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Environmental ammonia monitoring for urban and rural areas of Texas using optical EC-QCL based sensor platform

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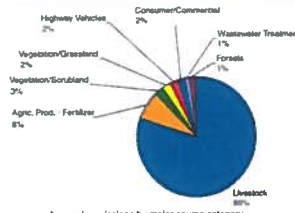
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- Motivation: Environmental monitoring of ammonia (NH₃) based on an optical sensor platform
- NH₃ sensor platform architecture
- NH₃ absorption line selection
- Detection results and long term measurements
- Summary and Future Directions

Special acknowledgement to Prof. Barry Lefer (University of Houston) for his hospitality

Ammonia (NH₃) facts

- NH₃ plays an important role in atmospheric chemistry.
- The total estimated global emission of NH₃ to the atmosphere is ~ 45 TgN/year
- Industrial and motor vehicles activities can be important in urban areas.
- In the atmosphere, NH₃ reacts with different acid pollutants forming ammonium particulates and aerosols.
- Ammonia is a potential source of atmospheric NO and N₂O.




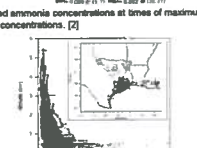
Ammonia emissions by major source category in Texas [1]

$\text{NH}_3 + \text{OH} \rightarrow \text{NH}_2 + \text{H}_2\text{O}$
 $\text{NH}_2 + \text{O}_3 \rightarrow \text{NH} + \text{HO}_2$
 $\text{NH}_2 + \text{NO}_2 \rightarrow \text{N}_2 + \text{H}_2\text{O}$

[1] R. T. Pavlovic, U. Nopmongkol, Y. Kimura and D. T. Allen, Atmospheric Environment 40, 536-551 (2006)

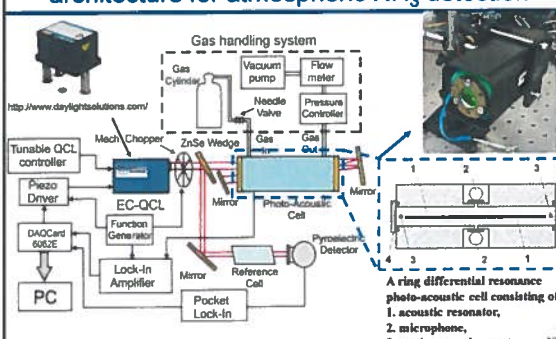
Estimated ammonia levels for Houston

- Atmospheric NH₃ concentrations for urban and industrial areas vary between 0.1 and 10 ppbv [1].
- The modeled ammonia concentration for Houston was estimated to range between 1 and 15 ppb [2].
- The typical NH₃ mixing ratios from area sources range from 0.2 to 3 ppbv, but for unexpected industrial accidents or events can increase even to 80 ppbv [3].

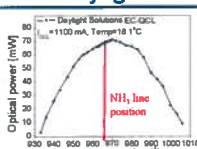

[1] J. H. Seinfeld, S. N. Pandis, Atmospheric Chemistry and Physics, John Wiley and Sons, Inc., Hoboken, NJ (2006)
 [2] R. T. Pavlovic, U. Nopmongkol, Y. Kimura and D. T. Allen, Atmospheric Environment 40, 536-551 (2006)
 [3] Brown, J. B., et al., J. Geophys. Res., 115, D27204, (2010)

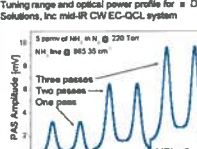
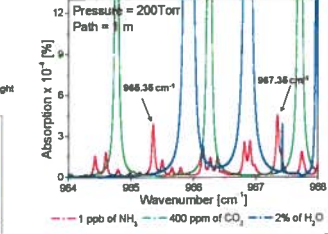
Mid-IR EC-QCL based AM-PAS sensor architecture for atmospheric NH₃ detection



A ring differential resonance photo-acoustic cell consisting of:
 1. acoustic resonator,
 2. microphone,
 3. gas input and output,
 4. window

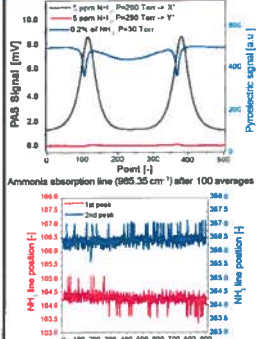
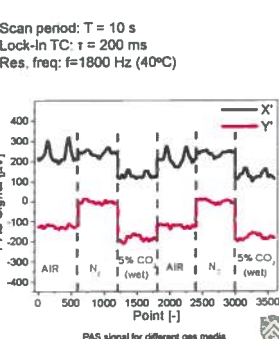
PAS based NH₃ measurements with a 10.34 μm Daylight Solutions CW EC-QCL source

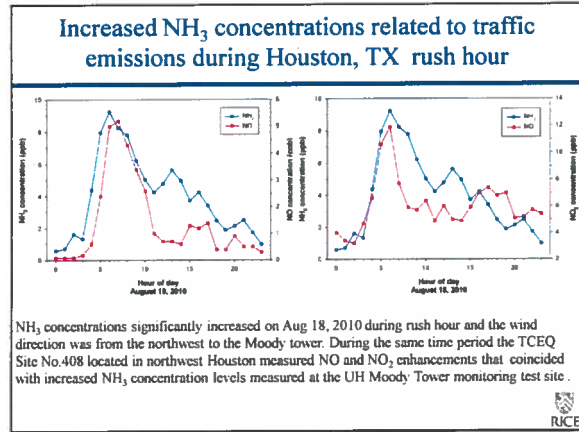
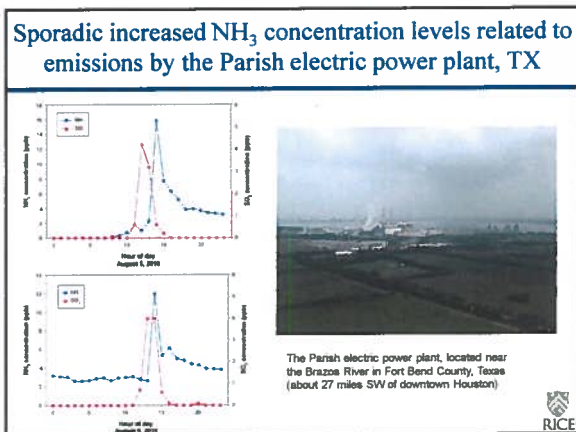
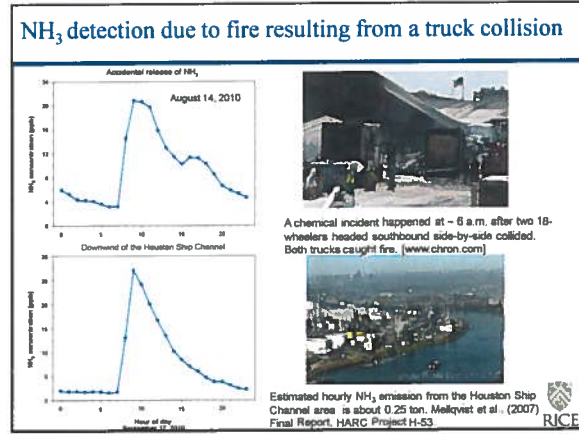
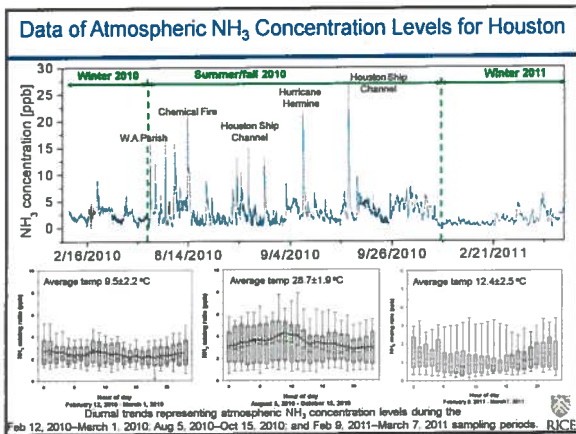
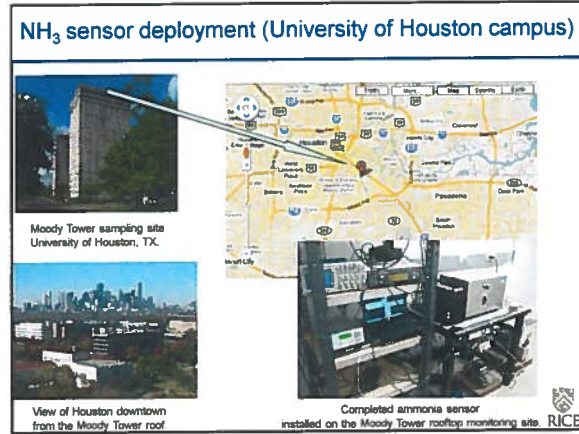
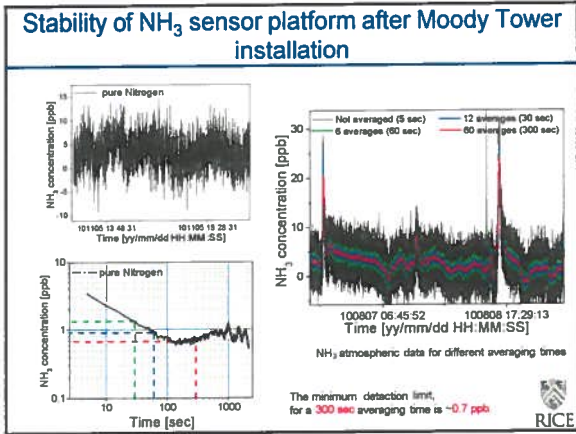



AM-PAS signal for 5 ppmv NH₃ in N₂ after one, two, and three QCL output beam passes through the photoacoustic cell

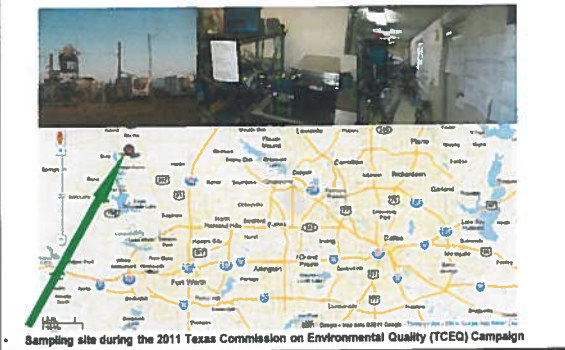
Data Acquisition Procedure

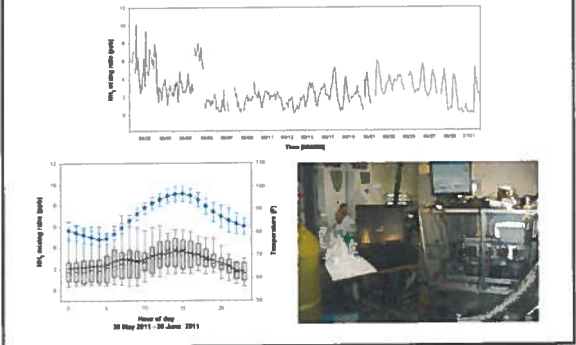
Scan period: T = 10 s
 Lock-In TC: τ = 200 ms
 Res. freq: f = 1800 Hz (40°C)



NH₃ sensor deployment in Fort Worth (CAMS 75)



Preliminary results for a May-June 2011 TCEQ campaign - Fort Worth, TX

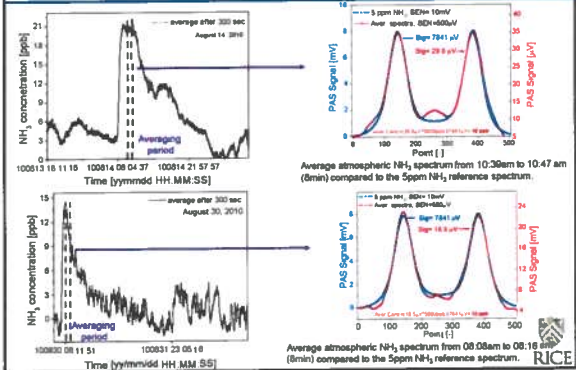


Summary

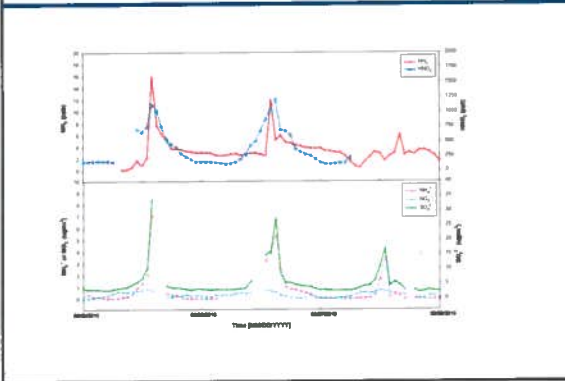
- A 10.4 μm EC-QCL based ammonia sensor, employing conventional photo-acoustic spectroscopy, was demonstrated.
- The minimum detection limit, obtained for the NH₃ absorption line at 965.35 cm⁻¹, was ~0.7 ppb for a 300 sec averaging time.
- The NH₃ sensor is capable of unattended operation and continuous data acquisition for extended periods of time.
- Remote access via internet, enabling real-time monitoring of the sensor performance was established.
- Several unexpected episodes of high NH₃ mixing ratios were observed.
- The Houston Ship Channel is a significant contributor. However, NH₃ source attribution is influenced by factors such as boundary layer dynamics, meteorology, and the lack of sufficient emission inventory data.
- The AM-PAS based NH₃ sensor was also field tested at the CAMS75 monitoring site in the Fort Worth/Dallas area, between May 31, 2011 and June 30, 2011, as part of the Summer 2011 Texas Commission on Environmental Quality (TCEQ) Campaign.



NH₃ Measurements in Air



Results – simultaneous measurements



Weekday-weekend differences in NH₃ concentrations

