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Mid-infrared laser based trace gas sensor technologies: recent advances and applications- I

OUTLINE

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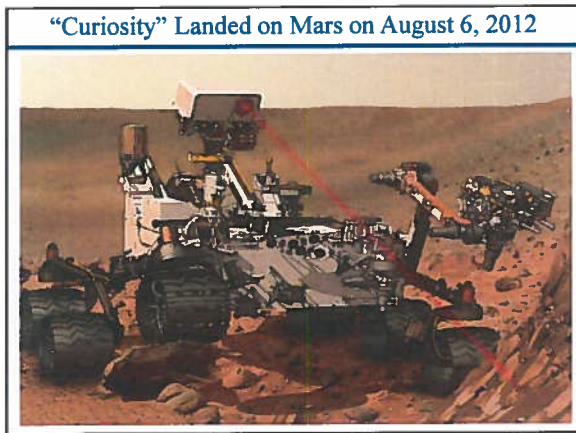
<http://www.ece.rice.edu/~lasersci/>

GE Global Research
Houston, TX
March 7, 2013

Wide Range of Trace Gas Sensing Applications

- **Urban and Industrial Emission Measurements**
 - Industrial Plants
 - Combustion Sources and Processes (e.g. fire detection)
 - Automobile, Truck, Aircraft and Marine Emissions
- **Rural Emission Measurements**
 - Agriculture & Forestry, Livestock
- **Environmental Monitoring**
 - Atmospheric Chemistry (e.g. isotopologues, climate modeling,...)
 - Volcanic Emissions
- **Chemical Analysis and Industrial Process Control**
 - Petrochemical, Semiconductor, Pharmaceutical, Metals Processing, Food & Beverage Industries, Nuclear Technology & Safeguards
- **Spacecraft and Planetary Surface Monitoring**
 - Crew Health Maintenance & Life Support
- **Applications in Medical Diagnostics and the Life Sciences**
- **Technologies for Law Enforcement, Defense and Security**
- **Fundamental Science and Photochemistry**

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Laser-Based Trace Gas Sensing Techniques

- **Optimum Molecular Absorbing Transition**
 - Overtone or Combination Bands (NIR)
 - **Fundamental Absorption Bands (Mid-IR)**
- **Long Optical Pathlength**
 - **Multipass Absorption Gas Cell** (e.g., White, Herriot, Chernin, Aeris Technologies, and Circular Cylindrical Multipass Cell)
 - Cavity Enhanced and Cavity Ringdown Spectroscopy
 - Open Path Monitoring (with retro-reflector or back scattering from topographic target): Standoff and Remote Detection
 - **Fiberoptic & Wave-guide Evanescent Wave Spectroscopy**
- **Spectroscopic Detection Schemes**
 - Frequency or Wavelength Modulation
 - Balanced Detection
 - Zero-air Subtraction
 - **Photoacoustic & Quartz Enhanced Photoacoustic Spectroscopy (QEPAS)**

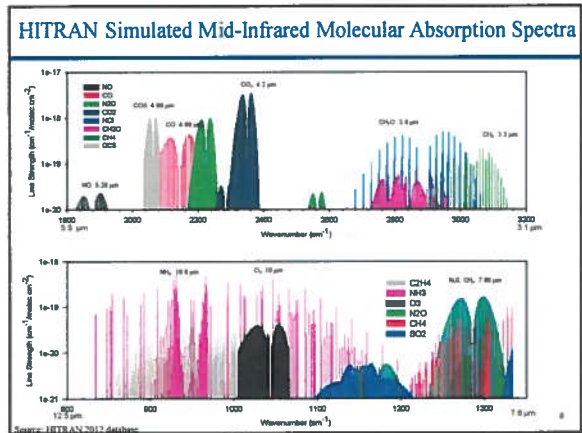
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Other Spectroscopic Methods

- Faraday Rotation Spectroscopy (limited to paramagnetic chemical species)
- Differential Optical Dispersion Spectroscopy (DODiS)
- Noise Immune Cavity Enhanced-Optical Heterodyne Molecular Spectroscopy (NICE-OHMS)
- Frequency Comb Spectroscopy
- Laser Induced Breakdown Spectroscopy (LIBS)

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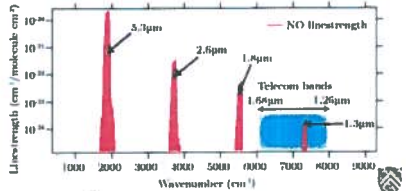
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Selection of Absorption lines in the mid-IR Spectral Range (3 – 5 μm)

- 2.5 μm < λ < 5 μm (4000 cm⁻¹ – 1900 cm⁻¹)
- Access to molecular fundamental rotational-vibrational states
- Atmospheric window (3.5-4.8 μm)

- Applications
- Medicine
 - Sensing
 - Emission monitoring
 - Process control
 - Free-space communication
 - Defense
 - Homeland security

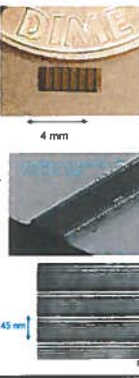


Mid-IR Source Requirements for Laser Spectroscopy

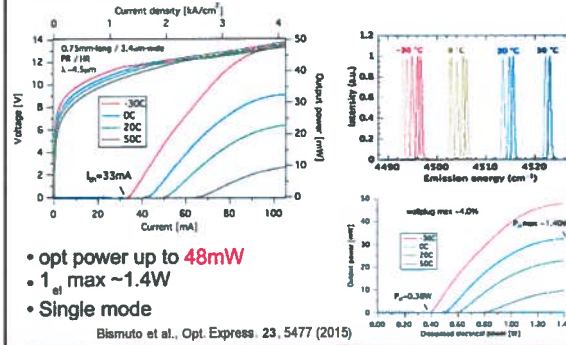
REQUIREMENTS	IR LASER SOURCE
Sensitivity (% to pptv)	Optimum Wavelength and Power
Selectivity (Spectral Resolution) or Specificity	Stable Single Mode Operation and Narrow Linewidth
Multi-gas Components, Multiple Absorption Lines, and Broadband Absorbers	Mode Hop-Free Wavelength Tunability
Directionality or Cavity Mode Matching	Beam Quality
Rapid Data Acquisition	Fast Time Response Time
Room Temperature Operation	High Wall Plug Efficiency, No Cryogenics or Cooling Water
Field Deployable in Harsh Environments	Compact and Robust

Key Characteristics of Mid-IR QCL & ICL Sources – March 2016

- **Band – structure engineered devices**
Emission wavelength is determined by layer thickness – MBE or MOCVD; QCLs operate in the 3 to 24 μm spectral region and ICLs can cover the 3 to 5 μm spectral range
 - Compact, reliable, stable, long lived, and commercially available
 - Fabry-Perot (FP), single mode (DFB) and multi-wavelength devices
- **Wide spectral tuning ranges in the mid-IR**
 - 1.5 cm⁻¹ using injection current control for DFB devices
 - 10-20 cm⁻¹ using temperature control for DFB devices
 - ~100 cm⁻¹ using current and temperature control for QCLs, DFB Array
 - ~525 cm⁻¹ (22% of c.w.) using an external grating element and FP chips with heterogeneous cascade active region design, also QCL, DFB array & Optical Frequency Combs (OFCs) > 100 to <450 cm⁻¹ with kHz to sub-kHz resolution and a comb spacing of > 10 GHz
- **Narrow spectral linewidths**
 - CW 0.1 - 3 MHz & <10kHz with frequency stabilization
 - Pulsed ~300 MHz
- **High pulsed and CW powers of QCLs and ICLs at TEC/RT temperatures**
 - Room temperature pulsed peak power of ~203 W with 10% wall plug efficiency for QCLs
 - ~1 W CW DFB TEC/RT QCL, wall plug efficiency 23% at 4.6 μm
 - ~5-10mW CW DFB ICLs at TEC/RT in the 3 to 4 μm spectral range

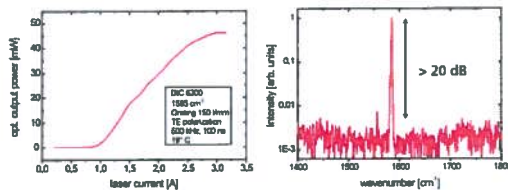


ALPES LASERS Low-dissipation DFB devices at 4.5 μm



A miniaturized External Cavity QCL with MEMS Technology

- Optical output power P > 45 mW @ 1585 cm⁻¹
- SMSR > 20 dB



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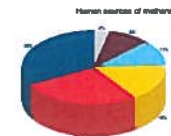


Methane Detection

Methane is one of the major atmospheric greenhouse gases contributing to global warming and climate change.

ARPA-E Monitor Aeris Technologies, Inc and Thorlabs, Inc

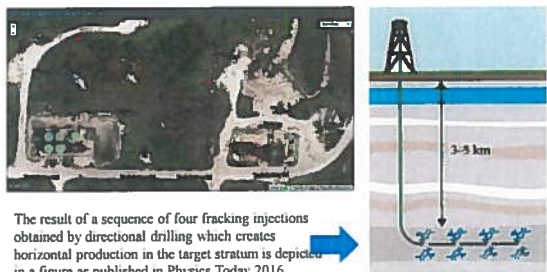
- Global warming potential (GWP) of 25 compared to GWP of 1 for CO₂ for a 100-year period)
- Short lifetime in the atmosphere (~12 yrs) compared to CO₂ and N₂O
- Atmospheric background concentration ~1.8 ppm



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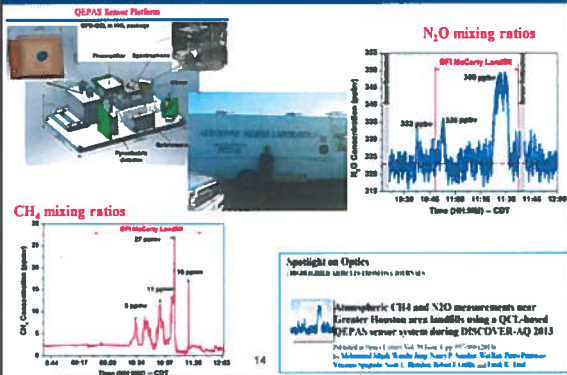
Typical Texas Oil & Gas Production Site near Houston



The result of a sequence of four fracking injections obtained by directional drilling which creates horizontal production in the target stratum is depicted in a figure as published in Physics Today 2016. A DOE-ARPA-E funded methane detection project at 3.33 μm was started in 2015. Texas located wellpad sites (Texas located well pad sites typically measure 10-30 m with a 1 m spatial resolution)



CH₄ and N₂O Measurements performed with a DFB-QCL based QEPAS Sensor installed in a Mobile Laboratory (operated by Aerodyne, Inc.)



Comparison of proposed Rice CH₄ Sensor System and current commercially available CH₄ Sensor Platforms

	Rice	Picarro	ABB-LGR I	ABB-LGR II	Aerodyne
Opt. Path length and method	MIR TDLAS: ~9 m	NIR CRDS: >2000m	NIR OA-ICOS: >1000m	NIR OA-ICOS: >2000m	MIR TDLAS: 70-100 m
Sensitivity/sec	< 5-10 ppb	1-2 ppb	5 ppb	2 ppb	<1 ppb
Accuracy (drift)	2 ppb	2 ppb	20 ppb, temp. stabilized	2 ppb	2 ppb
Cell Volume, cc	60	30	500	2000	2000
Pump Size (10 sec flush time)	~1 lpm	~0.5 lpm	~11 lpm	~45 lpm	~45 lpm
Cavity Mirror Reflectance	98.5%-99%	>99.99%	>99.99%	>99.99%	>99.99%
Power Consumption	2-20 W	200 W	70 W	200 W	400 W
Weight	~2-4 kg	~20 kg	~15 kg	~40 kg	~40 kg
Cost	~20-25K USD	~40-50K USD	~25K USD	~40K USD	~100K USD

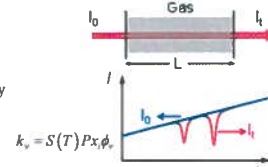
US Department of Energy Advanced Research Project Agency - Energy (ARPA-E) Methane Observation Networks with Innovative Technology to obtain Reductions (MONITOR)



Spectroscopy Fundamentals

Beer's Law: $I_t = I_0 \exp(-k \cdot L)$

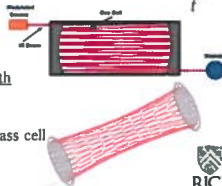
where: I_t is transmitted light intensity
 I_0 is incident light intensity
 k_v is absorption coefficient
 L is optical path



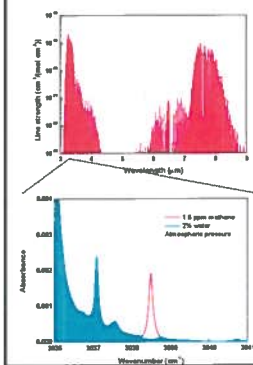
Multipass Gas Cell (MPGC):

The minimum detection limit can be improved by increasing the effective optical path without increasing the physical length.

3D Rice LSG simulation of a multipass cell
 Based on RLSG custom software
 (L=100mm R=100mm D=15mm)



CH₄ Absorption Line Selection

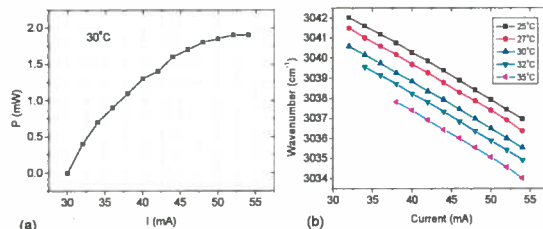


- The fundamental ν_1 and ν_3 CH₄ bands are located at ~7.7 μm and ~3.3 μm, respectively
- A high detection sensitivity for methane measurements using quantum cascade lasers (QCLs) at 7.7 μm was previously reported
- Compact, TEC, CW, DFB ICLs emitting at 3-4 μm wavelengths became recently commercially available
- An interference-free CH₄ absorption line located at 3038.5 cm⁻¹ was selected as the optimum target absorption line
- The 3.3 μm CH₄ absorption line can be used at atmospheric pressure

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ICL Characterization & Performance Evaluation

Nanoplus ICL, 3.291 μm center-wavelength

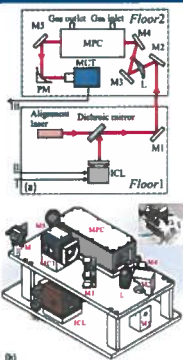


Performance evaluation for a 3.291-μm CW RT ICL at different operating temperatures and injection currents. (a) ICL output power response curves; (b) Emission wavenumber curves.

Current turning rate: -0.232308 cm⁻¹/mA; Temperature turning rate: -0.23994 cm⁻¹/C

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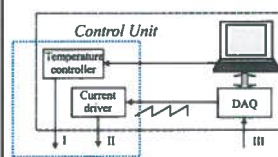
Schematic of current CH₄ Gas Sensor System



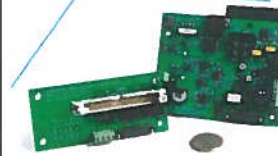
- ICL source (Nanoplus)
 - Current: 42 mA
 - Temperature: 30 °C
 - Power: 1.5 mW
- Multipass gas cell (Sentinel Photonics/Aeris Technologies, Inc)
 - 54.6 meter, 435-passes, sealed
 - Sampling volume: 220 mL
 - Dimensions: 16.9 x 6.6 x 5.3 cm³
- Sensor system platform
 - Two-floor design with folded optical path
 - Low power consumption: 6 W
 - Dimensions: 32 x 20 x 17 cm³



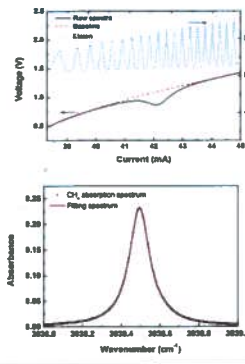
Current Electronics Controller for CH₄ Sensor System



- Control unit
 - Laptop+NI DAQ+OEM laser driver
 - Direct absorption spectroscopy
 - DAQ: acquiring data & scanning the ICL wavelength
- OEM laser driver for ICL
 - Neo Monitors, Oslo, Norway
 - Size: 10 x 8 cm²
 - Low noise characteristic: ≤ 1 nA/√Hz
 - On-board TEC driver: ± 3 A, 15 V
 - Single voltage power supply 12-24V



Data Processing for CH₄ Detection

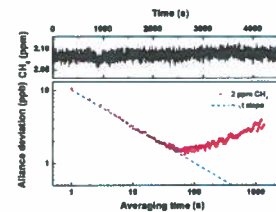


- A 4-step algorithm for CH₄ detection
- 150 spectra were averaged
 - Baseline of the spectral scan was fitted and eliminated
 - Linearized spectrum using fringe spacing of a germanium etalon
 - Lorentzian line shape fitting to retrieve concentration information

Interference-free absorption line of CH₄ at 3038.5 cm⁻¹ obtained from laboratory air at atmospheric pressure together with a fitted baseline and a transmission signal from a germanium etalon.



Allan-Werle Deviation Analysis



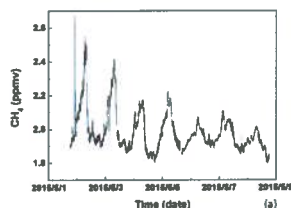
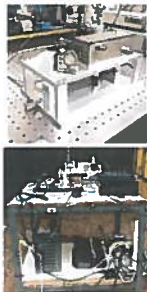
An Allan-Werle deviation plot was acquired in a time period of ~ 1.5 hours using a certified 2 ppm CH₄ cylinder with a 1 Hz sampling rate

1-s measurement precision is $\sigma=10.53$ ppb
60-s measurement precision is $\sigma=1.43$ ppb

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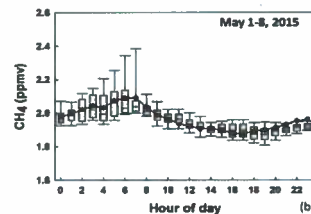
Stationary Laboratory Measurements



CH₄ concentrations measured over a 7-day period in ambient air on the Rice University campus during May 2015.



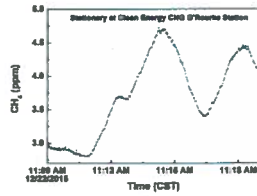
Laboratory Stationary Measurements



Diurnal variations of CH₄ mixing ratio. Bottom whisker, bottom box line, top box line and top whisker indicate the 5th, 25th, 75th and 95th percentile, respectively. Line inside the boxes and continuous solid line represent the hourly median and mean of the data respectively

The diurnal profile of the methane concentration shows an increase in concentration during the early morning with a subsequent gradual decrease after ~8:00 am CDT to its typical background level of ~1.87 ppm in the Greater Houston area in May 2015.

Recent mobile Field Tests: December 2015



CH₄ concentrations measured for a sampling period of ~10 minutes at a Clean Energy CNG O'Rourke Natural Gas Station in Houston, TX.

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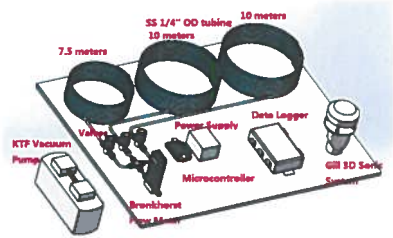


CH₄ Sensor System Summary (2015-2016)

- A **3.291 μm** CW room-temperature ICL based absorption sensor was developed for methane detection using a **54.6 m** optical path length multipass gas cell.
- A two-floor mechanical design with a folded optical path resulted in a sensor system dimension of **32 x 20 x 17 cm³**
- Good electrical power management resulted in a **low power consumption** of the CH₄ sensor system: **6 W**.
- A **minimum detectable Limit (MDL)** of **10.5 ppb** for CH₄ with a **1 sec** integration time was achieved.
- **Laboratory measurements and mobile-mode field tests** were conducted and results demonstrate the suitability of the sensor system to generate CH₄ spatial distributions in a typical U.S. urban area and at an oil and gas storage facility in Houston, TX.

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Portable three Line Methane Sampling System for Laboratory and Field Deployment

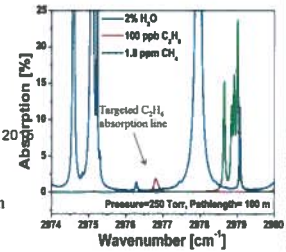


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Motivation for mid-infrared Ethane (C₂H₆) Detection

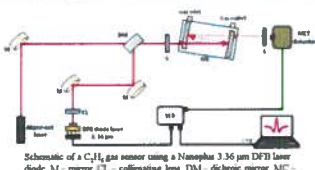
- **Application in medical breath analysis**
 - Asthma
 - Schizophrenia
 - Lung cancer
 - Vitamin E deficiency
- **Atmospheric chemistry and climate**
 - Fossil fuel and biofuel consumption
 - Biomass burning
 - Vegetation/soil
 - Natural gas loss



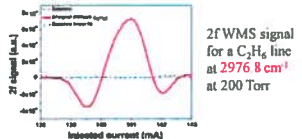
HITRAN absorption spectra of C₂H₆, CH₄, and H₂O



C₂H₆ Detection with a 3.36 μm CW DFB Diode Laser using a novel compact Multipass Absorption Cell and Control Electronics



Schematic of a C₂H₆ gas sensor using a Nanoptics 3.36 μm DFB laser diode. M - mirror, CL - collimating lens, DM - dichroic mirror, MCT - multipass cell, L - lens, SCB - sensor control board.



Minimum detectable C₂H₆ concentration: ~740 pptv (1σ; 1 s time resolution)

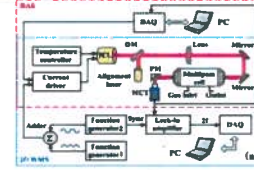


Innovative long path, small volume multipass gas cell: 57.6 m with 459 passes



MGC dimensions: 17 x 6.5 x 5.5 (cm) Distance between the MGC mirrors: 12.5 cm

Improved C₂H₆ sensor system using novel MPGC and a 3.337 μm CW, DFB ICL



The upper part surrounded by red dash line is the detection scheme using direct absorption spectroscopy (DAS) technique, and the lower part surrounded by blue dot line is the detection scheme using second harmonic wavelength modulated spectroscopy (2f-WMS) technique. ICL: interband cascade laser; DM: dichroic mirror; M: plane mirror; PM: parabolic mirror; MCT: mercury-cadmium-telluride detector.

Minimum detectable C₂H₆ concentration: 7.92 ppbv @ 1s for DAS and ~1.19 ppbv @ 4s for WMS were obtained, which are limited by interference fringes due to unwanted etalons. A detection limit of 299 pptv for a data acquisition time of 100s with the use of WMS compared with 777 pptv @ 166s with DAS can be realized.



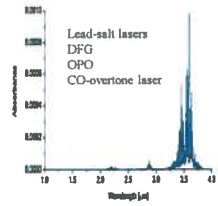
Innovative long path, small volume multipass gas cell: 57.6 m with 459 passes

2f WMS signal for a C₂H₆ line at 2976.8 cm⁻¹ and 200 Torr

MGC dimensions: 17 x 6.5 x 5.5 (cm) Distance between the MGC mirrors: 12.5 cm

Motivation for mid-infrared Formaldehyde Detection

- **Atmospheric chemistry and climate**
 - Important volatile organic compound (VOC) present in all regions of the atmosphere which reacts in the presence of sunlight to yield ozone.
 - Primary H_2CO sources are vehicle exhaust and fugitive industrial emissions
 - Secondary H_2CO sources originate from the breakdown of primary VOCs via photochemical oxidation
- **Industrial Applications**
 - Textile industry
 - Automobile industry
 - Adhesive resins for use in carpeting & plywood

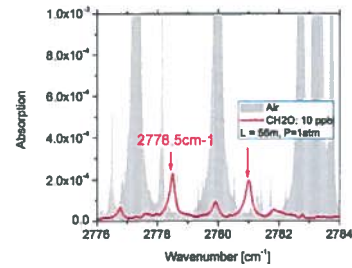


HITRAN absorption spectra of H_2CO

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Formaldehyde Line Selection in the 3-4 μm Spectral Region



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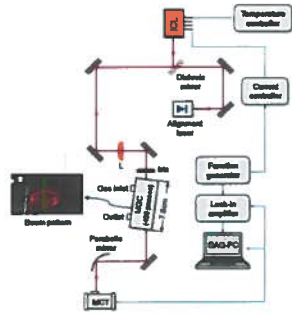
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H_2CO Sensor Configuration

Laser source
 Nanoplus ICL, 3.6 μm
 Injection current: 50 mA
 Output power: 3mW

Compact multipass cell
 Sentinel Inc.
 7.6 cm multipass cell length
 32 ml sampling volume
 3.7 m effective optical length

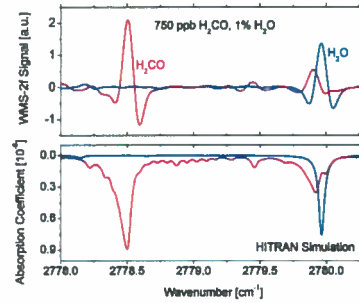
λ -modulation scheme



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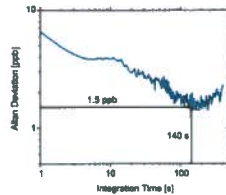
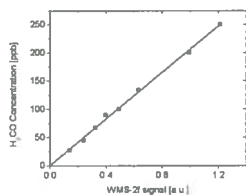
Representative H_2CO Sensor Calibration Results

- H_2CO gas standard: Kin-Tek gas standard generator



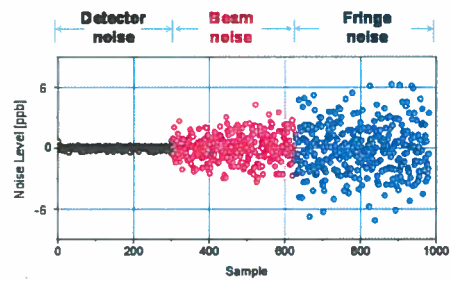
H_2CO Detection Sensitivity

- Minimum detection concentration: 1.5 ppb with a 140 sec averaging time



Noise Limitations

- Zero air measurements: 1s sampling rate



Summary and Future Work (2014-15)

- Development of laser-based absorption sensors for H_2CO detection using an [interband cascade laser](#) & a [compact xx m multipass absorption cell](#).
- A minimum detection concentration of **1.5 ppb with 140 sec averaging time** was achieved.
- Future work is planned to further improve the sensor detectivity to sub-ppb concentration level by using a multipass cell with an increased effective optical path length. Preliminary results show that a minimum detection concentration of **1 ppb with 10 sec averaging time** can be achieved.

Summary, Conclusions and Future Developments

- Development of robust, compact, sensitive, selective mid-IR trace gas sensor technology based on RT, CW high performance DFB ICLs & QCLs for detection of explosives and TICs as well as [environmental monitoring and medical diagnostics](#).
- Interband cascade and quantum cascade lasers were used in [QEPAS and TDLAS based sensor platforms](#).
- Performance evaluation of seven target trace gas species were reported. The minimum detection limit (MDL) with a 1 sec sampling time were:
 - C_2H_6 MDL of 24 ppbv at $\sim 3.36 \mu m$, CH_4 MDL of 13 ppbv at $\sim 7.28 \mu m$, N_2O MDL of 6 ppbv at $\sim 7.28 \mu m$.
- Development of Trace Gas Sensors for the monitoring of broadband absorbers: acetone (C_3H_6O), propane (C_3H_8), benzene (C_6H_6)

Hydrogen Peroxide (H_2O_2)

- Strong oxidant species in the atmosphere
- Associated with the formation of acid rain and atmospheric aerosols
- Employed in the synthesis of multiple chemical products & as bleaching agent in the pulp and paper industry
- Used for decontamination and sterilization of medical and pharmaceutical facilities
- Biomarker of lung and respiratory system diseases in exhaled breath



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Vapor-Phase Hydrogen Peroxide (VPH)

- VPH is used for:
 - Decontamination of health-care and pharmaceutical facilities
 - Sterilization of medical equipment and packing materials in the food industry
- VPH units: gas-phase H_2O_2 generated from concentrated liquid H_2O_2 solutions
- H_2O_2 concentrations between 200-1200 ppm are produced in the gas-phase and maintained for ~ 10 min
- After decontamination procedures, ambient H_2O_2 concentrations need to be monitored



Source: Bioquell UK Ltd

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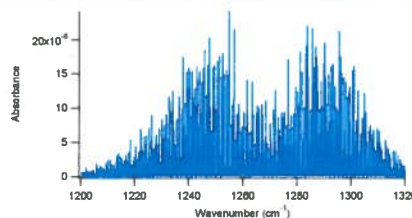
Techniques for H_2O_2 Detection

- Wet-chemistry methods based on fluorescence spectroscopy, colorimetric analysis and chemiluminescence
 - Transfer from gas to liquid phase required for subsequent analysis
 - Interference from other species and formation of sampling artifacts
- Mid-infrared laser based spectroscopy
 - Direct detection in the gas-phase
 - Real-time detection
 - High sensitivity and specificity

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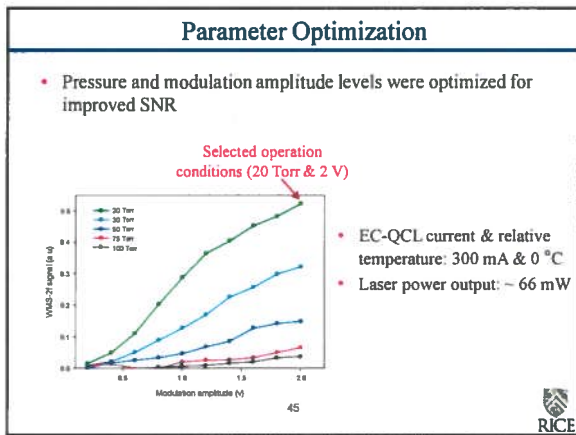
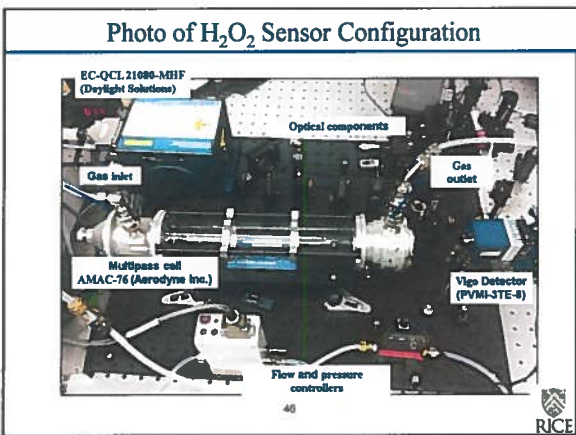
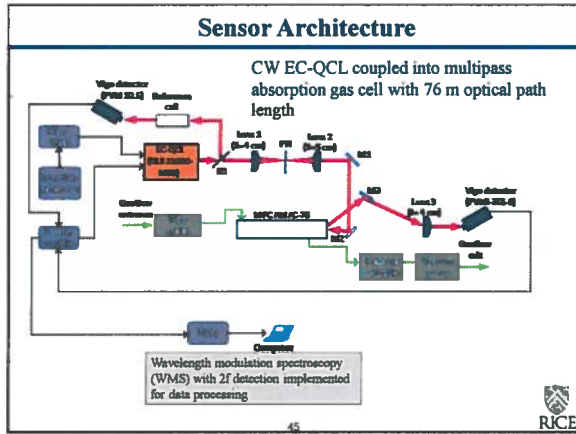
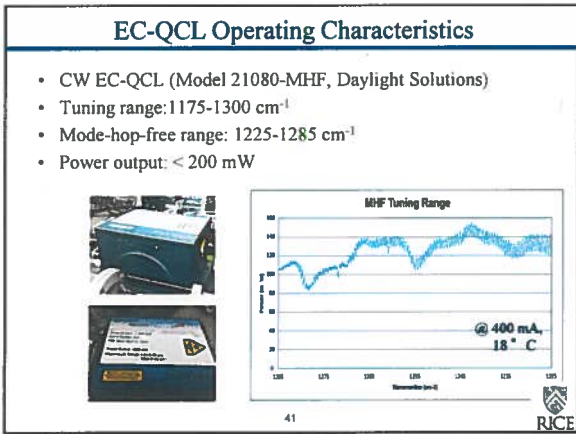
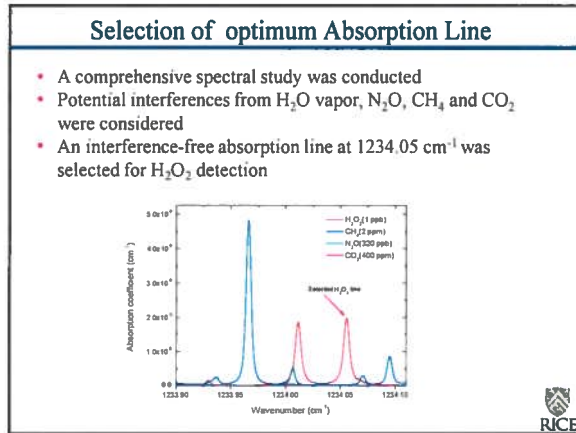
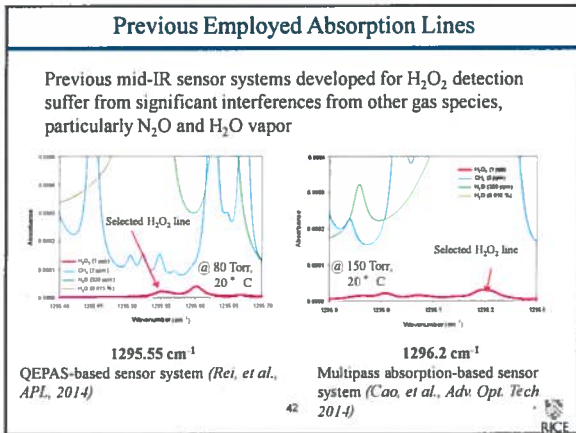
H_2O_2 Absorption in the mid-infrared spectral Region

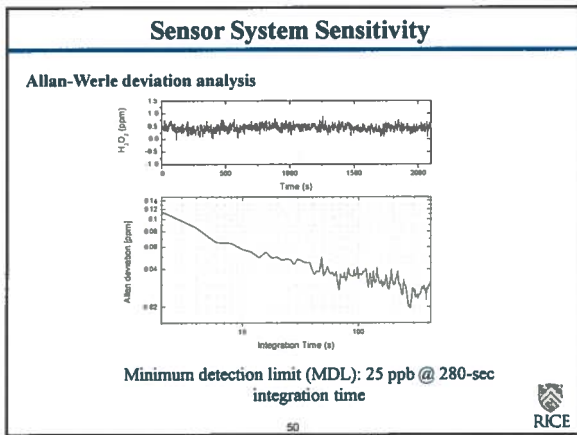
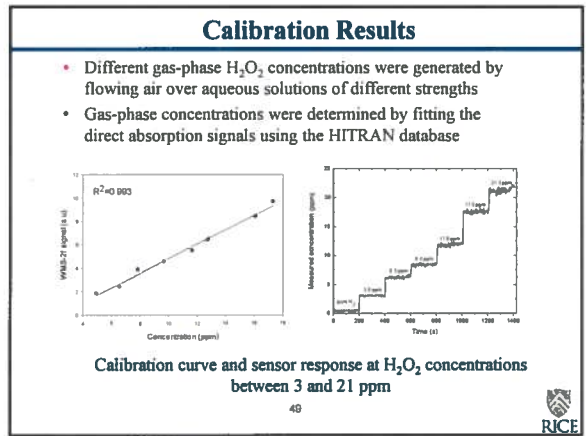
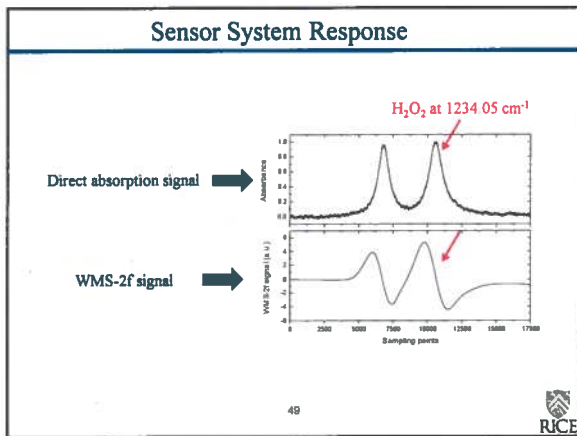


Fundamental ν_6 H_2O_2 band located at ~ 7.5 - $8.3 \mu m$

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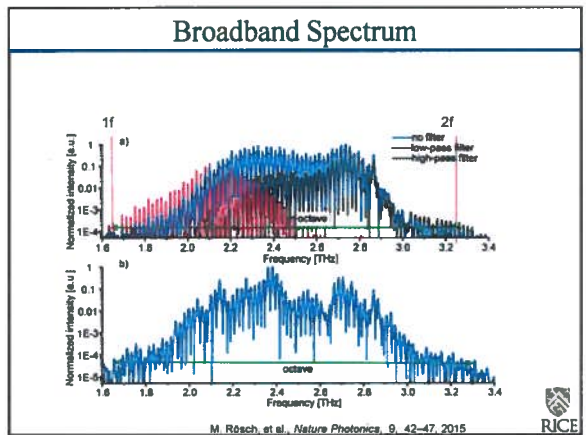
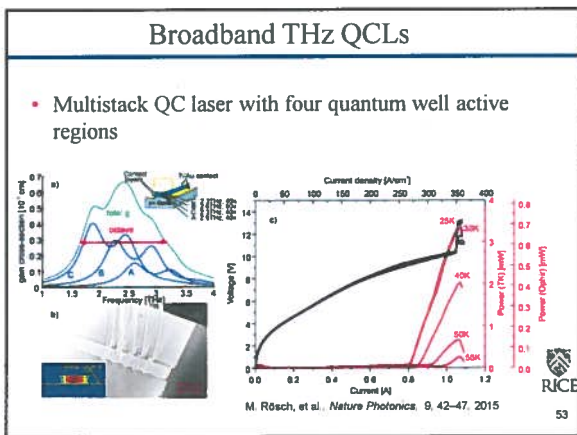




H_2O_2 Sensor System Summary

- Selected absorption line at 1234.05 cm^{-1} effectively alleviates interference issues identified previously for H_2O_2 detection
- MDL and ability to operate with no interference from water make our sensor system suitable for the monitoring of H_2O_2 in:
 - Industrial sites to establish possible exceedances of OSHA permissible exposure levels (PELs)
 - Decontamination/sterilization locations using VPHP
 - Exhaled breath as biomarker of lung-related diseases
- Further improvement of the MDL is necessary for application in other fields such as atmospheric monitoring

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Experimental setup

- Mix a DFB QCL with a source operating by DFG (widely tunable)

P. Friedli, et al., Appl Phys Lett, 102, 222104, 2013.

Use a Fabry-Perot as a spectroscopy tool

- Can be used to measure the spectrum
- S/N limited by stability

Cavity enhanced optical frequency comb spectroscopy

For an optimum build-up of pulses inside a sensing cavity resonator, three conditions must be met:

- Sensing cavity has to support equidistant frequency eigenmodes.
 - Intracavity dispersion compensation
- Separation of the cavity eigenmodes and the separation of the frequency comb modes must be equal.
 - Matching both cavity length
- entire comb must be shifted to overlap with the cavity eigenfrequencies.
 - Alignment of the eigenfrequencies

What is an optical frequency comb?

Array of N independent DFBs combined: line-to-line frequency noise is uncorrelated

$$\delta(\omega_i - \omega_j) \approx \sqrt{\delta(\omega)}$$

Optical frequency comb: line-to-line frequency noise is correlated

$$\delta(\omega_i - \omega_j) \ll \delta(\omega)$$

The beat tone has a very narrow linewidth in a comb

Still, the array source may still be very interesting for some applications!!!

Cavity design for ICL-comb source

HR mirrors radius of curvature calculation

Cavity stability criterion: $1 - \frac{L_{cav}}{R_1} < 0$

Specs for cavity mirrors design: For $L_{cav} = 30 \text{ mm}$

$$1 - \frac{L_{cav}}{R_1} = 1 - \frac{L_{cav}}{R_2} = -\frac{1}{2}$$

$$R_1 = R_2 = 20 \text{ mm}$$

- Plano-concave
- Diameter: $\frac{1}{2}$ "
- Radius of curvature: 20mm

R_1 : radius of curvature of mirror #1
 R_2 : radius of curvature of mirror #2

H.Kogelnik, Proc. IEEE 54,1312-1329, 1966
 N.J.Delmon & H. Weber, "Laser Resonators & Beam Propagation" Chapter 1, Springer 2005

Cavity design for ICL-comb source

Optical Cavity Performances:

- For effective length: $\zeta = \frac{\pi}{1 - R_{mirror}} \approx 15700$
- Number of half round-trip: $N = \frac{L_{eff}}{L_{cav}} = 6667$
- Fraction of residual optical intra-cavity power after N half round-trips: $R_{mirror}^N = 26\%$
- Intra-cavity power enhancement factor: $G = \frac{\zeta}{\pi} = 5000$
- Width of the cavity mode: $\Delta\nu = \frac{FSR}{\zeta} = 637 \text{ KHz}$

Cavity design for an ICL-comb source

Cavity length calculation

Frequency comb spacing:

$$\Delta f = \frac{c}{2nL}$$

For ICL cavity length:

$L = 2 - 4 \text{ mm}$

2mm	20GHz
3mm	15GHz
4mm	10GHz

Cavity mode spacing:

$$FSR = \frac{c}{2L_{cav}}$$

Mode matching between comb and cavity modes: $\Delta f = n \cdot FSR$

For $n=2$

20GHz	15 mm
15GHz	20 mm
10GHz	30 mm

The possibility to tune the comb modes spacing allows us to choose a cavity length of 30mm, maximizing the effective absorption length.

Sensing cavity design for ICL-comb source

Specifications for cavity mirror material

- Transmittance within the 3-4 μm spectral range
- Transparent in the visible (red diode laser for pump beam alignment)
- High surface quality (for HR and AR coatings to be applied)

AR coatings:

HR coatings:

$\lambda = 3590 \text{ nm}$

$R_{\text{mirror}} \geq 99.98\%$

LohnStar Optics was contacted on Jan. 14, 2016 for a quotation of AR & HR coatings on sapphire optics from Mellor Optics, Inc. | <http://www.lohnstaroptics.com>; www.melloroptics.com |

Proposed Sensing Cavity Locking to JPL-NRL ICL

Cavity locking system

Control cavity length

Locking feedback loop

Critical requirement: resonances of the optical cavity must coincide with the comb teeth of the optical source

↓

Cavity length must be adjusted

↓

Locking of feedback loop will be realized by means of a **piezo-actuator**

↓

Rice team will design a quasi Pound-Drever-Hall circuit to lock the output of the optical comb to the external cavity

Schematic of CE-OFC sensor system

CE-OFC-SENSOR

TARGET: detection and identification of toxic industrial chemicals with strong absorption features in the 3-4 μm range.

Sensor system will offer:

- High detection sensitivity
- High spectral stability and purity
- High selectivity
- Possibility of multi-gas detection

Two Sensor Configurations:

Comb laser is locked to a high finesse cavity

Cavity is locked to comb laser

Fast detector

- Use an intersubband Quantum Well Infrared Photoconductor (QWIP) detector

Detector: H.C. Liu
Ref: H. Schneider and H.C. Liu, Springer book

Is it a comb?

- Beat note measurement of the photocurrent (at 7.5GHz)
 - For modes of amplitude E_k , the photocurrent at $\Delta\omega$

$$I^{\Delta\omega} = \sum_k E_k E_{k+1} \cos(\phi_{k+1} - \phi_k)$$

Frequency noise of a comb

- Use a (matched) optical cavity as an optical discriminator
- Measures all the modes at once

Uncorrelated modes would yield no broader than $N\sqrt{N} \approx 300 \times$

$\sqrt{N} \delta \omega_i$

66 F. Cappelli, G. Villares et al, ArXiv RICE

Frequency noise power spectral density

- Single mode versus comb
- Above noise floor
- No significant limit

67 RICE

Beatnote spectrum

- The very narrow width confirms the correlations between modes
 - Uncorrelated lines could not be narrower than Schawlow-Townes (100's Hz)
 - However the signal is only about 2% of the c.w. photocurrent

67 RICE

Summary, Conclusions and Future Developments

- Development of robust, compact, sensitive, selective mid-IR trace gas sensor technology based on RT, CW high performance DFB ICLs & QCLs for detection of explosives and TICs as well as environmental monitoring and medical diagnostics
- Interband cascade and quantum cascade lasers were used in QEPAS and TDLAS based sensor platforms
- Performance evaluation of seven target trace gas species were reported. The minimum detection limit (MDL) with a 1 sec sampling time were :
 - C_2H_2 : MDL of .24 ppbv at $\sim 3.36 \mu m$; CH_4 : MDL of 13 ppbv at $\sim 7.28 \mu m$; N_2O : MDL of 6 ppbv at $\sim 7.28 \mu m$.
- I-QEPAS demonstration with a power enhancement factor of 240 providing a corresponding increase in detection sensitivity
 - CO_2 : for the P(42) absorption line located at $\sim 4.33 \mu m$ (2311.105 cm^{-1}), a MDL of 300 pptv at 50mbar was achieved for a 20 sec integration time.
- Development of "active" I-QEPAS system for CO and NO detection in the few ppt range
- Development of Trace Gas Sensors for the monitoring of broadband absorbers: acetone (C_3H_6O), propane (C_3H_8), benzene (C_6H_6)
- Development of Mid-IR Electrically pumped Interband Cascade Optical Frequency Combs (OFCs) with JPL, Pasadena, CA, NRL, Washington, and the U of Bari (Italy)

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- NSF-ANR (France) award for international collaboration in chemistry
- Robert Welch Foundation grant C-0586

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New QCL data at $1 \sim 3.3 \mu m$

72 RICE