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Absorption Spectroscopy with Quantum Cascade Lasers

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- Motivation, Design, and Technology Issues
- Infrared Diode Laser Based Gas Sensors
- Performance Characteristics of Compact IR Sensors
- Selected Applications of Trace Gas Detection
- Outlook and Summary

LPHYS
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Wide Range of Gas Sensor Applications

- Urban and Industrial Emission Measurements
 - Industrial Plants
 - Combustion Sources
 - Automobile
- Rural Emission Measurements
 - Agriculture
- Environmental Monitoring
 - Atmospheric Chemistry
 - Volcanic Emissions
- Spacecraft and Planetary Surface Monitoring
 - Crew Health Maintenance & Life Support
- Chemical Analysis and Industrial Process Control
 - Semiconductor Industry
- Medical Applications

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Air Composition

| Main Components | Trace Components |
|--------------------------|----------------------------|
| • Nitrogen 78% | • Methane 1.7 ppm |
| • Oxygen 21% | • CO 0.4 ppm |
| • Water 0.8% | • N ₂ O 0.3 ppm |
| • CO ₂ 0.03 % | • O ₃ 0.03 ppm |
| | • ... |

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Absorption Spectroscopy

Beer's Law

$$I_1(\nu) = I_0 \cdot e^{-\alpha(\nu) \cdot L}$$

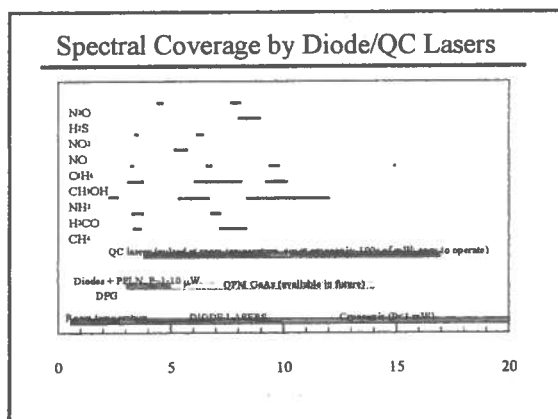
$\alpha(\nu)$ - absorption coefficient (cm⁻¹), L - path length (cm), ν - frequency (cm⁻¹)

Molecular Absorption Coefficient

$$\alpha(\nu) = C \cdot S \cdot g(\nu)$$

C - gas concentration (cm⁻³), S - absorption line strength (cm), $\Delta\nu$ - linewidth (cm⁻¹)
g(ν) - line shape function: Gaussian, Voigt, or Lorentzian profile

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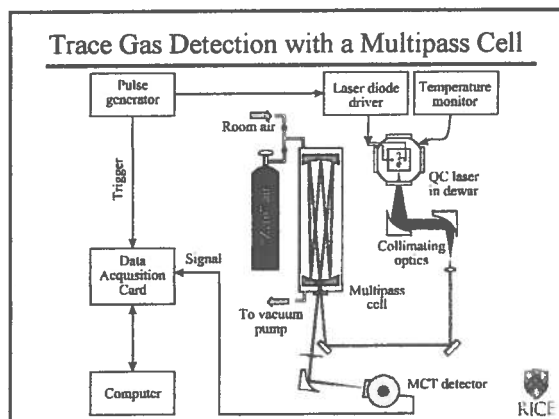


Key Characteristics of Quantum Cascade Lasers

- Laser wavelengths cover entire range from 3.4 to 17 μ m determined by layer thickness of same material
- Intrinsically high power lasers (determined by number of stages)
 - CW: 0.2W @ 80 °K, ~100 mW single frequency
 - Pulsed: 0.5W peak at room temperature, ~15 mW avg. @ 300 °K
- High Spectral purity (single mode)
- Wavelength tuning by current or temperature scanning
- High reliability: low failure rate, long lifetime, robust operation and extremely reproducible emission wavelengths

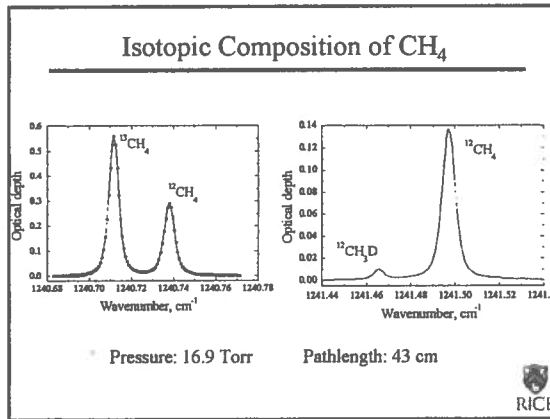
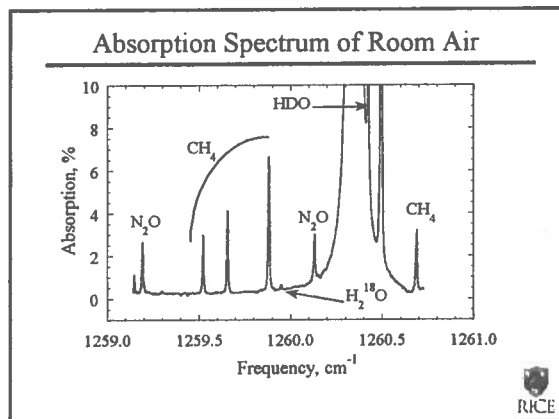
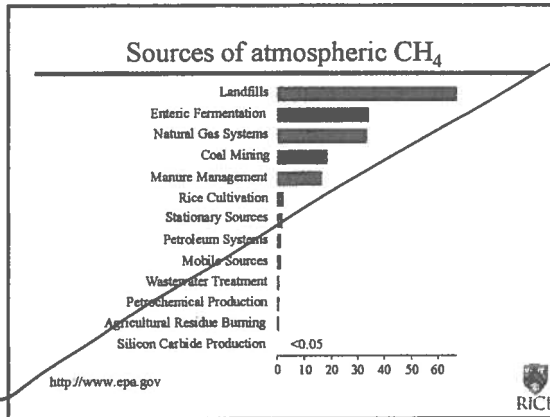
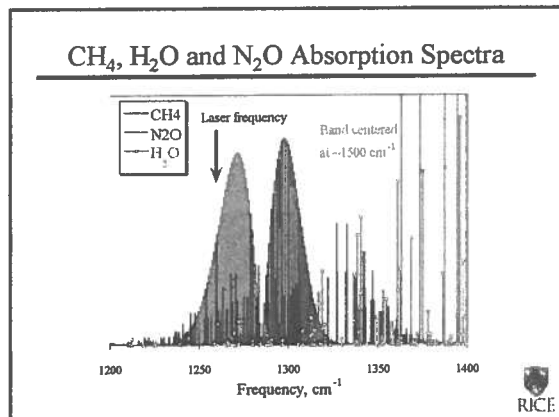
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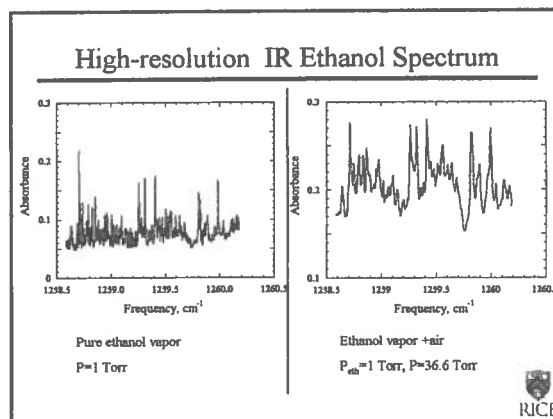
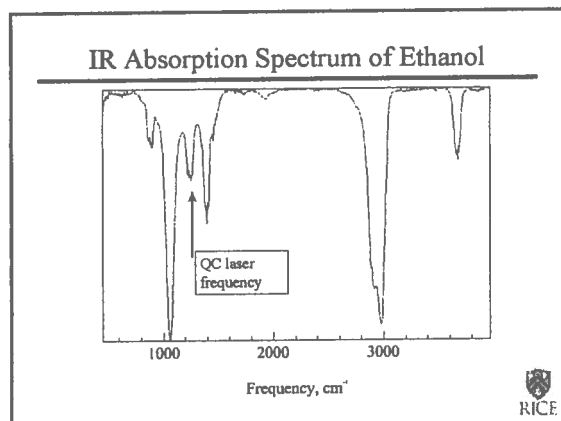
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Motivation for CH₄ and N₂O Detection

- Contribution to global warming
- Important in tropospheric and stratospheric chemistry
- Emitted by agricultural sources
- CH₄ leaks from gas pipelines

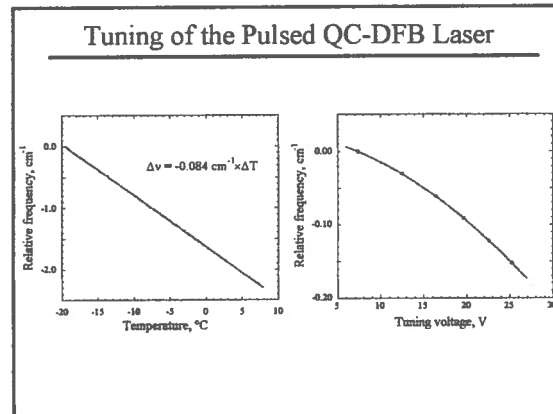
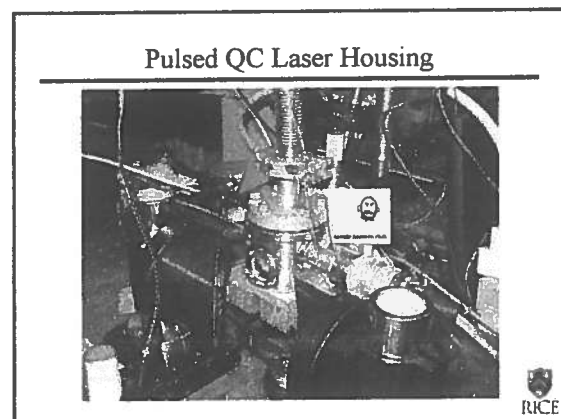
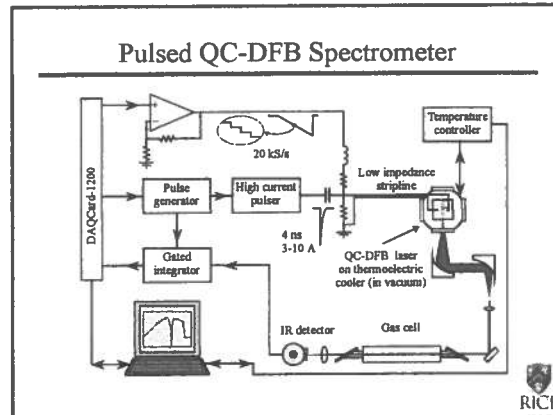


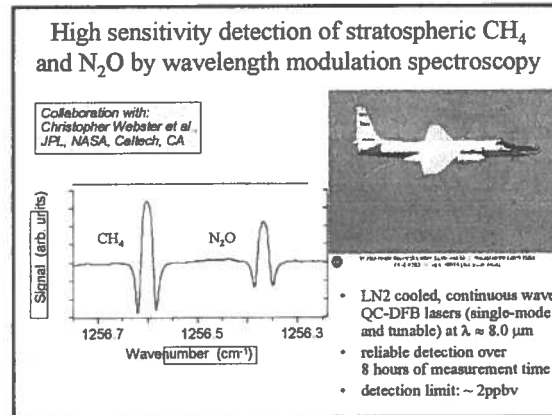
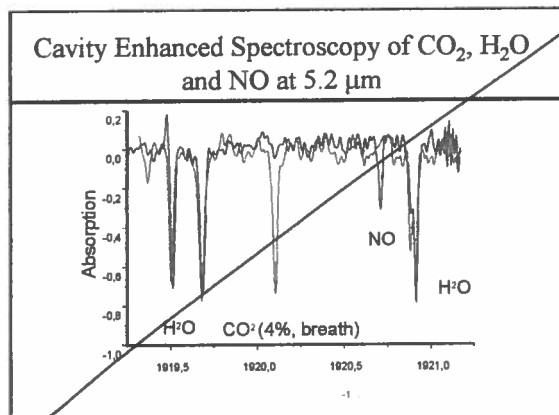
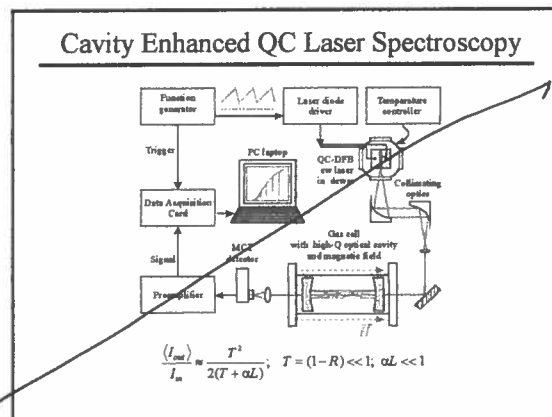
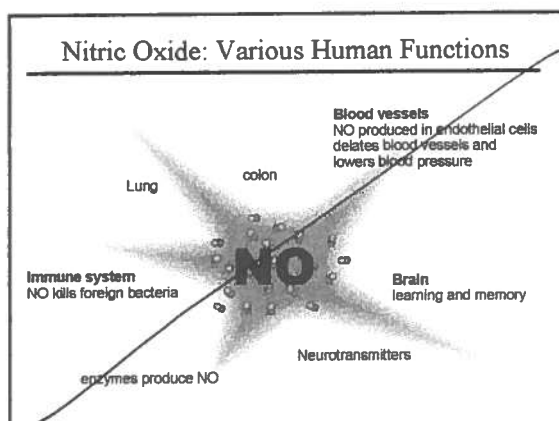


Pulsed Operation of a QC-DFB Laser

| ADVANTAGES | DISADVANTAGES |
|---|--|
| <ul style="list-style-type: none"> ♦ Laser can be operated at near-room temperature ♦ Easy temperature handling ♦ No consumables (liquid N₂) ♦ Compact | <ul style="list-style-type: none"> ♦ Broader linewidth (~400 MHz) ♦ Less average power ♦ More sophisticated electronics for driving QC laser and data acquisition |

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- ### Summary
- QC-DFB Laser Based Trace Gas Sensors
 - Compact, tunable, CW or pulsed, robust
 - High sensitivity (<10⁻⁴) and selectivity (<50 MHz)
 - Fast data acquisition and analysis
 - Detected trace gases: CH₄, N₂O, H₂O, NO, CO₂, NH₃, and C₂H₅OH
 - Isotopic Compositions
 - Current Applications in Trace Gas Detection
 - CH₄: NOAA, NASA-JPL, and gas industry
 - NH₃: NASA-JSC, semiconductor industry, combustion diagnostics
 - Future Directions
 - More efficient suppression of optical interference fringes
 - Pulsed quasi room temperature operations
 - Detection of complex molecules
 - Cavity enhanced spectroscopy
 - Medical Diagnostics: NO, CO, CO₂ and NH₃
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