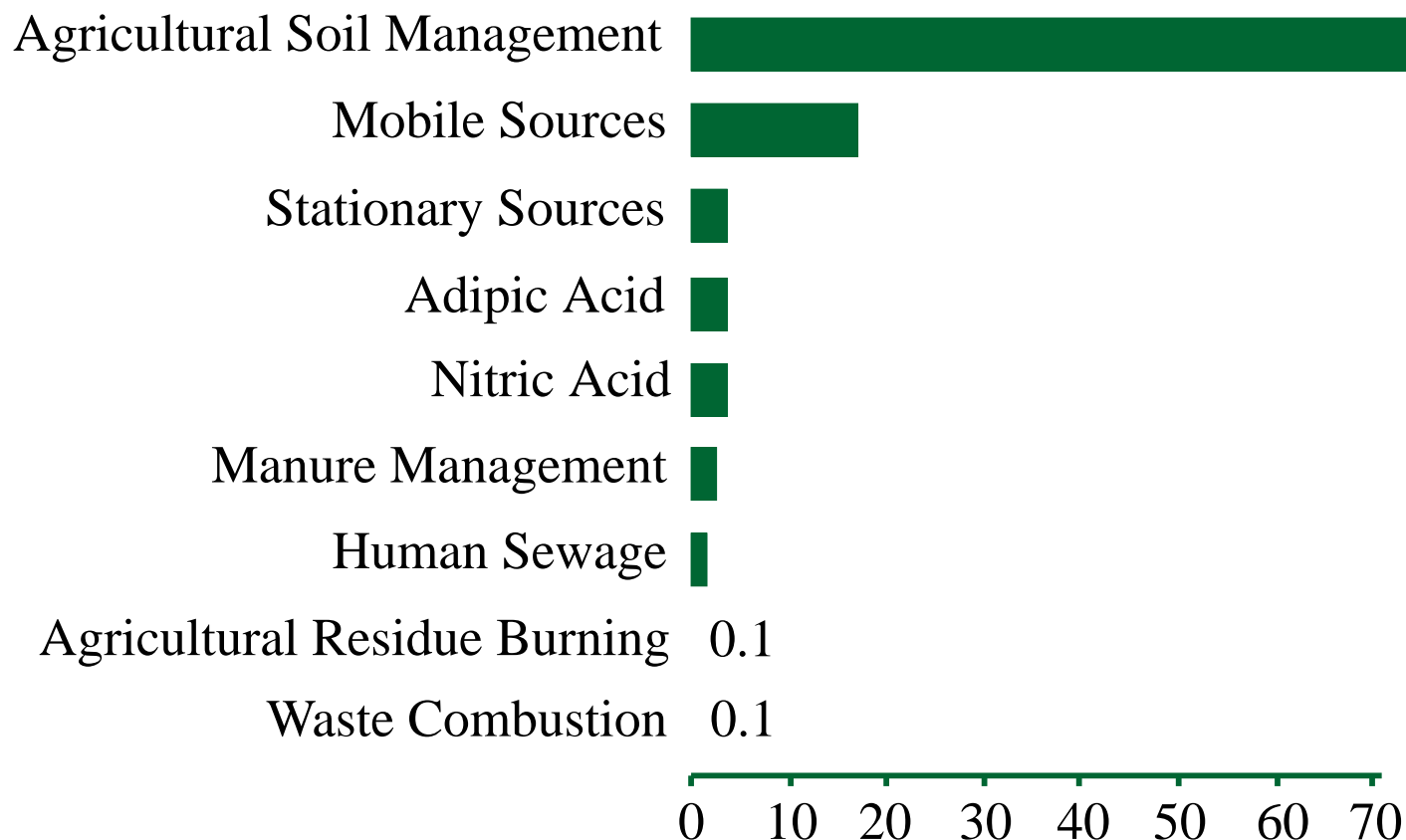
- ❖ Contribution to global warming
- ❖ Important in tropospheric and stratospheric chemistry
- ❖ Emitted by agricultural sources
- ❖ CH₄ leaks from gas pipelines

Sources of atmospheric CH₄

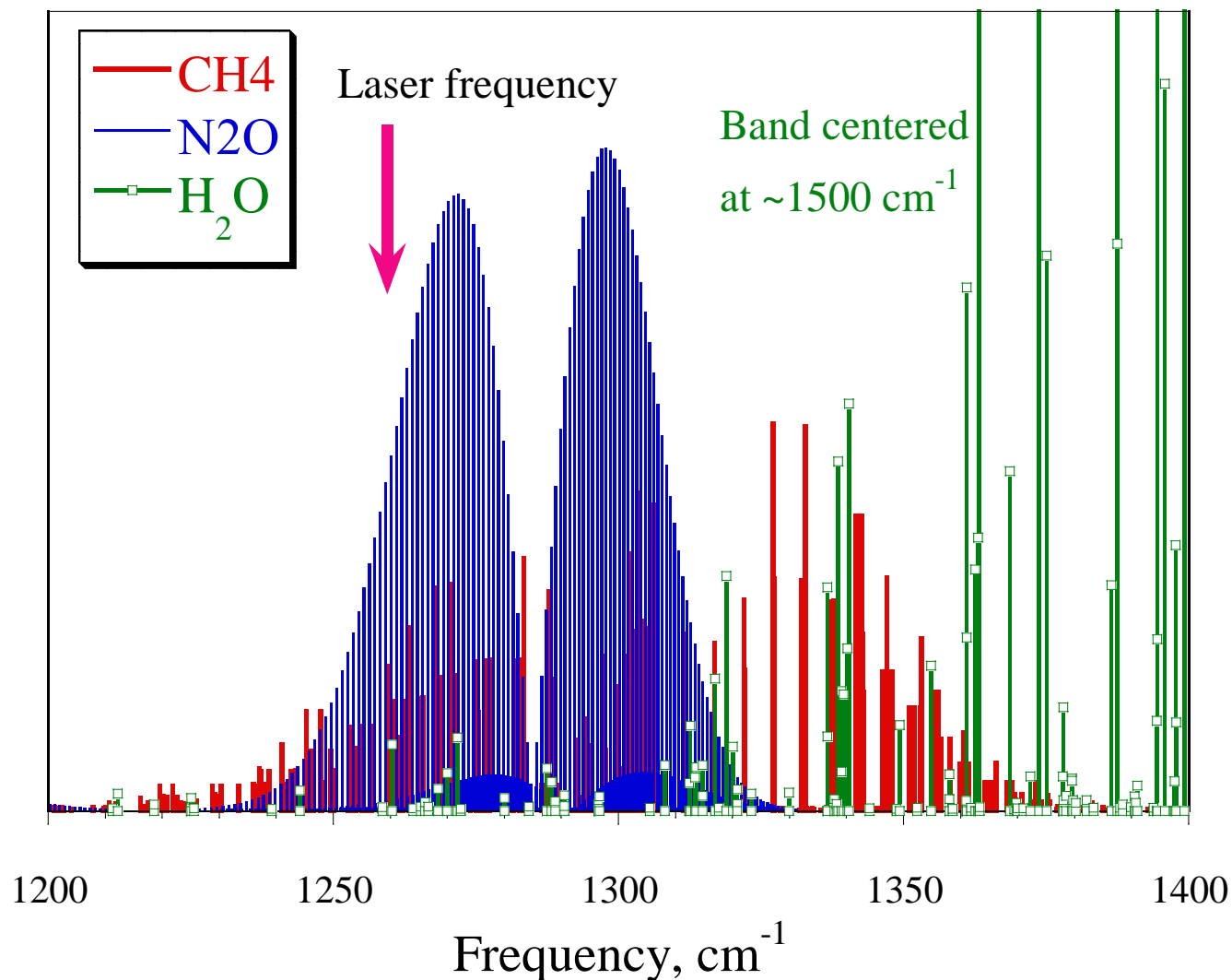


<http://www.epa.gov>

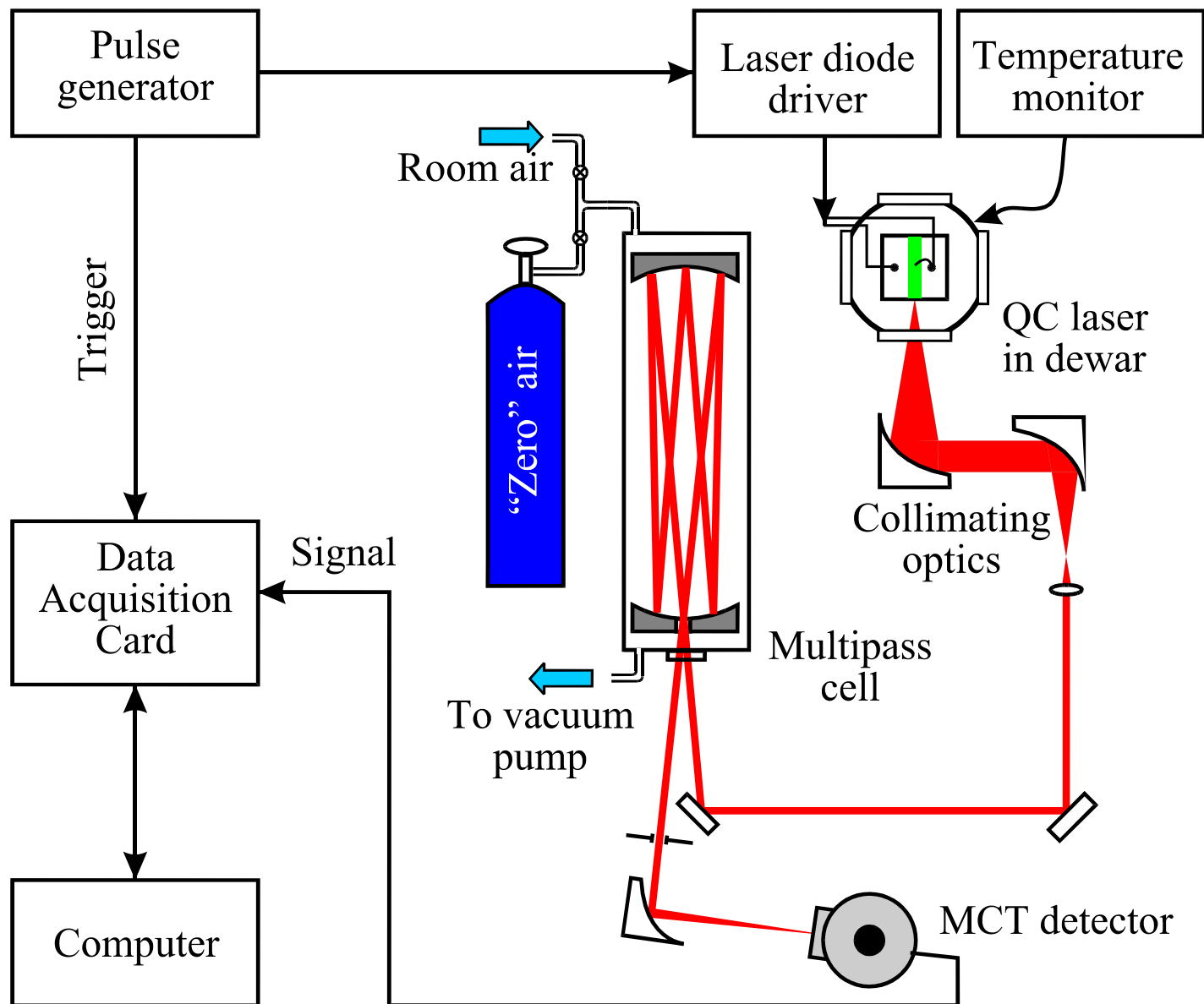
Sources of atmospheric N₂O



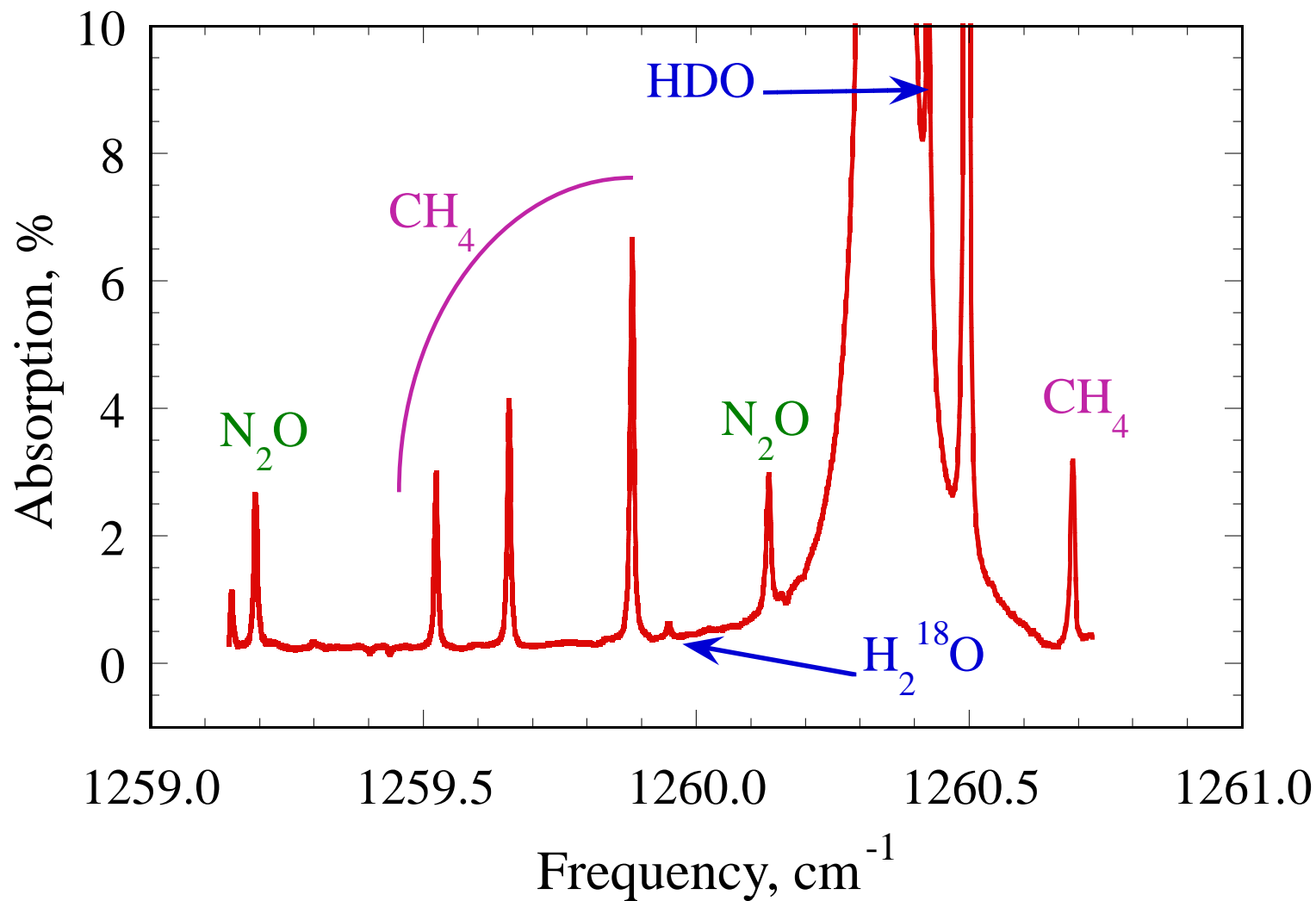
CH₄, H₂O and N₂O Absorption Spectra



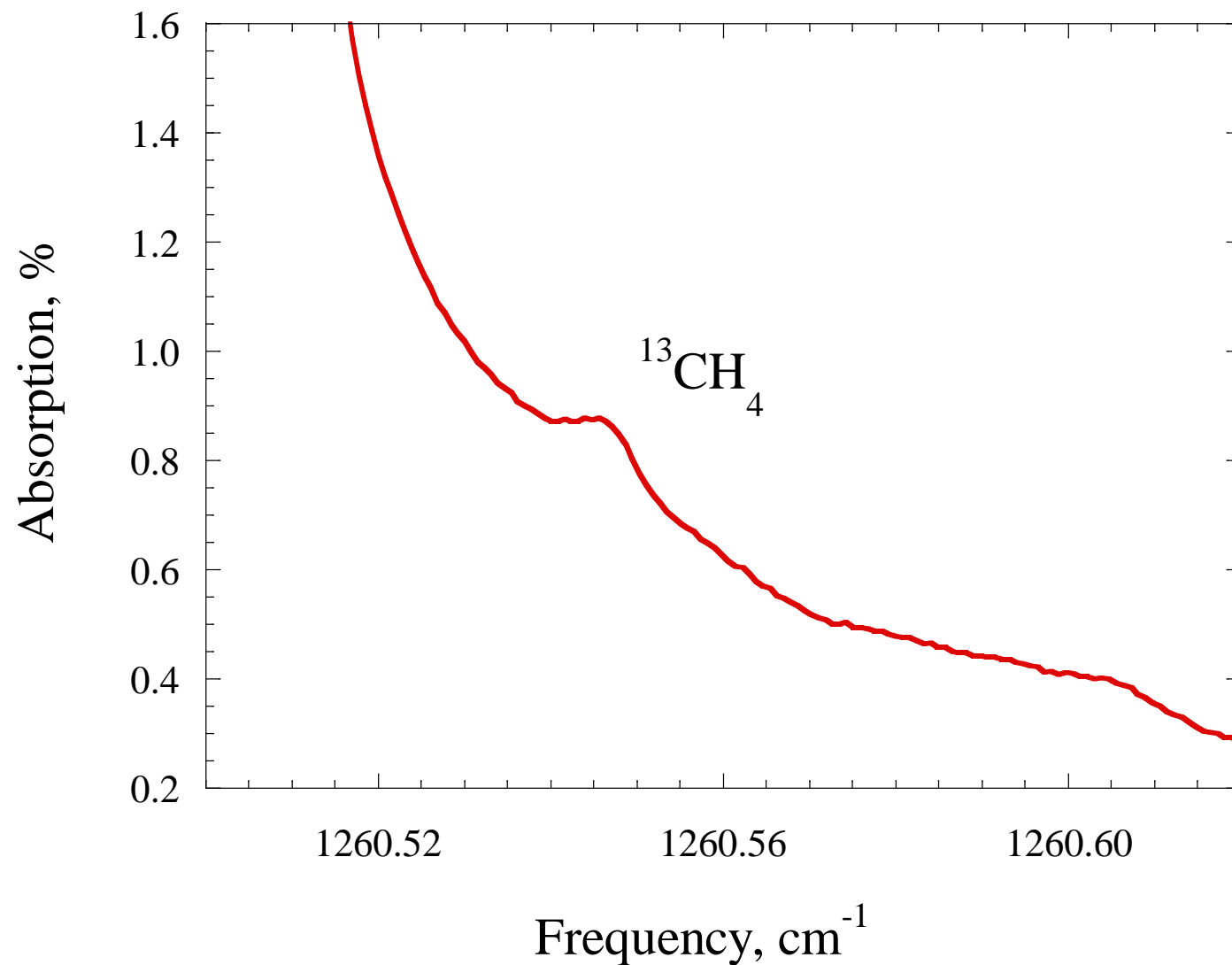
Trace Gas Detection with a Multipass Cell



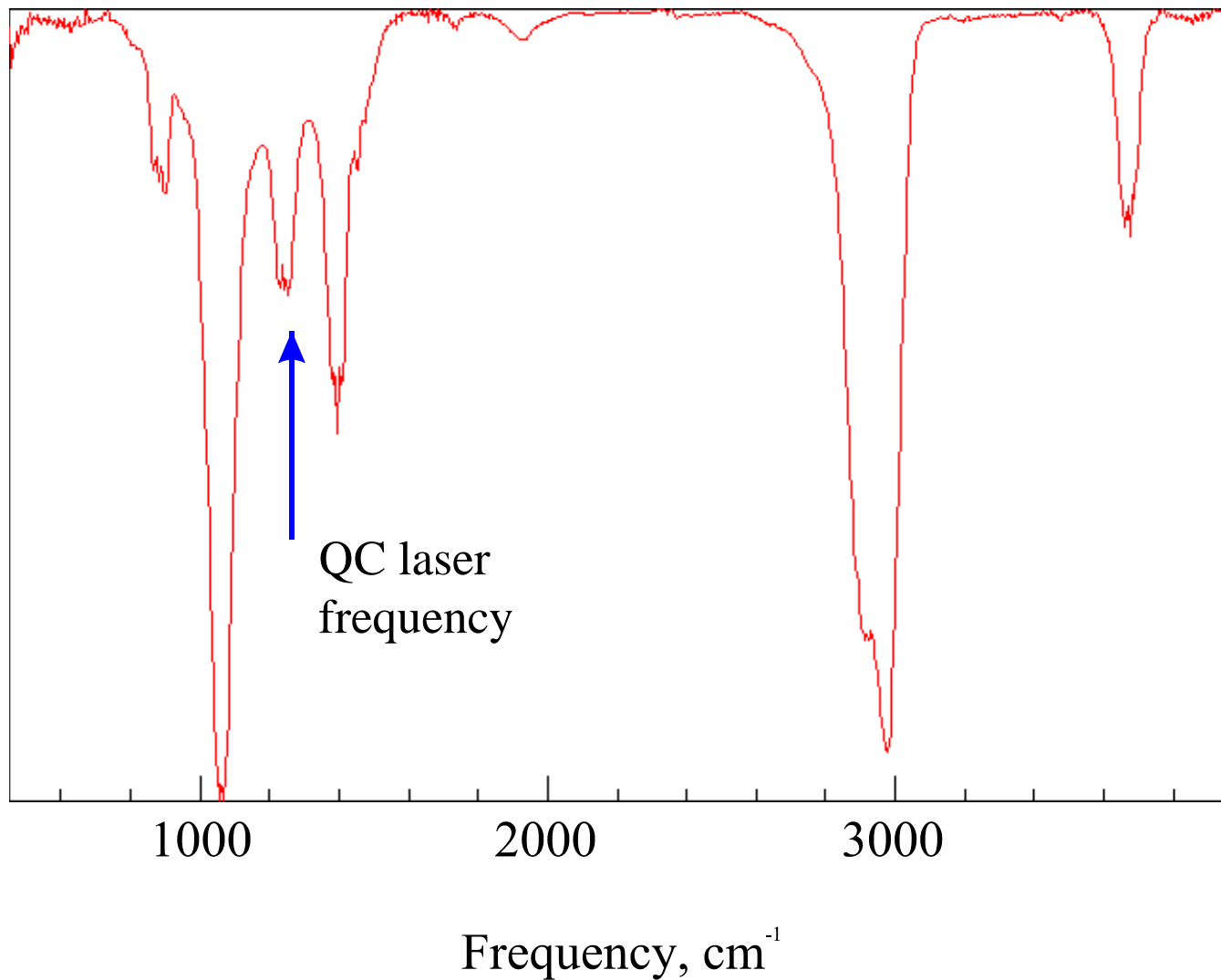
Absorption Spectrum of Room Air



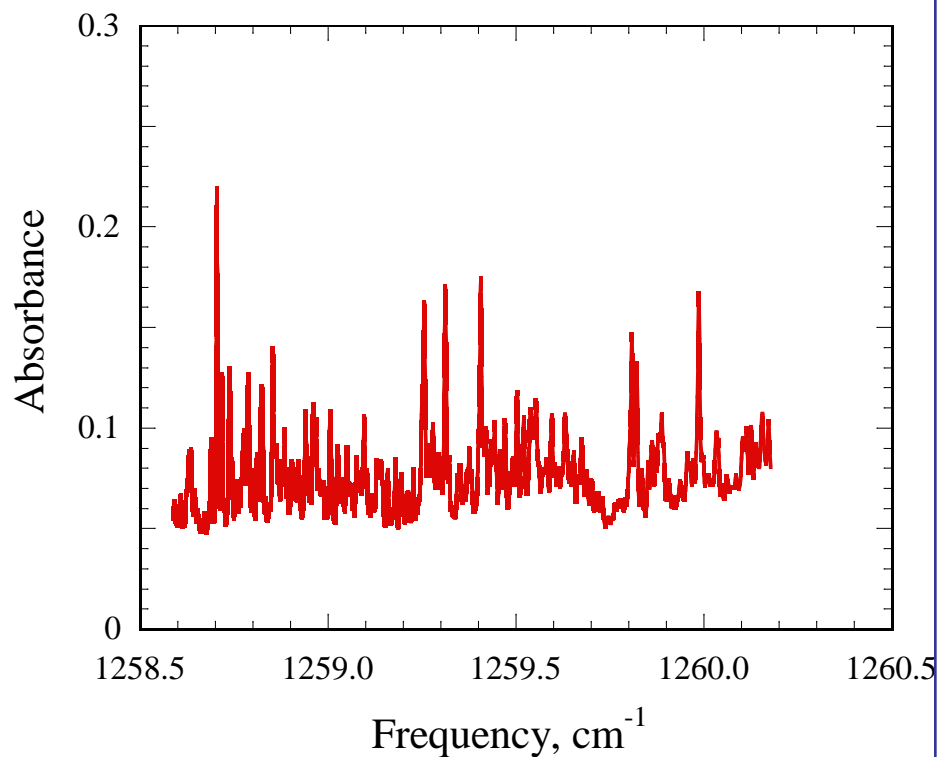
Detection of $^{13}\text{CH}_4$ in Ambient Air



IR Absorption Spectrum of Ethanol

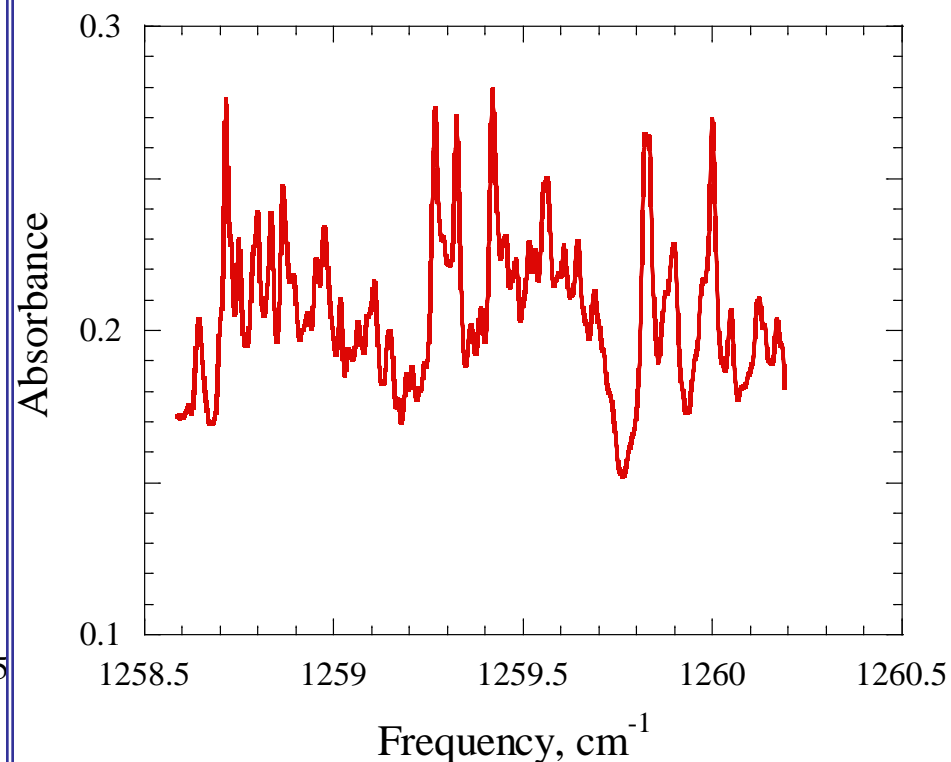


High-resolution IR Ethanol Spectrum



Pure ethanol vapor

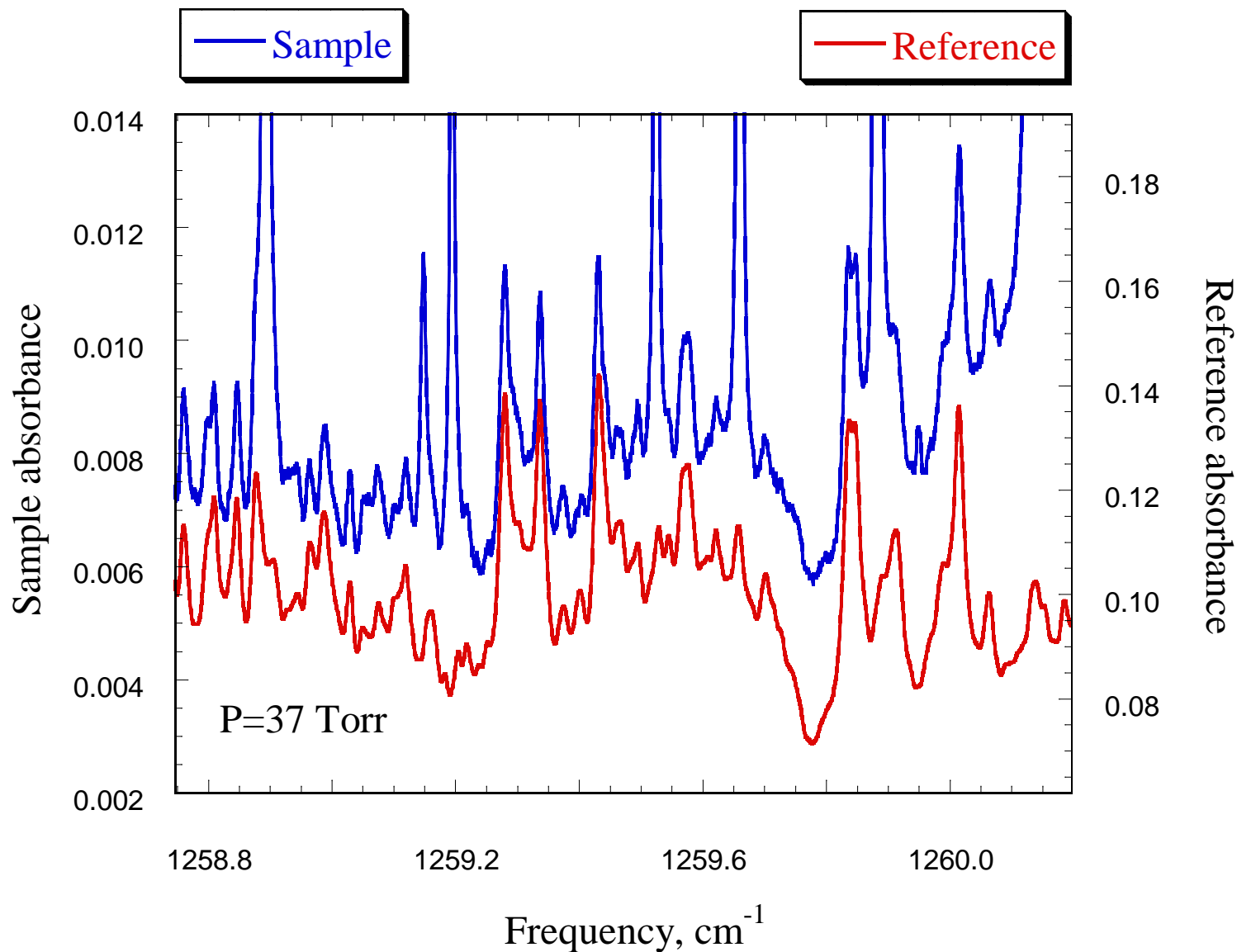
$P=1$ Torr



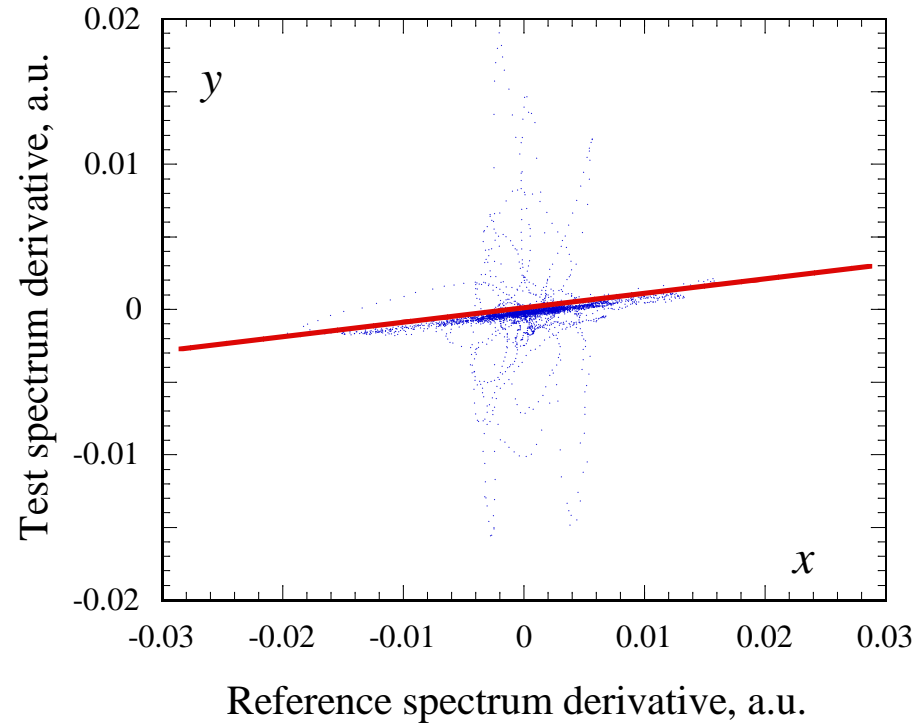
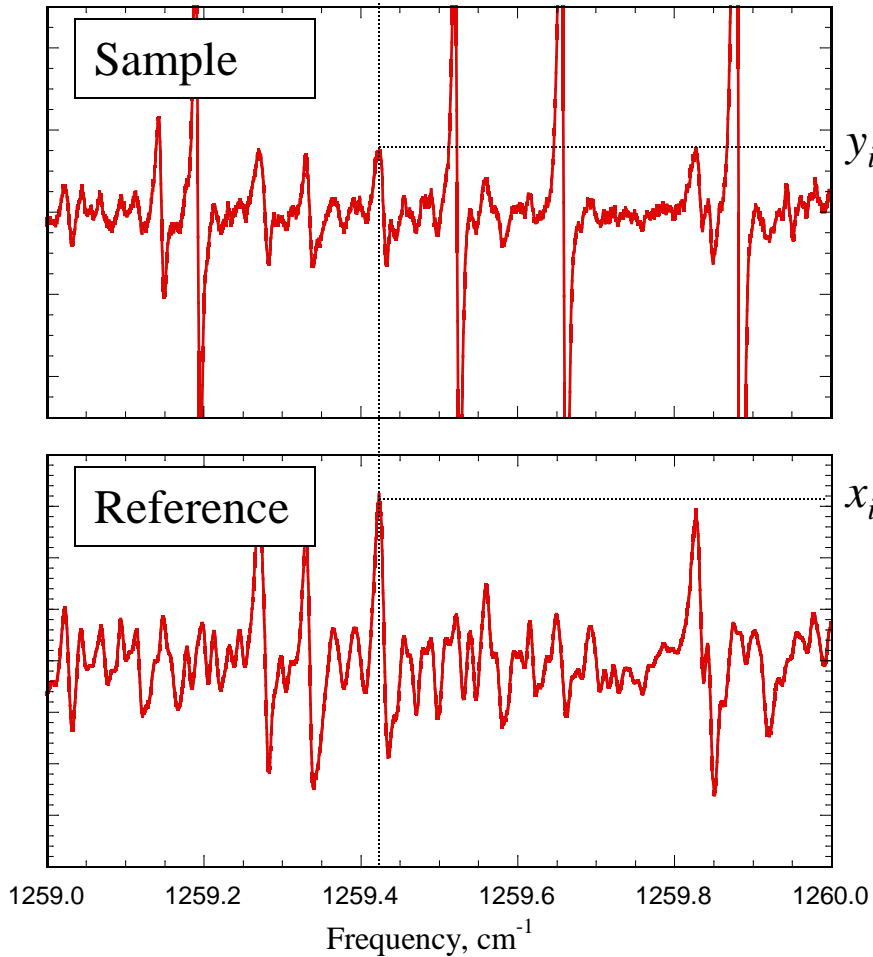
Ethanol vapor +air

$P_{\text{eth}}=1$ Torr, $P=36.6$ Torr

Reference and Sample Spectra of Ethanol in Air



Linear Regression Technique



$$1-D: \quad y_i = ax_i$$

$$\text{MLR:} \quad y_i = \sum_{k=1}^N a_k x_{ki}$$

Results of the Linear Regression Analysis

Species	Measured concentration – sample 1		Measured concentration - sample 2	
	MLR	1-D regression	MLR	1-D regression
C ₂ H ₅ OH	11.60×10^{-6}	12.12×10^{-6}	1.44×10^{-6}	1.41×10^{-6}
CH ₄	1.72×10^{-6}	-	1.70×10^{-6}	-
N ₂ O	0.302×10^{-6}	-	0.301×10^{-6}	-
H ₂ O	1.72×10^{-3}	-	1.73×10^{-3}	-

Pulsed Operation of a QC-DFB Laser

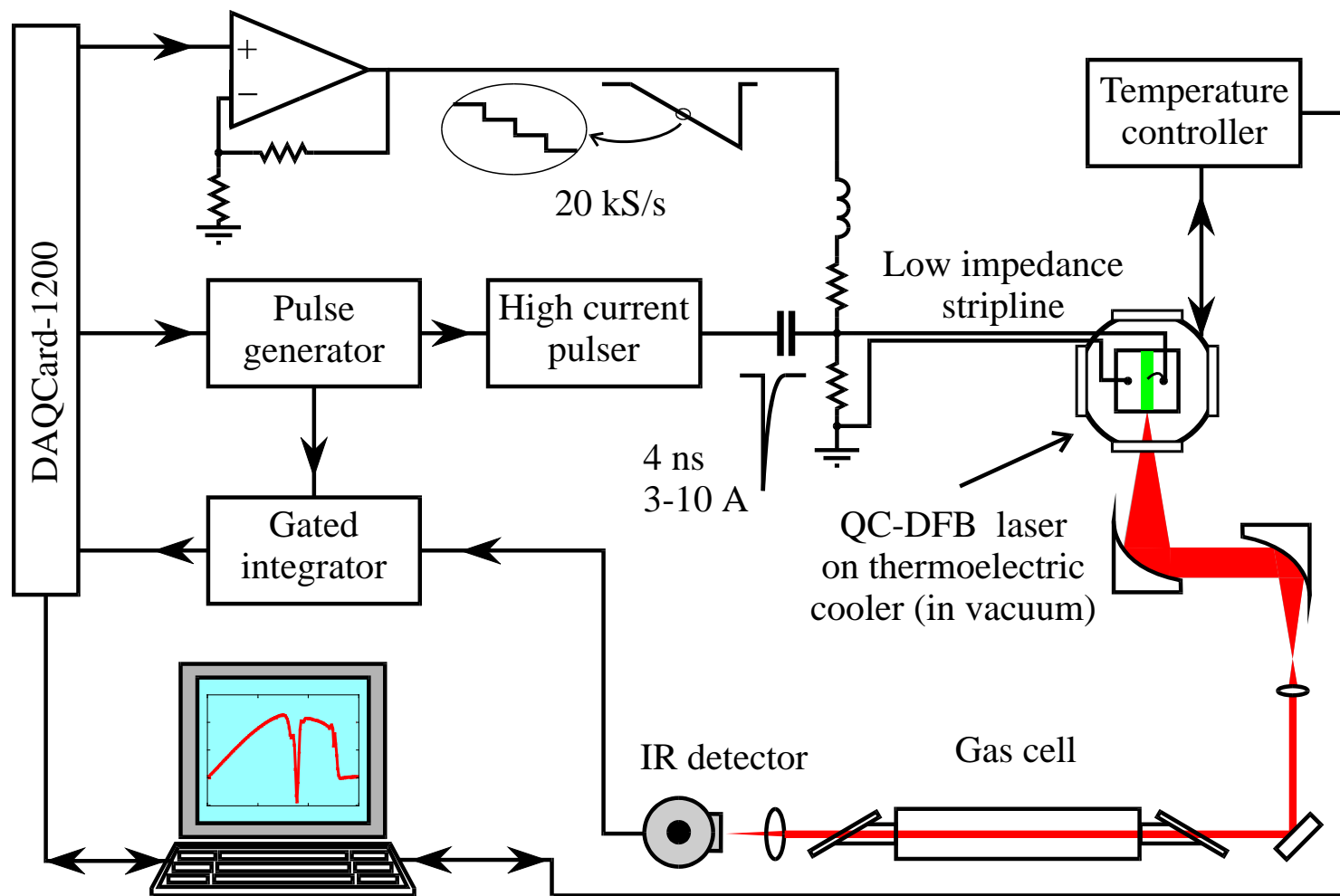
ADVANTAGES

- ♦ Laser can be operated at near-room temperature
- ♦ Easy temperature handling
- ♦ No consumables (liquid N₂)
- ♦ Compact

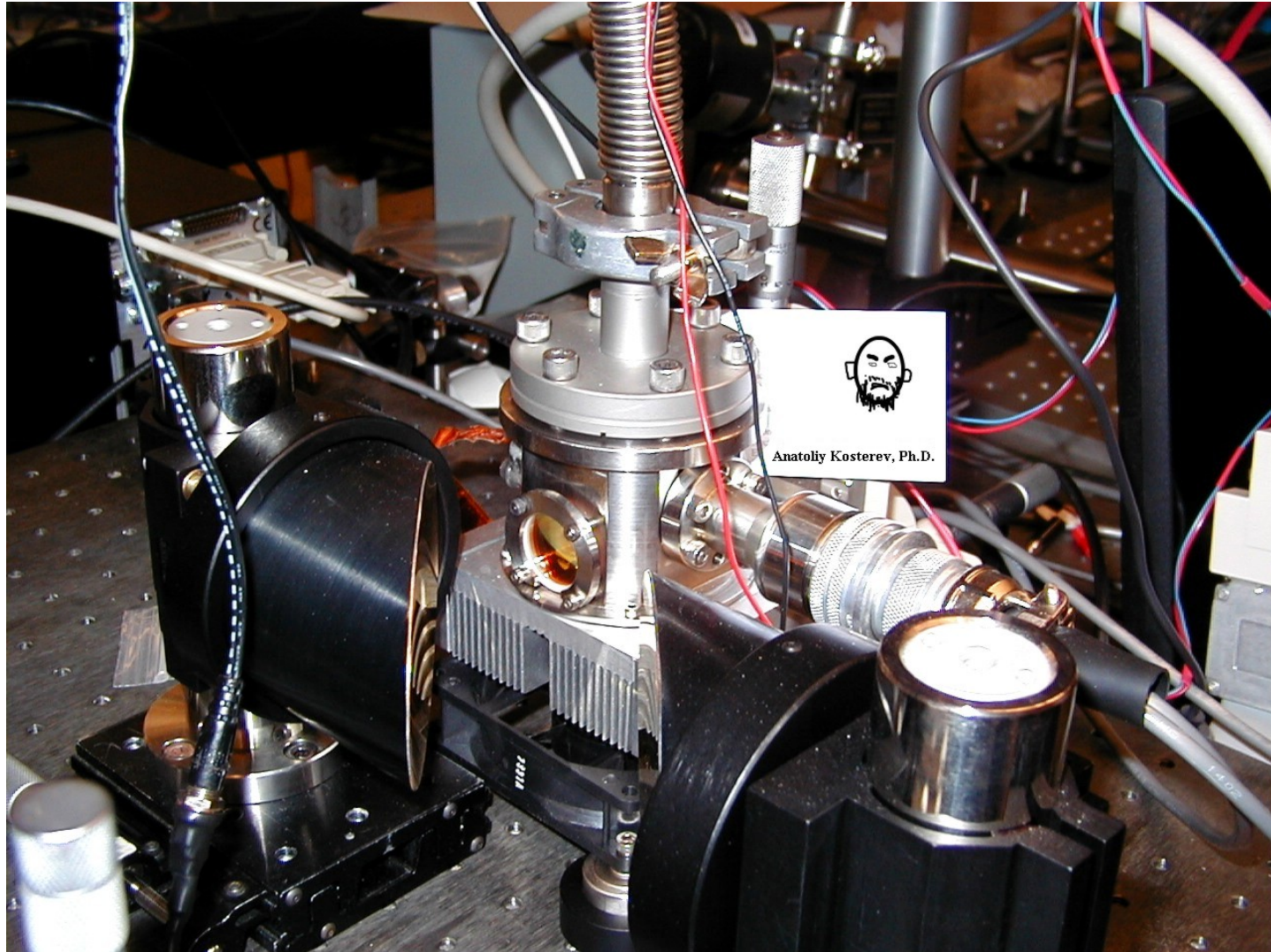
DISADVANTAGES

- ♦ Broader linewidth (~400 MHz)
- ♦ Less average power
- ♦ More sophisticated electronics for driving QC laser and data acquisition

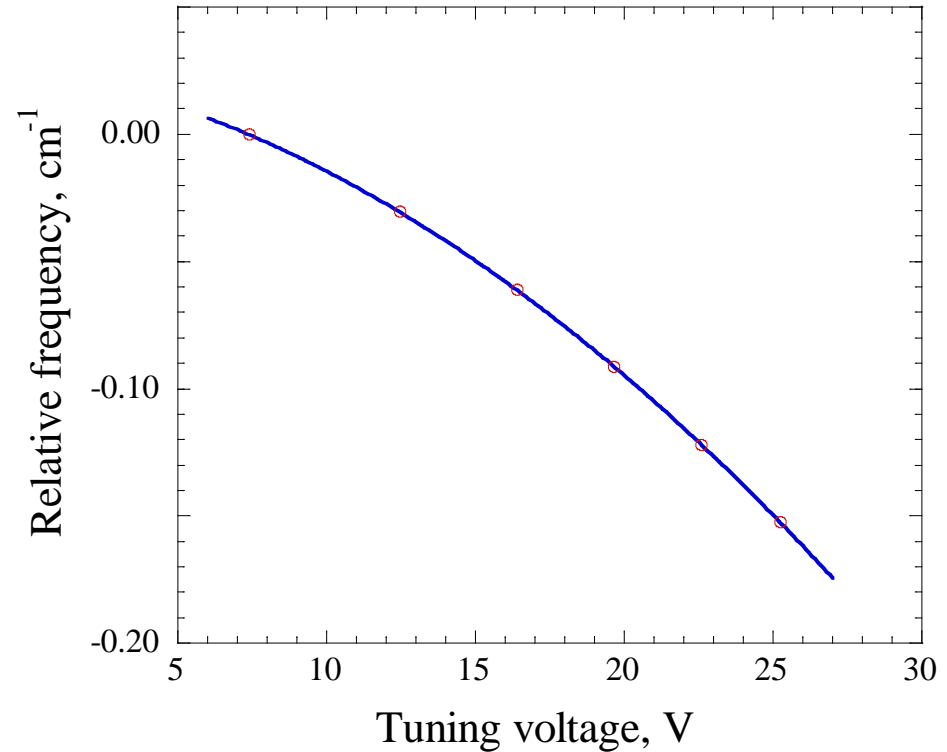
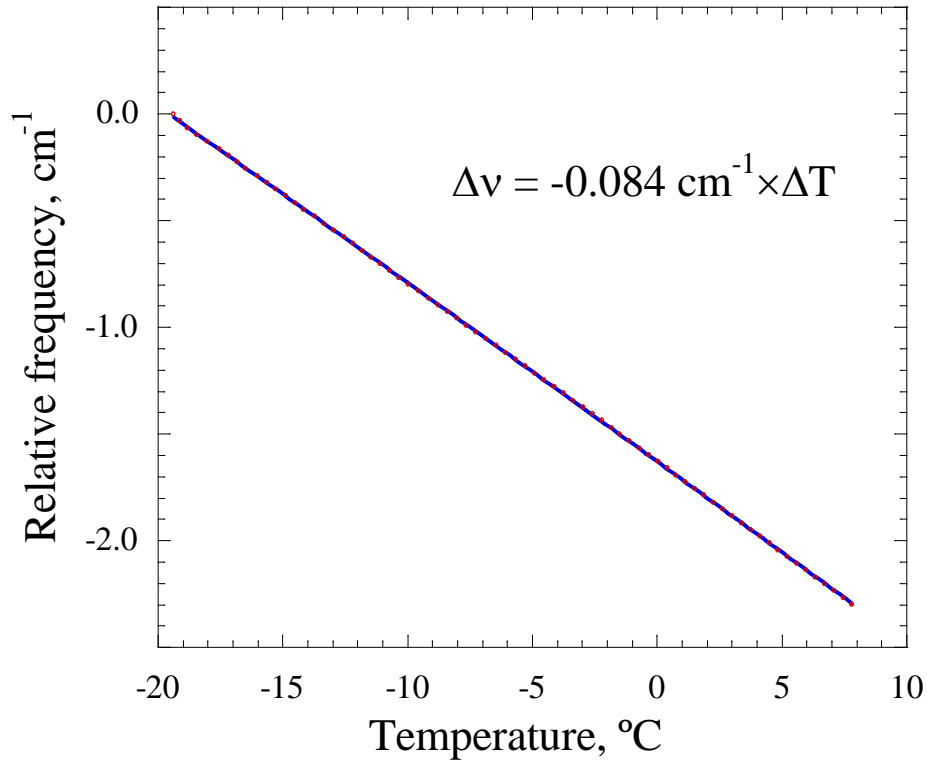
Pulsed QC-DFB Spectrometer



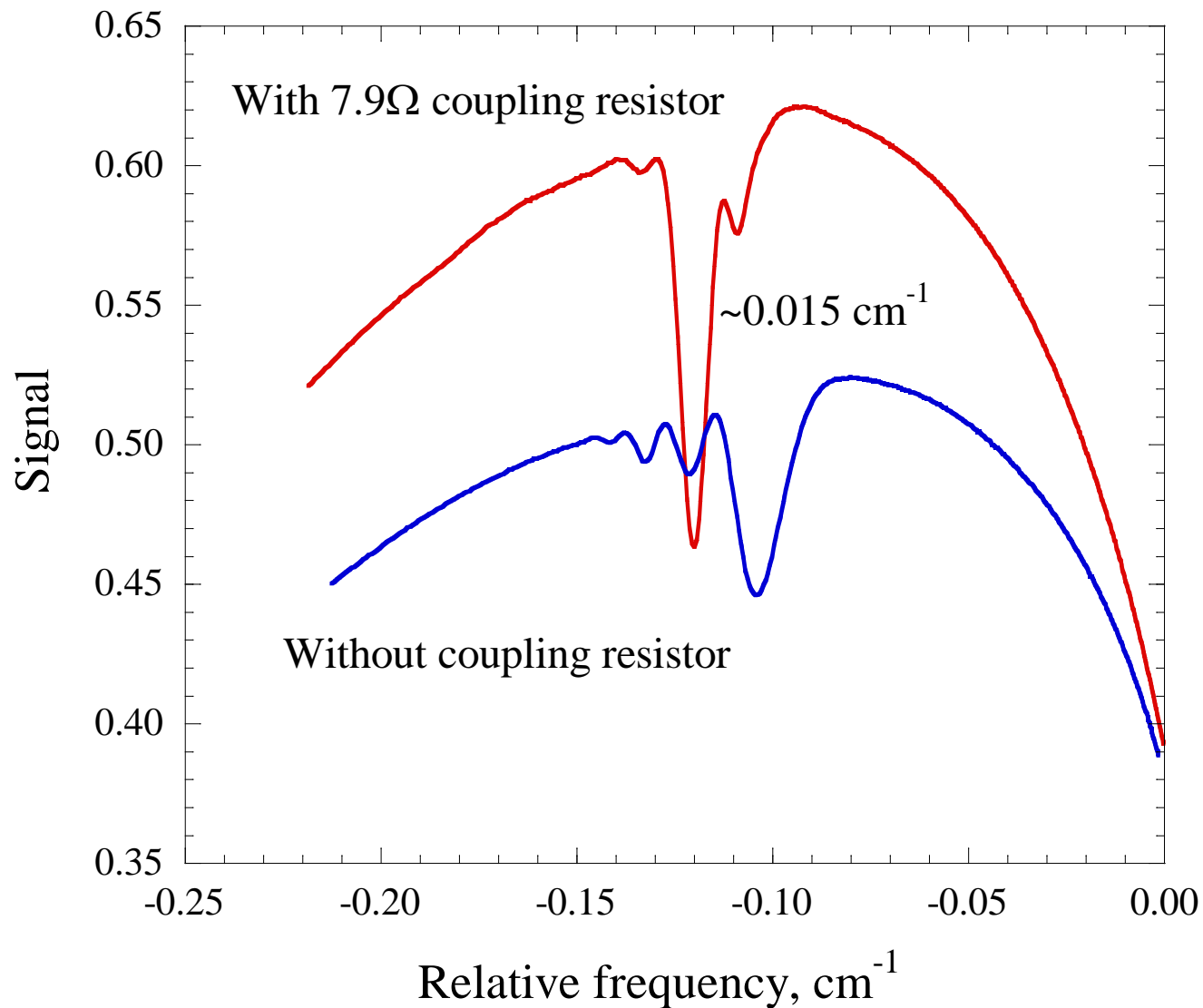
Pulsed QC Laser Housing



Tuning of the Pulsed QC-DFB Laser



Lineshape dependence on electric coupling



Summary and Future Outlook

- A cw QC-DFB laser based gas sensor at $7.9\text{ }\mu\text{m}$ was designed and tested for methane, nitrous oxide, water and ethanol detection in ambient air; a detection limit of 2.5 ppb for CH_4 , 1.0 ppb for N_2O , 60 ppb for H_2O and 125 ppb for $\text{C}_2\text{H}_5\text{OH}$ was achieved
- A new technique for processing congested spectra of VOC was developed and applied to ethanol detection
- A spectrometer based on the near-RT operated pulsed QC-DFB laser was designed and tested

Future development

- ◆ Medical applications - NO detection ($\sim 5\text{ }\mu\text{m}$)
- ◆ Field applications
- ◆ Development of cavity-enhanced detection methods