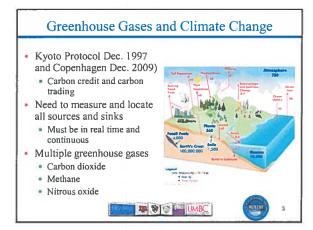
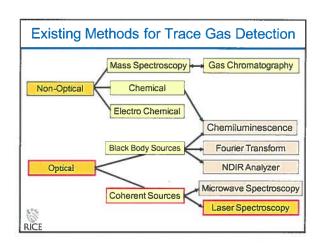


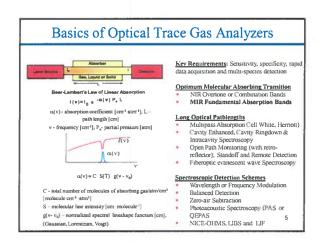
Wide Range of Trace Gas Sensing Applications Urban and Industrial Emission Measurements Industrial Plants Combustion Sources and Processes (e.g. fire detection) Automobile, Truck, Aircraft and Marine Emissions Rural Emission Measurements Agriculture & Forestry, Livestock Environmental Monitoring Atmospheric Chemistry Volcanic Emissions Chemical Analysis and Industrial Process Control Petrochemical, Semiconductor, Nuclear Safeguards, Pharmaceutical, Metals Processing, Food & Beverage Industries Spacecraft and Planetary Surface Monitoring Crew Health Maintenance & Life Support Applications in Biomedical and the Life Sciences

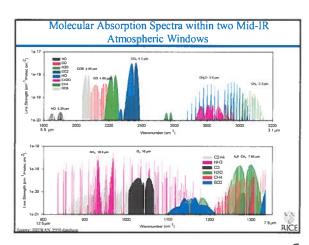
Technologies for Law Enforcement and National Security

Fundamental Science and Photochemistry











Mid-IR Source Requirements for Laser Spectroscopy

REQUIREMENTS	IR LASER SOURCE
Sensitivity (% to ppt)	Optimum Wavelength, Power
Selectivity (Spectral Resolution)	Single Mode Operation and Narrow Linewidth
Multi-gas Components, Multiple Absorption Lines and Broadband Absorbers	Tunable Wavelength
Directionality or Cavity Mode Matching	Beam Quality
Rapid Data Acquisition	Fast Time Response
Room Temperature Operation	No Consumables
Field deployable	Compact & Robust

•	Band – structure engineered devices (Emission wavelength is determined by layer thickness – MBE or MOCVD); mid-infrared OCLs operate from 3 to 24 µm (AlinAs/GalnAs)	
	Compact, reliable, stable, long lifetime, and commercial availability	
	Fabry-Perot (FP), single mode (DFB) and multi-wavelength	IIIIIII
• 2	= 10-20 cm ⁻¹ using temperature control for DER devices	4 mm
	Narrow spectral linewidth	
	□W 0.1 - 3 MHz & <10Khz with frequency stabilization (0,0004 cm ⁻¹)	1
	■ Pulsed: ~ 300 MHz	
•	High pulsed and cw powers of QCLs at TEC/RT temperatures	
	 Pulsed and CW powers of 34 W and 3 W respectively, high temperature operation ~300K 	9000
	* >280 mW, TEC CW DFB @ 5 μm	



Quantum Cascade (QC), Interband (IC) and GaSb Laser Availability in November 2009

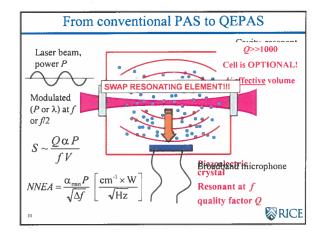
- · Commercial Sources
 - · Adtech, CA
 - Alpes Lasers, Switzerland & Germany

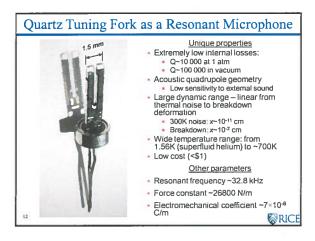
 - Alcatel-Thales, France
 Hamamatsu, USA & Japan
 - Maxion Technologies, Inc (Physical Sciences, Inc), MD
 Nanoplus, Germany
 Pranalytica, CA
- Research Groups

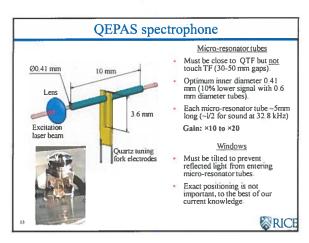
 - Harvard University
 Fraunhofer-IAF, Freiburg, Germany
 NASA-JPL, Pasadena, CA

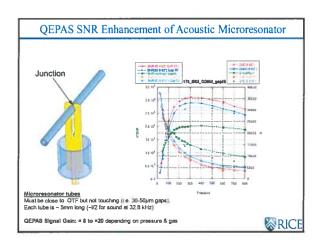
 - NASA-JPL, Pasadena, CA
 Naval Research Laboratories, Washington, DC
 Northwestern University, Evanston, IL
 Princeton University (MIRTHE), NJ
 State University of New York
 Technical University, Zuerich, CH
 University of Montpelier, France
 UK: Sheffield

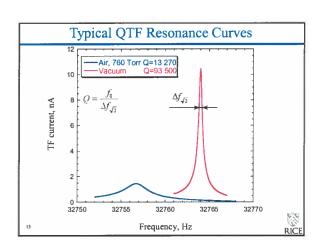
Quartz Enhanced Photoacoustic Spectroscopy

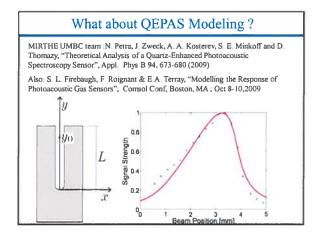


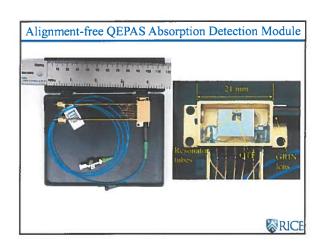


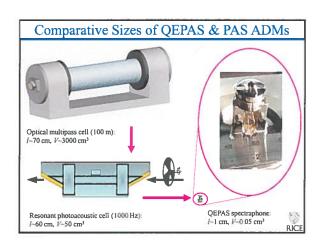












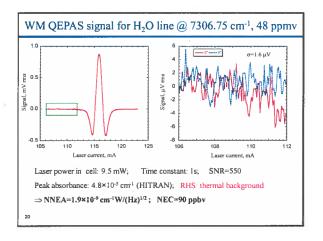
Merits of QEPAS based Trace Gas Detection

- Very small sensing module and the sample volume (a few mm³)
- · Optical detector is not required
- · Wide dynamic range
- Rugged transducer quartz monocrystal; can operate in a wide range of pressures and temperatures
- Immune to environmental acoustic noise, sensitivity is limited by the fundamental thermal TF noise: k_BT energy in the TF symmetric mode, directly observed
- Absence of low-frequency noise: SNR scales as \sqrt{t} , up to t=3 hours experimentally verified

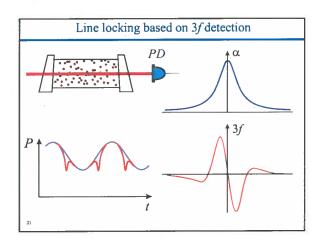
QEPAS: some challenges

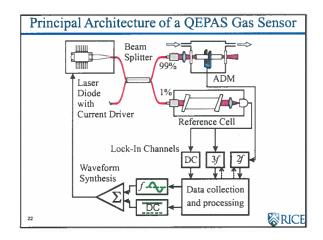
- Responsivity depends on the speed of sound and molecular energy transfer processes
- · Sensitivity scales with laser power

19

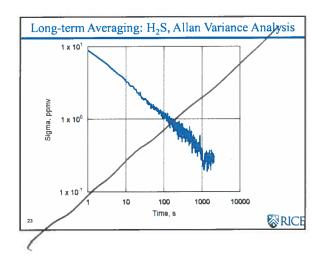


19





22



	Frequency,	Pressure, Torr	NNEA,	Power, mW	NEC (1º16)
(O (N))**	7306.73	60	19×10 ⁻⁴	9.5	0.09
HCN (alr. 58% RH)	6539 11	60	<43=10	50	0.16
Calla (Na)*	6523.83	720	41=10	57	0 03
NH ₃ (N ₃)*	6528.76	575	31×10*	60	0 06
CHL (Na)*	6177.07	715	5.4×10	15	1.7
CH4 (Nr+1.3% H/O)*	6037 09	760	3 71110	16	0.24
CO ₂ (broath ~100% RII)	6361 25	150	8.2×10	45	40
H,S (N ₂)*	6357 63	780	5.6=10	43	5
CO2 (N2+1.5% H3O) *	4991.36	50	1.4×10 ⁻⁴	44	18
CTI ₂ O (N ₂ 75 ** RII)*	2804 90	75	8.7×16™	7.2	0.12
CO (N ₂)	2196 66	50	53×10"	13	0.5
CO (propylene)	2196 66	50	7.4×10 ⁻⁴	6.5	0.14
N ₂ O (alr+5%SF ₄)	2195 63	50	. 15×10 ⁻⁸	19	0.007
C ₂ H ₂ OH (N ₂)**	1934.2	770	2.2×10°	10	90
CiHFs (Ni)***	1204 62	770	7.8×10 ⁻⁹	66	0.009
NH ₄ (N ₁)*	1046.39	110	1.6×10*	20	0.006

24